



## Retention of aroma compounds in basil dried with low pressure superheated steam

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Received 3 October 2003; accepted 4 January 2004

### Abstract

Basil leaves were dried using either conventional hot air (50, 60 and 70 °C) and low pressure superheated steam (LPSS) dryers. The effect of the drying method on the retention of some volatile compounds was evaluated. The extraction from the fresh and dried products was performed by the simultaneous distillation–extraction technique. Identification and quantification was performed by capillary gas chromatography–mass spectrometry (GC–MS) and gas chromatography (GC) respectively.

The identified compounds were 23, with 61% of monoterpenes, 26% of aromatic and 13% of aliphatic compounds, out of which those characteristic of this spice were detected (1,8-cineole, methyl chavicol, methyl cinnamate and linalool).

Results show that in the LPSS dried product, the original aroma profile of the fresh vegetable is kept almost constant, while air-dried product shows a significant variation in the relative proportions of aroma compounds.

The conclusion is that the LPSS drying technique renders a better product in terms of the aroma compounds content, than the conventional air-drying, with the advantage of being cheaper.

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### 1. Introduction

Dehydration is the most popular and cheap way of preserving foodstuffs. By reducing the water activity, micro-organisms growth is inhibited and adverse biochemical reactions are slowed down. The most ancient dehydration agent has been hot air, but it provokes undesirable changes in the physicochemical properties of the product, as a consequence of the thermal impact imposed on it and the presence of oxygen reacting with some components of the foodstuff.

Among these, volatile compounds are particular sensitive to those effects, and, being important as factors of quality and acceptance, the effect of drying on aromatic vegetables, spices and the like, has received attention by many authors (Huopalahti, Kesalahti, & Linko, 1985; Jerkovic, Mastelic, & Milos, 2001; Yousif, Durance, Scaman, & Girard, 2000; Yousif, Scaman, Durance, & Girard, 1999; Venskutonis, 1991, among many others).

Results have shown that changes in concentration of those compounds are always produced, the magnitude

of which depend, for a given product, on the drying method and conditions.

An interesting alternative to the use of hot air is low pressure superheated steam (LPSS). It has been proved (Elustondo, 2001; Iwotech Limited, 1993; Kumar & Mujumdar, 1990; Mujumdar, 1980; Nomura & Hyodo, 1984; Svensson, 1980; Yoshida & Hyodo, 1960), that it results in significantly less damage as it works at lower temperatures and there is no oxygen present. The resulting product keeps the original colour and shape and shows a highly porous structure which makes it easily re-hydrated. Since it has been proved that LPSS requires less energy than hot air-drying for the same duty, it makes the procedure attractive from the economical point of view also.

Thus far, there is no information about the impact of this procedure on the volatile compounds, both lost and retained in the dried product, hence this work is focused to study this point, i.e. the amount and quality of the original contents kept in the sample, as well as those captured from the withdrawn water vapor by means of cold traps. Also, a comparison is made between samples dried with hot air in a conventional drier and other dried with this technique.

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65 Basil (*Ocimum basiculum*) is one of the preferred  
66 aromatic herbs for cooking and seasoning, and it is  
67 commercialised either fresh or dried. It belongs to the  
68 botany family Lamiacea, comprising many different  
69 species. Out of them the *O. basilicum* is the commonly  
70 used for cookery, pharmaceutical and cosmetic purposes  
71 (Simon, Morales, Phippen, Vieira, & Hao, 1999). The  
72 composition of volatiles depends on the species, origin,  
73 and culture conditions, but most authors have men-  
74 tioned that 1,8-cineole, methyl cinnamate, methyl  
75 chavicol and linalool are the main responsible for the  
76 typical basil aroma (Backstroom-Sternberg, Stephen,  
77 Duke, & Wain, 1994; Díaz-Maroto, Pérez-Coello, &  
78 Cabezudo, 2002; Lachowicz et al., 1996, 1997; Simon et  
79 al., 1999; among others).

80 **2. Materials and methods**

81 *2.1. Raw material*

82 The experimented material was fresh basil (*O. basil-*  
83 *icum*). Leaves were separated from the stems and stored  
84 at 4 °C until usage. Initial water content was determined  
85 on triple samples by using a laboratory vacuum oven at  
86 50 °C where samples were dried to constant weight. The  
87 average of moisture and volatile matter contents was  
88 obtained to be 89% on wet basis.

89 *2.2. LPSS drying equipment*

90 A bench scale drying equipment was designed and  
91 built, its schematic is shown in Fig. 1. It consists of a

cylindrical steel camera, (0.32 m in height, ID = 0.58 m, 92  
wall thickness = 0.1 m) externally covered by a 0.15 m 93  
thick isolation blanket, and containing an annular bas- 94  
ket (0.06 m in height, OD = 0.33 m and ID = 0.21 m) 95  
made of wiremesh to contain the samples. A centrifugal 96  
fan, driven by a shaft connected to an electrical motor, 97  
enables the steam to flow through the sample as shown 98  
in Fig. 1. 99

The drying camera is provided with a pressure sensor 100  
Abs Sensyn, sensitivity = 1.5 mV/psia, span = 67.5 mV 101  
for a range of 0–1034 mbar, connected to a digital dis- 102  
play. Temperature is measured and controlled by a LAE 103  
instrument, model MTW12T9RD/1, with an input 104  
range = –19.9 to 99.9 °C from a semiconductor sensor, 105  
and an output to a 5 A triac. This device controls the 106  
input power to an electrical resistance (1300 W) 107  
installed to heat steam. Accuracy is of ±0.1 °C from the 108  
set point. 109

The vapor–aroma mixture leaving the sample is ex- 110  
tracted from the camera by the action of a vacuum 111  
system, Jet-1-automatic, Genser Scientific Instruments 112  
MFG. Co., D-8803 Rothenburg, OT, Germany (see 113  
Fig. 2). It goes through a condenser made of a 0.08 m 114  
in diameter and 0.30 m in height glass cylinder with 115  
a 0.008 m in diameter spiral tube built it, and fed 116  
with a countercurrent stream of tap water, where 117  
most of the water vapor is separated. The remaining 118  
exhaust gas is conducted to a cold trap made of a 119  
0.03 m in diameter and 0.40 m in height glass cylin- 120  
der, submerged in an ethyleneglycol–water mixture at 121  
–15 °C, where almost all compounds are also con- 122  
densed. 123

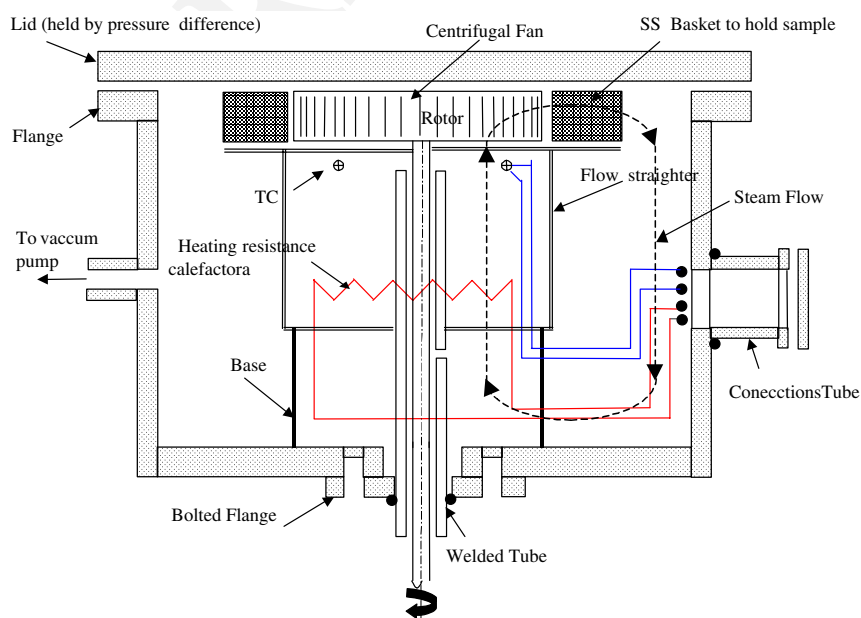


Fig. 1. Drying chamber cross-sectional view.

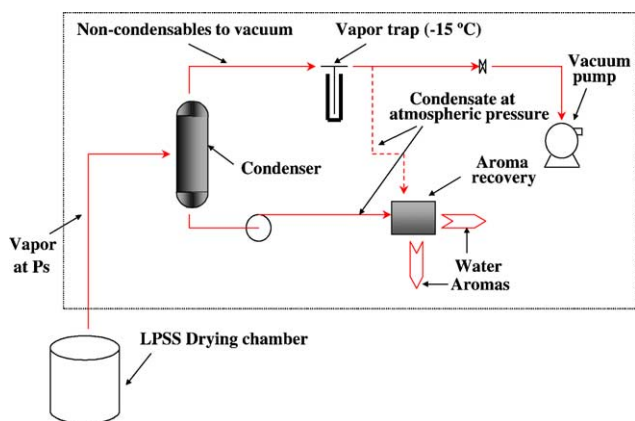


Fig. 2. Sketch of the experimental aroma recovery setup.

### 124 3. Experimental technique

#### 125 3.1. LPSS drying

126 A sample of 300 g of fresh basil was dried by super-  
 127 heated steam at 50 °C, at a working pressure of 5066 Pa  
 128 at which water equilibrium temperature is 33 °C. Steam  
 129 was recirculated at a linear speed of 4.5 ms<sup>-1</sup>. The  
 130 operation was performed during 6 h. Once finished the  
 131 volume of water and aromas trapped in the condenser  
 132 and the cool trap was collected and measured. The li-  
 133 quid from the condenser as well as the dried product  
 134 were stored at 4 °C for further extraction of volatile  
 135 compounds.

#### 136 3.2. Air-drying

137 Basil from the same source, though collected in the  
 138 following season, was air-dried using a conventional  
 139 convective drier, where air at 40, 50 and 60 °C was  
 140 blown at an air velocity of 1.2 ms<sup>-1</sup>. Temperature was  
 141 automatically controlled at ±2 °C.

#### 142 3.3. Extraction and concentration of volatiles

143 Samples of volatile compounds were obtained by the  
 144 simultaneous distillation-extraction technique (SDE)  
 145 (Likens & Nickerson, 1964; Shultz, Flath, Mon, Eg-  
 146 gling, & Teranishi, 1977), with ethyl-ether as extraction  
 147 agent. The same conditions were set for the extraction  
 148 from fresh and dried samples and from the water solu-  
 149 tion captured in the condenser and the trap. Solid  
 150 samples weighed 5 g and the liquid ones measured 100  
 151 ml. Samples of dehydrated and fresh basil were chopped  
 152 previously. All samples were mixed with 200 ml of bi-  
 153 distilled water in a 1000 ml flask. The distillation-  
 154 extraction process took 2 h at a pressure of 20,265 Pa.

155 The extract, once dried on NaSO<sub>4</sub>, was concentrated  
 156 by evaporating the solvent through a Vigreux micro-

column to a volume of approximately 0.6 ml. Samples  
 were finally stored in N<sub>2</sub> atmosphere at -18 °C.

#### 3.4. GC conditions

A Hewlett Packard 4890D chromatograph (Agilent  
 Technologies, USA) equipped with a fused silica capil-  
 lary column CW 20M (50 m, ID 0.22 mm, film thickness  
 0.2 μm) was used for GC analysis. Conditions were as  
 follows: sample size, 2 μl; initial column temperature,  
 46 °C, held for 2 min, followed by heating at a rate of  
 4 °C min<sup>-1</sup> to 200 °C, held at this temperature for 20  
 min; injector temperature, 240 °C, split ratio, 1/50;  
 FID detector temperature, 240 °C; carrier gas, H<sub>2</sub> (1.4  
 ml min<sup>-1</sup>).

#### 3.5. GC-MS conditions

For the identification of the volatiles compounds,  
 some samples were submitted to GC-MS analysis on a  
 Hewlett Packard 6890 gas-chromatograph (Agilent  
 Technologies, USA), equipped with a 5972 A Series  
 mass spectrometer, scan mode (ionization energy: 70 eV,  
 mass range: 35-500 u). An amount of 5 μl of extract was  
 analysed. The column was an HP5 fused silica (30 m, ID  
 0.25 mm, film thickness 0.25 μm); initial column tem-  
 perature was 46 °C; held for 2 min, followed by heating  
 at a rate of 4 °C min<sup>-1</sup> to 200 °C, held at this tempera-  
 ture for 20 min. Injector temperature was 250 °C;  
 detector temperature was 280 °C; carrier gas, He (1  
 ml min<sup>-1</sup>).

#### 3.6. Identification and quantitative analysis

The tentative identification of compounds was per-  
 formed by comparison of mass spectra of each compo-  
 nent with the corresponding standards in the data base,  
 and using authentic standards when available, results  
 were compared with those reported in literature  
 (Backstroom-Sternberg et al., 1994; Díaz-Maroto et al.,  
 2002; Lachowicz et al., 1996, 1997; Maarse, 1991; Simon  
 et al., 1999; Yousif et al., 1999, among others).

The presence of the compounds selected for the study  
 was confirmed by comparison of the retention times  
 with those of pure reference standards. The quantitative  
 determination was made by the internal standard  
 method, with response factors calculated from pure  
 reference compounds. The internal standard used was 2-  
 heptanol.

All experiments were repeated three times and results  
 were analysed statistically to obtain average values and  
 sample standard deviations, as parameters of results  
 reproducibility. Variance analysis (ANOVA, Origin v6)  
 was used to evaluate the significance of the difference in  
 the amounts of studied compounds. Values were con-  
 sidered significantly different when  $p < 0.05$ .

207 **4. Results and discussion**

208 *4.1. Identification of volatile compounds*

209 Fig. 3 shows a typical chromatogram obtained from a  
210 fresh basil sample, and Table 1 shows the percent  
211 composition of the identified compounds only, which  
212 turned out to be 23, as calculated from the respective  
213 chromatographic areas. They are mainly terpenes (61%),  
214 and in minor proportion aromatics (26%) and aliphatics  
215 (13%). Remaining peaks correspond to compounds that  
216 were separated but not identified.

217 The two major volatile compounds in our samples  
218 were linalool and 1,8-cineole, usually considered  
219 responsible for the basil typical aroma. Similar results  
220 were reported by Lachowicz et al. (1997) from a variety  
221 of species *O. basilicum* and by Díaz-Maroto et al. (2002).  
222 The area under the chromatographic peaks of most of  
223 the compounds is small and characteristic compounds  
224 like methyl cinnamate and methyl chavicol are present  
225 in very small quantities.

226 To study the impact of LPSS drying, eight com-  
227 pounds were selected: four of them are considered as  
228 character-impact of basil aroma, like 1,8-cineole, linal-  
229 ool, methyl chavicol and methyl cinnamate. The  
230 remaining four are also usually found in basil like  $\alpha$ -  
231 pinene, limonene, *p*-cymene and hexadecane contrib-  
232 ute to the aroma profile and standards were avail-  
233 able. Camphene, though in larger proportion, was not  
234 selected because it has not been usually cited by  
235 other authors and it is felt that the selected com-  
236 pounds are enough for a proper analysis of the drying  
237 effects.

Table 1  
Basil volatile components (Batch 1) (Percent by Gas Chromatography)

Peak <sup>a</sup>	Compound	Composition (%)
1	$\alpha$ -pinene <sup>b</sup>	0.42 ± 0.01
2	Camphene	3.97 ± 0.12
3	$\beta$ -pinene	0.38 ± 0.01
4	$\alpha$ -terpinene	0.07 ± 0.00
5	Limonene <sup>b</sup>	0.16 ± 0.00
6	1,8-Cineole <sup>b</sup>	6.34 ± 0.08
7	$\gamma$ -terpinene	0.41 ± 0.00
8	$\alpha$ -phellandrene	0.07 ± 0.00
9	<i>p</i> -Cymene <sup>b</sup>	0.05 ± 0.00
10	Terpinolene	0.09 ± 0.00
11	2-Heptanol (internal standard)	
12	Camphor	0.08 ± 0.00
13	Linalool <sup>b</sup>	28.89 ± 0.28
14	Bornyl acetate	0.19 ± 0.01
15	Linalyl acetate	1.00 ± 0.01
16	Hexadecane <sup>b</sup>	0.32 ± 0.02
17	$\alpha$ -caryophyllene	2.22 ± 0.03
18	$\beta$ -caryophyllene	0.49 ± 0.00
19	Methyl chavicol <sup>b</sup>	0.56 ± 0.01
20	1-Octen-3-ol	0.58 ± 0.00
21	Octadecane	1.45 ± 0.01
22	Methyl cinnamate <sup>b</sup>	0.06 ± 0.00
23	Eugenol	0.31 ± 0.01
24	Methyl eugenol	0.13 ± 0.01

Values are given as mean ± standard deviation (*n*:3).

<sup>a</sup>The peak numbers correspond to the identification numbers in Fig. 3.

<sup>b</sup>Identification confirmed by injection of standard compound.

238 *4.2. Effect of LPSS drying on volatile compounds*

The Table 2 shows the composition of fresh basil, 239  
dried product and condensate from the condenser and 240  
the cold trap, in terms of the mentioned compounds 241  
with the results of the statistical analysis. 242

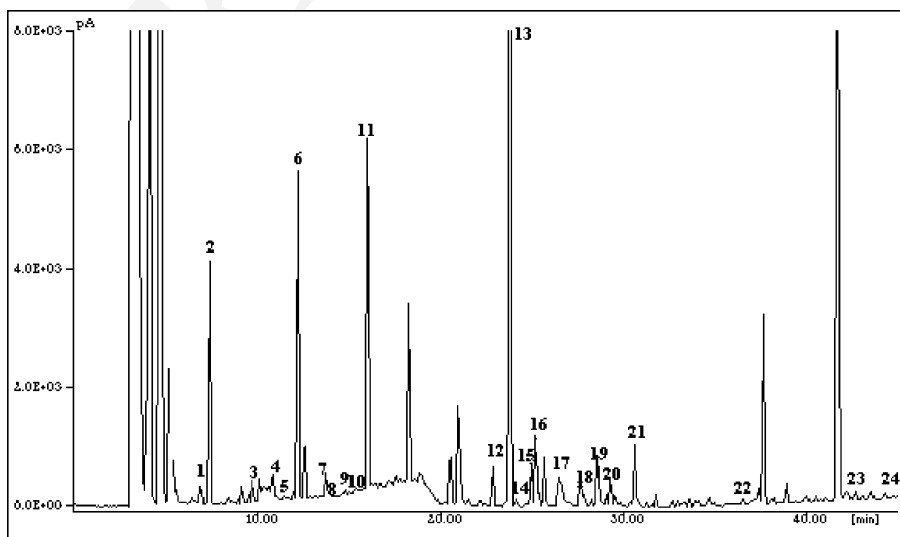


Fig. 3. Typical chromatogram of volatile compounds extracted from fresh basil from GC analysis (CW 20 M column).

Table 2

Volatile compounds contents in fresh product, LPSS dried product and extract from the condenser and the trap (Batch 1) [mg (kg of dry weight)<sup>-1</sup>]

Compound	Fresh	LPSS dried material	In condensate
$\alpha$ -pinene	4.47 ± 0.304	3.13 ± 0.313	0.719 ± 0.033
Limonene	1.61 ± 0.111	1.37 ± 0.011	0.191 ± 0.001
1,8-Cineole	86.7 ± 3.33	79.6 ± 1.61	4.66 ± 0.048
<i>p</i> -Cymene	0.242 ± 0.026	0.179 ± 0.002	0.110 ± 0.002
Linalool	292.4 ± 29.7 <sup>a</sup>	255.6 ± 11.3 <sup>a</sup>	51.3 ± 0.428
Hexadecane	2.89 ± 0.066	1.06 ± 0.028	0.430 ± 0.007
Methyl chavicol	4.88 ± 0.051	3.98 ± 0.075	0.507 ± 0.016
Methyl cinnamate	0.530 ± 0.006 <sup>a</sup>	0.443 ± 0.015 <sup>a</sup>	0.095 ± 0.003

Values are given as mean ± standard deviation (*n*:3).

<sup>a</sup> Not significant differences (*p* > 0.05).

243 In all compounds, with the exception of linalool and  
244 methyl cinnamate, a statistically significant reduction  
245 has been observed as a consequence of the LPSS dehy-  
246 dration.

247 However, it is apparent that retention of all the  
248 compounds in the LPSS-dried product was higher than  
249 70%, with the exception of hexadecane which was only  
250 37%, which is a good result in terms of the product  
251 acceptance. The higher losses were those of not oxy-  
252 genated terpenes, such as  $\alpha$ -pinene (29.8%) and *p*-cym-  
253 ene (26%), while methyl cinnamate, 1,8-cineole, linalool  
254 and methyl chavicol, which are considered character-  
255 impact of basil aroma, were affected in lower proportion  
256 by drying process, keeping 83%, 80%, 88% and 82% of  
257 the original contents, respectively.

258 Fig. 4 shows the relative amounts of the mentioned  
259 compounds retained in the LPSS dried product and  
260 captured in the condenser and the trap, both expressed  
261 as percentage of the original contents in the basil sam-  
262 ple.

263 The recovery of aroma compounds from the con-  
264 densate show different efficiency for the different com-  
265 pounds. But considering that some of them are present  
266 in the original product at very low concentrations,  
267 handling and measuring errors are unavoidable.

268 It is worth to note that the total quantity of linalool  
269 and methyl cinnamate kept in the dried product plus

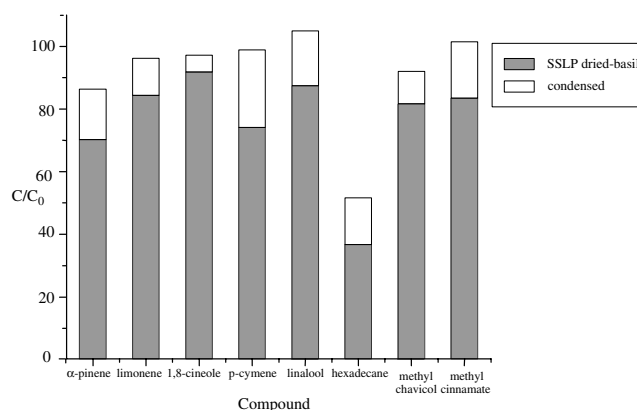


Fig. 4. Percent contents (relative to fresh product) of volatile compounds in LPSS dried basil and in condensed withdrawn vapor.

that recovered from the condensate is higher than the  
original one (see Table 2). This result, tough unexpected,  
has been reported by other authors (Lachowicz et al.,  
1996; Yousif et al., 2000; Yousif et al., 1999; Vensku-  
tonis, 1997) and has been attributed to the hydrolysis of  
glycosides and the conversion of linalyl acetate into  
linalool during drying process.

On the other hand, compounds with losses higher  
than 10%, like hexadecane (48%) and  $\alpha$ -pinene (13%),  
are not regarded as character-impact of basil aroma.

#### 4.3. Effect of air-drying on volatile compounds 280

Table 3 shows the concentration (in mg kg<sup>-1</sup> dry  
basis) of studied compounds in fresh basil and hot air-  
dried at different temperatures, as well as the results of  
the statistical analysis.

The difference in composition of the fresh vegetable  
used for air-drying and for LPSS drying is due to the  
fact that the samples, though from the same origin, are  
not the strictly the same. However, this is not a problem  
for the purpose of the work, as it is to compare the effect  
of each technique in terms of the final composition of  
the resulting product, that is, in what extent the original  
aroma profile is kept after undergoing different methods  
of drying. To identify the different samples, they are

Table 3

Volatile compounds contents in fresh basil and air-dried samples at 3 different temperatures (Batch 2) [mg (kg of dry weight)<sup>-1</sup>]

Compound	Fresh	Air-dried 40 °C	Air-dried 50 °C	Air-dried 60 °C
$\alpha$ -pinene	11.0 ± 0.452 <sup>a</sup>	6.96 ± 0.146 <sup>b</sup>	5.20 ± 0.030 <sup>c</sup>	5.76 ± 0.017 <sup>d</sup>
Limonene	8.24 ± 0.337 <sup>a</sup>	6.56 ± 0.05 <sup>b</sup>	5.32 ± 0.004 <sup>b</sup>	5.28 ± 0.006 <sup>b</sup>
1,8-Cineole	998.5 ± 113.4 <sup>a</sup>	396.9 ± 20.6 <sup>b</sup>	283.4 ± 27.3 <sup>c</sup>	347.9 ± 0.965 <sup>c</sup>
<i>p</i> -Cymene	2.14 ± 0.105 <sup>a</sup>	2.17 ± 0.053 <sup>a</sup>	0.952 ± 0.021 <sup>b</sup>	1.67 ± 0.226 <sup>c</sup>
Linalool	2258.4 ± 141.5 <sup>a</sup>	1162.0 ± 89.0 <sup>b</sup>	1626.8 ± 133.0 <sup>c</sup>	1583.2 ± 111.4 <sup>c</sup>
Hexadecane	6.72 ± 0.679 <sup>a,b</sup>	7.50 ± 0.131 <sup>a</sup>	6.90 ± 0.117 <sup>b</sup>	3.34 ± 0.061 <sup>c</sup>
Methyl chavicol	71.3 ± 5.89 <sup>a</sup>	16.4 ± 0.366 <sup>b</sup>	18.4 ± 0.191 <sup>b</sup>	110.6 ± 5.90 <sup>c</sup>
Methyl cinnamate	1.56 ± 0.003 <sup>a</sup>	0.851 ± 0.008 <sup>b</sup>	0.639 ± 0.059 <sup>c</sup>	0.123 ± 0.013 <sup>d</sup>

Values are given as mean ± standard deviation (*n*:3).

Different letters (a, b, c, d) in the same row indicate statistical differences according to ANOVA analysis (*p* < 0.05).

294 named “batch 1” and “batch 2” respectively, keeping  
295 the denomination “sample” for those of the same batch  
296 submitted to different treatments.

297 Results show marked differences between the com-  
298 position of the dried product as compared with the fresh  
299 one, and, exception made of the above mentioned me-  
300 thyl chavicol (dried at 60 °C), there is a decrease in the  
301 content of the studied compounds.

302 As the air temperature increases, the differences are  
303 greater, though the effect is not the same for all com-  
304 ponents and no trend can be established.

305 1,8-Cineole underwent a pronounced decrease, in the  
306 order of 65% for all of these air-drying temperatures.  
307 The amount of methyl cinnamate showed a significant  
308 decrease (45–60%) from original levels present in the  
309 fresh samples when basil was air-dried at 40–50 °C  
310 respectively, and it almost disappeared at 60 °C (8%).  
311 Linalool, other character-impact compound was also  
312 affected by air-dried.

313 The observed increase in the content of methyl  
314 chavicol upon air-drying (60 °C) was reported by Yousif  
315 et al. (1999).

316 The percentage of retention of most of the analysed  
317 compounds is higher for the LPSS dried samples than in  
318 those dried by hot air, at all temperatures. Moreover,  
319 and what is more relevant, the percent distribution of  
320 the retained compounds is similar to that of the fresh  
321 samples (see Fig. 5), while air-dried samples show a  
322 different percent composition (see Fig. 6). Since the he-  
323 donic appreciation of an aroma results from the simul-  
324 taneous and combined perception of the individual  
325 components of the mixture, it is expected that the dry  
326 product exhibiting the same distribution profile than the  
327 fresh one, should have the same degree of acceptance by  
328 the consumer.

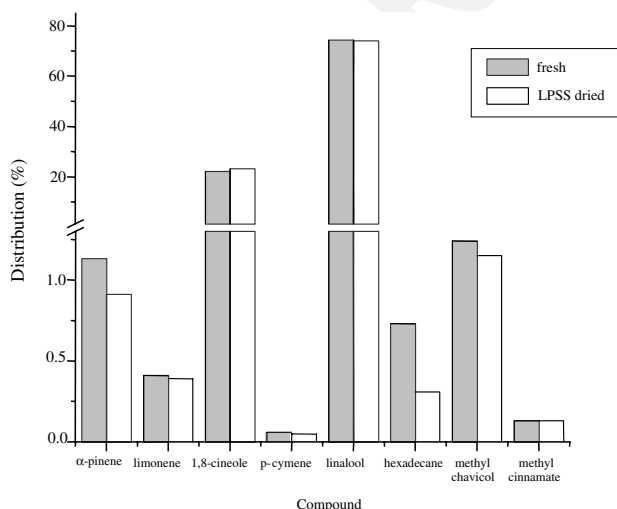


Fig. 5. Composition profile of fresh basil and LPSS dried sample.

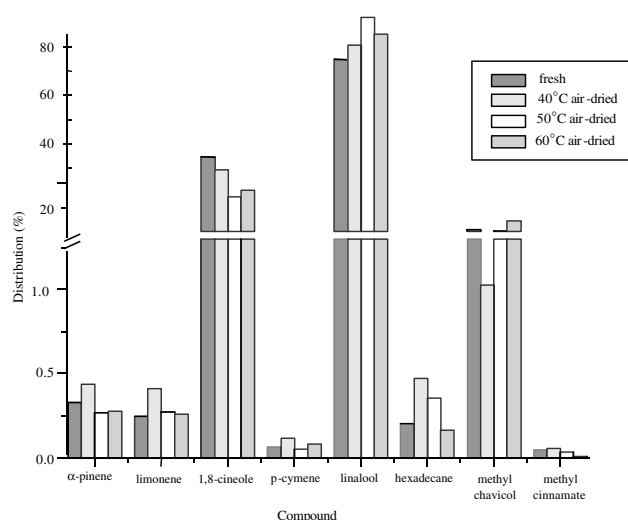


Fig. 6. Composition profile of fresh basil and air-dried samples at three different temperatures.

## 5. Conclusions

329

The technique for extraction of volatile compounds 330  
used in this work (SDE) proved to be efficient in sepa- 331  
rating components regarded as characteristics of basil 332  
aroma mentioned in literature. The extract obtained was 333  
constituted mainly by terpenes. 334

The use of low pressure superheated steam as drying 335  
agent rendered a product with a higher retention of the 336  
original volatiles compounds and a lower modification 337  
of the percent composition. 338

The LPSS drying technique allows the recovery of a 339  
substantial amount of the aroma compounds withdrawn 340  
from the product, increasing the economical advantages 341  
of the new method. 342

Regarding air-drying, it was observed a significant 343  
reduction in the content of most of the analysed com- 344  
pounds in the product. Also, changes in air temperature 345  
have different effect on different compounds: while 346  
concentration of some increase as temperature rise, 347  
other decrease. 348

From an overall analysis of the results obtained, it 349  
can be concluded that LPSS drying is a method that 350  
guarantees the preservation of main basil aroma com- 351  
pounds. 352

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