

# Evolution of Reactive mPEG Polymers for the Conjugation of Peptides and Proteins

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**Abstract:** The covalent attachment of methoxy-poly(ethylene glycol) (mPEG) is a well established strategy used to improve the pharmaceutical properties of several biomolecules. Since the pioneering work of Abuchovsky, PEGylation has emerged as a powerful technology of significant relevance, not only for the development of new and better drugs, but also for application in material science. Peptides and proteins are the most traditional targets for PEGylation due to their intense and diverse biotechnological applications. The terminal amino group, as well as the  $\epsilon$ -amino group of lysine and the thiol group of cysteine, are all well known nucleophilic sites that have traditionally been used to couple peptides and proteins to mPEG derivatives. Advances in the methods for preparation of the mPEG starting materials, together with a careful selection of new mPEG functional end-groups have allowed new reactive mPEGs to emerge, which show narrow polydispersity and controlled reactivity, providing more homogeneous conjugates. In the last few years the trend has moved towards site-selective, reversible and enzymatic PEGylation using a new generation of tailor-made reagents and strategies. The main goal of this article is to present some of the most relevant achievements obtained in the PEGylation of peptides and proteins. The chemistry underlying the current methods used for the preparation of mPEG reagents, as well as the chemistry involved in the PEGylation reactions are presented in detail, in order of stimulating the synthetic and polymer chemist to turn their attention in this fascinating multi and interdisciplinary field of research.

**Keywords:** PEGylation, poly(ethylene glycol), PEG reagents, Conjugation, Protein PEGylation, Controlled drug release, Activated PEG.

## 1. INTRODUCTION

In the last 30 years the attachment polyethylene glycol (PEG) has been used extensively to improve the physicochemical, biomedical and pharmacological properties of several therapeutic molecules, such as peptides, proteins, antibodies, antibody fragments, oligonucleotides, and small drugs [1]. PEG is an FDA approved diol compound that has historically been used as an excipient in pharmaceutical and cosmetic formulations. Moreover, PEGs of different molecular weights are well known solvents, widely used in organic synthesis. PEGylation strategies using PEG diol derivatives have been rather limited to the conjugation of small molecules, which require increased drug loads per polymer chain. Indeed, PEG dendrimers and dendromers are preferred to linear PEG diol derivatives for small drug delivery, owing to the fact that higher drug ratios can be obtained. In the PEGylation of biomacromolecules such as peptides and proteins, the usefulness of PEG diol reagents for conjugation is relatively low, as they often lead to cross-linking of the target molecules, leading to the formation of aggregates with decreased bioactivity. On other hand, methoxy-poly(ethylene glycol) (mPEG) is a protected version of PEG, from which monofunctional reactive polymers can be prepared. These are ideal for the conjugation of proteins as they do not lead to the cross-linking of the macromolecules. mPEG is a biocompatible and non-biodegradable class of polymer that has distinctive properties that make it particularly attractive for protein-polymer conjugation: i) mPEG is soluble in water and also in many typical organic solvents. As a distinctive feature, mPEG is soluble in methylene chloride, chloroform, toluene, acetonitrile and other

solvents; while it is not soluble in ethyl ether, *iso*-propanol and aliphatic hydrocarbon solvents. This unique solubility behaviour allows the purification of mPEG and mPEG derivatives from low molecular weight organic impurities via a simple precipitation-centrifugation work-up. ii) mPEG is not antigenic, immunogenic or toxic [2,3]. However, the occurrence of antibodies against PEG (anti-PEG) in healthy blood donors has been reported when high polymer loads were employed [4,5]. iii) mPEG has been approved by the FDA for human use in intravenous, oral and dermal applications. iv) mPEG chains are flexible, and in aqueous solution coordinate 2-3 water molecules per oxyethylene unit. The flexibility of the polymer chain, together with the ability of coordinating several water molecules, means that mPEGs have an apparent molecular weight that is 5 to 10 times higher than that of globular proteins of comparable molecular weights. v) Nowadays, commercially available mPEG polymers have narrow polydispersities, typically ranging from 1.01 for low molecular weight mPEGs ( $\approx$ 5000 Da) to 1.1 for mPEG of up to 50 kDa, which are adequate for pharmacological applications.

The covalent attachment of mPEG to bioactive polypeptides and proteins increases the plasmatic half-life of the biomacromolecules by increasing the hydrodynamic ratio, which in turn reduces excretion rate. Furthermore, the PEGylation of proteins: i) decreases the immunogenicity by shielding antigenic epitopes, a major drawback with heterologous proteins; ii) reduces the degradation by proteolytic enzymes and antibody recognition by covering the recognition sites [6]; iii) increases water solubility, which can be an extremely important for low molecular weight drugs [7,8]; iv) modifies the biodistribution; promoting the accumulation of the conjugated bio-drugs into tumors [9,10]. Additionally PEG can promote accumulation in specific organs, tissues or cells by the EPR effect [11,12]; v) enhances bioavailability and plasmatic half-life, owed to the increment of hydrodynamic volume, which re-

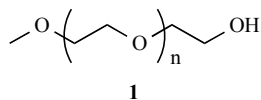
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duces renal clearance; vi) stabilizes labile drugs from chemical degradation.

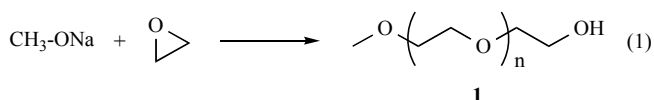
The hydroxyl group of mPEG is not reactive enough to allow direct coupling of the polymer to biomacromolecules. Thus, to make mPEG useful for the conjugation of proteins, the terminal hydroxyl group should be converted into a group that can react with some of the functional groups present in the biomacromolecule. In this regard, most of the functional groups present in the polypeptidic chains that are amenable for PEGylation are nucleophilic groups, namely, primary amino groups (such as terminal amino group and the  $\epsilon$  amino group of lysine), the secondary amino group of histidine, the thiol group of cysteine, and, to a lesser extent, the terminal carboxyl, the carboxyl group of aspartic and glutamic acid and the hydroxyl group of tyrosine, serine and threonine. Hence, and as will be shown later, the vast majority of the reactive mPEGs used in bioconjugation are electrophilic derivatives. From a synthetic chemistry standpoint, proteins could be considered as sensitive compounds whose bioactivity would easily be altered even under relatively mild reaction or purification conditions. Thus, the reactive mPEGs should be able to react with the macromolecules under mild and usually aqueous conditions, in a relatively narrow pH range. Furthermore, the conjugate and the unreacted protein should be purified from the unreacted mPEG derivatives, the poly-PEGylated protein and other reaction by-products under mild and strictly controlled purification conditions.

## 2. GENERAL AND PRACTICAL CONSIDERATIONS ABOUT FUNCTIONAL PEGs REAGENTS

mPEG **1** is a neutral and linear polyether alcohol with the general structure shown below.



mPEG is obtained by anionic polymerization of ethylene oxide using a methoxide to initiate nucleophilic attack of the epoxide ring (equation 1) [13,14].



Today, mPEGs of different molecular weights and of variable quality are commercially available. The diol content of the starting mPEG raw materials, which arises from trace amounts of water, oxygen or other impurities during polymer synthesis, is a key factor that determines the mPEG quality as upon synthetic manipulation, mPEG diols give rise to bifunctional reactive polymers, ultimately affording undesirable high degrees of cross-linking [15]. Usually,

the diol content increases with the molecular weight of the polymer and, generally, the molecular weight of the diol polymer is twice than that of mPEG (as propagation occurs at both ends of the polymer chain). Diol contents as high as 10-15% have been observed in high molecular weight mPEGs [16,17]. It has been shown that diol-free mPEG can be obtained by methylation of crude benzyloxy-PEG, followed by hydrogenation. This reaction renders mPEG and inert dimethoxy-PEG that can ultimately be removed after biomacromolecule conjugation [18].

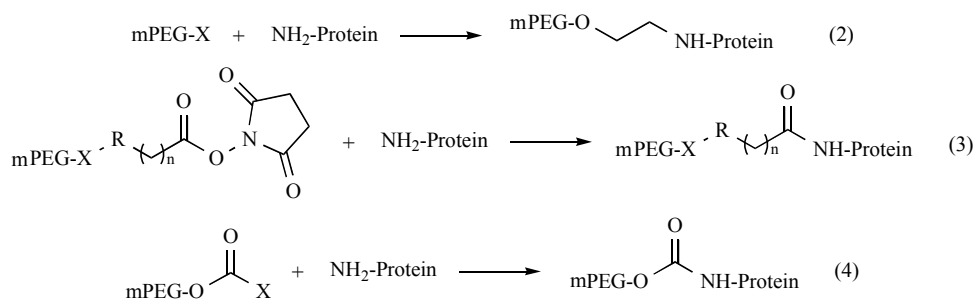
The polydispersity index (PDI, a measure of distribution of molecular masses, defined as weight average molecular weight ( $M_w$ ) divided by number average molecular weight ( $M_n$ )) is another important parameter of the starting mPEG raw material, since it can critically influence the homogeneity of the conjugates. Nowadays, mPEG raw materials with PDI of nearly 1.01 for PEGs of 2-10 kDa and up to 1.1 for PEGs with higher molecular weight (>20 kDa) are commercially available.

The free hydroxyl group of mPEG is not reactive enough to allow conjugation with bioactive molecules and materials; as result, mPEG has to be converted to more reactive species, most usually to electrophilic derivatives, for protein and peptide PEGylation. In high molecular weight mPEGs, the structural changes introduced upon modification of the terminal hydroxyl groups can be negligible when compared with the structure of the repeat oxyethylene unit. With the aim of avoiding demanding purification techniques, great efforts have been made in order of providing synthetic methodologies that affords the mPEG derivatives with conversions that are as high as possible. As mentioned earlier, mPEG and mPEG derivatives can be purified via precipitation from diethyl ether or other suitable solvents. Purification and spectroscopic characterization issues of PEG derivatives have all been reviewed in detail, however it is worth mentioning that most chromatographic and spectroscopic methods used within organic chemistry are also well suited for PEGylation chemistry [19-21]. Particularly useful techniques include  $^{13}\text{C}$  and  $^1\text{H}$  NMR as well as chromatographic methods (gel permeation, reverse phase and ion exchange). Moreover, polyacrylamide gel electrophoretic methods, capillary electrophoresis and HPLC with ELSD detection can be used for the characterization of branched mPEGs and mPEGs conjugates [22].

Reactive mPEG derivatives should be able to react with peptides and proteins under mild and buffered aqueous conditions. Hence, the reactivity of mPEG derivatives should be high enough to give acceptable yields of conjugated products, but without affording undesirable high degrees of poly-PEGylated bioconjugates, compromising the protein activity, or reacting too quickly with the (aqueous) reaction solvent. As expected, most commercially available reactive mPEGs are not completely stable and it is advisable to store them in an environment protected from moisture, light and air. A summary of advantages and drawbacks associated with use of mPEG in bioconjugation are presented in Table 1.

**Table 1. Advantages and Drawbacks Associated with the Use of mPEG.**

Advantages	Drawbacks
<ul style="list-style-type: none"> <li>• Non-toxic.</li> <li>• Broad molecular weight availability.</li> <li>• Stable if stored properly.</li> <li>• Solubility in organic solvents and water; precipitation with ethyl ether.</li> </ul>	<ul style="list-style-type: none"> <li>• High diol content (cross-linking).</li> <li>• Spectroscopic, chromatographic and purification issues would be difficult to resolve.</li> <li>• Reactive mPEGs are not stable.</li> </ul>



### 3. GENERAL FEATURES OF THE CHEMISTRY OF PEGYLATION

In a series of papers from 1977, Davies and co-workers demonstrated that not only was it possible to covalently attach mPEG to bovine serum albumin and bovine liver catalase, but that the enzyme conserved 95% of its activity, and that both proteins showed altered immunogenicity and prolonged circulation half-lives [23,24]. The chemistry of these pioneering PEGylations involved the reaction of the nucleophilic  $\epsilon$ -amino group of lysine amino acids with electrophilic cyanuric chloride-mPEG derivatives. Very high levels of amino group conjugation were achieved with these extremely reactive mPEG reagents.

From the 20 proteinogenic amino acids, alanine, valine, leucine, isoleucine, proline and phenylalanine do not have side chains that could react with activated mPEG reagents, and hence, are not amenable for PEGylation. In the same fashion, glycine does not have a functional side chain, and asparagine and glutamine have relatively non-reactive amide side chains. All the remaining amino acids have side chains with pendant functional groups that could be manipulated for chemical conjugation with adequately substituted mPEG reagents. In this regard, it has been well documented in the last thirty years that the most suitable amino acids for PEGylation are (in order of suitability): Lys>Cys>Tyr>His>Asp>Glu>Arg>Trp>Ser>Thr>Met. The terminal amino group and  $\epsilon$ -amino group of lysine amino acids of peptides and proteins have traditionally been exploited for the attachment of mPEG using different electrophilic reagents. In fact, and as will be shown later, a good part the chemistry of PEGylation involves the reaction of these nucleophilic amino groups to yield stable amide, amine and urethane bonds. The primary  $\epsilon$ -amino group of lysine is good nucleophile at pH above 8.5 ( $pK_a$  9.4), while terminal amino groups are less reactive, since they are close to the electrophilic carboxyl group ( $pK_a \approx 1$  to 2 pH units lower than the  $\epsilon$  amino group of lysine), but they are able to react with electrophiles at pH of around 7. This  $pK_a$  difference has been exploited to achieve more site-selective PEGylation reactions. In addition, lysine is widely present in most proteins, making it likely for the conjugation of polypeptides from very different origins and structures. The free thiol group of cysteine is an excellent nucleophile, especially at high pH, that has intensively been used for the conjugation of proteins by mean of formation of stable thioether bonds. However, only a limited number of cysteine residues are usually available in most proteins and, for this reason, PEGylation of native thiol groups has been explored to a lesser extent in earlier developments.

The conjugation of carboxyl groups with mPEG derivatives has scarcely been used for bioconjugation. The conjugation of carboxyl groups with mPEG amines by diimide activation is difficult since this strategy can lead to the undesired cross-linking of the biomacromolecule. This problem has been surpassed using mPEG hydrazide under a specific set of reaction conditions [25]. Moreover, the

conjugation at carboxylic groups has been mostly applied for the PEGylation of small drugs. The hydroxyl groups of serine, threonine and tyrosine, although nucleophilic, are not as reactive as amino group towards electrophilic mPEG derivatives. In fact, some of the reactions of these hydroxyl groups with electrophilic mPEG derivatives may lead to the formation of ester linkages, which are unstable under physiologic conditions. The guanidine moiety of arginine is a strongly nucleophilic group whose application in PEGylation with electrophilic mPEG reagents is limited, due to its high  $pK_a$ ; *i.e.* the conjugation reaction should be performed at too high pH values in order of having significant amounts of the unprotonated nucleophilic group. However, a site-selective method for the conjugation of arginine has been recently accomplished (see section 7 of this article).

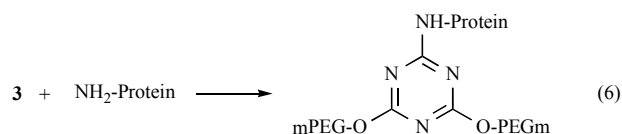
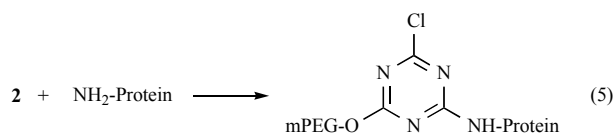
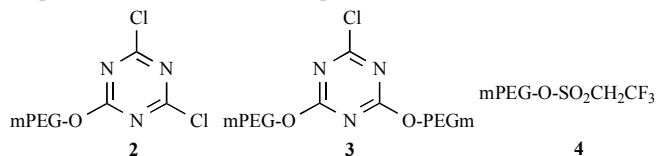
In the forthcoming pages we present some of the historically most important reactive mPEGs that, owing to their intense or relevant application, are considered nowadays as standard reagents for PEGylation chemistry. Key references on laboratory synthesis of mPEG reagents are provided, together with the chemistry supporting the PEGylation reaction, as well as exemplary applications. Actual and future trends on PEGylation are also discussed.

### 4. FIRST GENERATION MPEG REAGENTS

The first generation of PEGylation reagents were restricted to commercially available low molecular weight PEGs. Furthermore, the PDI of the starting mPEG raw materials were high, yielding ultimately heterogeneous conjugates. The diol content was also high, which afforded homobifunctional reagents that led to the cross-linking of biomacromolecules, and to the formation of inactive protein aggregates. Although novel, the chemistry of the PEGylation reactions was simple, and the high reactivity mPEG derivatives made the conjugation reactions essentially non-selective, affording heterogeneous and multi-PEGylated conjugates, many of which involved the formation of hydrolytically unstable bonds. In spite of these difficulties, the emerging technology quickly allowed to two PEGylated products to reach the market, namely, Adagen<sup>®</sup> and Oncaspar<sup>®</sup>. Adagen<sup>®</sup> is a PEGylated form of bovine adenosine deaminase approved by the FDA in 1990 for the treatment of severe combined immunodeficiency. This conjugate is obtained by random PEGylation with mPEG-succinimidyl carbonate (5 kDa). In the same fashion, Oncaspar<sup>®</sup> is a PEGylated form of asparaginase obtained by random PEGylation with the same reactive mPEG, which is used for the treatment acute lymphoblastic leukemia [26].

The first generation of reactive mPEG reagents for protein and peptide modification were directed towards the attachment of the polymer chain at nucleophilic amino groups. In this regard, the PEGylation reactions involved most often alkylation reactions to form stable secondary amine (equation 2) and acylation reactions to form amide (equation 3) or urethane (equation 4) bonds.

In the PEGylation of albumin and bovine liver catalase, mPEG-dichlorotriazine **2** was used as an electrophilic mPEG reagent [23,24]. Compound **2**, which can be obtained by the reactions of mPEG or mPEG alkoxide with cyanuric chloride, reacts in a non-selective fashion with the nucleophilic groups present in proteins, such as lysine, tyrosine, histidine, cysteine and serine, yielding linkages that retain the same charge of the native polypeptide (equation 5). PEGylation reactions with **2** and related compound **3** are performed at neutral or basic pH (7.4-10.0).



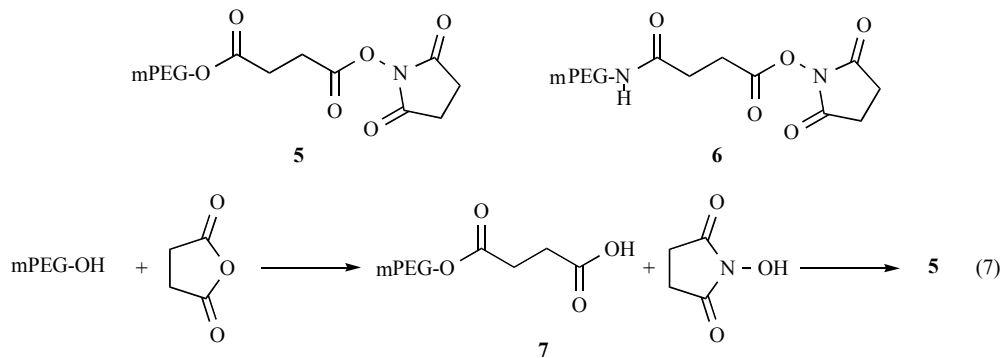
Although the remaining chloride of **2** is less reactive than the first for a second substitution, the reactivity is high enough to obtain undesirable cross-linking of the proteins. Examples of proteins that have been PEGylated using reactive mPEG **2** includes ovalbumin [27,28], bovine serum albumin [23], superoxide dismutase [29], elastase [30], acyl-plasmin-streptokinase complex [31], trypsin [32], silk fibroin [33] and immunoglobulin [34] amongst others. The Y-shaped, branched mPEG **3** is a more convenient reagent than **2** as the decreased reactivity obtained upon replacement of two chlorine atoms yields more selective PEGylation reagents [35]. Reactive mPEG **3** has two polymer chains on one reagent molecule, allowing the attachment of two mPEG chains on each conjugation site, decreasing the possibility of inactivation of the biomacromolecule activity (equation 6). Branched polymer **3** was initially prepared by reaction of mPEG with cyanuric chloride in benzene using  $\text{Na}_2\text{CO}_3$  as base. The activated polymer was successfully used in the conjugation of *E. coli* asparaginase [36]. It was later shown that this method affords a mixture of linear and branched polymers **2** and **3**. An improved method that affords **3** as the only product has also been provided [37]. Some further examples of use of branched mPEG **3** includes the PEGylation of ovalbumin and phenylalanine ammonia-lyase [38,39].

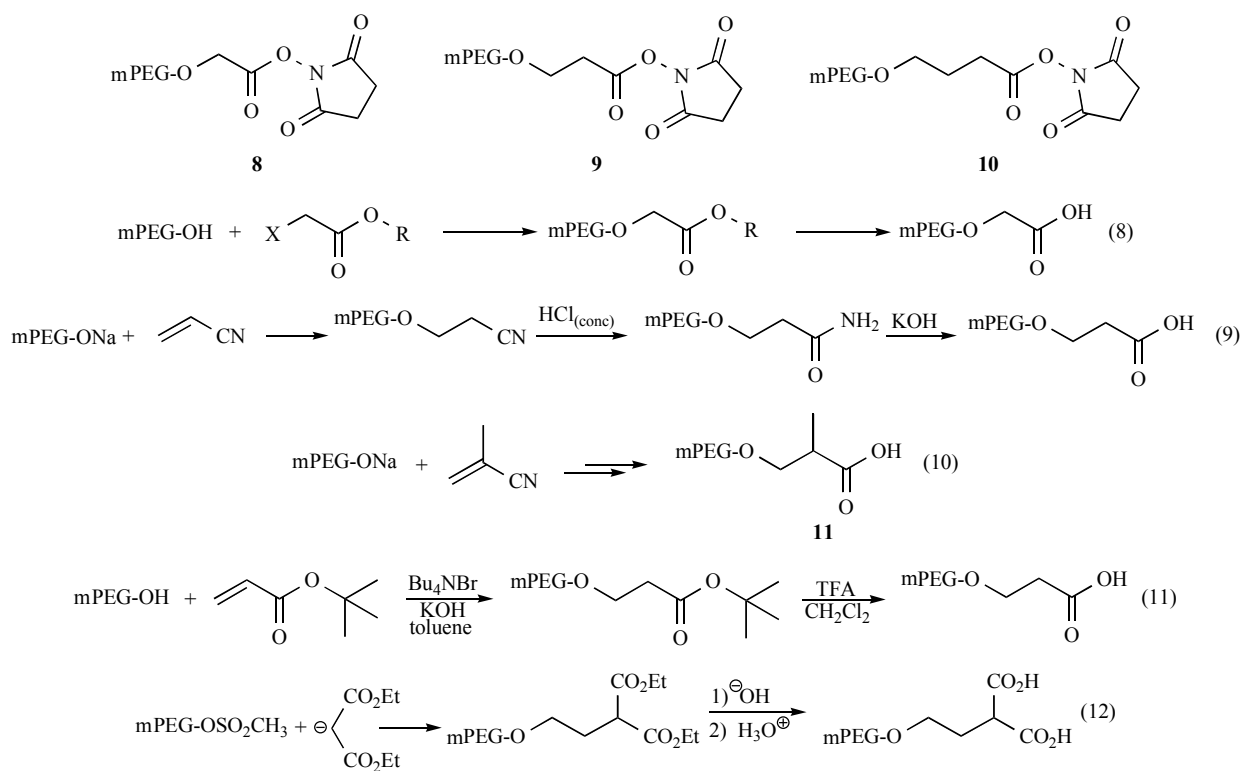
mPEG tresylate (mPEG-2,2,2-trifluoroethanesulfonate) **4** has successfully been employed as an electrophilic mPEG reagent for the conjugation of proteins by mean of alkylation reactions (equation 2). Activated mPEG **4** can easily be prepared by the reaction of mPEG with tresyl chloride in the presence of pyridine [40]. The alkylation of the amino groups of peptides and proteins proceed under mild conditions, rendering conjugates that retain the net charge of the native biomacromolecule [41]. A detailed kinetic study on the coupled hydrolysis-nucleophilic substitution reactions of PEG bis-tresylate with amines was performed by Griffith and co-workers [42]. In this work it was indicated that low temperatures and pH around 8.0 are advisable for secondary amino group PEGylation, while the pH should be raised to around 8.8 for the conjugation to the  $\epsilon$  amino group of lysine. Although these PEGylation reactions yield stable secondary amine bonds, it has been indicated that **4** reacts in non-specific fashion, yielding a mixture of products that can involve unstable bonds, originating non-well defined mixture of conjugated adducts [43]. For this reason, the use of mPEG **4** has been almost abandoned. Representative proteins that have been conjugated to mPEG by mean of tresylate **4** include lipases [44], alkaline phosphatase [45], albumin [40], granulocyte-macrophage colony stimulating factor [46], insulin [47] and B-deleted recombinant coagulation factor VIII [48]. Related mPEG tosylates are well known activated polymers that have been used, for example, in the PEGylation of trypsin [49].

mPEG hydroxysuccinimidyl succinate **5** is a reagent that has widely been used for the conjugation of several different proteins. Reactive mPEG **5** is more selective than derivatives **2-4** and it reacts selectively with primary amino groups yielding stable amide bonds (equation 3). A half-life of about 20 minutes at pH 8 for **5** has been reported [50].

Succinimidyl succinate **5** can easily be obtained by nucleophilic ring opening of succinic anhydride to give succinic acid derivative **7**, followed by diimide mediated esterification with *N*-hydroxysuccinimide (equation 7) [51]. The nucleophilic ring opening reaction can be accelerated using DMAP as nucleophilic acylation catalyst. A melt synthetic process for the preparation of this reagent has also been reported [52].

The ester linkage between the polymer chain and the succinate moiety is very sensitive to hydrolysis under physiological conditions, and the beneficial effect obtained by PEGylation can be lost due to hydrolysis of the ester bond. In fact, the conjugation of activated **5** with proteins is nowadays recognized as an early development in releasable PEGylation. In addition, it has been indicated that the succinate moiety remains linked after hydrolysis of the polymer chain, and that it can behave as a hapten, increasing the immunogenicity of the protein [53]. Succinimidylation of mPEG-NH<sub>2</sub> affords reactive mPEG **6**, in which the polymer chain is at-





tached to the succinic acid residue by hydrolytically stable amide bonds [20]. In another approach, mPEG was attached to different amino acids, usually by means of stable urethane bonds, and an appropriately activated carboxyl group was employed to link the polymer (with spacer) to the macromolecule [16]. In this regard, the activated mPEG bearing a norleucine moiety is particularly important since it can be used to study the degree of PEGylation by amino acid analysis of the mPEG-protein conjugates [54]. Other amino acids, including  $^{14}\text{C}$  or tritium labelled glycine, tryptophan and  $\beta$ -alanine, have also been employed in bioconjugation for different applications. In addition, Met-NLe and Met- $\beta$ Ala dipeptides were used for the identification of the PEGylation sites [55].

The ease of the preparation of reactive mPEG **5** stimulated its evaluation for conjugation to several different proteins. To date, the most relevant application of **5** is its use in the preparation of commercially available Adagen<sup>®</sup> (adenosine deaminase [56]) and Oncaspar<sup>®</sup> (asparaginase [57]). Further examples include the PEGylation of peptides [58], alkaline phosphatase [45], rHuG-CSF [59], arginine deiminase [60] and other proteins [61].

Activated polymers **8-10** are a related family of mPEG derivatives to **5**, on which the polymers are attached to the spacers by hydrolytically stable ether bonds. Compounds **8-10** react with primary amino groups, such as the  $\epsilon$  amino group of lysine, yielding stable amide bonds (equation 3). Reactive mPEGs **8-10** are obtained using diimide activated esterification of the corresponding mPEG-acetic, propionic and butyric acid derivatives with *N*-hydroxysuccinimide (Steglich type esterifications). The syntheses of these useful derivatives are not easy and, given their utility for the PEGylation of therapeutically relevant proteins, most of the methods for their preparation have been published in the patent literature. mPEG-acetic acid has early been employed in PEGylation technology. Although mPEG-propionic and butyric acid derivatives are usually considered as second generation reagents, they are included here for comparison purposes.

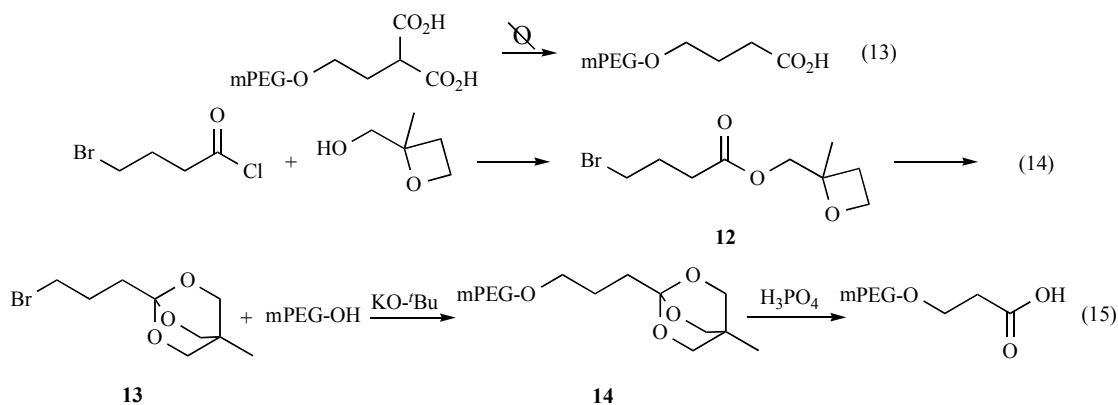
The direct oxidation of mPEG with  $\text{KMnO}_4$ , as well as the more selective stepwise oxidation with  $\text{MnO}_2$ , followed by  $\text{H}_2\text{O}_2$ , have both been proposed as synthetic methods for the preparation of mPEG-acetic acid **8**. However, a straightforward and more convenient synthesis of mPEG-acetic acid involves the reaction of mPEG alkoxide with alkyl 2-haloacetate, followed by hydrolysis (equation 8). mPEG alkoxides can be obtained by reaction of the polymer with sodium or potassium *tert*-butoxide, sodium naphthalenide or butyllithium [20]. Most often, mPEG alkoxides are obtained by an acid-base reaction with *tert*-butoxide in *tert*-butanol, and then reacted with *tert*-butyl 2-bromoacetate (equation 8, where X: Br and R:  $^t\text{Bu}$ ). Although the hydrolysis of intermediate mPEG-ester can be done with trifluoroacetic acid, it has been indicated that the reaction is best performed under standard basic conditions that prevent the degradation of the polymer chain. For this method, yields of up to 100% have been reported [62].

mPEG propionic acid has been prepared by conjugate addition of mPEG alkoxide to acrylonitrile, followed by hydrolysis in strongly basic conditions [63,64]. Using mPEG of 20 kDa it was shown that this method affords 68% of the required polymer derivative (equation 9) [65].

Following a similar strategy, but employing  $\alpha$ -methylacrylonitrile as acceptor,  $\alpha$ -methylpropionic mPEG acid derivative **11** can be obtained (equation 10) [66,67].

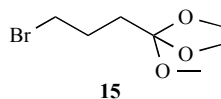
A related method involves the use of *tert*-butyl acrylate as electrophilic acceptor (equation 11). The method works well with mPEGs of 20 and 30 kDa (68%) and it has been indicated that the reaction does not proceed in the absence of the quaternary ammonium salt as phase transfer catalyst (equation 11) [65].

mPEG-butyric acid has been prepared by nucleophilic substitution reaction using diethylmalonate and mPEG-methanesulfonate as substrate (equation 12). Hydrolysis in alkaline media followed by decarboxylation rendered the target mPEG-acid derivative (equation 13) [63].



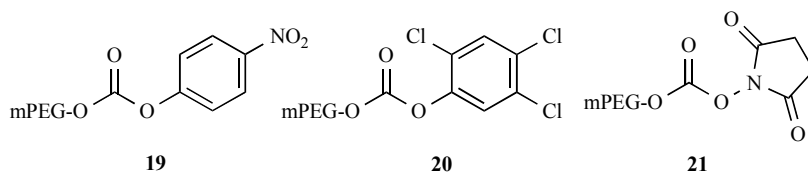
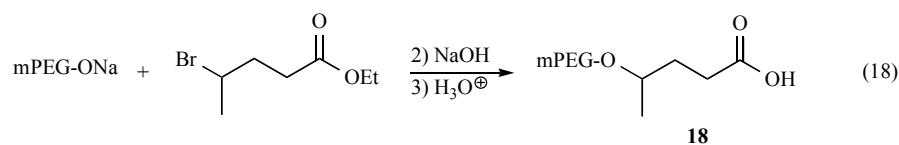
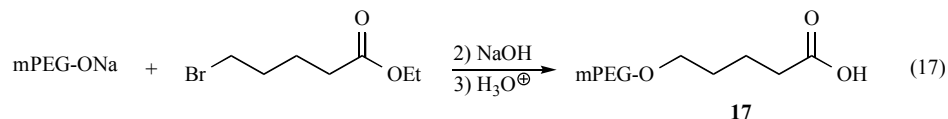
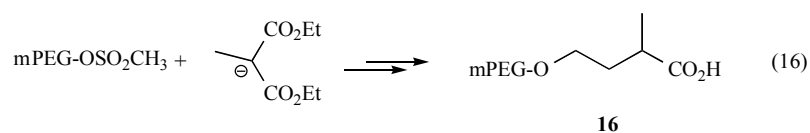
Other interesting syntheses of mPEG butyrinic acid have also been published in the patent literature [68]. The esterification reaction of 4-bromobutanoyl chloride with 3-methyl-3-oxetanemethanol affords ester **12**, which was converted to the bicyclic compound **13** via  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  complex. Nucleophilic displacement with an mPEG alkoxide yielded protected acid **14** that was converted to mPEG-butyrinic acid under acidic conditions (equations 14-15).

Commercially available trimethyl 4-bromo-*ortho*butyrate **15** was used as electrophilic starting material in a similar substitution reaction. Halides **13** and **15** are synthetically equivalent to 4-bromobutyrate esters, but they lack acidic protons in the  $\alpha$  position that may compete through acid-base reaction with the nucleophilic substitution



It has been indicated that related mPEG-butyrinic acid derivatives bearing a methyl group in the  $\alpha$ -position (**16**) can be prepared following the reaction sequence presented in equations 12-13 (equation 16).

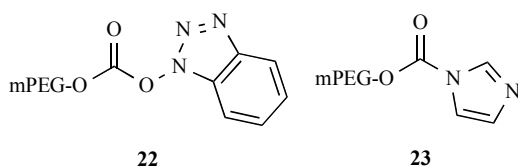
In addition, it has been shown that mPEG-valeric acid (**17**, equation 17) and mPEG-butyrinic acid derivative **18** (equation 18) can be obtained by nucleophilic substitution reactions using mPEG alkoxides with ethyl 5-bromovalerate and ethyl 3-bromobutyrate as substrates [69].



The reactivity of activated mPEG acids **8-10** towards the amino groups of peptides and proteins depends on the length of the hydrocarbon spacer moiety. Thus, the longer the the hydrocarbon chain between the polymer and the reactive center, the lower the reactivity towards the amino group and hence, the higher the selectivity of the PEGylation reaction. The hydrolysis rates of derivatives **8-10** and of *N*-hydroxysuccinimide esters of **11** and **18** and related compounds correlates well with this reactivity tenet. For example, the half-lives of compounds **8-10** and of *N*-hydroxysuccinimide esters of **11** and **18** are 0.75, 16.5, 23.3, 33 and 44 minutes respectively (25°C, pH 8) [63,65].

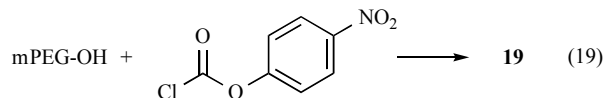
mPEG-acetic acid **8** and mPEG-propionic acid derivative **9** have extensively been employed in bioconjugation of relevant proteins, such as methioninase [70], insulin [71], murine GM-CSF [72], glucagon-like peptide 1 [73], protamine [74], salmon calcitonin [75],  $\alpha$ -lactalbumin [76], recombinant human arginase I [77], concanavalin A [78], pretreatment of allografts [79], adenovirus vectors [80,81], platelet CD42a [82], red blood cells [83], formate dehydrogenase [84] and antibodies against glutamic acid decarboxylase [85-87].

An example of the use activated mPEG-butyrinic acid **10** includes the PEGylation of erythropoietin [88]. This conjugate, commercially available as Mircera<sup>®</sup>, is used for the treatment of anaemia associated with chronic kidney disease.



mPEG carbonates **19-22** and mPEG-carbonylimidazole **23** belong to a closely related family of activated polymer derivatives that react with primary amino groups of peptides and proteins forming stable urethane bonds (equation 4). The reactivity of these compounds towards the amino groups depends on nature of the leaving group, and this can affect not only the selectivity, but also the degree of conjugation. Kinetic data for hydrolysis rates of some of these derivatives have been published. For example, the half-life of compounds **21-23** are as follow: **21**, 20.4 min (*pH* 8, 25°C), **22**, 13.5 min (*pH* 8, 25°C), and of around 10 h for **23** (*pH* 8, 25 °C). From a practical point of view, the reactivity order of derivatives **19-24** is **22>21>19>20>23**. Although all these mPEG reagents have intensively been used in early developments, the actual trend in PEGylation reactions involving urethane bond formation has moved towards derivatives **22** and **21**, since they show a good balance between the reactivity and selectivity, but lacking from the toxicity problems found with reagents **19** and **20**.

mPEG-4-nitrophenyl carbonate **19** has been prepared by the reaction of mPEG with 4-nitrophenylchloroformate under different conditions (equation 19). Veronese and co-workers initially prepared **19** in acetonitrile using pyridine as base [89]. Variations of this method include the reaction performed in dichloromethane with pyridine and triethylamine [90,91]. Although useful, it has been indicated that these methods afford an unknown mPEG-amine as impurity. It was suggested that pure **19** can be prepared upon reaction of mPEG with 4-nitrophenylchloroformate in toluene with tripropylamine [92] or with 4-dimethylaminopyridine in dioxane [93].

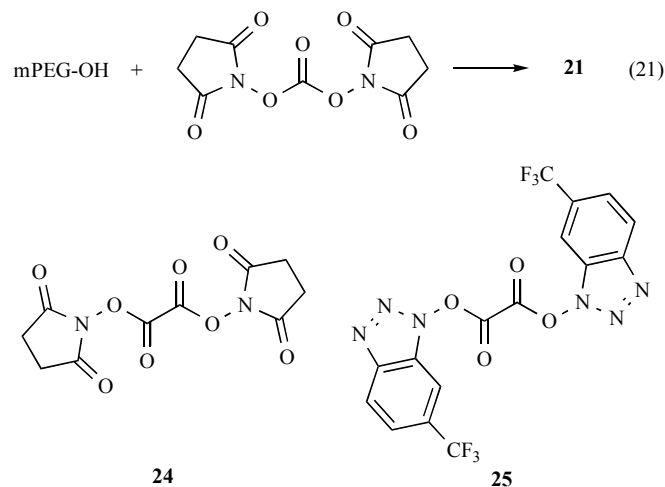


Representative applications of the use of **19** include the PEGylation of ribonuclease and superoxide dismutase [89], doxorubicin [94], recombinant mammalian urate oxidase [95], uricase [96] and lysozyme [97].

Reactive mPEG **20** can be prepared following the same method used for the synthesis **19**, but employing commercially available 2,4,5-trichlorophenylchloroformate [89].

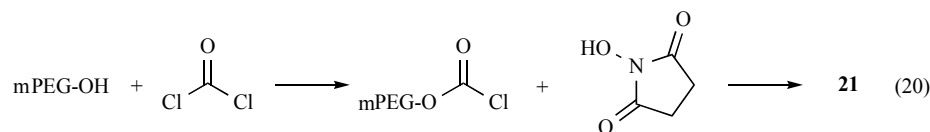
mPEG-succinimidylcarbonate **21** is a well known activated mPEG that has been employed in the conjugation of several different peptide and proteins, such as asparaginase [98], trypsin [99], chymotrypsin [100], bacteriorhodopsin [101], salmon calcitonin [102], interferon  $\alpha$ -2b [103], arginine deiminase [104] and recombinant human arginase I [77], among other bio-macromolecules [105]. Compound **21** reacts with primary amino groups yielding stable urethane linkages and with the hydroxyl group of tyrosine, threonine and serine to afford hydrolytically unstable carbonate bonds. Moreover, it has been indicated that, depending on the *pH* of

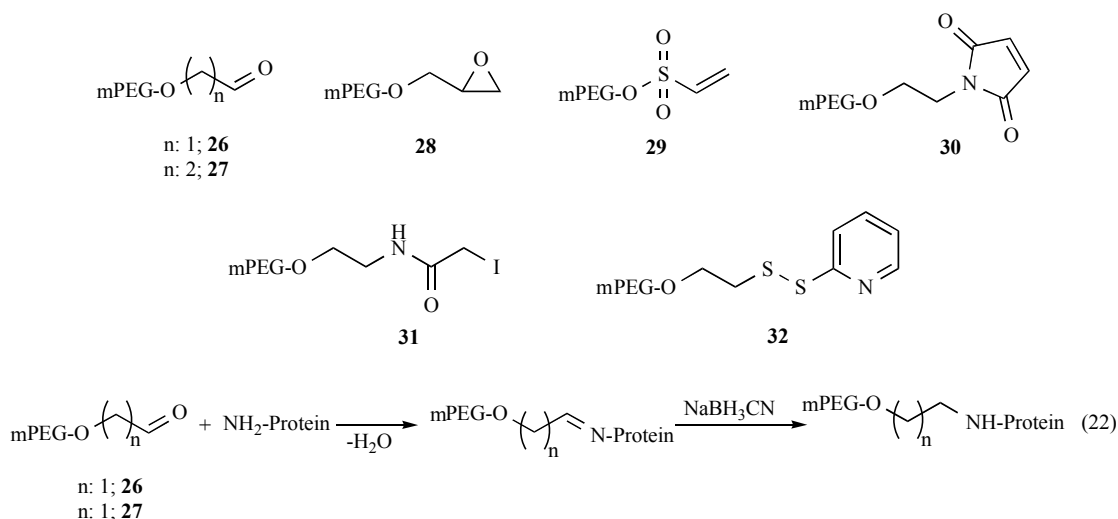
the conjugation medium, **21** can react with the imidazole ring of histidine to yield unstable urethane linkages. For example, activated mPEG **21** reacts with interferon  $\alpha$ 2-b at *pH* 6.5 to give a conjugate that has 47% of PEGylation at histidine residues and 32% at lysine amino acids [106]. The bioconjugate obtained in this conditions and employing mPEG of 12 kDa is commercially available (PEGintron<sup>®</sup>) and is actually used for the treatment of hepatitis C and melanoma (Sylatron<sup>®</sup>). PEGylation at histidine residues to yield unstable urethane bond is also considered as a releasable or reversible conjugation method. Activated mPEG **21** is more reactive than derivatives **19** and **20** and it can be obtained by reaction of mPEG with phosgene followed by *N*-hydroxysuccinimide (equation 20) [56]. It has been proposed that this reaction is complete affording **21** in almost quantitative yields [107]. A variation of this method, in which toxic phosgene was replaced by more friendly triphosgene, has been published in the patent literature [108]. In addition, activated mPEG **21** can be prepared by the reaction of mPEG with commercially available *bis-N*-hydroxysuccinimidylcarbonate (equation 21) [109]. An alternative method, in which mPEG is reacted with *N,N'*-disuccinimidyl oxalate (**24**) in the presence of pyridine, has also been published. The same reaction, but using 1,1'-bis[6-(trifluoromethyl)benzotriazolyl]oxalate (**25**), was used for the synthesis of an activated mPEG closely related to hydroxybenzotriazolyl derivative **22**.



1-Benzotriazolyl derivative **22** is a reactive mPEG that has found application in the PEGylation of D-glucosamine, sodium heparin, human serum albumin [110], fibrin biomatrix [111], lysozyme [112] and peptides among other targets [113]. In similar fashion than **21**, activated mPEG **22** reacts with histidine and tyrosine residues to afford unstable urethane and carbonate linkages. mPEG-benzotriazolyl carbonate **22** can be prepared following similar strategies than those depicted in equations **20** and **21**, *i.e.* the reaction of mPEG with phosgene followed by reaction with 1-hydroxybenzotriazole [110], or the reaction with *bis*-1-benzotriazole carbonate in presence of pyridine as base [112].

mPEG-carbonylimidazole **23** is a mild PEGylating reagent that has been used since the mid 1980's for the conjugation of proteins, amongst other targets. It has been claimed that, due to its low reac-





tivity, activated mPEG **23** behave as more selective reagent than **21** and **22**. In agreement with this reactivity, usually low to moderate yields of conjugation are obtained with reagent **23**. Compound **23** was initially prepared by reaction of mPEG with *N,N'*-carbonyl-bis-imidazole (CDI) in dioxane at 37°C [114]. It was later demonstrated that the yield of the activated polymer obtained following this method was around 42% [115]. An improved method that affords activated mPEG **23** with quantitative yields, which involves the use 4-dimethylaminopyridyne in toluene, has also been published [115]. Similarly, an excellent yield of activation was obtained for the reaction of mPEG (20 kDa) with CDI in THF at 60°C [116].

Activated polymer **23** is a mild PEGylating reagent that has been applied in the conjugation of superoxide dismutase [114],  $\alpha$ 2-macroglobulin [117], lactoferrin [114], streptokinase [117], alkaline phosphatase [45], IgG [118] and urokinase [119], among others.

## 5. SECOND GENERATION mPEG REAGENTS

The syntheses of first generation mPEG reagents involve usually simple and straightforward reaction steps with very reactive reagents that were aimed to convert the relatively unreactive hydroxyl group of mPEG to electrophilic derivatives. However, it has been indicated that most of the first generation synthetic methodology are problematic with respect to high molecular weight mPEGs, affording heterogeneous mPEG reagents that contain variable amounts of unreacted polymer. In addition, the high diol content of the initially available mPEGs yielded high molecular weight bifunctional polymers that ultimately led to undesirable cross-linking of the proteins. The vast majority of first generation chemistries were directed towards the conjugation of electrophilic mPEG reagents with amino groups and, mainly, with the primary  $\epsilon$  amino group of ubiquitously present lysine amino acids. Furthermore, the high reactivity of the first generation reagents conducted to unselective random PEGylation reactions, which, in addition, can afford poly-PEGylated conjugates with decreased biological activity, or to polymer conjugates bearing unstable links.

Nowadays, the improvement on the methods for the preparation of mPEGs allows for the easy commercial availability of high molecular weight mPEGs with adequate polydispersities and low diol contents. This fact, together with the huge efforts made in the development of more selective PEGylation reactions, allowed the emergence of a second generation of mPEG reagents that lack of some of the drawbacks associated to the first generation chemis-

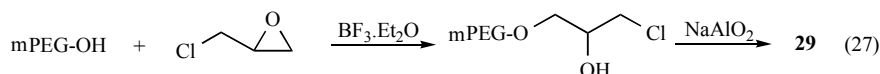
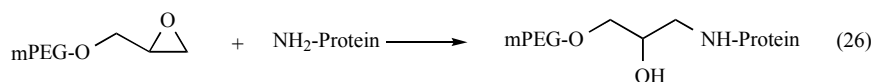
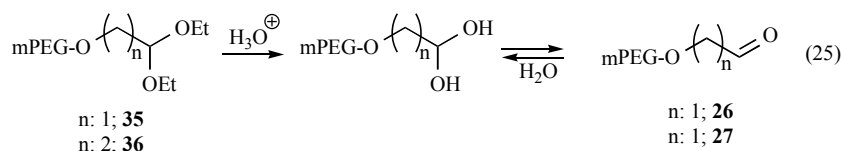
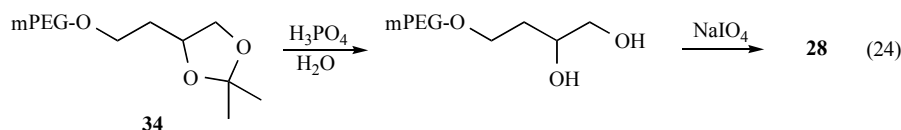
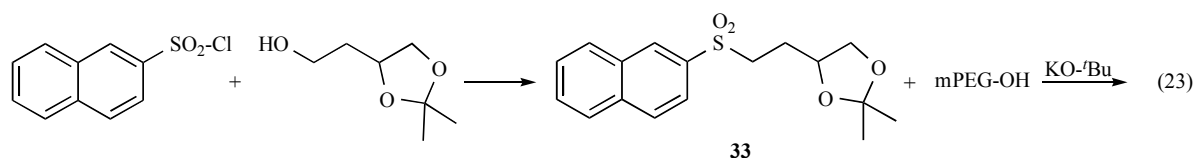
tries. Although nucleophilic amino groups remained as important targets for the attachment of electrophilic PEGs, some second generation mPEG derivatives were designed to anchor PEG to the thiol group of cysteine amino acids, which are amenable residues for site-selective PEGylation.

The reductive alkylation of primary amino groups is a well known two-step method used in organic synthesis for the preparation of amines without over-alkylation. Under slightly acidic conditions (*pH* 4.5-5.0) a carbonyl compound reacts with a nucleophilic amine to yield a Schiff base that can be isolated and converted to the amine by mean of an adequate reducing reagent, such a NaBH<sub>4</sub> or LAH. One-pot versions of the same reaction using NaBH<sub>3</sub>CN and NaBH(OAc)<sub>3</sub> are also well established methods. In PEGylation chemistry, the reductive alkylation reaction has successfully been used to achieve mild and quite site-selective conjugation reactions using mPEG-acetaldehyde **26** and mPEG-propanaldehyde **27** with NaBH<sub>3</sub>CN as the reducing reagent (equation 22). mPEG-propanaldehyde **27** is preferred over acetaldehyde **26** since the later is unstable and dimerizes by aldol condensation reaction.

The reaction of electrophilic mPEG **26** and **27** with amino groups depends upon i) the nucleophilicity of the particular amino group involved in the condensation reaction, and ii) the *pH* at which the condensation reaction is performed. In this regard, the  $\epsilon$  amino group of lysine is more nucleophilic and basic than most of terminal (and  $\alpha$  carboxyl) amino groups. Under the slightly acidic conditions required for the condensation reaction, the more reactive  $\epsilon$  amino groups are in the protonated and non-nucleophilic form, allowing for relatively high site-specific PEGylations at terminal amino group. As with previous alkylating reagent, the net charge of the protein remains unchanged after the conjugation reaction.

mPEG-acetaldehyde **26** has been prepared by the oxidation of mPEG with pyridinium chlorochromate in methylene chloride, albeit in low yield (30%) [120]. Aldehyde **26** can also be prepared by Moffat-type oxidation reaction or by the nucleophilic substitution reaction of mPEG potassium alkoxide with chloro- or bromoacetaldehyde diethylacetal followed by acid hydrolysis [121-123]. In a similar approach, aldehyde **27** can be obtained by nucleophilic substitution reaction of mPEG alkoxide with 3-chloropropanaldehyde diethyl acetal followed by deprotection (50%) [124]. In the patent literature it has been claimed that 2-naphthalenesulfonyl chloride reacts with 2,2-dimethyl-1,3-dioxolane-4-ethanol, yielding an intermediate sulfonate **33** (equation 23). Sulfonate **33** is then reacted with the mPEG alkoxide to afford a protected mPEG-diol derivative





**34**, which after hydrolysis in acid conditions, followed by oxidation with  $\text{NaIO}_4$  gives mPEG-propionaldehyde **27** (equation 24) [125].

It has been indicated that acetals **35** and **36** are more stable and can be obtained with higher purities than the corresponding mPEG-aldehydes **26** and **27**. The diethylacetals **35-36** are converted to aldehyde hydrates under acidic conditions, and after adjusting the pH, they can directly be used for the conjugation reaction (equation 25) [126].

Examples of the use of aldehydes **26** and **27** includes the PEGylation of horseradish peroxidase [127], CD4 immunoadhesin [121], doxorubicin [128], recombinant human granulocyte colony-stimulating factor (Filgrastim, r-metHuG-CSF) [129,130], human megakaryocyte growth and development factor (MGDF) [131], lysozyme [132], tumor necrosis factor receptor type I [133], interferon  $\beta$ -1a [134], octreotide [135], recombinant human arginase I [77], recombinant erythropoietin [136], porcine follicle-stimulating hormone [137] and others [124,125].

mPEG-epoxide **28** is a mild PEGylating reagent that reacts with the amino groups of peptides and proteins, producing stable secondary amine bonds (equation 26). Due to the low reactivity of this electrophilic mPEG reagent, its application for the mPEGylation of biomacromolecules has been limited. In addition, it has been suggested that this reagent is not selective for amino group conjugation, since it reacts with hydroxyl, thiol and imidazole groups. However, mPEG-epoxides have found application in material and polymer science among other uses [138-142]. Proteins that have been PEGylated with this reagent include glutathione, bovine serum albumin, protein A, and monoclonal antibodies [143-145].

Epoxide **29** can be prepared by nucleophilic reaction of mPEG with the oxyrane ring of epichlorohydrin in presence of  $\text{BF}_3\cdot\text{Et}_2\text{O}$  complex. The intermediate alcohol product is then dehalogenated with sodium aluminate to afford the required mPEG reagent (equation 27) [146].

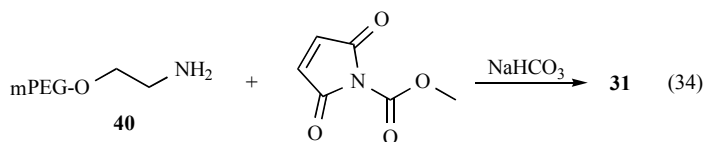
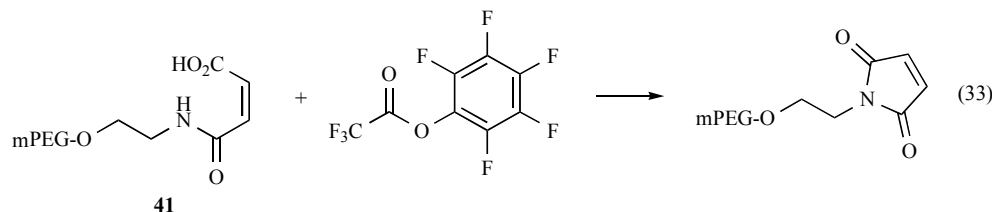
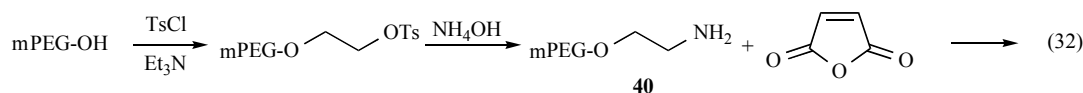
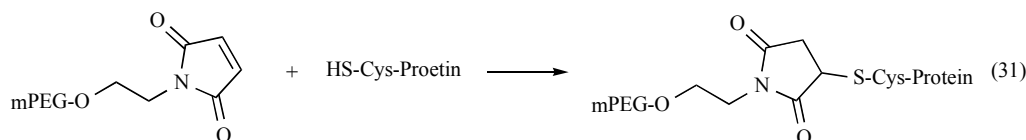
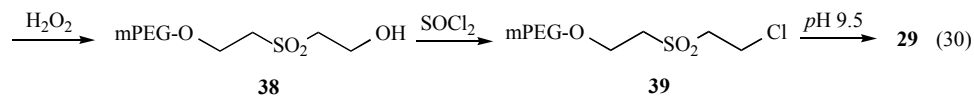
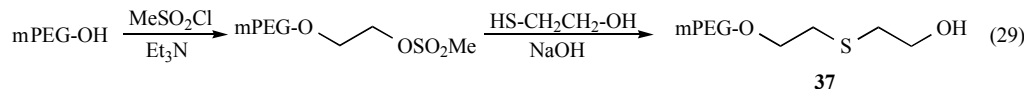
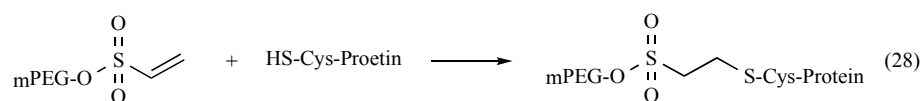
In the last 15 years, thiol group conjugation has evolved as a major approach for site-selective PEGylation. In this regard, cysteine amino acids are scarcely present in peptides and proteins and

the availability of this nucleophilic group at the surface of the protein can be limited. In addition, the SH moiety is one of most reactive nucleophilic groups present in biomolecules, that can react with electrophilic mPEG reagents at slightly acidic or neutral pH (6-7). At these pH values, nucleophilic amino groups are present in the protonated form that cannot compete with thiol nucleophiles. Cysteine is a hydrophobic amino acid that is usually located in the internal part of the protein, limiting its accessibility for conjugation. On the other hand, thiol group PEGylations at cysteine residues located in the surface of the protein usually do not compromise the protein bioactivity, taking full advantage of the benefits of PEGylation. In the case of the absence of cysteine residues in the native protein, one or more of these amino acids can be added by genetic engineering methods [147,148]. Although useful, this approach can lead to the formation of unwanted disulfide linkages in addition to protein dimerization. Furthermore, another approach to increase the amount of available cysteine amino acids for conjugation involves the reduction of protein disulfide bridges [149]. This strategy proved to be particularly well suited for the site-selective PEGylation of antibodies, the activity of which can critically be decreased by terminal amino group conjugation, among other proteins. Of course, these conjugation methods are site-selective when the protein possesses only one disulfide bridge or when only one of two or more of the disulfides bridges can be reduced selectively.

Activated mPEGs **29-31** are commercially available reagents that have been designed to selectively react with the thiol group of cysteine.

Vinyl sulfone **29** reacts selectively with SH of cysteine in slightly basic conditions (pH 7-8) yielding a stable thioether linkage (equation 28). However, it has been indicated that lysine conjugation also occurs at pH of around 9.3 using large excess of the PEGylating reagent.

mPEG-vinylsulfone **29** can be prepared following the method presented in equations 29-30. Mesylation of the mPEG starting material under standard conditions renders the corresponding methanesulfonate, that is then reacted with 2-mercaptoethanol under basic conditions to give sulfide **37** (equation 29). Oxidation



with  $\text{H}_2\text{O}_2$  yields sulfone **38** that by halogenation reaction with  $\text{SOCl}_2$  produces halide **39**. Under very mild basic conditions chlorosulfone **39** is dehalogenated to the target reagent with quantitative yield (equation 30) [150].

Recently, an extremely simple one-step “click” method for the preparation of the vinyl sulfones of mPEG, among other hydroxyl-terminated polymers, has been published. This reaction involves the solubilization of mPEG in alkaline solution followed by Michael-type addition of the alkoxide ions using divinylsulfone as acceptor to afford directly the desired reactive polymer [151].

mPEG-maleimide derivative **30** is a well known reagent that has been widely used for selective thiol group conjugation. In neutral or slightly basic conditions (pH 7.5-8.5) mPEG-maleimide reacts with the thiol group in a Michael addition reaction, yielding hydrolytically stable thioether linkages (equation 31). As with mPEG-sulfone **29**, side-reactions that arise from the addition reaction of amino groups are observed when the pH is increased, although at very low rates. It has been suggested that the rate of amino conjugation also increases if an organic co-solvent is added [16].

Activated mPEG **30** can be prepared as depicted in equations 32-33. Tosylation of mPEG under standard conditions followed by nucleophilic displacement with aqueous ammonia solution gives mPEG-amine **40** (equation 32). Acylation of **40** with succinic anhydride in a polar aprotic solvent yields acid derivative **41**, which

under harsh reactions conditions, is ultimately converted to the desired maleimide (equation 33) [152].

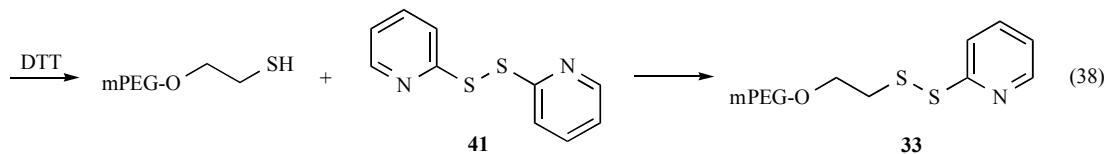
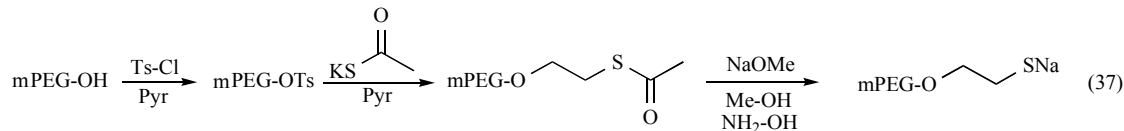
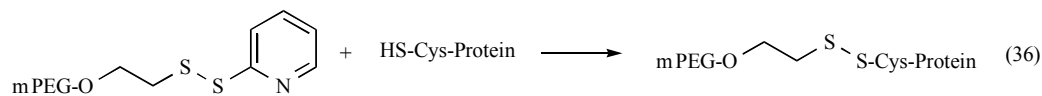
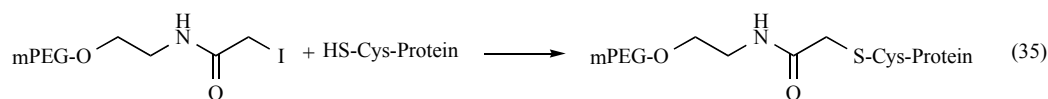
Although closely related methodology for synthesis of **30** has earlier been published, it has been indicated that these methods affords undesirable high amounts of side-products [153-155].

In the patent literature it has been claimed that mPEG-amine **40** reacts with *N*-methoxycarbonylmaleimide in aqueous basic solution to yield mPEG derivative **30** in one step (equation 34) [156].

Although **30** is, probably, one of the most widely used mPEG for thiol group conjugations, it has been shown that this compound is unstable in aqueous solution. Furthermore, it has been indicated that it cannot be stored for a prolonged period of time. Selected examples of PEGylation with **30** include the conjugation of interleukin-2, certolizumab pegol, staphylokinase, rHuGM-CSF, erythropoietin analogues, cyanovirin-N, prolidase, recombinant human arginase I, si-RNA, bone morphogenetic protein-2, exenatide analogues and aptamers, among many other examples [77,157-167].

Related to **29** and **30**, mPEG acrylate has also been used as Michael acceptor for selective cysteine PEGylation [168].

mPEG-iodoacetamide **31** is a mild PEGylating reagent that reacts with thiol group of cysteine aminoacids of peptides and proteins, yielding stable thioether linkages (equation 35). In the same fashion as shown for derivatives **30** and **31**, some loss in the selectivity of the conjugation reaction is observed as the pH is increased.



PEGylation reactions with derivative **31** are usually performed using excess of iodine in a darkened flask in order to limit the formation of iodine, which can react with other amino acids. mPEG-iodoacetamide can be prepared by the reaction of mPEG-amine **40** with iodoacetic anhydride in dioxane [154]. Among other examples, reagent **31** has been used for the PEGylation of acyl-coenzyme A and cholesterol acyltransferase [169].

mPEG-*ortho*pyridyldisulfide **32** (mPEG-OPSS) is the most selective reagent for thiol group conjugation. This derivative reacts with SH groups in acid and alkaline media via a thiol exchange reaction to afford a disulfide linkage that is unstable under reducing conditions (equation 36).

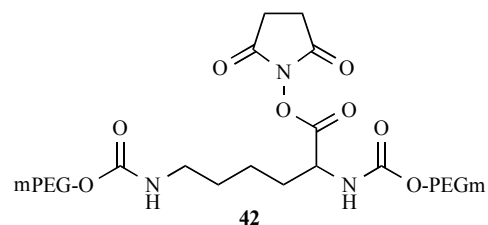
Several other mPEG-OPSS polymers bearing different spacers between the polymer chain and the pyridyl-disulfide moiety have been published, mainly in the patent literature [170].

mPEG-OPSS **32** has been prepared by the tosylation of mPEG followed by nucleophilic displacement with potassium thioacetate to give an intermediate thioester (equation 37). After methanolysis and treatment with dithiothreitol, the free thiol group was converted to the final product by reaction with *ortho*-bis-dithiopyridine **41** (equation 38) [171].

Examples of the application of **32** includes the PEGylation of interferon  $\beta$  and staphilokinase [158,171,172].

Branched mPEGs have received particular attention due the well documented improvements in the biopharmaceutical properties of several therapeutically relevant proteins. In this regard, it has been showed that branched mPEGs behave as if they have higher molecular weights than linear mPEGs of identical molecular weight. In addition, when branched mPEGs are used, more than one polymer chain can be linked onto each attachment site. The net result of these two effects translates into a better shielding of the protein, while at the same time the possibility of deactivation of the active site of the protein is diminished. Among the several different branched mPEG structures that have been prepared and evaluated, a branched mPEG bearing a lysine linker (**42**) proved to be particularly well suited for the conjugation of several different proteins, including: ribonuclease, catalase, asparaginase, trypsin [173], interferon  $\alpha$ -2a [174-176], interferon  $\alpha$ -2b [177], interferon  $\beta$ -1b [178], TNYL-RAW - Fc portion of human IgG1 [179], peptides targeting the human neonatal Fc receptor [180],  $\alpha$ -momorcharin [181,182],

erythropoietin [136], organophosphorus hydrolase [183] and lactoferrin [184] among several others.

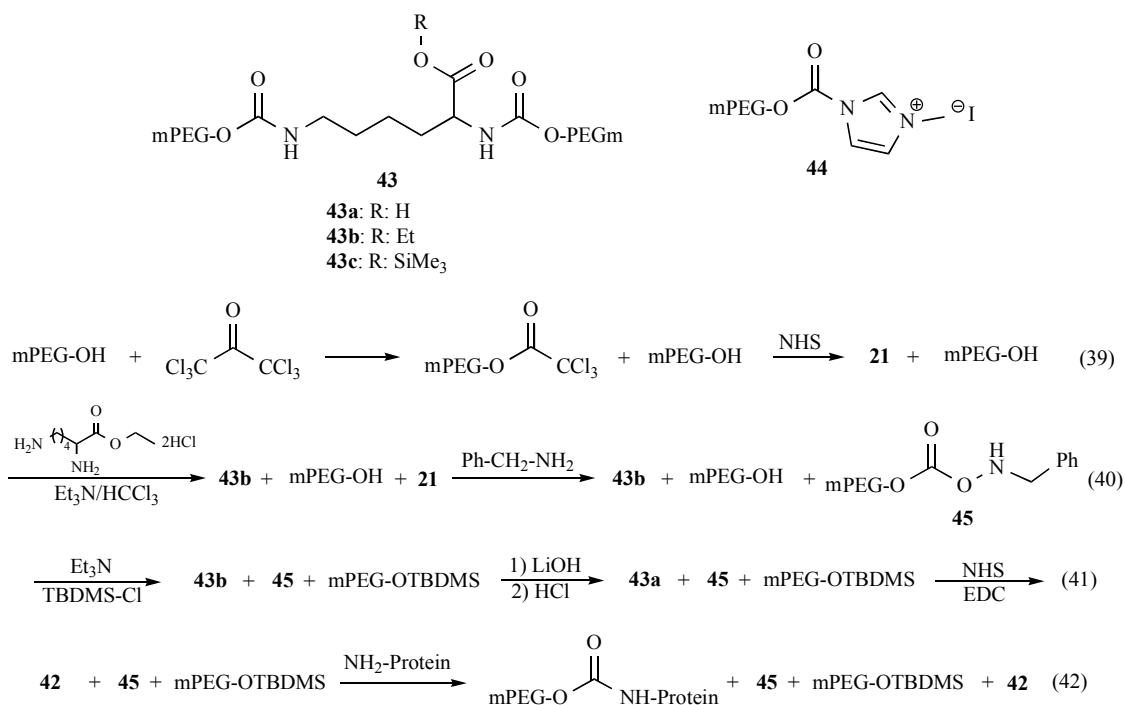


As with the *N*-hydroxysuccinimidyl esters **8-10**, mPEG derivative **42** reacts with the amino group of peptides and proteins yielding stable amide bond (equation 3).

Several methods for the synthesis of **42** have been published or patented. mPEG-*p*-nitrophenylcarbonate **19** was reacted with lysine in buffered aqueous media to yield mPEG-monosubstituted lysine, that was then treated with the same reagent in methylene chloride to yield **43a** in a two-step procedure. Given that the  $\epsilon$  amino group of lysine is more reactive than the  $\alpha$  amino group, this two-step procedure allows the preparation of branched **43a** bearing mPEG chains of different molecular weight [173]. A more convenient method proved to be the reaction of mPEG-*N*-hydroxysuccinimidyl carbonate **21** with lysine in buffered aqueous solution [173]. In these methods **43a** was purified by ion exchange chromatography and converted to **42** by diimide mediated esterification with *N*-hydroxysuccinimide [185]. The related **43b** has been prepared by reaction of lysine with **19** using pyridine as base [186].

A straightforward method for the synthesis **43a** has recently been published. *N*-Carbamoylimidizaolium iodide **44**, which can easily be obtained by the methylation of mPEG-carbonylimidazole **23**, was reacted with an organic soluble silylated lysine in DMSO/acetonitrile using di-*isopropyl*ethylamine as base, to yield **43c** in a single step. Compound **43c** was converted to **43a** during the aqueous workup of the reaction, and the product was purified by hydrophobic interaction chromatography [116].

An elaborate but useful method for the preparation and use of **42** in the PEGylation of proteins that avoids intermediate purification steps has been published in the patent literature [108]. mPEG was reacted with triphosgene to yield the activated mPEG derivative together with unreacted mPEG. The crude product mixture was



treated with *N*-hydroxysuccinimide (NHS) affording the required mPEG-NHS carbonate **21** (equation 39). The reaction mixture containing of **21** and unreacted mPEG was reacted with lysine ethyl ester di-hydrochloride to give a complex reaction mixture that contains protected branched mPEG **43b**, unreacted mPEG and activated **21** (equation 40). In order to deactivate **21**, the mixture was treated with benzylamine, which converted **21** into an unreactive carbamate **45** (equation 40). Upon reaction of this mixture with *tert*-butyl-dimethylsilane chloride (TBDMS-Cl), unreacted mPEG was transformed to the inert silyleter mPEG-OTBDMS (equation 41). Finally, saponification followed by esterification with NHS under diimide activation yields a mixture of products containing reagent **42**, which can directly be used for the PEGylation of proteins (equation 41 and 42). A noteworthy point is that only one purification step is performed after the conjugation reaction.

It is worth mentioning that four conjugates available in the marketplace are prepared using branched mPEG **42** (Pegasys<sup>®</sup> (interferon  $\alpha$ -2a), Macugen<sup>®</sup> (aptanib), Cimzia<sup>®</sup> (anti-TNF fab) and Omontis<sup>®</sup> (dimeric synthetic peptide for the stimulation of erythropoiesis).

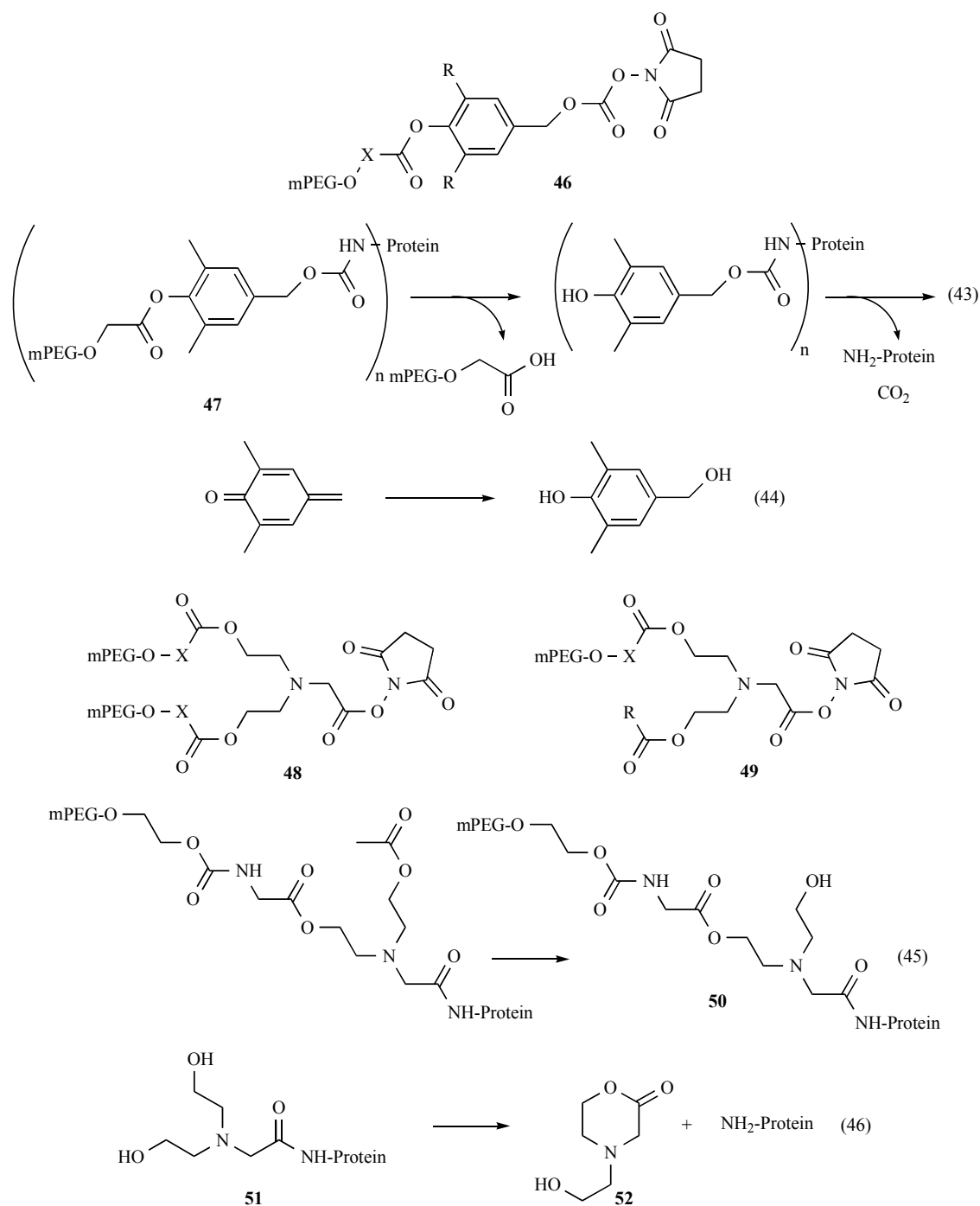
## 6. REVERSIBLE OR RELEASABLE PEGYLATION

As described earlier, one of the main goals of PEGylation is to covalently attach mPEG polymer chains to biomacromolecules. Such attachment should be stable under physiological conditions in order to increase the time of residence of the conjugate in the blood and, in this way, take full advantage of the benefits of PEGylation. On the other hand, the attachment of the polymer chain can decrease the activity of the protein by conjugation at the active site or by steric hindrance. These effects may be so profound that they can lead to drug inactivation. In order to overcome the drug inactivation by polymer conjugation, in the last few years different strategies for releasable or reversible PEGylation have been developed [187,188]. In releasable PEGylation, the polymer is bound to the drug or biomolecule through an unstable bond or a degradable linker which can be cleaved under specific conditions, releasing the native un-

modified drug at a controlled rate. In fact, in releasable PEGylation the drug-polymer conjugate is a prodrug from which the unmodified active drug is delivered after polymer release. In the particular case of proteins, this conjugate prodrug may show reduced activity or be completely inactive. In addition, since PEG is water soluble, this approach can be used to deliver poorly soluble drugs. Moreover, the careful design and selection of specific linkers allows tissue or organ site-specific drug release.

Esters, carbonates, carbamates, hydrazones and amides have all been used as hydrolytically or enzymatically degradable linkages for reversible PEGylation. These approaches proved to be particularly useful for the reversible PEGylation of small drugs, such as paclitaxel, camptothecin and podophyllotoxin, amongst others [189-191]. It has been shown that the design of the linker which brings together the protein and the polymer chain, forming a tripartate prodrug, is a key issue to provide specific initiation and accurate rates of release. As a result, releasable prodrugs from daunorubicin, vancomycin, amphotericin B, and doxorubicin with bicine and linkers for 1,4- or 1,6- benzyl elimination (BE) have all been prepared [192-195].

Benzyl elimination has been found to be particularly attractive for reversible protein PEGylation. In this approach, the polymer is initially released by the cleavage of a hydrolytically unstable linkage, followed by a well known fast 1,4- or 1,6- elimination reaction that renders the unmodified native protein (equations 43-44) [196,197]. The chemistry of the PEGylation reaction with these linkers involves the formation of urethane bonds between free amino groups (preferably  $\epsilon$  amino groups of lysine amino acids) of proteins and a reactive mPEG *N*-hydroxysuccinimidyl carbonate, such as **46**, in a similar fashion to that shown in equation 4. The kinetics of the release reaction can be finely tuned by the introduction of substituents in the *ortho* position to the hydroxyl group. For example, it has been shown that the presence of two methyl groups *ortho* to the hydroxyl functionality affords ester linkages that are relatively more stable than the unsubstituted counterparts. Furthermore, a wide variety of compounds related to **46** where the mPEG



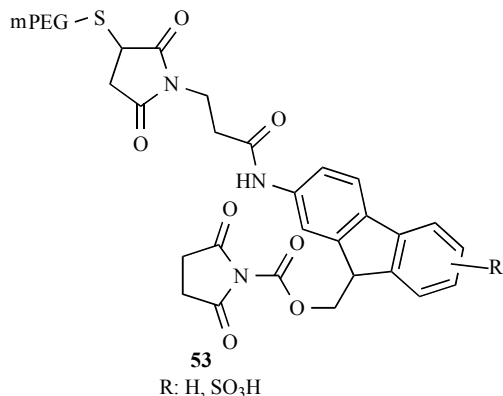
chains are linked by groups of different stability (trigger groups), and derivatives having different substitution patterns at the *ortho* position, have all been prepared and evaluated in reversible PEGylation. These trigger groups may influence the reactivity of the parent reactive mPEG with  $\epsilon$  amino groups of the specific protein. The derivative bearing two *ortho* methyl groups and the mPEG chain linked by an ester moiety (**47**) has proven to be particularly attractive for releasable PEGylation, as the hydrolysis rate is slower than with unsubstituted compounds.

Releasable PEGylation with branched (**48**) and linear (**49**) bicine linkers has emerged in the last few years as powerful method for the controlled delivery of small drugs and proteins [198,199]. The chemistry of the delivery with these linkers, using a linear de-

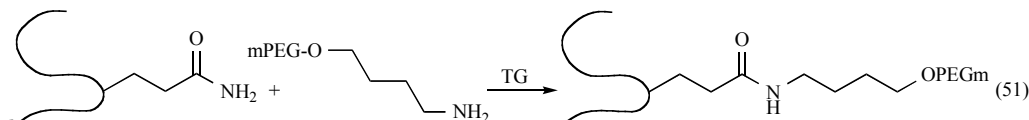
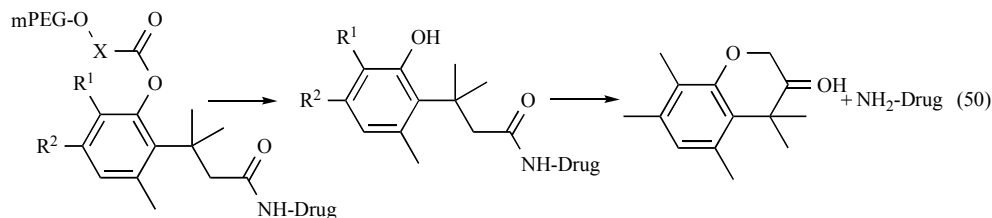
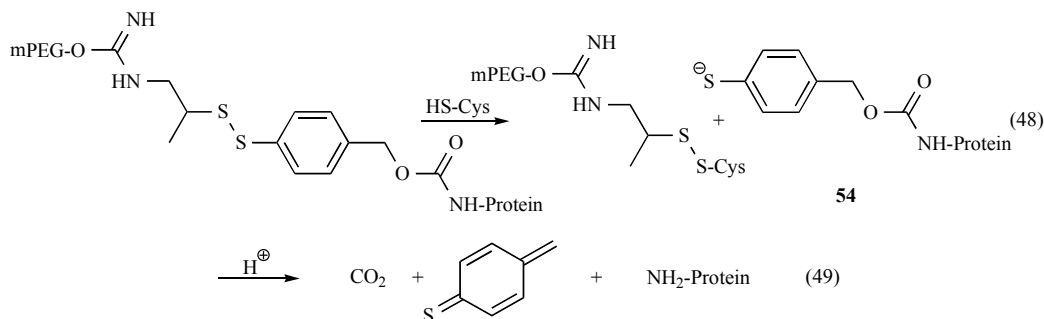
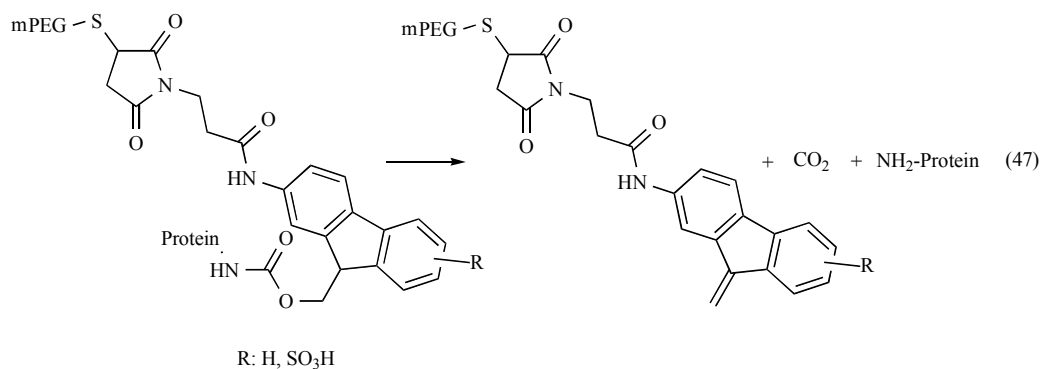
riivative **49**, is depicted in equations 45-46. The fast hydrolysis of the ester group affords the corresponding carboxylate and the hydroxyethyl derivative **50**, which further reacts with water at a controlled rate to yield *bis*-hydroxyethyl intermediate **51**. Intermediate **51** undergoes cyclization affording bicine **52** and delivering the native protein (equation 46).

It has been shown that the rate of release can be modulated upon modification of the heteroatom of the trigger (X in compounds **48** and **49**), by introduction of additional side chains or by increasing the steric hindrance. Branched bicine linkers **48** have proven to be interesting structures due to the advantages that branched structures have for protein conjugation.

The bifunctional linker **53** has been used for the attachment of mPEG of up to 40 kDa to proteins by means of urethane bond formation. The fluorenyl moiety releases the native unmodified protein at a slow controlled rate via hydrolysis (equation 47) [200]. This approach has been used, for example, for the reversible PEGylation of exendine-4 [200], human growth hormone, interferon  $\alpha 2$  [201], enkephalin [202], atrial natriuretic peptide [203] and insulin [204].



Cleavable disulfide bridges have also been exploited for the reversible PEGylation of cysteine residues of proteins. The underlying

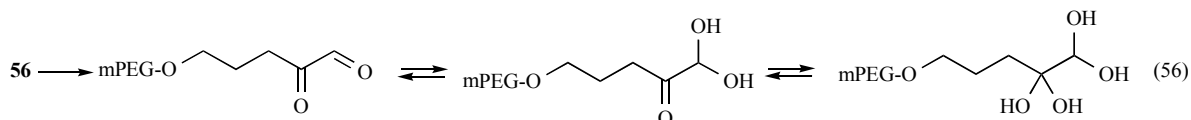
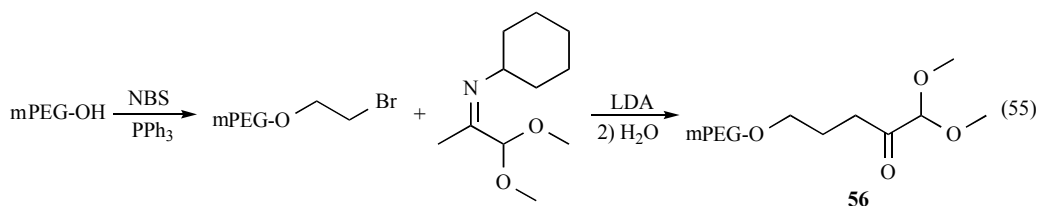
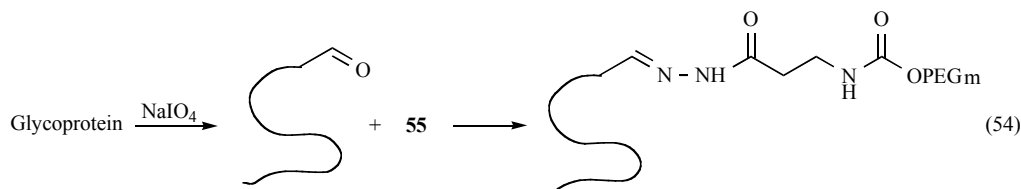
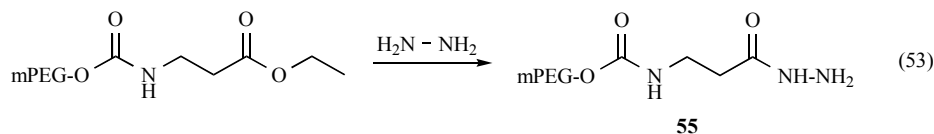
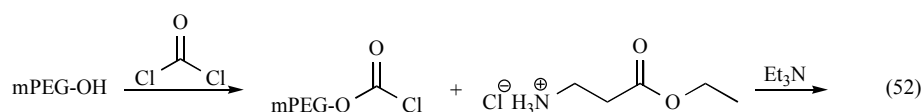


chemistry of the process using a crowded carbamate is presented in equations 48 and 49. Reduction of the disulfide bridge by a cysteine residue of the protein, or any other entity containing a thiol group, such as glutathione, releases the polymer and delivers sulfide **54**. Finally, 1,6-elimination delivers the unmodified native protein (equation 49) [205,206].

Reversible PEGylation based on trimethyl lock lactonization is well established strategy that has mainly found applications for the preparation of small compound prodrugs [207]. The process begins by the controlled hydrolysis of the ester linkage by endogenous enzymes (equation 50, where X= spacer), followed by nucleophilic attack of a free hydroxyl group to form the lactone and delivering the drug.

## 7. ENZYMATIC PEGYLATION

Transglutaminase has been used for site-selective PEGylation. The concept of these conjugations is based in the enzymatic catalyzed acyl transfer reaction between the carboxamide group of glutamine and the amino group of PEG-amine to afford a new and stable amide linkage (equation 51). The approach can be used for the PEGylation of native or chimeric proteins, yielding homogeneous conjugates.



Human granulocyte colony-stimulating factor [208],  $\alpha$ -lactalbumin, interleukin-2 [209], salmon calcitonin and IgGH [210], among other proteins, have all been PEGylated using transglutaminase.

## 8. PEGYLATION AT CARBOHYDRATES RESIDUES

Chemical or enzymatic oxidation of carbohydrate residues of glycoproteins generates aldehyde groups that can react with hydrazides to yield hydrazones, which may be reduced to more stable alkyl hydrazides [211]. For example, hydrazide **55**, which can be prepared from mPEG and the ethyl ester of  $\beta$ -alanine (equations 52 and 53), was reacted with oxidized ovalbumin and immunoglobulin G to yield diacyl-hydrazone linkages (equation 54) [212].

Although PEGylation of the reactive aldehyde groups with PEG-amines has been proposed, it has been suggested that this process is difficult, given that the amino groups of proteins show a similar reactivity to those of PEG-amine, leading to cross-linking of the protein. In this regard, the  $pK_a$  of hydrazides is around 3, allowing for selective conjugation at acidic  $pH$ s, at which most of protein amino acids are protonated ( $pK_a$  around 10).

In a related method, serine or threonine terminated proteins were oxidized to the corresponding glyoxylyl groups, which were then reacted with aminoxy mPEG derivatives to achieve site-selective PEGylation. Furthermore, the reactive terminal carbonyl groups can be introduced by metal catalyzed transamination [213].

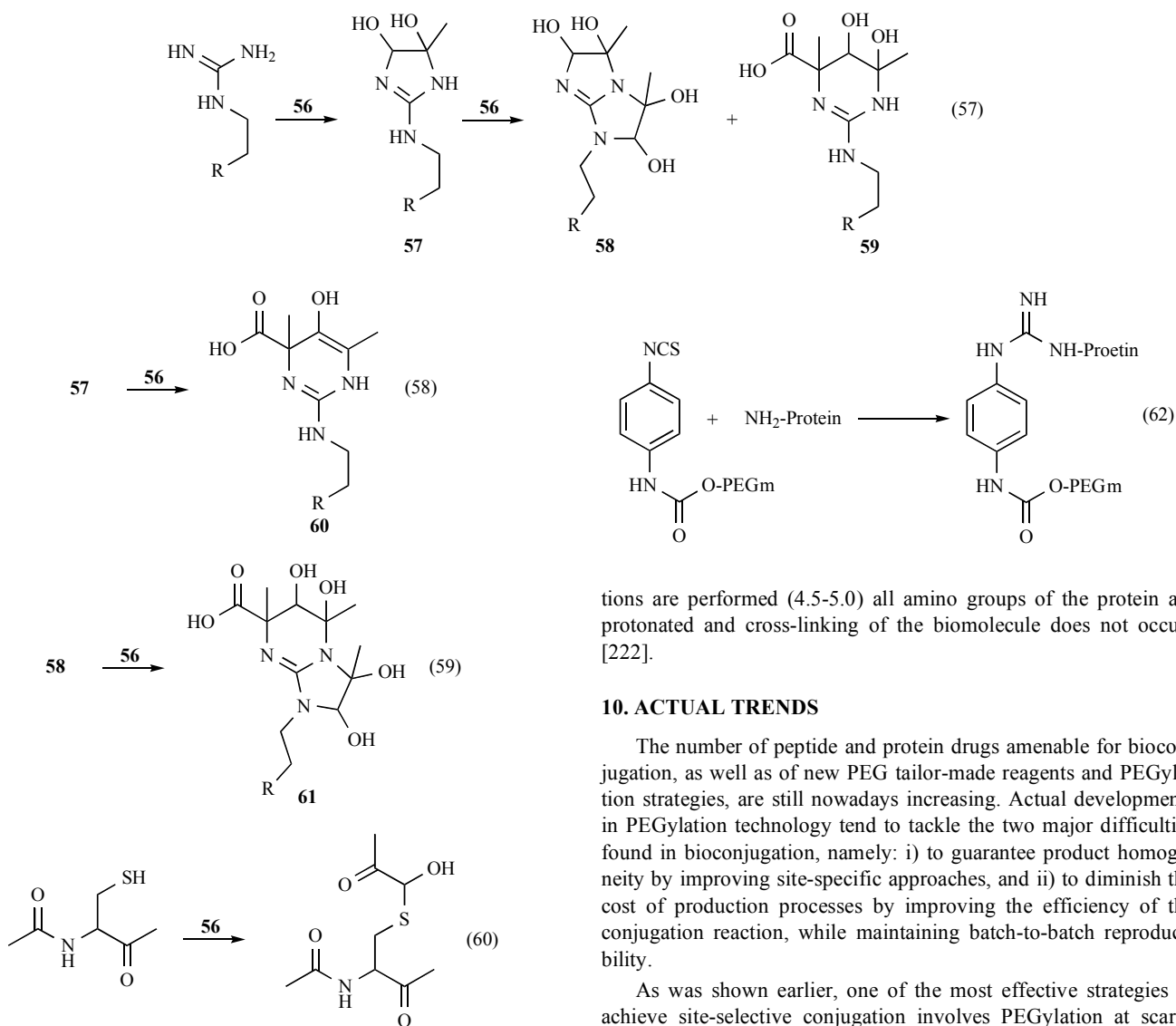
## 9. MISCELLANEOUS MPEG REAGENTS AND REACTIONS

The PEGylation of arginine residues with mPEG phenylglyoxal has been reported. The usefulness of this methodology is limited

due to the long reaction times that are required to achieve good yields of conjugation. In addition, the PEGylation with mPEG phenylglyoxal is not selective and conjugation at histidine and lysine residues is also usually observed [214,215].

The site-selective and permanent arginine PEGylation has recently been disclosed [216,217]. The process relies on the thermodynamic selectivity obtained upon reaction of the protein with methyl glyoxal analog **56**. The dimethyl acetal derivative **56** which can easily be obtained as shown in equation 55, is in equilibrium with unprotected forms in aqueous media, and this mixture was used as the actual PEGylating reagent (equation 56).

A detailed study of the reaction of methyl glyoxal with *N*-acetyl arginine and *N*-acetyl cysteine at  $pH$  9.0 (borate buffer) and 7.4 (phosphate buffer) (equations 57-60) suggest that although cysteine reacts quickly with the dicarbonyl compound (equation 60), the formed adduct can be hydrolyzed at  $pH$  3. Moreover, this adduct is unstable under dilution or by re-suspension of the purified product at  $pH$  7.0. On the other hand, at  $pH$  9.0 adducts **57** and **58**, which arise from arginine modification with one and two molecules of **56**, were the only observed products. At  $pH$  7.4 compounds **60** and **61** (equations 58 and 59) were also formed, indicating that complexation of diols **57** and **58** prevented the reaction with another molecule of **56**. In addition, the formation **61** indicates that, at some degree, **59** was also obtained. Hydrolysis studies suggested that the adducts obtained upon reaction with **56** are more stable than those obtained with *N*-acetyl cysteine, and also that the adducts obtained in the reaction with *N*-acetyl arginine are more stable than the adducts obtained with the same compound and phenyl glyoxal. Importantly, it has been demonstrated that with reagent **56**, PEGylation at lysine residues does not occur.



The site-selective conjugation at tyrosine residues of salmon calcitonin using diazonium salt **62** has recently been accomplished (equation 61) [218]. This PEGylation approach is reversible, and the starting materials can be obtained by reaction with suitable reducing reagents.

PEGylation of amino groups of human serum albumin and human keratinocyte growth factor 2 using mPEG isothiocyanate has been reported [219,220]. The conjugation with this reagent affords hydrolytically stable thiourea linkages by means of a simple and easily obtained PEG derivative (equation 62) [221].

Finally, the selective PEGylation of terminal carboxyl groups has been reported by Zalipsky. In this method mPEG-hydrazide reacts selectively with the carboxyl group by means of carbodiimide activation at low pH. At the pH values at which the conjugation

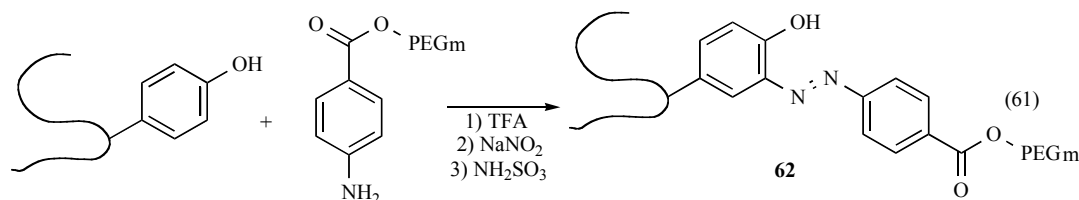
are performed (4.5-5.0) all amino groups of the protein are protonated and cross-linking of the biomolecule does not occur [222].

## 10. ACTUAL TRENDS

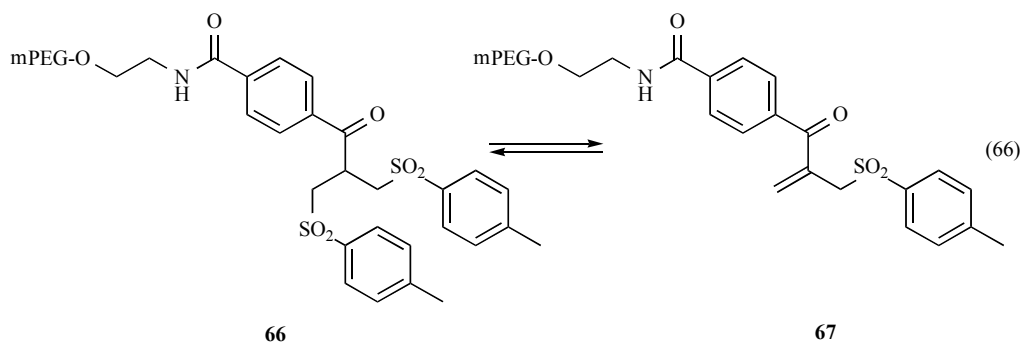
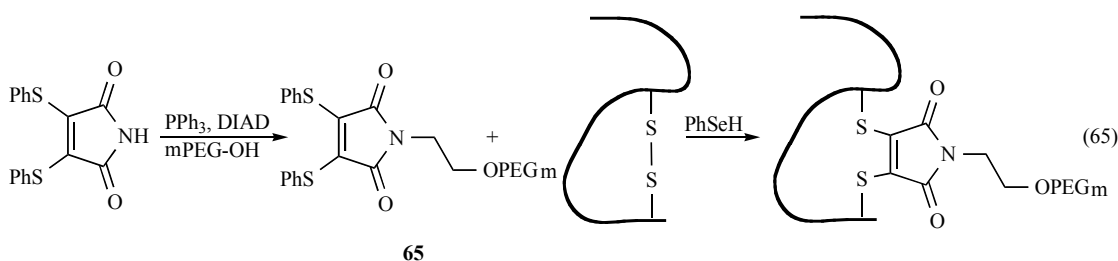
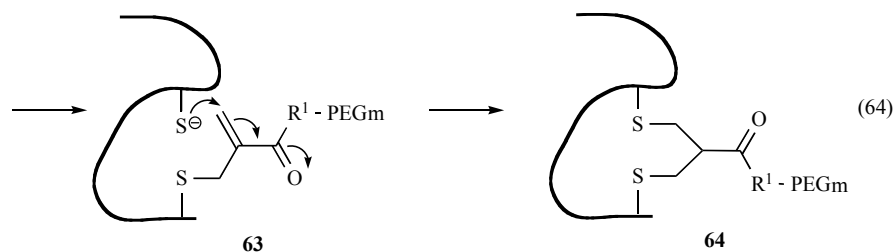
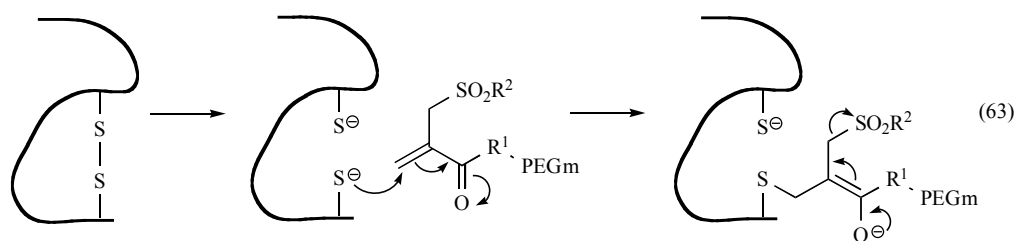
The number of peptide and protein drugs amenable for bioconjugation, as well as of new PEG tailor-made reagents and PEGylation strategies, are still nowadays increasing. Actual developments in PEGylation technology tend to tackle the two major difficulties found in bioconjugation, namely: i) to guarantee product homogeneity by improving site-specific approaches, and ii) to diminish the cost of production processes by improving the efficiency of the conjugation reaction, while maintaining batch-to-batch reproducibility.

As was shown earlier, one of the most effective strategies to achieve site-selective conjugation involves PEGylation at scarce cysteine residues located at the surface of the protein. If these residues are not available at suitable location they may be introduced by genetic engineering methods. In connection, PEGylation of cysteine involved in accessible disulfide bridges is nowadays well established for site-specific conjugation. Based on these concepts, Brocchini has developed the TheraPEG™ technology for site-selective PEGylation (equations 63-64) [149]. The disulfide bridge of the protein is reduced under mild reaction conditions to afford nucleophilic thiol group. The sulfide group reacts with the double bond of the acceptor in a Michael-type 1,4-addition reaction (equation 63), which releases a sulfonate anion to give intermediate **63**. A second Michael 1,4-addition reaction with the sulfide anion finally yields hydrolytically stable cyclic sulfide **64** (equation 64).

It has been demonstrated that the conjugation reaction does not affect the tertiary structure of the protein and, importantly, that, due







to steric hindrance at the conjugation site, only one polymer chain is attached at each disulfide bridge.

A related approach using dithiophenolmaleimides **65** has been published. The conjugation reaction involves the generation of the free thiol group, which is quickly trapped *in situ* by added **65** (equation 65). It has been shown that this simple and fast method restrict the unfolding, aggregation and disulfide scrambling of the protein [223]. A related method for disulfide site-specific PEGylation that employs dibromomaleimide derivatives has recently been devised [224].

Recombinant proteins expressed in several prokaryotic expression systems are often produced with histidine-tags (HisTag), which allows the purification of proteins using affinity chromatography with matrices that contains nickel or other metal ions. Tailor made mPEG reagents have been created to selectively react with these tags and, importantly, as the tags are expressed at the N or C terminal of proteins, the biological activity is not compromised [225]. The concept of the HisTag PEGylation is closely related to that of the TheraPeg<sup>®</sup> technology, namely, a *bis*-alkylation through a double 1,4-Michael-type addition. Monosulfone **67** is formed by elimi-

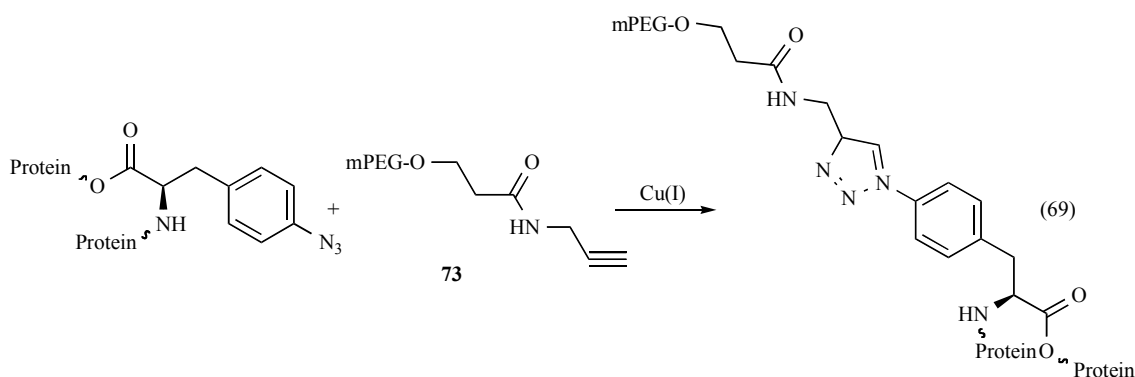
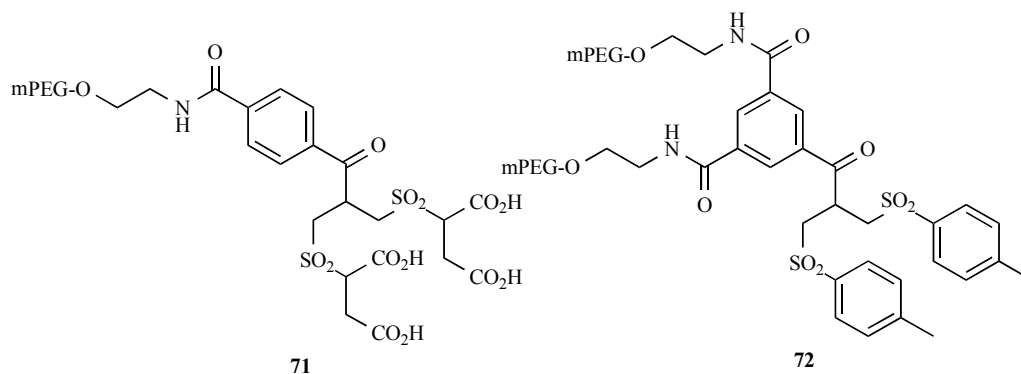
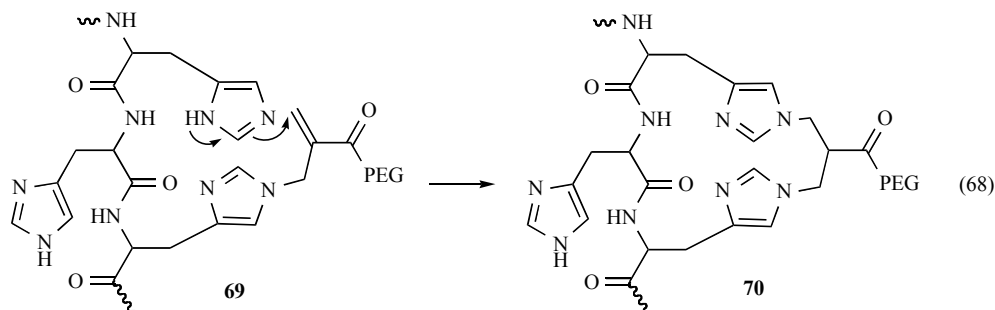
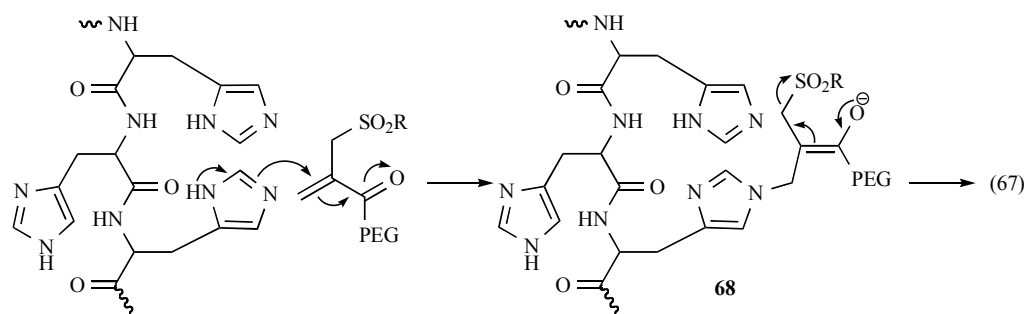
nation of sulfenic acid from mPEG bis-sulfone **66**. Intermediate **67** is the actual PEGylating reagent (equation 66).

The proposed mechanism of conjugation reaction involves the nucleophilic addition of the heterocyclic nitrogen of the aromatic ring to the double bond to deliver intermediate **68** (equation 67), that upon elimination of sulfenic acid forms an  $\alpha,\beta$ -unsaturated ester **69** (equation 68). A second Michael addition finally gives cyclic *bis*-alkylated product **70** (equation 68).

*Bis*-sulfone **71** and branched *bis*-sulfone **72** have also been prepared and carefully evaluated in HisTag PEGylation [225].

Proteins that have been PEGylated by HisTag technology include cytokines (IFN  $\alpha$ -2a), antibody fragments (anti-TNF, scFV), endostatin and pro-brain natriuretic peptide.

The PEGylation of non-natural amino acids incorporated to the protein by genetic engineering methods is an increasingly important approach for site-specific conjugation. For example, the genetically introduced *para*-azidophenylalanine residue was used to achieve the site-specific conjugation of SOD by mean of a [3+2] cycloaddition reaction with mPEG-alkyne **73** (equation 69) [226].



Other related methods for the conjugation of non-natural amino acids includes the PEGylation of non-natural *p*-acetylphenylalanine with mPEG-oxyamine [227,228], pyrrolysine and pyrroline-carboxy-lysine with mPEG carbonyl compounds [229], among several other methods [230-232].

A new two-step process, known as glycoPEGylation, has become increasingly important in the last years for the site-specific bioconjugation of proteins [233]. The sequence consists of a first step of enzymatic glycosylation with *N*-acetylgalactosamine and *N*-acetylgalactosamine transferase at specific threonine and serine residues. In the second step, the protein is PEGylated at the *N*-acetylgalactosamine residue with a sialic acid mPEG derivative us-

ing sialyltransferase. Examples of proteins that has been conjugated by this strategy includes factor VIIa [234,235], G-CSF, IFN  $\alpha$ -2b and GM-CSF [233]. In a related approach, a transpeptidase (sortase A), has recently been employed for the site-selective attachment of mPEG to cytokines [236].

The solid-phase PEGylation methodology involves the conjugation of protein to a tethered PEG derivative that is attached to a solid matrix. The process has been aimed at overcoming some of the drawbacks of the commonly used solution-phase methods. Several approaches have been developed, including site-specific PEGylation of keratinocyte growth factor 1 [237] and the use of heparin-spharose columns to conjugate recombinant human fibroblast

growth factor 2 to PEG-butyraldehyde through reductive alkylation [238]. Other biomolecules that have been PEGylated using solid-phase procedures include albumin and staphylokinase [239], lipopeptides [240], interferon  $\alpha$ -2a [241] and hemoglobin among other [242].

Further innovative PEGylation strategies include: i) C-terminal PEGylation of recombinant proteins expressed as intein-fusion proteins, using hydrazone-forming ligation reactions [243]; ii) PEGylation of latex surfaces beads for immunocamouflage [244], iii) PEGylation of immunonanoparticles [245], and iv) monoPEGylation using an adapted Dock and Lock (DNL) method [246].

## 11. CONCLUSION

PEGylation is undoubtedly a diverse and hot field, which is embedded within emerging technologies for the development of improved and new protein therapeutics among other applications. PEGylation takes place at the interface of chemistry, biotechnology, polymer, pharmaceutical and biomedical sciences. Thus, it is synonym of inter- and multi-disciplinary research, a key factor for success that is not new but scarce in a situation where finding a common language is sometimes difficult. The combination of these and other factors, such as the ever growing demand and increasing complexity emerging from the need for more effective therapeutic targets, still today stimulates the development of research projects aimed towards the design, synthesis and evaluation of tailor-made mPEG reagents and technologies. We believe that these efforts will provide, in the near future, further interesting and technologically relevant results.

## CONFLICT OF INTEREST

The author(s) confirm that this article content has no conflict of interest.

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