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Palladium-catalyzed stereoselective hydrostannation of substituted propargyl alcohols with trineophyltin hydride

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Abstract

This paper reports results obtained in a study on the palladium-catalyzed hydrostannation of substituted propargyl alcohols with the bulky trineophyltin hydride (1). The reaction of 1 with 10 propargyl alcohols containing one up to three substituents, was carried out in THF at room temperature leading to the corresponding allylstannanes following in all cases a *syn* addition stereochemistry. These additions took place in good to excellent yields and, mostly, with a high degree of stereoselectivity. The results obtained suggest that the observed α/β regioselectivity might be ascribed to the steric bulk of the *proximal* substituents rather than to electronic effects. Full ¹H-, ¹³C-, and ¹¹⁹Sn NMR characteristics are included.

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1. Introduction

Vinylstannanes are very useful intermediates in organic synthesis and due to their great versatility as building blocks for synthesis, considerable effort has been devoted to their regio- and stereocontrolled synthesis. They can be condensed with a large array of electrophiles including carbonyl compounds, enones, acyl chlorides, vinyl, aryl, allyl, and benzyl halides or triflates [1]. The vinylstannanes are also valuable partners in Stille reactions, i.e., the palladium-catalyzed cross coupling reaction between organic halides and organostannanes, one of the more important methods used to make carbon–carbon bonds [2]. In total synthesis of natural products, the use of vinylstannanes enabled the stereoselective building of some unsaturated fragments [3].

The addition of organotin hydrides to alkynes is still the more simple, direct, and economical method for the preparation of vinylstannanes. Thus, palladium-catalyzed hydrostannation of propargylic alcohol derivatives with tri-n-butyltin hydride takes place with syn stereoselectivity leading to mixtures of the two possible stannylated allylic alcohols regioisomers in good to excellent yields [4]. It was observed that steric effects upon substitution at the propargylic position could change the regioselectivity of these reactions [5]. It should be noted that although the reported analytical yields of these hydrostannations were always nearly quantitative, we have found that the yields of the mixtures of adducts were substantially lower due to loss of material during chromatographic purification, and in most cases it was not possible to isolate a regioisomer in pure form.

In previous studies [6], we reported that hydrostannation of substituted alkynes with trineophyltin hydride (1) leads stereoselectively and in high yields to vinylstannanes which showed enhanced stability in comparison with that of the vinylstannanes resulting from the additions of the

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more common triorganotin hydrides (Me, Bu, and Ph) [4,5]. Now, we considered it of interest to carry out a study on the palladium-catalyzed addition of trineophyltin hydride (1) to substituted propargyl alcohols in order to determine the effect of the bulk of the neophyl ligands attached to the tin atom combined with that of various substituents in the organic substrate on the regioselectivity of these reactions as well as on the stability of the new adducts.

2. Results and discussion

The addition of trineophyltin hydride (1) to various mono-, di-, and trisubstituted propargyl alcohols, at r.t. in THF containing 2% of bis(triphenylphosphine) palladium(II) chloride led, after 30-120 min of reaction, in all cases to mixtures of adducts as shown in Scheme 1. The yields were in the range 60-90% and in all cases the regioisomers could be isolated in pure form.

The results obtained are summarized in Table 1.

This Table shows that all the regioisomers formed were those resulting from a *syn* attack (Scheme 1, *syn* α and *syn* β). Full ¹H and ¹³C NMR characteristics of compounds **2–21** are summarized in Tables 2 and 3.

The assignment of the stereochemistry of the new stannyl derivatives 2-21 was carried out making use of their NMR characteristics, especially ⁿJ(¹H, ¹¹⁹Sn) and ${}^{n}J({}^{13}C, {}^{119}Sn)$ coupling constants. Thus, the ¹H NMR spectra of the adducts resulting from the α -svn attack, compounds 2, 4, 6, 8, 10, 12, 14, 16, 18, and 20 (Table 2), show in all cases signals between 4.90 and 5.65 ppm corresponding to the proton attached to C(3) (H_A). The values of the ${}^{3}J(Sn,H)$ coupling constants range from 85 to 64 Hz indicating that this proton is *cis* with respect to the trineophyltin moiety attached to C(2). In the case of compounds 2, 4, 6, 8, and 10, the ¹H NMR spectra also show that attached to C(3) there is another proton (H_B) with ${}^{3}J(Sn,H)$ coupling constants values of around 140 Hz, indicating a trans relationship between this proton and the triorganotin group attached to C(2). Similarly, the ¹H NMR spectra of the adducts resulting from the β -syn attack, compounds 3, 5, 7, 9, 11, 13, 15, 17, 19, and 21 (Table 3), also show in all cases signals between 5.10 and 6.20 ppm corresponding to the proton attached to C(2) (H_A) with ³J(Sn,H) coupling constants values between 66 and 77 Hz indicating that this proton is *cis* with respect to the organotin substituent attached to C(3). Also, in the case of the adducts resulting from the hydrostannation of the terminal



Scheme 1. Palladium catalyzed hydrostannation of substituted propargyl alcohols.

Table 1 Hydrostannation of substituted propargyl alcohols with trineophyltin hydride $(1)^a$



Entry number	Compound number	R	\mathbb{R}^1	\mathbb{R}^2	syn Adducts		Yield ^b (%)	Ratio β/α	¹¹⁹ Sn ^c (ppm)	
					$\alpha (\%)^{b}$	β (%) ^b				
1	2 and 3	Н	Н	Me	28 (2)	55 (3)	83	1.96	2 : -84.2, 3 : -82.4	
2	4 and 5	Н	Н	Ph	23 (4)	35 (5)	58	1.52	4 : -79.1, 5 : -82.0	
3	6 and 7	Н	Me	Me	7 (6)	80 (7)	87	11.4	6 : -82.8, 7 : -79.7	
4	8 and 9	Н	Me	Ph	24 (8)	60 (9)	84	2.5	8 : -77.7, 9 : -79.3	
5	10 and 11	Н	Ph	Ph	38 (10)	44 (11)	82	1.15	10 : -75.1, 11 : -78.6	
6	12 and 13	Bu	Н	Н	59 (12)	19 (13)	78	0.32	12 : -79.5, 13 : -74.4	
7	14 and 15	Me	Н	Me	8 (14)	55 (15)	63	6.88	14 : -82.5, 15 : -73.4	
8	16 and 17	Bu	Н	Me	18 (16)	50 (17)	68	2.78	16 : -80.7, 17 : -72.9	
9	18 and 19	Bu	Н	Ph	18 (18)	65 (19)	83	3.61	18 : -77.4, 19 : -71.8	
10	20 and 21	Bu	Me	Ph	8 (20)	48 (21)	56	5.87	20 : -77.0, 21 : -70.4	

^a $PdCl_2(PPh_3)_2$ in dry THF; alkyne/hydride 1 ratio = 1.

^b Yields of products isolated from column chromatography.

^c In CDCl₃; in ppm with respect to Me₄Sn.

propargyl alcohols, i.e., compounds **3**, **5**, **7**, **9**, and **11**, the ¹H NMR spectra show signals corresponding to a proton attached to C(3) (H_B) with ²J(Sn,H) coupling constants values between 76 and 72 Hz indicating that this proton and the organotin moiety are *geminal*. It should be noted that due to the multiplicity of the signals and the overlap between stannyl satellites and other signals, in many cases it was not possible the measurement of ³J(Sn,H) coupling constants (see Tables 2 and 3).

The ¹³C NMR chemical shifts (Tables 2 and 3) were assigned through the analysis of the multiplicity of the signals by means of DEPT experiments and taking into account the magnitude of ${}^{n}J({}^{13}C,{}^{119}Sn)$ coupling constants. Thus, the values of ${}^{2}J({}^{13}C,{}^{119}Sn)_{\text{geminal}}$ coupling constants for adducts 12, 14, 16, 18, and 20 (see C(1), Table 2), within the range 31.7-22.7 Hz indicate that the configuration of these compounds is (E) [7]. The (E) configuration of these compounds is confirmed by the fact that the values of their ${}^{3}J({}^{13}C, {}^{119}Sn)$ coupling constants lie between 61.2 and 54.8 Hz, indicating that the R substituent at C(3) is trans with respect to the the organotin moiety attached to C(2). Also, the ${}^{2}J({}^{13}C,{}^{119}Sn)_{geminal}$ coupling constants with values ranging from 40.0 to 23.7 Hz for adducts 2, 4, 6, 8, and 10 (see C(1) in Table 2,) confirm that the vinyl groups are terminal. Similarly, ³*J*(¹³C, ¹¹⁹Sn)_{trans} coupling constant values of compounds 3, 5, 7, 9, 11, 13, 15, 17, 19, and 21 range from 67.8 to 60.2 Hz (see C(1) in Table 3) demonstrating the (E) stereochemistry of these adducts [7].

As shown in Table 1, the *syn* hydrostannation of substituted propargyl alcohols with trineophyltin hydride leads in all cases but one (Table 1, entry 6), to mixtures of regioisomers in which the β -isomer is always formed in higher proportion. These results could be explained [8] taking into account that, due to steric effects the reactions might take place through an intermediate complex A (Scheme 2) in which the palladium residue adopts the distal position to minimize the interaction between the metal and the R¹ and/or R², as shown in Scheme 2. In this way, the distal stannane is obtained even when an additional R substituent is attached to the C- γ position.

On the other hand, in the case of the hydrostannation of 2-heptyn-1-ol (Table 1, entry 6) with trineophyltin hydride (1) i.e., a compound in which the C- α is not substituted, the proximal stannylated derivative is obtained as the major regioisomer *via* the intermediate complex **B** (Scheme 2). This result together with the results reported previously [6a], clearly indicates that the proximal adduct will be the major isomer whenever the C- α is not substituted and whether or not the C- γ is substituted.

The chemical reactivity of the new stannylated allyl alcohols is similar to that of other vinyltriorganostannanes. Thus, the Stille reactions of vinylstannanes 9 and 11 with *p*-bromoanisole lead to the corresponding coupling products in 62 (22) and 65% (23) yield.

The major advantages of the hydrostannation of propargyl alcohols with trineophyltin hydride are not only the higher yields but the greater stability of the resulting adducts compared with that of their tributyl- and trimethylstannyl analogues [6b,9]. The stability of the trineophylstannyallyl alcohols enable their separation and purification by column chromatography using silica gel 60 (in most cases) as adsorbent, this resulting in a dramatic improvement of the yields of pure isolated compounds.

3. Experimental

3.1. General methods

NMR spectra were recorded on a Bruker ARX 300 instrument, using CDCl₃ as solvent; chemical shifts (δ) are reported in ppm with respect to TMS, ¹H and ¹³C, and with respect to Me₄Sn in the case of ¹¹⁹Sn NMR spectra. Mass spectra were obtained using a Finnigan MAT Incos 50 Galaxy System (DIP-MS) at Cologne University (Germany). Elemental analyses (C, H) were performed at Cologne University (Germany). High resolution mass spectra (HRMS) were recorded on a Finnigan Mat. 900 (HR-EI-MS). All the solvents and reagents used were analytical reagent grade. Trineophyltin hydride (1) was prepared as described [10]. 1-Phenyl-2-heptyn-1-ol was obtained by reduction of 1-phenyl-2-heptyn-1-one [11]. One experiment is described in detail in order to illustrate the methods used.

3.2. Addition of trineophyltin hydride (1) to substituted propargyl alcohols catalyzed by bis(triphenylphosphine)palladium(II) chloride

All the reactions were carried out following the same procedure. One experiment is described in detail in order to illustrate the methods used.

To a solution of 2-phenyl-3-butyn-2-ol (0.11 g, 0.77 mmol) and bis(triphenylphosphine) palladium(II) chloride (0.010 g, 0.015 mmol) in dry THF (3 mL) under argon was added trineophyltin hydride (0.40 g, 0.77 mmol), and the mixture was stirred at room temperature for 2 h. Then, the solvent was distilled off under reduced pressure. The ¹¹⁹Sn NMR spectrum of the crude product showed a mixture of two compounds: α -2-trineophyltin-1-methyl-1-phenyl-2-propen-1-ol (**8**; 28%) and (*E*)- β -3-trineophyltin-1-methyl-1-phenyl-2-propen-1-ol (**9**; 72%). Column chromatography (silica gel 60) of the mixture afforded **8** (0.10 g, 0.16 mmol, 24%) and **9** (0.26 g, 0.39 mmol, 60%) in the fractions eluted with 98:2 and 96:4 hexane–diethyl ether, respectively.

3.3. Stille coupling reactions

All the reactions were carried out following the same procedure. One experiment is described in detail in order to illustrate the methods used.

To a mixture of 1-bromo-4-methoxybenzene (0.12 g, 0.74 mmol), $PdCl_2(PPh_3)_2$ (0.011 g, 2%), and some crystals of 2,6-di-*tert*-butyl-4-methylphenol under argon was added a solution of **9** (0.25 g, 0.37 mmol) in dry THF (2 mL) at



Compound number	2	4	6	8 1	0	12	14	16	18	20
R	H _B	H _B	H _B	H _B I	HB	<i>n</i> -Bu	Me	<i>n</i> -Bu	<i>n</i> -Bu	<i>n</i> -Bu
\mathbb{R}^1	H	H	Me	Me	Ph	Н	Н	Н	Н	Me
R^2	Me	Ph	Me	Ph I	Ph	Н	Me	Me	Ph	Ph
$\delta C_n [{}^n J ({}^{119} Sn - {}^{13} C)] \delta H - C_n [{}^n J ({}^{119} Sn - {}^{1} H)] and [{}^n J ({}^{1} H - {}^{1} H)]$	C_n Compound 2 $H^{-1}H$]		Compound 4		Compound 6			Compound 8	Compound 10	
$\overline{C_1(^2J_{SnC})}$	¹³ C: 73.38 (3	39.3)	¹³ C: 79.54	(40.0)	¹³ C: 75	5.49 (23.7)		¹³ C: 78.94 (24.1)	¹³ C: 85.36 (24.1))
$C_2 \left({}^{1}J_{Sn C} \right)$	¹³ C: 161.69 (363.0)		¹³ C: 158.92 (354.9)		¹³ C: 166.84 (380.5)			13 C: 163.68 (382.0) 13 C: 164.		.2)
$C_3(^2J_{Sn C});$	¹³ C: 123.40 (22.7)		¹³ C: 126.38 (23.7)		¹³ C: 122.26 (25.3)			¹³ C: 124.42 (25.5)	¹³ C: 128.39 (25.8)	
$H_{A}: ({}^{3}J_{Sn,H})_{cis} ({}^{2}J_{H,H});$ $H_{B}: ({}^{3}J_{Sn,H})_{trans.} ({}^{2}J_{H,H})$	${}^{1}H_{A}$: 4.91 (t)	(67.6) (1.3)	¹ H _A : 5.11 (t) (64.9) (1.6)	${}^{1}\text{H}_{A}$: 4.91 (s) (72.0) ${}^{1}\text{H}_{B}$: 5.41 (s) (151.6)			${}^{1}\text{H}_{\text{A}}$: 5.11 (s) (70.0) ${}^{1}\text{H}_{\text{B}}$: 5.55 (s)(145.0)	${}^{1}\text{H}_{\text{A}}$: 5.65 (s) (64.8) ${}^{1}\text{H}_{\text{B}}$: 5.39 (s) (139.0)	
\mathbf{R}^1	${}^{1}\text{H}_{\text{B}}$: 5.53 (t) ${}^{1}\text{H}$: 3.92 (q,	(146.3) (1.3) 1H) (42.3) (6.3)	¹ H _B : 5.49 (¹ H:4.81 (s,	t) (140.3) (1.6) 1H) (34.7)	13 C: 30 35 (9.7)			¹³ C: 31.02 (16.4) ¹ H: 1.24 (s. 3H)	13 C: 145 83 (12 0) (inso)	
R ²	¹³ C: 23.63 (9	9.5) 211) (6.2)	¹³ C: 142.46 (10.5) (<i>ipso</i>)		1 H: 1.04 (s, 7H) ^d		¹³ C: 146.67 (NO) (<i>ipso</i>)		5) (<i>ips</i> 5)	
Other signals	н. 1.01 (d, b	50) (0.5)	с		e			f	g	
	Compound 12		Compound 14		Compo	Compound 16		Compound 18	Compound 20	
$C_1 (^2 J_{C,Sn})$	¹³ C: 62.73 (3	31.7)	¹³ C: 68.57	(23.1)	¹³ C: 68	3.84 (24.0)		¹³ C: 73.06 (22.7)	¹³ C: 78.47 (26.5))
$C_2 (^1 J_{C,Sn})$	¹³ C: 145.93 ((404.6)	¹³ C: 153.12 (411.2)		¹³ C: 152.14 (406.9)			¹³ C: 149.01 (NO)	¹³ C: 155.88 (NO)	
$C_3(^2J_{Sn,C});$	¹³ C: 142.02 ((22.5);	¹³ C: 133.19	(26.1);	¹³ C: 139.64 (21.4);			¹³ C: 142.44 (21.8); ¹³ C: 140.35		2)
$H_{A}: ({}^{3}J_{Sn H})_{cis} ({}^{4}J_{H H});$	${}^{1}H_{A}$: 5.35 (tt	t) (74.7) (6.7) (1.5)	${}^{1}\text{H}_{A}$: 5.17 (dq) (75.7) (6.7) (1.5)	${}^{1}\text{H}_{\text{A}}$: 5.20 (dt) (77.8) (6.7) (1.5)			${}^{1}\text{H}_{A}$: 5.47 (t) (73.8) (6.8) ${}^{1}\text{H}_{A}$: 5.39 (t		5.3) (7.2)
$C_4 ({}^{3}J_{Sn C}); H ({}^{3}J_{H H})$ ${}^{13}C: 28.94 (54.8);$		54.8);	¹³ C: 15.00	(61.2);	¹³ C: 29.13 (56.4)			¹³ C: 31.46 (NO)	¹³ C: 30.61 (NO);	
4 (51,0) (11,11)	¹ H: 1.86 (dd) (13.5)		¹ H: 1.33–1.	51 (m.4H) ⁱ	¹ H: below neophyl			¹ H: 1.16–1.37 $(m.3H)^{1}$	¹ H: below neopl	nvl
R^1	¹ H:4.13			. 1H) (64.4)	1 H:4.16 (q, 1H) (65.2) (6.3)			1 H: 5.23 (m. 1H) (68.6)	¹³ C: 29.62 (9.9):	-
	1 H: 3.76 (s. 1	2H) (44.8)	(/ / / / / /					1 H: 1.54 (s. 3H)	
\mathbb{R}^2		, (,	¹³ C: 22.71	(NO): ¹ H: below neophyl	¹³ C: 13	3.99 (13.90): ¹ H: below ne	ophyl	¹³ C: 142.89 (NO) (<i>inso</i>)	¹³ C: 148.15 (10.2	2) (<i>ipso</i>)
Other signals	h		j	(k			m	n	, (T~~)

^a In CDCl₃ solution; chemical shifts, δ , in ppm with respect to TMS; ^{*n*}J(¹¹⁹Sn,¹H) coupling constants in Hz (in brackets); multiplicity (in brackets): d = doublet, dd = doublet, t = triplet, dt = double triplet, tt = triplet triplet, q = quartet, dq = double quartet, m = multiplet; NO = not observed.

^b Neophyl: ¹³C: 31.50 (328.7), 33.23 (36.7), 33.42 (36.7), 38.22 (19.5), 125.43, 125.63, 127.97, 151.31 (19.0); ¹H: 1.03 (s, 3H, ² $J_{Sn,H} = 50.0$), 1.04 (s, 3H, ² $J_{Sn,H} = 50.0$), 1.13 (s, 18H), 7.05–7.20 (m, 15H). OH: 0.66 (s, 1H).

^c Neophyl: ¹³C: 31.48 (328.2), 33.25 (35.7), 38.15 (19.0), 125.42, 125.61,127.97, 151.41 (19.6); ¹H: 0.92 (s, 6H, ² $J_{Sn,H} = 50.3$), 1.08 (s, 19H) (includes OH proton), 6.97–7.27 (m, 20H); Phenyl: ¹³C: 127.15, 127.25, 128.11; ¹H: below neophyl.

^d Probably includes the OH proton.

^e Neophyl: ¹³C: 32.25 (326.7), 33.36 (36.2), 38.42 (19.2), 125.40, 125.71, 128.01, 151.63 (19.4); ¹H: 1.08 (s, 6H, ${}^{2}J_{Sn,H} = 49.9$), 1.14 (s, 18H), 6.90–7.20 (m, 15H).

f Neophyl: ¹³C: 31.92 (326.0), 33.06 (35.8), 33.26 (19.3), 125.34, 125.70, 127.94, 151.49 (19.8); ¹H: 0.69-1.10 (m, 25 H) (probably includes OH proton), 6.97-7.32 (m, 20H); Phenyl: ¹³C: 25.63, 126.48, 127.82; ¹H: below neophyl.

^E Neophyl: ¹³C: 32.03 (326.3), 33.20 (35.8), 38.22 (19.2), 125.33, 127.69, 127.94, 151.58 (22.0); ¹H: 1.30 (s, 7H, ²J_{Sn,H} = 50.5) (probably includes OH proton), 1.52 (s, 18H), 7.30–7.80 (m, 25H); Phenyl: ³C: 125.57, 126.89; ¹H: below neophyl.

[25.38, 125.57, 127.94, 151.54 (19.1); ¹H: 1.03 (s, 7H, ²J_{Sn,H} = NO) (probably includes OH proton), 1.13 (s, 18H), 6.90–7.30 (m, 15H). *n*-Bu: ¹³C: 13.95, 22.39, 31.57; ¹H: 0.83 (t, 3H, ${}^{3}J_{H,H} = 7.1$), 1.18–1.27 (m, 4H). 38.18 (18.0), Ę. ^h Neophyl: ¹³C: 31.60 (325.5), 33.25 (36.

ⁱ Probably includes the OH proton.

^j Neophyi. ¹³C: 32.14 (325.7), 32.42 (31.0), 34.22 (41.1), 38.42 (18.4), 125.38, 125.82, 127.85, 151.53 (18.1); ¹H: 0.84–1.00 (m, 6H), 1.05–1.30 (m, 21H), 6.85–7.40 (m, 15H). ^k Neophyl: ¹³C: 32.11 (324.9), 32.50 (31.6), 34.13 (42.1), 38.40 (18.8), 125.78, 127.86, 151.56 (18.6); ¹H: 0.77–0.99 (m, 10H), 1.03–1.32 (m, 25H), 6.95–7.00 (m, 15H). *n*-Bu: ¹³C: 22.58, 23.22, 31.58; ¹H: 1.68–1.96 (m, 2H), other signals together with neophyl signals. ^k Neophyl: ¹³C: 32.11 (324.9), 32.50 (31.6),

¹ Probably includes the OH proton.

^m Neophyl. ¹³C: 32.02 (326.0), 32.98 (34.8), 33.56 (38.3), 38.29 (18.0), 125.33, 125.72, 127.86, 151.56 (19.2); ¹H: 0.71–1.04 (m, 11H), 1.09 (s, 18H), 7.00–7.28 (m, 20H). *n*-Bu: ¹³C: 13.96, 22.58, 29.75; ¹H: 1.71-2.00 (m, 2H), other signals together with neophyl signals. Phenyl: ¹³C: 126.49, 126.71, 128.00; ¹H: together with neophyl signals.

ⁿ Neophyl: ¹³C: 32.10 (324.5), 33.28 (34.3), 33.41 (35.8), 38.42 (18.5), 125.25, 127.95, 152.03 (20.1); ¹H: 0.95-1.09 (m, 11H) (probably includes the OH proton.), 1.14 (s, 18H), 6.97-7.43 (m, 20H). *n*-Bu: ¹³C: 13.86, 22.39, 31.42; ¹H: 0.68 (t, 3H, ³*H*,H</sub> = 6.8), 1.25-1.70 (m, 2H), other signals together with neophyl signals. Phenyl: ¹³C: 126.25, 126.83, 128.08; ¹H: together with neophyl signals.

3.4. Mass spectra, and elemental analyses of the new compounds

3.4.1. α -3-Trineophyltin-3-buten-2-ol (2)

MS (m/z, relative intensity): 590 (M⁺, Sn-pattern); 519 $(40\%, [SnNph_3]^+); 457 (10\%, [M-Nph]^+, Sn-pattern); 403$ (15%, Sn-pattern); 253 (7%, [SnNph]⁺); 197 (20%, Sn-pattern); 133 (15%, $[Nph]^+$); 118 (10%, $[Sn]^+$); 105 (11%, $[C_8H_9]^+$; 91 (100%, $[C_7H_7]^+$); 55 (18%, $[C_3H_3O]^+$). HR-MS (ESI) calcd for C34H46OSn 590.2571, found 590.2563. Anal. Calc. for C₃₄H₄₆OSn: C, 69.28; H, 7.87. Found: C, 69.39; H, 7.74%.

3.4.2. (E)- β -4-Trineophyltin-3-buten-2-ol (3)

MS (m/z, relative intensity): 590 (M⁺, Sn-pattern); 519 $(30\%, [SnNph_3]^+); 457 (35\%, [M-Nph]^+, Sn-pattern); 253$ $(4\%, [SnNph]^+); 197 (18\%, Sn-pattern); 133 (14\%,$ $[Nph]^+$; 118 (10%, $[Sn]^+$); 105 (12%, $[C_8H_9]^+$); 91 $(100\%, [C_7H_7]^+); 55 (20\%, [C_3H_3O]^+).$ HR-MS (ESI) calcd for C₃₄H₄₆OSn 590.2571, found 590.2565. Anal. Calc. for C₃₄H₄₆OSn: C, 69.28; H, 7.87. Found: C, 69.37; H. 7.76%.

3.4.3. α -2-Trineophyltin-1-phenyl-2-propen-1-ol (4)

MS (m/z, relative intensity): 652 (M⁺, Sn-pattern); 519 $(45\%, [SnNph_3]^+); 403 (37\%, Sn-pattern); 253 (8\%,$ [SnNph]⁺); 197 (22%, Sn-pattern); 133 (16%, [Nph]⁺); 118 (11%, $[Sn]^+$); 105 (12%, $[C_8H_9]^+$); 91 (100%, $[C_7H_7]^+$; 55 (21%, $[C_3H_3O]^+$). HR-MS (ESI) calcd for C₃₉H₄₈OSn 652.2727, found 652.2735. Anal. Calc. for C₃₉H₄₈OSn: C, 71.90; H, 7.43. Found: C, 71.81; H, 7.54%.

3.4.4. (E)-3-Trineophyltin-1-phenyl-2-propen-1-ol (5)

MS (m/z, relative intensity): 652 (M⁺, Sn-pattern); 519 $(45\%, [SnNph_3]^+); 403 (37\%, Sn-pattern); 253 (8\%),$ $[SnNph]^+$; 197 (22%, Sn-pattern); 133 (16%, $[Nph]^+$); 118 $(11\%, [Sn]^+)$; 105 $(12\%, [C_8H_9]^+)$; 91 (100%, $[C_7H_7]^+$; 55 (21%, $[C_3H_3O]^+$). HR-MS (ESI) calcd for C₃₉H₄₈OSn 652.2727, found 652.2738. Anal. Calc. for C₃₉H₄₈OSn: C, 71.90; H, 7.43. Found: C, 71.79; H, 7.35%.

3.4.5. α -3-Trineophyltin-2-methyl-3-buten-2-ol (6)

MS (m/z, relative intensity): 604 (M⁺, Sn-pattern); 519 $(54\%, [SnNph_3]^+); 403 (32\%, Sn-pattern); 253 (9\%,$ $[SnNph]^+$; 197 (20%, Sn-pattern); 133 (16%, $[Nph]^+$); 118 (10%, $[Sn]^+$); 105 (13%, $[C_8H_9]^+$); 91 (100%, $[C_7H_7]^+$; 55 (23%, $[C_3H_3O]^+$). HR-MS (ESI) calcd for C35H48OSn 604.2727, found 604.2737. Anal. Calc. for C35H48OSn: C, 69.66; H, 8.02. Found: C, 69.74; H, 8.10%.



Compound number	3	5	7	9	11	13	15	17	19	21
R	H_B	H _B	H_B	H _B	H_B	<i>n</i> -Bu	Me	<i>n</i> -Bu	<i>n</i> -Bu	<i>n</i> -Bu
\mathbb{R}^1	Н	Н	Me	Me	Ph	Н	Н	Н	Н	Me
\mathbb{R}^2	Me	Ph	Me	Ph	Ph	Н	Me	Me	Ph	Ph
$\delta C_n [{}^n J ({}^{119} Sn - {}^{13} C)] \delta H - C_n [{}^n J ({}^{119} Sn - {}^{1} H)] and [{}^n J ({}^{1} H - {}^{1} H)]$	Compour	nd 3	Comp	ound 5		Compound 7	Compound 9		Compound 11	
$\overline{C_1({}^3J_{\operatorname{Sn.C}})}$	¹³ C: 70.7	2 (65.1)	¹³ C: 7	7.23 (67.8)		¹³ C: 71.89 (62.3)	¹³ C: 75.56	5 (60.4)	¹³ C: 80.30 (60.6)	
$C_2({}^2J_{Sn.C});$	¹³ C: 149.47 (NO);		¹³ C: 147.22 (NO);			¹³ C: 152.92 (NO);	¹³ C: 151.39 (NO);		¹³ C: 149.74 (NO);	
$H_{A}: ({}^{3}J_{Sn,H})_{cis} ({}^{3}J_{H,H})$	${}^{1}\text{H}_{A}$: 5.51 (dd) (68.6) (18.6) (5.2)		$^{1}H_{A}$: 5	5.64 (dd) (66.4) (18.7)	(5.2)	${}^{1}\text{H}_{\text{A}}$: 6.00 (d) (71.3) (19.0)	${}^{1}\text{H}_{\text{A}}$: 6.21 (d) (67.6) (19.0)		¹ H _A : 6.16 (d) (69.6) (19.0)	
$C_3 ({}^1J_{Sn,C});$	¹³ C: 128.75 (388.0);		¹³ C: 1	30.71 (369.0);		¹³ C: 124.98 (393.0);	¹³ C: 126.72 (382.2);		¹³ C: 128.57 (373.3);	
$H_{B}: ({}^{2}J_{Sn,H})_{gem}; ({}^{3}J_{H,H})$	${}^{1}\text{H}_{\text{B}}: 5.19$	9 (d) (75.3) (18.6)	¹ H _B : 5	.37 (dd) (74.1) (18.7)	(1.4)	${}^{1}\text{H}_{\text{B}}$: 5.54 (d) (76.3) (19.0)	${}^{1}\text{H}_{\text{B}}$: 5.71 (d) (73.8) (19.0)		${}^{1}\text{H}_{\text{B}}$: 5.34 (d) (72.2) (19.0)	
R ¹	¹ H: 3.83 (qui, 1H) (5.9)		¹ H: 4.	77 (d, 1 H) (5.2)			¹³ C: 28.91;			
						¹³ C: 28.91;	¹ H: 1.78 ((s, 3H)	¹³ C: 145.93 (<i>ipso</i>)
\mathbb{R}^2	¹³ C: 22.3	9;	¹³ C: 14	42.58 (<i>ipso</i>)		¹ H: 1.45 (s, 7H) ^d	¹³ C: 146.6	65 (<i>ipso</i>)		
	1 H: 1.01	(d, 3H) (6.2)								
Other signals	b		с			e	f		g	
	Compour	nd 13	Comp	ound 15		Compound 17	Compour	nd 19	Compound 21	
$C_1 ({}^3J_{Sn,C})$	¹³ C: 59.0	4 (64.9)	¹³ C: 6	3.52 (61.1)		¹³ C: 63.41 (63.3)	¹³ C: 69.28	8 (60.2)	¹³ C: 75.09 (65.7)	
$C_2 ({}^2J_{Sn,C}); H_A:$	¹³ C: 140.	16 (28.2);	¹³ C: 14	44.63 (23.6) 1 H _A :		¹³ C: 144.95 (25.0); ¹ H _A :	¹³ C: 142.6	65 (27.6);	¹³ C: 147.26 (27.0));
$({}^{3}J_{\text{Sn},\text{H}})_{cis}, ({}^{3}J_{\text{H},\text{H}})$	${}^{1}H_{A}$: 5.14(t)(73.1) (6.2)		5.24 (0	5.24 (d) (72.5) (7.9)		5.12 (d) (73.3) (8.4)	${}^{1}\text{H}_{\text{A}}$: 5.31 (s) (72.1)		${}^{1}\text{H}_{\text{A}}$: 5.63 (s) (77.3)	
$C_3 (^1 J_{Sn,C})$	¹³ C: 148.51 (NO)		¹³ C: 14	¹³ C: 141.54(399.1)		¹³ C: 147.00 (NO)	¹³ C: 148.29(380.9)		¹³ C: 150.94 (388.8)	
$C_4 ({}^{3}J_{Sn,C}); H: ({}^{3}J_{Sn,H}) ({}^{3}J_{H,H})$	¹³ C: 32.80 (37.9) H:		¹³ C: 1	¹³ C: 18.65 (41.9); ¹ H:		¹³ C: 32.83 (NO)	¹³ C: 32.84 (NO);		¹³ C: 33.46(NO);	
	1.94 (t, 2	H) (30.3) (7.9)	1.42 (s	, 3H) (46.6)		¹ H: 1.96 (m, 2H)	¹ H: 2.09 ((t, 2H) (60.1) (8.1)	¹ H: see other sig	nals
R ¹	1 H: 3.90 (d,2H) (6.2)		¹ H: 4.4	¹ H: 4.42 (qui, 1H) (6.6)		¹ H: 4.40 (m, 1H)	¹ H: 7.24 (s, 1H)		¹³ C: 32.89; ¹ H: 1.47 (s, 3H)	
\mathbb{R}^2			¹³ C: 2	2.70;		¹³ C: 23.04;	¹³ C: 143.6	63 (<i>ipso</i>)	¹³ C: 148.95 (<i>ipso</i>	·)
			¹ H: 1.	06 (d, 3H) (6.3)		¹ H: below neophyl				
Other signals	h		i			j	k		1	

^a In CDCl₃ solution; chemical shifts, δ , in ppm with respect to TMS; ^{*n*}J⁽¹¹⁹Sn, ¹H) coupling constants in Hz (in brackets); multiplicity (in brackets); d = doublet, dd = doublet, t = triplet, dt = double triplet, tt = triple triplet, q = quartet, dq = double quartet, qui = quintet, m = multiplet; NO = not observed.

^b Neophyl: ¹³C: 31.24 (339.1), 33.05 (35.5), 33.14 (36.5), 38.03 (18.7), 125.32, 125.36, 127.94, 151.24 (19.8); ¹H: 0.94 (s, 6H, ²J_{H,Sn} = 50.0), 1.13 (m, 19H) (probably includes OH proton), 6.90–7.30 (m, 15H).

^c Neophyl: ¹³C: 31.33 (340.3), 33.03 (35.2), 33.16 (36.5), 38.07 (18.9), 125.42, 128.02, 151.26 (19.9); ¹H: 0.93 (s, 6H, ² $J_{H,Sn} = 50.1$), 1.10 (s, 18H), 6.80–7.40 (m, 20H). OH: 1.50 (s, 1H). Phenyl: ¹³C: 126.28, 127.32, 128.27; ¹H: below neophyl.

^d Probably includes the OH proton.

^e Neophyl: ¹³C: 31.25 (339.1), 33.14 (35.4), 38.09 (18.5), 125.35, 125.40, 127.99, 151.32 (19.9); ¹H: 1.38 (s, 6H, ² $J_{H,Sn} = 50.1$), 1.56 (s, 18H), 7.30–7.70 (m, 15H). ^f Neophyl: ¹³C: 31.24 (340.0), 33.07 (36.0), 33.16 (36.0), 38.04 (19.8), 125.36, 127.99, 151.19 (19.2); ¹H: 1.39 (s, 6H, ² $J_{H,Sn} = 50.0$), 1.56 (s, 18H), 7.40–7.80 (m, 20H). OH: 1.88 (s, 1H). Phenyl: ¹³C: 125.15, 126.51, 127.88; ¹H: below neophyl.

^g Neophyl: ¹³C: 31.33 (339.6), 33.07 (35.6), 38.03 (18.8), 125.37, 125.40, 128.04, 151.17 (19.9); ¹H: 0.92 (s, 6H, ² $J_{H,Sn} = 49.8$), 1.08 (s, 18H), 6.90–7.30 (m, 25H). OH: 1.83 (s, 1H). Phenyl: ¹³C: 126.89,

27.00, 127.85; ¹H: below neophyl.

^h Neophyl: ¹³C: 31.02 (317.9), 33.36 (35.2), 38.10 (18.9), 125.37, 125.52, 127.99, 151.41 (18.0); ¹H: 0.97 (s, 3H, ²J_{H,Sn} = 49.6), 1.00–1.40 (m, 26H), 6.95–7.30 (m, 15H). *n*-Bu: ¹³C: 13.98, 22.86, 32.81; ¹H: 0.84 (t, 3H, ${}^{3}J_{\rm H,H} = 7.2$), other signals below neophyl

¹ Neophyl: ¹³C: 30.90 (318.1), 33.33 (35.3), 38.08 (18.1), 125.38, 125.47, 127.99, 151.42 (18.4); ¹H: 0.88-1.30 (m, 32H), 6.75-7.30 (m, 15H). *n*-Bu: ¹³C: 14.00, 22.96, 33.15; ¹H: 0.85 (t, 3H, ³J_{HH} = 6.9), proton), 6.75–7.50 (m, 15H). (s, 19H) (includes OH $^{2}J_{\rm H,Sn} = 49.4$), 1.11 151.31 (18.2); ¹H: 0.95 (s, 6H, 125.43, 127.95, 33.32 (37.5), 37.97(18.4), 125.36, 33.19 (36.0), ⁱ Neophyl: ¹³C: 30.79 (322.8), other signals below neophyl

= 49.0), 1.00–1.60 (m, 21H) (includes OH proton), 6.90–7.35 (m, 20H). *n*-Bu: ¹³C: 14.00, 23.03, 32.98; ¹H: 0.85 (t, 3H, ³J_{H,H} = 7.1), 1.90–2.30 (m, 2H), other signals below neophyl. Phenyl: ¹³C: 125.96, 127.07, 128.22; ¹H: together with neophyl signals. ^k Neophyl: ¹³C: 30.96 (319.1), 33.26 (36.1), 38.00 (18.4), 125.37, 125.44, 127.98, 151.32 (19.2); ¹H: 0.91 (s, 6H, ²J_{H:Sn}) (19.8), ¹³C: 30.96 (319.1), ³³C: 30.96 (319.1), ³³C: 30.96 (319.1), ³³C: 30.96 (319.1), ³³C: 30.96 (319.1), ³⁴C: 30.96

= 49.1), 1.11–1.12 (split signal, 19H) (includes OH proton), 6.90–7.50 1 H: 0.66 (t, 3H, $^{3}J_{H,H} = 7.6$), 1.52–2.24 (m, 6H). Phenyl: 13 C: 125.13, 126.25, 127.85; 1 H: together with neophyl signals. ¹ Neophyl: ¹³C: 31.24 (NO), 33.29 (NO), 33.47 (NO), 38.14 (18.7), 125.43, 125.50, 128.05, 151.49 (19.8); ¹H: 0.93 (s, 6H, ²J_{H:Sn}) ¹³C: 13.82, 23.08, 33.68; ¹ m, 20H). *n*-Bu:

3.4.6. (E)-4-Trineophyltin-2-methyl-3-buten-2-ol (7)

MS (m/z, relative intensity): 604 (M⁺, Sn-pattern); 519 (54%, [SnNph₃]⁺); 403 (32%, Sn-pattern); 253 (9%, [SnNph]⁺); 197 (20%, Sn-pattern); 133 (16%, [Nph]⁺); 118 (10%, [Sn]⁺); 105 (13%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (23%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₃₅H₄₈OSn 604.2727, found 604.2734. Anal. Calc. for C₃₅H₄₈OSn: C, 69.66; H, 8.02. Found: C, 69.72; H, 8.08%.

3.4.7. α-3-Trineophyltin-2-phenyl-3-buten-2-ol (8)

MS (m/z, relative intensity): 666 (M⁺, Sn-pattern); 519 (54%, [SnNph₃]⁺); 403 (40%, Sn-pattern); 385 (9%, [SnNph₂]⁺); 253 (8%, [SnNph]⁺); 197 (24%, Sn-pattern); 133 (17%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (10%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (19%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₀H₅₀OSn 666.2884, found 666.2875. Anal. Calc. for C₄₀H₅₀OSn: C, 72.19; H, 7.57. Found: C, 72.25; H, 7.51%.

3.4.8. (E)-4-Trineophyltin-2-phenyl-3-buten-2-ol (9)

MS (*m*/*z*, relative intensity): 666 (M⁺, Sn-pattern); 533 (25%, [M–Nph]⁺, Sn-pattern); 253 (6%, [SnNph]⁺); 197 (25%, Sn-pattern); 133 (14%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (12%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (18%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₀H₅₀OSn 666.2884, found 666.2878. Anal. Calc. for C₄₀H₅₀OSn: C, 72.19; H, 7.57. Found: C, 72.22; H, 7.50%.

3.4.9. a-2-Trineophyltin-1,1-diphenyl-2-propen-1-ol (10)

MS (m/z, relative intensity): 728 (M⁺, Sn-pattern); 519 (31%, [SnNph₃]⁺); 253 (8%, [SnNph]⁺); 197 (23%, Sn-pattern); 133 (12%, [Nph]⁺); 118 (10%, [Sn]⁺); 105 (13%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (17%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₅H₅₂OSn 728.3040, found 728.3031. Anal. Calc. for C₄₅H₅₂OSn: C, 74.28; H, 7.20. Found: C, 74.21; H, 7.30%.

3.4.10. (*E*)-*3*-*Trineophyltin-1,1-diphenyl-2-propen-1-ol* (*11*)

MS (*m*/*z*, relative intensity): 728 (M⁺, Sn-pattern); 595 (22%, [M–Nph]⁺, Sn-pattern); 253 (10%, [SnNph]⁺); 197 (21%, Sn-pattern); 133 (13%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (12%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (15%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₅H₅₂OSn 728.3040, found 728.3033. Anal. Calc. for C₄₅H₅₂OSn: C, 74.28; H, 7.20. Found: C, 74.20; H, 7.27%.

3.4.11. (E)-2-Trineophyltin-2-hepten-1-ol (12)

MS (*m*/*z*, relative intensity): 632 (M⁺, Sn-pattern); 519 (13%, [SnNph₃]⁺); 403 (28%, Sn-pattern); 253 (9%, [SnNph]⁺); 197 (20%, Sn-pattern); 133 (15%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (13%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (18%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₃₇H₅₂OSn 632.3040, found 632.3031. Anal. Calc. for C₃₇H₅₂OSn: C, 70.37; H, 8.30. Found: C, 70.29; H, 8.21%.



Scheme 2. Palladium catalyzed addition of trineophyltin hydride (1) to substituted propargyl alcohols.

3.4.12. (E)-3-Trineophyltin-2-hepten-1-ol (13)

MS (m/z, relative intensity): 632 $(M^+, \text{ Sn-pattern})$; 519 $(19\%, [\text{SnNph}_3]^+)$; 403 (24%, Sn-pattern); 253 $(8\%, [\text{SnNph}]^+)$; 197 (22%, Sn-pattern); 133 $(13\%, [\text{Nph}]^+)$; 118 $(10\%, [\text{Sn}]^+)$; 105 $(12\%, [\text{C}_8\text{H}_9]^+)$; 91 $(100\%, [\text{C}_7\text{H}_7]^+)$; 55 $(15\%, [\text{C}_3\text{H}_3\text{O}]^+)$. HR-MS (ESI) calcd for $\text{C}_{37}\text{H}_{52}\text{OSn}$ 632.3040, found 632.3032. Anal. Calc. for $\text{C}_{37}\text{H}_{52}\text{OSn}$: C, 70.37; H, 8.30. Found: C, 70.30; H, 8.19%.

3.4.13. (E)-3-Trineophyltin-3-penten-2-ol (14)

MS (m/z, relative intensity): 604 (M⁺, Sn-pattern); 519 (23%, [SnNph₃]⁺); 471 (12%, [M–Nph]⁺); (24%, Sn-pattern); 403 (27%, Sn-pattern); 253 (10%, [SnNph]⁺); 197 (19%, Sn-pattern); 133 (16%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (13%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (18%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₃₅H₄₈OSn 604.2727, found 604.2719. Anal. Calc. for C₃₅H₄₈OSn: C, 69.66; H, 8.02. Found: C, 69.59; H, 8.10%.

3.4.14. (E)-4-Trineophyltin-3-penten-2-ol (15)

MS (m/z, relative intensity): 604 (M⁺, Sn-pattern); 519 (5%, [SnNph₃]⁺); 471 (45%, [M–Nph]⁺); (24%, Sn-pattern); 253 (7%, [SnNph]⁺); 197 (27%, Sn-pattern); 133 (15%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (12%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (22%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₃₅H₄₈OSn 604.2727, found 604.2720. Anal. Calc. for C₃₅H₄₈OSn: C, 69.66; H, 8.02. Found: C, 69.57; H, 8.13%.

3.4.15. (E)-3-Trineophyltin-3-octen-2-ol (16)

MS (m/z, relative intensity): 646 (M⁺, Sn-pattern); 519 (6%, [SnNph₃]⁺); 403 (18%, Sn-pattern); 253 (6%, [SnNph]⁺); 197 (18%, Sn-pattern); 133 (14%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (11%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (18%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₃₈H₅₄OSn 646.3197, found 646.3192. Anal. Calc. for C₃₈H₅₄OSn: C, 70.70; H, 8.43. Found: C, 70.78; H, 8.37%.

3.4.16. (E)-4-Trineophyltin-3-octen-2-ol (17)

MS (m/z, relative intensity): 646 (M⁺, Sn-pattern); 519 (35%, [SnNph₃]⁺); 253 (6%, [SnNph]⁺); 197 (32%, Sn-pat-

tern); 133 (18%, $[Nph]^+$); 118 (11%, $[Sn]^+$); 105 (12%, $[C_8H_9]^+$); 91 (100%, $[C_7H_7]^+$); 55 (27%, $[C_3H_3O]^+$). HR-MS (ESI) calcd for $C_{38}H_{54}OSn$ 646.3197, found 646.3192. Anal. Calc. for $C_{38}H_{54}OSn$: C, 70.70; H, 8.43. Found: C, 70.78; H, 8.37%.

3.4.17. (E)-2-Trineophyltin-1-phenyl-2-hepten-1-ol (18)

MS (m/z, relative intensity): 708 (M⁺, Sn-pattern); 575 (18%, [M–Nph]⁺); 519 (35%, [SnNph₃]⁺); 253 (6%, [SnNph]⁺); 197 (32%, Sn-pattern); 133 (18%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (12%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (27%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₃H₅₆OSn 708.3353, found 708.3345. Anal. Calc. for C₄₃H₅₆OSn: C, 72.99; H, 7.98. Found: C, 72.87; H, 7.90%.

3.4.18. (E)-2-Trineophyltin-1-phenyl-2-hepten-1-ol (19)

MS (m/z, relative intensity): 708 (M⁺, Sn-pattern); 575 (27%, [M–Nph]⁺); 519 (7%, [SnNph₃]⁺); 253 (7%, [SnNph]⁺); 197 (30%, Sn-pattern); 133 (11%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (10%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (21%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₃H₅₆OSn 708.3353, found 708.3351. Anal. Calc. for C₄₃H₅₆OSn: C, 72.99; H, 7.98. Found: C, 72.89; H, 7.91%.

3.4.19. (E)-3-Trineophyltin-3-octen-2-ol (20)

MS (m/z, relative intensity): 722 (M⁺, Sn-pattern); 519 (7%, [SnNph₃]⁺); 403 (26%, Sn-pattern); 253 (8%, [SnNph]⁺); 197 (21%, Sn-pattern); 133 (15%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (11%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (18%, [C₃H₃O]⁺). HR-MS (ESI) calcd for C₄₄H₅₈OSn 722.3510, found 722.3519. Anal. Calc. for C₄₄H₅₈OSn: C, 73.23; H, 8.10. Found: C, 73.29; H, 8.17%.

3.4.20. (E)-4-Trineophyltin-3-octen-2-ol (21)

MS (m/z, relative intensity): 722 (M⁺, Sn-pattern); 589 (30%, [M–Nph]⁺); 519 (6%, [SnNph₃]⁺); 253 (7%, [SnNph]⁺); 197 (35%, Sn-pattern); 133 (14%, [Nph]⁺); 118 (11%, [Sn]⁺); 105 (15%, [C₈H₉]⁺); 91 (100%, [C₇H₇]⁺); 55 (20%, [C₃H₃O]⁺). HR-MS (ESI) calcd for $C_{44}H_{58}OSn$ 722.3510, found 722.3517. Anal. Calc. for $C_{44}H_{58}OSn$: C, 73.23; H, 8.10. Found: C, 73.27; H, 8.19%.

3.4.21. (*E*)-*3*-(*4*-*Methoxyphenyl*)-*1*,*1*-*diphenyl*-*2*-*propen-1*- *ol* (**22**)

¹H NMR (CDCl₃) δ 189 (s, 1H); 3.85 (s, 3H); 6.63 (d, 1H, ³*J*(H, H) 16.1 Hz); 6.89–7.23 (m, 14H); 7.28 (d, 1H, ³*J*(H,H) 16.1 Hz). ¹³C NMR (CDCl₃) δ 54.84; 78.82; 113.03; 125.90; 127.20; 127.35; 128.01; 129.07; 129.95; 133.65; 148.01; 160.59. HR-MS (ESI) calcd for C₂₂H₂₀O₂ 316.1463, found 316.1453.

3.4.22. (*E*)-4-(4-Methoxyphenyl)-2-phenyl-3-buten-2-ol (23)

¹H NMR (CDCl₃) δ 1.17 (s, 3H); 2.02 (s, 1H); 3.83 (s, 3H); 6.50 (d, 1H, ³*J*(H,H) 16.1 Hz); 7.12 (d, 1H, ³*J*(H,H) 16.1 Hz); 7.15–7.45 (m, 9H). ¹³C NMR (CDCl₃) δ 30.78; 54.85; 74.77; 113.41; 123.05; 125.66; 127.39; 127.65; 127.72; 129.73; 135.01; 148.04; 160.49. HR-MS (ESI) calcd for C₁₇H₁₈O₂ 254.1307, found 254.1301.

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