

Contents lists available at ScienceDirect

# Food Research International



journal homepage: www.elsevier.com/locate/foodres

# Cheese milk low homogenization enhanced early lipolysis and volatiles compounds production in hard cooked cheeses



## María A. Vélez \*, Erica R. Hynes, Carlos A. Meinardi, Verónica I. Wolf, María C. Perotti

Instituto de Lactología Industrial (INLAIN, Universidad Nacional del Litoral/CONICET), Santiago del Estero 2829, Santa Fe, Argentina

#### A R T I C L E I N F O

Article history: Received 1 November 2016 Received in revised form 14 February 2017 Accepted 17 February 2017 Available online 20 February 2017

*Keywords:* Free fatty acids Cheese ripening Flavour

### ABSTRACT

Homogenization applied to cheese milk has shown to increase lipolysis but its use is not spread as it can induce detrimental effects. The aim of this work was to assess the effect of low-pressure homogenization of the cream followed by pre-incubation of cheese milk on the composition, ripening index, lipolysis and volatile profiles of hard cooked cheeses. For that, control and experimental miniature Reggianito cheeses were made and analyzed during ripening (3, 45 and 90 days). Homogenization had no impact on composition and proteolysis. An acceleration of the lipolysis reaction was clearly noticed in cheeses made with homogenized milk at the beginning of ripening, while both type of cheeses reached similar levels at 90 days. We found the level of several compounds derived from fatty acid catabolism were noticeably influenced by the treatment applied: straight-chain aldehydes such as hexanal, heptanal and nonanal and methylketones from  $C_5$  to  $C_9$  were preferentially formed in experimental cheeses.

© 2017 Elsevier Ltd. All rights reserved.

#### 1. Introduction

Lipolysis is one of the main biochemical events that occurs during cheese ripening, and is caused by hydrolytic enzymes resulting in the hydrolysis of milk lipids to free fatty acids (FFA), partial glycerides and glycerol (McSweeney & Sousa, 2000). Further, FFA undergoes catabolic reactions that conduct to the formation of aroma compounds, such as esters, methylketones, lactones and secondary alcohols; moreover, short-chain fatty acids contribute themselves to cheese flavour. These compounds give characteristic aromatic/taste notes to some cheese varieties, being of great importance in mold and Grana cheeses (Collins, McSweeney, & Wilkinson, 2003). Inhard cooked cheeses as Reggianito, moderate to extensive lipolysis is an advantage for genuine flavour development. Reggianito cheese is an Argentinian product which manufacture has been inspired in Italian cheeses such as Parmigiano Reggiano and Grana Padano (Candioti et al., 2002; Vélez et al., 2011).

Enzymes responsible for liberation of FFA in cheese come from six main sources; the milk itself (mainly the indigenous lipoprotein lipase, LPL), rennet paste, starter bacteria, secondary organisms, non-starter lactic acid bacteria (NSLAB) and exogenous lipases (Collins et al., 2003). In milk, lipolysis does not occurs spontaneously, as the LPL is electrostatically associated to the casein micelle and the substrate is in the form of globules protected by the milk fat globule membrane (MFGM). However, physical treatments applied to milk prior to cheese making (agitation, pumping, homogenization) may decrease the protective action of the MFGM (Evers, 2004).

The basic principle of homogenization consists in the disruption of milk fat globules to smaller ones, achieved by forcing the milk at high pressure through small holes (Walstra, Geurts, Noomen, Jellema, & Boekel, 1999). During this process surface area of fat is considerable increased, which is stabilized by casein micelles and whey proteins. Thus, the homogenization would also favor the accessibility of lipolytic enzymes to fat (Kelly, Huppertz, & Sheehan, 2008). Indeed, homogenization is performed in cheeses where lipolysis is desired to enhance flavour (mainly in blue cheeses). On the other hand, it has also been used for reduced-fat cheese varieties in order to improve texture obtaining higher moisture and smoother or creamier bodied cheeses (Johnson, 2011). However, homogenization is not widespread in cheese making technologies, and no previous reports of this procedure applied in hard cooked cheeses is available. The aim of the present work was to assess the effect of low-pressure homogenization of the cream followed by a pre-incubation step of cheese milk on composition, proteolysis, lipolysis and the volatile profiles of Reggianito type cheeses.

#### 2. Materials and methods

#### 2.1. Pre-treatment of cheese milk

Fig. 1 shows the experimental scheme performed. A volume of 40 L of bulk raw milk (pH 6.7  $\pm$  0.05) supplied by a nearby dairy plant (Milkaut S.A., Santa Fe, Argentine) was centrifuged (500 g, Alfa Laval, Lund, Sweden) at a flow of 40 L/h and at 40 °C, and skim milk (<0.5% of fat content) and cream (>40% fat content) were obtained. These fractions were mixed to obtain a cream with 20% fat which was split into two portions, unhomogenized and one-stage homogenized at 9 MPa

<sup>\*</sup> Corresponding author. *E-mail address:* mvelez@fiq.unl.edu.ar (M.A. Vélez).





Fig. 1. Flow diagram depicting the process of manufacture of Reggianito cheeses, as well as the experimental treatment added (homogenization).

and 45 °C (Homogenizer 31 M-3TA, Gaulin Corporation, Boston, MA). Each portion was subsequently blended with raw skim milk to obtain cheese milk standardized at 3% fat content. Cheese milks were incubated for 12 h at 12 °C prior to cheese making day and destined to the manufacture of control (C) cheeses and cheeses made homogenized cream (H).

#### 2.2. Total and free fat analysis

Total fat content of cream and skim milk samples were determined by Gerber method (Bradley et al., 1992). Free fat was analyzed on homogenized and unhomogenized cream samples (20% fat) to assess the damage of the MFGM. For that purpose, samples were centrifuged in standardized conditions (600 g 10 min 60 °C) and the layer of free fat was measured with a caliber (Vélez et al., 2011).

#### 2.3. Phase contrast microscopy

Cream samples (20%) were viewed and photographed using a Contrast-phase Microscope (Jenned 2 - Carl Zeiss-Jena, Jena, Germany) attached to a camera (nit-AKS  $24 \times 36$ , automatic 2).

#### 2.4. Cheese manufacture

Reggianito-type cheeses were made at laboratory scale (Fig. 1) using an equipment composed by 4 vats of 5 L each, operated in parallel (Vélez, Perotti, Wolf, Hynes, & Zalazar, 2010). C and H raw cheese milks were heated to 33 °C and CaCl<sub>2</sub> was added (Merck, Darmstadt, Germany) to final concentration of 0.14 g/L. Lactic acid (15 g/L) was added until milk reached a pH of 6.3–6.4. Then, two DVS commercial starters of *Lactobacillus helveticus* (LH-B02) and *Lactobacillus bulgaricus*  (Lb-12) (Chr. Hansen, Buenos Aires, Argentina) were added in a dose to reach a total starter bacterial count of  $10^6$  cfu/mL of milk. After manual stirring, 70 mg of chymosin (Maxiren 150, France) previously suspended in 10 mL of distilled water was added. Once coagulum acquired the required firmness (18–20 min), curd was cut to the size of a rice grain. Mixture of curd particles and whey was stirred and heated until 52 °C (approximately 1 °C/min); stirring was stopped to allow the curd grains to go to the bottom and separate whey. Curds were put into molds and pressed for 20 h, brined for 12 h in saturated brine at 12 °C. Cheeses of approximately 500 g were obtained and ripened for 90 days at 12 °C.

After cheese making, whey samples were analyzed for total fat by Gerber method (Bradley et al., 1992).

After two weeks of ripening, cheeses were vacuum wrapped in plastic films. Sampling of cheeses were performed at 3, 45 and 90 days (ISO, 2008b) and samples were stored at -18 °C until analysis. Two cheeses makings trials (A and B) were performed in different days: per day, two cheeses of each type (C and H) were performed.

#### 2.5. Cheese composition and ripening index

Samples were analyzed at 3 and 90 days for moisture by oven drying (ISO, 2004) and pH by potentiometric method (Bradley et al., 1992); fat and protein contents were analyzed at the beginning of ripening (3 days) by Gerber-Van Gulik method (ISO, 2008a) and Kjeldahl method (ISO, 2011) respectively. Nitrogen content was also assessed at 90 days in the fraction of the cheese soluble extract at pH 4.6 (SN), and expressed as percentage of total nitrogen (SN/TN, ripening index).

#### 2.6. Microbial counts

Thermophilic lactic bacteria populations were assessed at 3 and 90 days by plating sample dilutions on skim milk agar and counting colonies after 48 h of incubation at 37 °C (Candioti et al., 2002).

#### 2.7. FFA analysis

Extraction of fat matter, isolation of FFA, derivatization to ethyl esters and quantification of FFA by internal standard method (using C7:0 and C17:0 as standards), were performed in duplicate as described by Perotti, Bernal, Meinardi, and Zalazar (2005). A PerkinElmer model GC-9000 series gas chromatograph (PerkinElmer Corp., Waltham, MA) equipped with a flame ionization detector (FID) and with a split/ splitless injector was utilized. Ethyl esters FFA were separated on a fused silica capillary column (60 m  $\times$  0.25 mm; HP-INNOWax, Agilent J&W, USA) coated with a bonded polyethylene glycol stationary phase (0.25 µm layer thickness); carrier gas H<sub>2</sub> flow at 2 mL/min; 1 µL injection; split mode injection at 1:50 splitting ratio; injector and detector temperatures at 250 and 300 °C, respectively; oven temperatures running from 75 °C (1.5 min) up to 150 °C (10 min) at 8 °C/min, then increased to a final temperature of 245 °C (15 min) at 10 °C/min. The concentrations of FFA were determined at 3, 45 and 90 days and were expressed as µmol of FFA per 100 g of fat. Besides, the degree of lipolysis was calculated as the sum of individual free fatty acids (total free fatty acids TFFA, mg/kg cheese).

#### 2.8. Volatile compounds

Volatile compounds were isolated by headspace solid-phase microextraction (HS-SPME) and analyzed by GC/FID-MS. Samples of 5 g of cheese were weighed in a 40 mL screw-top glass vial closed with a cap with a Teflon-linedsilicone rubber septum (Supelco Inc., Bellefonte, PA). Samples were preheated for 10 min at 40 °C. Then, a 1 cm 50/30  $\mu$ m DVB/Car/PDMS fiber (Supelco Inc., Bellefonte, PA) was exposed to the vial headspace during 15 min. After that, the fiber was withdrawn and inserted into the injection port of the GC (same

equipment above mentioned) for 5 min at 250 °C and the volatile compounds were desorbed in splitless mode. Chromatographic analysis, identification of volatile compounds and semi-quantitative analysis (peak area values) were performed according to Wolf, Vénica, and Perotti (2015).

#### 2.9. Statistical analysis

One-way ANOVA was applied to compositional data, lipolysis and volatile compounds profiles, using a general linear model procedure with least significant difference pairwise comparison at 95% confidence level to test differences between both types of cheeses at each sampling time as well as to determine changes during ripening, if appropriate. Besides, Principal Component Analysis (PCA) was conducted on the values of the individuals FFA through ripening. After that, Discrimant Analysis (DA) was applied to PC scores to establish those variables capable of discriminating the samples from different months and treatment (Hair, Anderson, Tatham, & Black, 1999).SPSS software (v 10.0, SPSS Inc., Chicago, IL) was employed.

#### 3. Results and discussion

#### 3.1. Microscopy and free fat analysis

The microscopic observations of C and H cream samples (20% fat) are shown in Fig. 2. In homogenized cream samples (H), small fat globules joined forming clusters. In this sense, it have been reported that fat globules in cream homogenized can be subject to several kinds of aggregation (Walstra, Geurts, Noomen, Jellema, & Boekel, 1999). By contrast, in unhomogenized samples (C) larger and unclustered fat globules were observed, which is expected for milk fat in its native state.

Free fat was not observed, neither in samples submitted to homogenization treatment nor in C samples, confirming an appropriated performance of the homogenization step. It is known that the formation of visible oil in emulsions - oiling off- can be caused by either an extensive coalescence or ineffectiveness of the homogenization process (Evers, 2004). The validity of our method for assessing free fat was previously checked, as we reported a free fat layer (approx. 8 mm)in cream samples with significant damage of MFGM (by mechanical agitation) and without absorption of surface-active compounds on the fat globulemilk plasma interface agitated samples (Vélez et al., 2010).

On the other hand, H samples were visible whiter than C samples due to an increase in the number of light scattering centers in homogenized samples (Walstra, Geurts, Noomen, Jellema, & van Boekel, 1999).



Fig. 2. Microscopic observations of control cream (C) and homogenized (H) samples. Lines represent 10  $\mu m$ 

### 218 Table 1

mnosition data	of control and	evnerimental	cheeses Significan	ice of the main effect

	Control samples	Experimental samples	Significance
Moisture (%) (3 days)	$42.40 \pm 2.57$	$43.95\pm0.62$	NS
Moisture (%) (90 days)	$30.78 \pm 1.83$	$32.44 \pm 1.16$	NS
FDM% (3 days)	$50.64 \pm 1.85$	$49.35 \pm 4.45$	NS
Protein % (90 days)	$29.19 \pm 1.29$	$26.73 \pm 1.63$	NS
pH (3 days)	$5.42\pm0.03$	$5.30\pm0.04$	S
pH (90 days)	$5.43 \pm 0.05$	$5.33 \pm 0.06$	S
SN-4.6/TNc (90 days)	$15.82\pm2.07$	$13.48 \pm 0.23$	NS

FDM Fat in dry matter.

PDM Protein in dray matter. Soluble nitrogen content (SN) at pH 4.6 as a percentage of total nitrogen.

NS not significant differences (p > 0.05).

S Significant differences (p < 0.05).

#### 3.2. Global composition, microbial counts and proteolysis

Composition data from cheeses is shown in Table 1. At the end of ripening (90 days) the cheeses accomplished the requirements of Argentinean legislation for moisture (maximum value of 35.9%) and fat content (between 45 and 59% in dry matter) (ANMAT, 2014) for this cheese variety.

All cheeses had similar moisture content at the beginning of ripening, followed by a significant (p < 0.05) and equivalent decrease due to water evaporation prior to vacuum packaging. Other authors, as Brito, Manríquez, Molina, and Pinto (2003), Madadlou, Mousavi, Khosrowshahi, Emam-Djome, and Zargaran (2007), Rowney, Hickey, Roupas, and Everett (2003) and Zamora et al. (Zamora, Ferragut, Juan, Guamis, & Trujillo, 2011) found higher moisture in cheeses made with homogenized milk fat, in comparison with control counterparts. They adjudicated the differences to the poor syneresis of curds prepared from homogenized milk as homogenized fat globules interact with para-casein matrix (Kelly et al., 2008).

Considering fat content, H and C cheeses had similar values, although differences (p < 0.05) were detected in whey from both types of cheeses. The whey derived from homogenized cheeses had lower content of fat (0.17%) than that from control cheeses (0.5%). Even if differences in fat content of whey were noticed, values were low and this effect was not reflected in cheese fat contents. The lower fat losses in whey from homogenized cheeses could be due to the size reduction of fat globules and to the modified MFGM, which is associated with proteins. Fat content was similar to Vélez et al. (2010) for the same type of cheese, prepared with raw milk (3% fat). Different results were reported in relation to the effect of low-pressure homogenization on fat content of cheeses and no previous reports on hard cooked cheeses are available. Nair, Mistry, and Oommen (2000) found that fat content (dry matter) of Cheddar cheeses made with homogenized cream at 10.4/3.5 MPa was lower than cheeses obtained from cream homogenized at 3.5/3.5 and 6.9/3.5 MPa, but similar to control cheeses. In this case, the results were correlated with whey fat contents. Rowney et al. (2003) and Zamora et al. (2011) studied Mozarella and fresh cheeses manufactured with homogenized milk at 2.6 and 15 MPa, respectively, showing an increase in their fat content compared to unhomogenized samples. Deegan and McSweeney (2013) found a diminution in fat (dry matter) for Cheddar cheeses made using homogenized milk at different pressures (5, 10, 15, 20, 25 MPa) compared to controls, but they reported a decrease in fat content of whey with increasing pressure. Metzger and Mistry (Metzger & Mistry, 1994) described an increase in fat in dry matter in Cheddar cheeses made with homogenized cream at 17.26/3.43 MPa compared to no-treated cheeses; fat in whey decreased for cheeses made with homogenization treatment. Likewise, Madadlou et al. (2007) reported higher fat content in cheeses manufactured with homogenized cream at 6.0/2.5 and 9.0/2.5 MPa.

The protein content of cheeses at the end of ripening showed similar values between treatments, and were comparable to the values reported for Reggianito cheeses (Candioti et al., 2002; Vélez et al., 2010).

As for pH, there were no significant differences between cheeses (p > 0.05). Variable results were reported in relation to the effect of milk homogenization on cheese pH. Deegan et al. (2013) found no influence of milk homogenization (10 MPa) in Emmental cheeses. On the contrary, other authors showed an increase in pH values in Cheddar made with homogenized milk (Deegan & McSweeney, 2013; Madadlou et al., 2007; Metzger & Mistry, 1994; Nair et al., 2000).

In respect to microbiology, at 3 days microbial counts were  $10^8$  CFU/g in all cheeses, and decreased one log order during ripening.

On the other hand, proteolysis levels also kept constant (p > 0.05) between cheeses, and they were comparable to those reported for matured Reggianito cheeses (Candioti et al., 2002; Wolf, Perotti, Bernal, & Zalazar, 2010). Similar results were obtained by other authors (Metzger & Mistry, 1994; Nair et al., 2000).

#### 3.3. *Lipolysis in cheeses*

The degree of lipolysis showed differences between both types of cheeses: for control cheeses at 3, 45 and 90 days of ripening the values were 4595, 9588 and 19,644 mg FFA/kg cheese, respectively; instead, for H cheeses the values were 16,297, 19,740 and 23,013 mg/kg cheese. TFFA average value was significantly (p < 0.05) higher in cheeses from homogenized cream than in control cheese sat 3 and 45 d of ripening, but at 90 d their levels matched (p > 0.05). In particular, TFFA were 72, 51 and 15% higher in H cheeses than in C cheeses at 3, 45 and 90 days, respectively. At the end of ripening, the global level of lipolysis for all cheeses was around 21,000 mg/kg, which was higher than the values reported for 6 month-old Reggianito cheeses. Perotti et al. (2005) and Wolf et al. (2010) found lipolysis levels of about 1400–3000 mg/kg for Reggianito prepared with pasteurized milk, while values of 6500 mg/kg were detected by Vélez et al. (2010) in cheeses made with milk sanitized by gravitational creaming.

At 3 days, C cheeses had 1.5% fat hydrolyzed, while H cheeses had 5.9%. Through ripening, both types of cheeses reached similar values, approximately 6% hydrolysis. In comparison with grana Italian cheeses, lipolysis progress was higher: Parmigiano Reggiano and Grana Padano cheeses (18–24 months ripening) have approx. 2% fat hydrolyzed (Battistotti & Corradini, 1993) with lipolysis degree between 4993 and 13,697 ppm (Gobetti & DiCagno, 2003). In this type of cheese, raw milk is sanitized by gravitational creaming (milk is left in shallow separation basins for about 8–12 h), which promotes the action of LPL (Wilkinson, 2007). The advance of lipolysis found in the present work is comparable to that in mold cheeses: 5–10% fat hydrolysis in Camembert and up to 20% in blue-vein cheeses (Collins et al., 2003). In particular, in Danablu, a semi-hard cow's cheese prepared with homogenized cream and skim milk and maturated for 3 months (Ardö, 2011) the concentration of FFA can exceed 30,000 ppm.

On the other hand, the accumulation of TFFA followed different rates in both cheese types. In H cheeses, the production of FFA was accelerated immediately after cheesemaking, and then it increased by 1.2 and 1.4 times towards the middle and the end of ripening. In C cheeses, the accumulation – from a much lower starting point – increased progressively up to the end of ripening, by 2 times from 3 days up to 45 d, and by 4.3 times up to 90 d, which allowed to match the TFFA of H cheeses.

The increase trend found for C cheeses over ripening have been described for numerous cheese varieties (Malacarne et al., 2009; O'Mahony, Sheehan, Delahunty, & McSweeney, 2006; Voigt, Chevalier, Qian, & Kelly, 2010). As for cheeses made with low-pressure homogenized milk, an early lipolysis was also reported by Brito et al. (2003), Michalski et al. (2004), Deegan et al. (2013) and Deegan and McSweeney (2013) for Chanco, Emmental and Cheddar cheeses, respectively. In particular, Deegan et al. (2013)and Deegan and McSweeney (2013)incubated the milk after homogenization during 1 h at 37 °C prior to cheese making in order to allow lipolysis to proceed unhindered.

In the present work the results indicate that the technological approach was successful at accelerating hydrolysis of TG. During the incubation of milk (for 12 h at 12 °C) before cheese making, lipolytic enzymes from milk and microbial lipases acted on acylglycerides mainly in H cheeses, as homogenization improved fat accessibility. The progressive increase of lipolysis over 90 d in C cheeses and the fact that C and H cheeses reached similar levels at the end of ripening, show that enzymes remained active during the storage period. Though, fat accessibility seems to be a limiting factor in fat lipolysis, but the extension of the reaction would also depend on the preservation of enzyme activities during ripening and the susceptibility of enzymes to product inhibition. Some authors reported that lipolytic activity of cheeses made with raw and pasteurized milk remained constant during ripening, but the higher activity in raw cheeses was adjudicated to LPL and enzymes from NSLAB (Svensson, Hernández, Virto, & De Renobales, 2006). NSLAB have weak to moderate lipase and esterase activities which contribute to FFA release (Holland et al., 2005). As for LPL, there are no studies of its specific activity in cheese matrix but it is known that is susceptible to product inhibition (Olivecrona, Vilaro, & G, 2003).

As for the percentage of FFA groups in relation to total FFA, some differences were detected between cheeses. For H cheeses, the percentage of SCFFA (C4:0-C8:0) increased from 3% at 3 d to 6.6% at 45 d and it remained constant until the end of ripening, while in C cheeses the percentage increased from 3% at 3 d to 9% at 45 d and then it diminished to 6% at 90 d. Thereby, at 90 d, C and H cheeses had similar SCFFA percentages. The percentage of medium-chain free fatty acids (MCFFA, C10:0-C14:0) was around 25% at the three sampling times for H cheeses, contrary to what happened in C cheeses in which it remained constant up to the middle of ripening and then it increased equalizing the values found in H cheeses (20, 18 and 25% at 3, 45 and 90 days, respectively). For C cheeses, the percentage of long-chain free fatty acids (LCFFA, C16:0-C18:2) represented approx. 77% of TFFA at 3 d and then diminished until 72% and 68% at 45d and 90d, respectively. For H cheeses, lower percentages were obtained, being of about 71% at 3 d and then diminished to 69% at 45 d and remained constant up to 90 days. Again, the values matched for both types of cheeses at the end of ripening. This point could be due to the fact that both, H and C cheeses, have similar enzymatic specificity along ripening.

The FFA groups' percentages during ripening of cheeses subject to low-pressure homogenization have not been reported as such to the best of our knowledge. Yet calculations made from published values indicate that FFA composition does vary in some cases. Deegan and McSweeney (2013) did not find differences in %SCFFA during ripening (180 d) between Cheddar cheeses made with homogenized milk (at 5 and 10 MPa) and their controls; although a significant increase of this fraction was verified for all cheeses during ripening (180 d). A similar trend was found for %MCFFA. As for LCFFA group, no clear pattern was found through ripening time. Meanwhile, Deegan et al. (2013)reported an increase of %SCFFA but a decrease of %MCFFA and %LCFFA through time, for Emmental cheeses manufactured with homogenized milk (at 5 and 10 MPa) and their respective un-treated cheese controls. In particular, at the beginning of ripening, %SCFFA and %MCFFA were lower in homogenized cheeses compared to controls; while %LCFFA were similar among cheeses. In all cases, the percentage values at the end of ripening were matched.

The amounts of individual FFA, in C and H cheeses at each sampling time are shown in Table 2. Palmitic and myristic were the main saturated FFA and oleic was the major unsaturated FFA found in all cheese samples. FFA profile is similar to the distribution found in milk fat (MacGibbon & Taylor, 2006) and in Reggianito cheeses (Perotti et al., 2005; Vélez et al., 2010).

At 3 and 45 d, all FFA concentrations were higher in H than in C cheeses (p < 0.05), whereas at 90 d the behavior was similar except for C6:0, C14:0, C18:1 and C18:2, where concentrations remained without differences.

FFA data were subjected to PCA (Fig. 3) using the correlation matrix (mean-centred and scaled). The value of Kaiser-Meyer-Olkin (KMO) measure of sampling adequacy was 0.89 which is greater than the minimum value 0.6 indicating that the size of sample is adequate to consider PCA as appropriate tool to reduce the data and the Bartlett's test of sphericity was significant indicating that the intercorrelation among the variables was strong to be considered in PCA (Hair et al., 1999). The first two principal components (PC) explained 89% of the total variance, PC1 and PC2 describing 78% and 11% of the variation, respectively. PC1 was positively correlated to all fatty acids analyzed, while PC2 was positively associated with short-chain FFA (butyric, caproic and caprylic), oleic and linoleic acids and negatively associated with medium-chain FFA (capric, lauric and myristic), palmitic and stearic acids. Globally, H cheese samples were clearly distinguished from C samples and within each cheese type were grouped by ripening time. E samples of 3 d were placed on the lower right quadrant characterized with higher concentrations of palmitic and stearic acids, while samples of 45 d and 90 d could not be noticeably separated between them and were grouped along positive PC2 axis. By contrast, unripened C samples appeared in the lower left quadrant, and samples of 45d and 90d were located in the upper left and in the upper right quadrants, respectively. The latter was characterized by higher contents of SCFFA and C18:1 and C18:2.All E samples (E-t3, E-t45, E-t90) and C samples of 90 d (C-t90) were placed in the positive semi-plane of PC1axis, as their lipolysis levels were similar, and differed from the other C samples (C-t3 and C-t45) which were situated in the negative semi-plane of PC1 axis. PC1 separated cheese samples with cheese age, from negative to positive as the ripening progressed, and PC2 differenced samples based on

Table 2

FFA conce	entrations of	control (0	2) and	experimental	(E) (	cheeses di	uring	ripening	g, expressed	l as µmol	/100 g	fat. I	Mean values o	f two independ	ent cheese	e making tr	ials.
-----------	---------------	------------	--------	--------------	-------	------------	-------	----------	--------------	-----------	--------	--------	---------------	----------------	------------	-------------	-------

Free fatty acids	3 days		45 days		90 days		
	С	E	С	E	С	E	
C4:0	$245^{b} \pm 45$	$603^{a} \pm 92$	1838 <sup>b</sup> ± 150	$2327^{a} \pm 338$	$2234^{a} \pm 168$	2025 <sup>b</sup> ± 185	
C6:0	$128^{b} \pm 34$	$427^{a} \pm 97$	$539^{b} \pm 42$	$741^{a} \pm 31$	$1046^{a} \pm 36$	$1033^{a} \pm 123$	
C8:0	$105^{b} \pm 22$	$508^{a} \pm 62$	$357^{b} \pm 63$	$618^{a} \pm 52$	$571^{b} \pm 25$	$730^{a} \pm 37$	
C10:0	$203^{b} \pm 62$	$913^{a} \pm 118$	$471^{b} \pm 55$	$1175^{a} \pm 132$	$687^{b} \pm 23$	$1221^{a} \pm 103$	
C12:0	$209^{b} \pm 72$	$601^{a} \pm 44$	$480^{b} \pm 53$	$907^{a} \pm 95$	$542^{b} \pm 42$	$845^{a} \pm 57$	
C14:0	$1030^{b} \pm 350$	$5410^{a} \pm 306$	$1518^{b} \pm 58$	$4558^{a} \pm 204$	$5027^{a} \pm 969$	$5424^{a} \pm 1505$	
C16:0	$2124^{b} \pm 452$	$8650^{a} \pm 190$	3235 <sup>b</sup> ± 144	$7411^{a} \pm 422$	$6759^{b} \pm 609$	$9347^{a} \pm 1354$	
C18:0	$514^{a} \pm 42$	$1828^{b} \pm 78$	$834a \pm 63$	$2146^{b} \pm 150$	$1097^{b} \pm 112$	$1918^{a} \pm 90$	
C18:1	$1574^{b} \pm 223$	$4434^{a} \pm 289$	$3016^{b} \pm 353$	$4384^{a} \pm 203$	$5242^{a} \pm 250$	$4826^{a} \pm 730$	
C18:2	$289^{b} \pm 71$	$748^{a} \pm 93$	$449^{b} \pm 39$	$707^{a} \pm 40$	$1044^{a}\pm130$	$853^{a} \pm 107$	

<sup>a</sup> Numbers represent mean  $\pm$  standard deviation.

<sup>b</sup> For each time of ripening, means in a row without common letters differ (p < 0.05).



Fig. 3. Loading (A) and score (B) plots of the first and second principal components after Principal Component Analysis (PCA) based on the free fatty acids (FFA) of hard cheeses made from homogenized milk (9 MPa) (Hcheeses) and untreated milk (C cheeses) at 3, 45 and 90 days (t3, t45, t90) of storage at 12 °C. Cheeses were carried out in duplicate (1 and 2) from two independent cheese making trials (A and B).

the relative differences among individual FFA by the treatment applied to cheese milk. In particular, the samples C-t3, C-t45 and C-t90 showed the most visible separation with ripening time, indicating a progressive

increase in lipolysis for this type of cheeses; in contrast, the samples Et3, E-t45 and E-t90 were nearly placed describing, for this case, similar lipolysis patterns. Finally, in order to validate the visual interpretation of PCA, discriminant analysis (DA) was applied on PC1 and PC2 scores using ripening time and cheese treatment as classification factors. There was no difference between the actual group of cheese sample and assigned group of cheese sample; in fact, all E samples were clearly distinguished of all C samples (100% of cases), while when the time of ripening was used as classification factor, 96% of cases were correctly grouped.

#### 3.4. Volatile profiles

A total of 37 volatile compounds, including 5 aldehydes, 12 alcohols, 8 ketones, 5 esters and 7 acids were identified in cheese samples. They were found in both types of cheeses and have been reported as typical components of volatile profile of grana-type cheeses, including Reggianito (Barbieri et al., 1994; Ceruti, Zorrilla, & Sihufe, 2016; Wolf et al., 2010). Volatiles identified in the analyzed samples and their mean peak area values are listed in Table 3 for individual trials (A and B) at the different ripening times (3, 45 and 90 days). Besides, in order to have an overall impression of the volatile profiles of cheeses, the mean percentage values (average of two trials) of the different chemical groups, calculated with respect to the total peak area values of the identified compounds, were also compared between both types of cheeses (Fig. 4).

#### 3.4.1. Acids

The acidic fraction was mainly composed of even linear-chain (fatty) acids from  $C_2$  to  $C_{12}$ . Lipolysis of milk fat seems to be the metabolic pathway of formation of carboxylic fatty acids with >4 carbon atoms. Shortand medium-chain free fatty acids are important contributors to the flavour in a wide variety of cheeses (Addis, Pirisi, Di Salvo, Podda, & Piredda, 2005; Collins et al., 2003). As seen in Fig. 4, regardless of the ripening time, the volatile fraction of both types of cheeses was composed of >50% of acidic components, representing the main group among all volatiles. Butanoic and hexanoic acids were the most abundant throughout ripening, reaching percentual values above 65% of total acidic compounds (data not shown). This finding has also been observed in samples of Reggianito cheese (Ceruti et al., 2016; Wolf et al., 2010).

The comparison of the peak area values of single fatty acids derived from lipolysis revealed that at the beginning of maturation and at 45 days, levels of the major acids such as butanoic, hexanoic and octanoic acids were higher in H than in C cheeses (p < 0.05) whereas at the end of ripening, there were not statistical differences among cheeses (p > 0.05). Overall, the data confirmed the trends found for lipolysis analysis: both, lipolysis and volatile profiling results showed that the step of homogenization applied to milk had a strong influence on the production of the FFA immediately after cheese making, and this originated great changes with regards to control cheeses. These differences were significant up to middle of ripening, but then decreased and levelled up towards 90 days.

Concerning the behavior of fatty acids over the ripening period, the data in Table 3 revealed that butanoic, hexanoic and octanoic acids shown a trend to increase (p < 0.05). Among minority acids, decanoic acid increased after 45 days (p < 0.05) whereas dodecanoic acid remained unchanged in almost all samples.

In addition to fatty acids derived from lipolysis, the presence of acetic acid and 3-methylbutanoic acid was detected. Acetic acid reached a higher level in cheeses made from unhomogenized milk than those made from homogenized milk at the beginning of ripening (p < 0.05) whereas at 45 days, differences between cheeses were not observed (p > 0.05). At 90 days, the data recorded were inconsistent; a higher level in C than in H cheeses was detected (p < 0.05) in trial A, while statistical differences were not observed in trial B (p > 0.05). On the other hand, the evolution of acetic acid with ripening time did not show a defined trend. These results reveal that the homogenization of milk did not produce in any case an increase in acetic acid content, which is not

surprising taking into account that its biosynthesis in cheeses includes pathways such as lactate/citrate metabolism and amino acid catabolism rather than lipolysis (Ott, Germond, & Chaintreau, 2000). For 3methylbutanoic acid (or isovaleric acid), a typical compound derived from leucine catabolism (McSweeney & Sousa, 2000), the values did not show differences between cheeses at each sampling time and these were not modified during maturation (p > 0.05).

#### 3.4.2. Alcohols

Alcohols were the most diversified chemical group of compounds. Among them, linear-chain primary alcohols (from  $C_2$  to  $C_7$ ), linearchain secondary alcohols (from  $C_3$  to  $C_7$ ) and branched-chain alcohols (2-methyl-1-propanol and 3-methyl-1-butanol) were detected. From Fig. 4, it can be observed that alcohols were the second group of volatile compounds in all cheeses at any ripening time, ranging from 11% to 30% and showing a decrease during ripening for both types of cheeses.

Regardless of the sampling time, primary alcohols represented >80% of total compounds in all samples (data not shown). In particular, ethanol was the most abundant (from 50% to 70% of total alcohols) and its level remained almost steady throughout ripening in all cheeses (p > 0.05). The prevalence of ethanol has been reported in a wide variety of cheeses (Kondyli, Katsiari, Masouras, & Voutsinas, 2002; Wolf et al., 2010). In individual trials, C cheeses had significantly higher contents of ethanol than H cheeses at each sampling time (p < 0.05). As happened in the case of acetic acid, this observation could be explained by the origin of ethanol in cheeses as it is a common terminal end-product in the breakdown of glucose or it is produced from amino acid catabolism such as alanine (Kondyli et al., 2002). For the remaining primary alcohols, statistical differences between cheeses were not detected at any ripening time (p > 0.05). No general pattern of increase or decrease of the levels of these compounds with ripening time was detected.

Secondary and branched-chain alcohols represented a minor fraction, among which 2-pentanol and 3-methyl-1-butanol prevailed. The occurrence of linear-chain primary alcohols longer than ethanol and linear-chain secondary alcohols in cheeses seems to be related, at least partially, with the fatty acid catabolism (Barbieri et al., 1994; Curioni & Bosset, 2002) whereas the presence of branched-chain alcohols is clearly associated to amino acid catabolism. In general, no significant changes attributable to homogenization were found, as high variability with the ripening time or the trial was verified.

#### 3.4.3. Aldehydes

Within the chemical family of aldehydes, acetaldehyde, 3methylbutanal, hexanal, heptanal and nonanal were detected. At any ripening time, the proportion of aldehydes was low with respect to other chemical classes (from 2% to 4%) (Fig. 4). Acetaldehyde constituted the most abundant aldehyde, ranging from 40% to 75% of total aldehydes, depending on the sampling time (data not shown). The levels found were higher in C than in H cheeses until 45 days (p < 0.05), and then, statistical differences were not detected (p > 0.05). This result is in agreement with the origin of acetaldehyde in cheeses, which does not appear to be related to fatty acid catabolism. In fact, it has been suggested that could derive from the breakdown of threonine, from the lactose metabolism or by the oxidation of ethanol (McSweeney & Sousa, 2000). Besides, at this point, it is important to highlight that for the three metabolically related compounds, namely acetaldehyde, ethanol and acetic acid, homogenized cheeses had similar or lower levels than unhomogenized counterparts, indicating that homogenization does not favor their production. Regarding to the evolution of acetaldehyde during ripening, the levels did not change in C cheeses (p > 0.05), whereas in H cheeses an unclear trend was observed.

In relation to 3-methylbutanal, the levels were not affected by the type of cheese or the ripening time, which was in accordance with the origin of this compound; it derives from leucine catabolism (Curioni & Bosset, 2002). Straight-chain aldehydes such as hexanal, heptanal and nonanal were higher (p < 0.05) in H than in C cheeses at 90 days.

#### Table 3

Volatile compounds at different stages of ripening of cheeses made with non-homogenised milk (C cheeses) and homogenised milk (H cheeses).<sup>a</sup>

		Trial A		Trial B	
Volatile compounds	Time of ripening	С	Н	С	Н
Aldehvdes					
Acetaldehyde	3	$177.7\pm8.8$ aA	$93.8\pm9.1$ bA	$172.5\pm16.1$ aA	$110.5 \pm 11.9 \mathrm{bB}$
	45	$201.3 \pm 13.6 \text{aA}$	$129.0\pm10.7\text{bA}$	$167.4\pm10.3\textrm{aA}$	$118.4\pm5.5bB$
	90	$166.5 \pm 12.5$ aA	$173.1 \pm 8.4aA$	$143.1\pm12.1\mathrm{aA}$	$156.9 \pm 7.3$ aA
3-methylbutanal	3	$18.3 \pm 1.1aA$	$12.7 \pm 0.8 \text{bA}$	$19.5 \pm 0.6aA$	$11.0 \pm 1.9 \text{bB}$
	45	$22.2 \pm 6.2aA$	$18.4 \pm 5.4aA$	$25.2 \pm 0.1aA$	$23.5 \pm 3.0aA$
Hexanal	3	$98 \pm 15aB$	$21.2 \pm 5.0$ a A $7.7 \pm 0.9$ a B	$23.0 \pm 2.7 dA$ 10.8 + 1.4 aB	$20.0 \pm 2.4$ arc $8.6 \pm 1.4$ arc $R$
Textilui	45	$6.8 \pm 0.2aB$	$5.7 \pm 1.1 aB$	$7.0 \pm 2.7 aB$	$10.8 \pm 1.3 aB$
	90	$16.4 \pm 1.6$ aA	$29.6\pm3.4aA$	$21.9 \pm 2.2 \text{bA}$	$35.9\pm3.3$ aA
Heptanal	3	$23.3\pm2.1aB$	$19.5\pm3.0\mathrm{aB}$	$23.4 \pm 3.1 \mathrm{aB}$	$16.7\pm1.7aB$
	45	$25.6 \pm 3.3 aB$	$30.6 \pm 2.1 aB$	$18.1 \pm 3.0 \mathrm{aB}$	$16.2\pm2.9aB$
	90	48.2 ± 5.7bA	$82.3 \pm 4.0$ aA	51.5 ± 4.5bA	$85.8\pm6.7aA$
Nonanal	3	$12.6 \pm 0.6aB$	$11.5 \pm 1.6aB$	$12.8 \pm 0.6aB$	$10.9 \pm 0.8aB$
	45 90	$18.1 \pm 3.7 \text{dB}$ 51.6 ± 8.9bA	$21.9 \pm 0.9$ dB 100.3 $\pm 13.6$ aA	$9.5 \pm 0.2 \text{ ab}$ 53.0 $\pm$ 7.0 bA	$9.5 \pm 1.9$ dB 101 7 $\pm$ 13 5 a A
Ketones	50	51.0 <u>+</u> 0.55M	100.5 ± 15.00/1	55.0 ± 7.00M	$101.7 \pm 15.5 d M$
Propanone	3	$16.6\pm0.7 \mathrm{aB}$	$13.4 \pm 1.5$ aA	$15.7 \pm 1.4$ aB	$10.4\pm2.0\mathrm{aB}$
-	45	$51.6 \pm 4.7 aA$	$13.0 \pm 1.9 \text{bA}$	$48.7\pm4.3$ aA	$30.1\pm3.0\text{bA}$
	90	$51.0\pm9.0aA$	$50.7\pm2.9aB$	$43.1\pm6.2aA$	$38.1\pm4.7aA$
Butanone	3	$13.5\pm0.6aB$	$9.2\pm1.2$ bA	$15.0 \pm 0.3 aB$	$8.1 \pm 0.5 bC$
	45	$45.9 \pm 2.6$ aA	17.9 ± 3.9bA	$52.7 \pm 5.4$ aA	31.8 ± 1.1bA
	90	$46.5 \pm 2.4aA$	$14.4 \pm 3.4bA$	$20.3 \pm 1.2aB$	$15.3 \pm 0.8 \text{bB}$
2,3-butanedione + 2-pentanone	3	$118.2 \pm 4.9aA$ 122.0 $\pm 7.1aA$	$95.1 \pm 8.1 \text{ as}$	$109.7 \pm 9.6aA$	$98.2 \pm 7.5 a$ A
	90	$132.9 \pm 7.1a$ A	$125.2 \pm 4.3ab$ 207.0 $\pm$ 9.1 a	$112.9 \pm 0.2a$ A 77.4 + 7.6bA	$100.2 \pm 2.2a$ A
2-hexanone	3	73.1 + 7.2aA	$53.6 \pm 3.1aA$	$70.5 \pm 8.0aA$	$57.2 \pm 11.5aA$
	45	5.6 + 1.9bC	30.2 + 3.2aB	7.6 + 0.6bC	$28.3 \pm 4.6aB$
	90	$30.4 \pm 0.9 \text{bB}$	$57.3 \pm 4.7aA$	$41.3 \pm 6.3 \text{bB}$	$65.2 \pm 4.1$ aA
2-heptanone	3	$121.6\pm9.2aA$	$104.3\pm7.4aA$	$133.1\pm7.7$ aA	$117.2\pm13.7\mathrm{aA}$
	45	$32.6 \pm 2.8 \text{bB}$	$167.7\pm14.6\mathrm{aB}$	$45.0\pm6.5 \mathrm{bB}$	$192.3\pm7.6\mathrm{aB}$
	90	$55.3 \pm 7.8 \text{bB}$	$386.6 \pm 23.8aC$	$128.9 \pm 12.2 \text{bA}$	$334.8 \pm 18.6aC$
3-hydroxy-2-butanone	3	$416.9 \pm 30.5 aA$	$274.1 \pm 22.6 \text{bA}$	$374.8 \pm 46.5aA$	$203.5 \pm 18.1$ bA
	45	$123.3 \pm 9.1aB$	$94.8 \pm 5.2aB$	$160.5 \pm 12.1aB$	$52.7 \pm 9.70B$
2-00-000	3	$93.7 \pm 7.0$ dB	$55.7 \pm 4.908$ $7.7 \pm 0.954$	$122.8 \pm 8.5 dB$ $10.2 \pm 0.9 \Delta$	$40.0 \pm 9.40B$ $9.9 \pm 0.554$
2-110118110116	45	$9.0 \pm 0.00$ A	$7.7 \pm 0.9a$ A $77.2 \pm 13.6a$ B	$9.2 \pm 0.9 a$ A	$3.9 \pm 0.3 a$ A
	90	$26.5 \pm 3.3 \text{bB}$	$166.2 \pm 11.7aC$	$44.4 \pm 5.0$ bB	$150.0 \pm 10.8 aC$
Alcohols					
2-propanol	3	$36.5 \pm 2.4 \mathrm{aAB}$	$27.1\pm5.1aAB$	$31.0\pm2.8aA$	$23.4\pm4.3$ aA
	45	$29.8\pm1.7aB$	$36.4\pm4.6aA$	$29.6\pm3.1\text{aA}$	$20.2\pm2.2a\text{A}$
	90	$49.4\pm7.3$ aA	$13.0 \pm 3.8 \text{bB}$	$27.3 \pm 1.1$ aA	$26.6 \pm 3.7 aA$
Ethanol	3	$1349.7 \pm 30.3$ aA	$1098.0 \pm 55.1$ bA	$1460.3 \pm 65.5 aA$	$996.0 \pm 71.8 \text{bA}$
	45	$1513.8 \pm 91.2aA$	$1009.7 \pm 96.50A$	$1365.4 \pm 84.8aA$	$1044.7 \pm 77.6 \text{ DA}$
2-butanol	3	$97 \pm 193$	$640.1 \pm 65.90A$ 7 3 $\pm$ 1 1 aB	$1311.9 \pm 33.7 dA$ $9.2 \pm 1.4 aB$	$1070.0 \pm 47.00A$ $77 \pm 0.32B$
2-Dutation	45	$16.6 \pm 2.9aA$	$152 \pm 1.1ab$	$16.1 \pm 0.6$ bA	$34.6 \pm 2.5aA$
	90	$5.5 \pm 0.1$ bB	$34.0 \pm 5.2aA$	$7.8 \pm 1.9 \text{bB}$	$33.2 \pm 1.4aA$
1-propanol	3	$85.9\pm10.4\text{aB}$	$82.1\pm6.8$ aA	$94.7\pm4.4aB$	$86.2\pm6.7aA$
	45	$613.4\pm24.6aA$	$83.1 \pm 1.2 \text{bA}$	$276.9\pm34.7aA$	$106.0\pm2.3\text{bA}$
	90	$651.3 \pm 49.2$ aA	$64.9 \pm 4.3 \mathrm{bA}$	$98.2 \pm 2.5 aB$	$109.5 \pm 4.7 \mathrm{aA}$
2-methyl-1-propanol	3	$13.5 \pm 1.1$ aA	$14.3 \pm 2.0$ aA	$11.4 \pm 0.6aA$	$9.2 \pm 0.5$ aA
	45	$17.7 \pm 1.5aA$	$19.4 \pm 0.8aA$	$13.6 \pm 3.3aA$	$12.6 \pm 1.4aA$
2-pentapol	3	$14.7 \pm 1.2 dA$ $02.8 \pm 3.5 cB$	$10.7 \pm 2.5 dA$ 68.6 ± 4.3bB	$15.1 \pm 0.4$ dA	$14.4 \pm 2.0$ A
2-pentanoi	45	$92.8 \pm 3.3 \text{ ab}$ 91.0 + 3.3 aB	$71.9 \pm 2.9$ bB	$57.7 \pm 2.4$ and $57.1 \pm 12.5$ aB	$57.8 \pm 7.9aA$
	90	264.9 + 28.4aA	$116.9 \pm 10.0$ bA	$69.3 \pm 2.5aAB$	73.3 + 3.4aA
1-butanol	3	$128.9 \pm 2.2$ aA	$122.9 \pm 21.8$ aA	$126.6 \pm 6.3$ aA	$113.7 \pm 8.2 aA$
	45	$88.7\pm3.7\mathrm{aB}$	$86.3\pm8.7aB$	$48.7\pm8.1\mathrm{aB}$	$43.7\pm7.7aB$
	90	$60.5\pm14.7 \mathrm{aB}$	$59.0 \pm 15.0 \mathrm{aB}$	$39.7\pm9.9aB$	$35.2\pm9.5aB$
3-methyl-1-butanol	3	$235.6\pm5.7\mathrm{aA}$	$202.4 \pm 18.4 \mathrm{aA}$	$225.9 \pm 17.1 \mathrm{aA}$	$218.1\pm17.6\mathrm{aA}$
	45	$143.2 \pm 22.0 aB$	$132.9 \pm 6.7 aB$	$149.3 \pm 13.5 aB$	$155.5 \pm 8.6aB$
1 poptanol	90	$65.1 \pm 7.1aC$	$69.5 \pm 7.1 aC$	$83.5 \pm 3.0aC$	$79.1 \pm 5.5aC$
i-pentanoi	5 45	$10.2 \pm 1.3 dC$ 38.8 $\pm 6.6 pP$	$14.5 \pm 1.1 d$ $43.6 \pm 1.2 p$	$10.1 \pm 1.3 dB$ $30.5 \pm 2.4 pB$	$11.3 \pm 2.2 \text{ as}$
	90	$50.0 \pm 0.0ab$ 69.8 + 5.1hA	$-1.0 \pm 1.3aD$ 967 + 19aA	مد 2.440 x 2.4400 x 2.44	$23.2 \pm 4.0$ dD 88.4 + 18.1 a
2-heptanol	3	13.3 + 2.1aB	7.7 + 1.8aB	11.8 + 1.2aB	$11.9 \pm 0.4aB$
r	- 45	$25.6 \pm 1.6 \text{bB}$	$56.2 \pm 1.2$ aA	$16.0 \pm 4.0 \mathrm{aB}$	$17.9 \pm 0.4aB$
	90	$92.2\pm9.6$ aA	$50.8\pm5.3$ bA	$31.0\pm2.6aA$	$33.7\pm4.1$ aA
1-hexanol	3	$30.6\pm1.5 \text{aB}$	$32.1\pm2.4aB$	$20.9\pm4.1\text{aB}$	$23.0\pm3.4\text{aB}$
	45	$27.2\pm1.5aB$	$26.6\pm2.1 \text{aB}$	$15.5\pm3.5aB$	$11.2\pm2.2aB$
	90	$65.6 \pm 6.4aA$	$69.2 \pm 10.3$ aA	$55.3 \pm 10.4$ aA	$60.2 \pm 10.9$ aA
i-neptanoi	3 45	$9.8 \pm 1.0aC$	$7.5 \pm 0.8aC$	$9.1 \pm 0.9aC$	$12.0 \pm 0.3aC$
	40	JO.∠ ± ∠.JdĎ	J2.0 ± 1.9dB	23.3 ± 2.2dD	23.0 ± 2./dB

#### Table 3 (continued)

		Trial A		Trial B	
Volatile compounds	Time of ripening	С	Н	С	Н
	90	$81.5\pm5.7$ aA	$85.4\pm8.1$ aA	$78.0\pm0.9$ aA	$84.9\pm6.2\text{aA}$
Esters					
Ethyl acetate	3	$67.7 \pm 4.9aC$	$39.4 \pm 2.7 \text{bB}$	$65.9 \pm 5.7 aC$	$43.9 \pm 2.2 \text{bB}$
	45	$160.9\pm6.2aB$	$47.8 \pm 0.7 \text{bB}$	$107.4\pm6.6aB$	$61.2 \pm 8.1 \text{bB}$
	90	$218.0\pm14.3\text{aA}$	115.7 ± 5.5bA	$169.5\pm10.1$ aA	$108.6\pm9.0\text{bA}$
Ethyl butanoate	3	$183.5 \pm 9.1 \mathrm{aC}$	123.0 ± 17.6bB	$196.0 \pm 8.1 \mathrm{aC}$	$140.8\pm12.0\text{bB}$
	45	$504.3 \pm 42.9 \mathrm{aB}$	137.6 ± 2.3bB	$382.6 \pm 16.6 aB$	$232.9\pm30.7\text{bB}$
	90	$794.6 \pm 65.6$ aA	$540.2 \pm 8.4 \text{bA}$	$659.9 \pm 36.1 aA$	503.1 ± 75.5bA
Ethyl hexanoate	3	$87.1 \pm 8.6$ aC	$66.0\pm8.8\mathrm{aB}$	$84.1 \pm 6.2aC$	$70.4\pm6.3aC$
	45	$228.1 \pm 6.7 \mathrm{aB}$	$107.5 \pm 9.1 \text{bB}$	$180.7 \pm 2.5 \mathrm{aB}$	$173.8\pm20.3aB$
	90	$541.9\pm70.3$ aA	$282.1 \pm 43.2 \text{bA}$	$344.4 \pm 12.4$ aA	$383.5\pm33.8$ aA
3-methylbutyl butanoate	3	$6.5\pm0.8 \mathrm{aB}$	$5.4 \pm 0.7$ aB	$7.8\pm2.1$ aB	$6.0 \pm 1.4$ aB
	45	$16.3 \pm 2.8$ aB	$17.1 \pm 3.2$ aB	$8.7\pm2.2$ aB	$9.7\pm1aB$
	90	$50.4 \pm 12.9$ aA	$52.4 \pm 8.5$ aA	$31.6 \pm 3.3$ aA	$36.1 \pm 1.7$ aA
Ethyl octanoate	3	$38.1 \pm 3.9 \mathrm{aB}$	$25.6 \pm 3.6$ aC	$34.4 \pm 2.2$ aA	$24.9 \pm 4.4$ aA
-	45	$36.7 \pm 4.2 \mathrm{aB}$	$45.8 \pm 5.1$ aB	$29.6 \pm 4.0$ aA	$18.5\pm2.1$ aA
	90	$59.8\pm5.5$ aA	$71.3 \pm 7.4$ aA	$13.8 \pm 1.3$ aB	$16.9\pm2.7$ aA
Acids					
Acetic acid	3	$886.1\pm91.3aAB$	683.7 ± 61.1bA	$862.0 \pm 97.2$ aA	$579.2 \pm 30.1 \text{bB}$
	45	$626.4\pm21.0$ aB	$649.3 \pm 56.7 aA$	$805.3 \pm 43.9$ aA	$687.9 \pm 27.0$ aAB
	90	$1138.5 \pm 115.3$ aA	654.1 ± 77.7bA	$856.2 \pm 43.3$ aA	$742.1 \pm 46.0$ aA
Butanoic acid	3	$1448.3 \pm 108.1 \text{bC}$	$1806.7 \pm 98.9 {\rm aC}$	1898.2 ± 137.2bC	$2476.7\pm60.2\mathrm{aC}$
	45	$2632.7 \pm 89.1 \text{bB}$	$4075.3\pm80.9\mathrm{aB}$	$3257.4 \pm 210.0 \text{bB}$	$4279.9 \pm 183.3 aB$
	90	$5111.7 \pm 66.4$ aA	$5370.8\pm93.8\mathrm{aA}$	$5292.0 \pm 374.6$ aA	$5977.4 \pm 250.9$ aA
3-methyl butanoic acid	3	$22.2\pm0.7$ aA	$29.4 \pm 3.1$ aA	$19.4\pm1.8$ aA	$13.0\pm2.1$ aA
	45	$25.3 \pm 1.5$ aA	$28.3 \pm 2.5$ aA	$17.0 \pm 3.0$ aA	$13.1\pm2.8$ aA
	90	$28.9 \pm 2.4$ aA	$31.3 \pm 1.9$ aA	$20.3 \pm 3.7$ aA	$19.0\pm4.0$ aA
Hexanoic acid	3	$716.0 \pm 67.3 \text{bC}$	$1107.9 \pm 92.1 aC$	937.9 ± 41.8bC	$1541.4 \pm 81.6aC$
	45	$1444.2 \pm 23.5 \text{bB}$	$2320.2\pm14.2\mathrm{aB}$	$1663.4 \pm 79.3 \text{bB}$	$2281.3 \pm 161.6 aB$
	90	$2788.9 \pm 79.5$ aA	$3012.4\pm80.4$ aA	$2826.7 \pm 122.4$ aA	$3146.9 \pm 62.4$ aA
Octanoic acid	3	$147.6 \pm 18.3 \text{bC}$	$272.6 \pm 28.9aC$	170.8 ± 25.7bC	$462.1\pm19.5 \mathrm{aC}$
	45	$280.9 \pm 31.2 \text{bB}$	$422.5 \pm 17.7 aB$	$369.4 \pm 28.5 \text{bB}$	$603.3\pm27.6aB$
	90	$592.2\pm51.9$ aA	$701.1 \pm 29.9$ aA	$653.3 \pm 37.0$ aA	$727.8\pm29.8$ aA
Decanoic acid	3	$49.6 \pm 9.6 \mathrm{bB}$	$108.4 \pm 13.4$ aB	$47.9 \pm 3.4 \text{bB}$	$176.6 \pm 14.8$ aB
	45	$66.1 \pm 16.6 \text{bB}$	$138.6 \pm 4.6 aB$	$83.5 \pm 12.0 \text{bB}$	$193.2\pm15.8\mathrm{aB}$
	90	$118.5 \pm 5.0 \text{bA}$	$253.8 \pm 7.8$ aA	152.5 ± 10.3bA	$254.3 \pm 5.3$ aA
Dodecanoic acid	3	$10.6 \pm 1.0$ aA	$16.5\pm2.0$ aA	$9.8\pm0.7bB$	$21.1 \pm 2.4$ aA
	45	$11.0 \pm 1.1$ aA	$13.2\pm1.2$ aA	$12.6 \pm 0.9 \text{bB}$	$23.7\pm0.6$ aA
	90	$13.2 \pm 1.3$ aA	$17.0 \pm 1.4$ aA	$19.4 \pm 2.3$ aA	$25.2 \pm 1.3$ aA

a,b: type of cheese factor: for each trial, the same lowercase letters within a row denote no significant differences (p > 0.05) between both types of cheeses.

A–C: time of ripening factor: For each cheese, the same capital letters within a column denote no significant differences (p > 0.05) between values obtained at different days of ripening according to Tukey's ANOVA.

<sup>a</sup> Peak area values (×10<sup>3</sup>) of volatile compounds identified in C and H cheeses from A and B trials. Peak areas are mean values of two cheeses analyzed in duplicate.

They can be formed by  $\beta$ -oxidation of unsaturated fatty acids (Barbieri et al., 1994; Curioni & Bosset, 2002), and our results showed a significant effect of the pretreatment of milk. Differences were only at the end of ripening, suggesting that transformation of FFA followed the initial accelerated lipolysis found in H cheeses.

#### 3.4.4. Ketones

Among ketones group were detected methylketones (propanone, butanone, 2-pentanone, 2-hexanone, 2-heptanone and 2-nonanone), diketones (2,3-butanedione or diacetyl) and hydroxyketones (3-hydroxy-2-butanone or acetoin). As seen in Fig. 4, they ranged from 3% to 11%, depending on type of cheese and ripening time. The most abundant ketones identified in all samples were diacetyl + 2-pentanone (unresolved peak), acetoin and 2-heptanone. At the beginning of ripening, the proportion of acetoin ranged from 40% to 50% of total ketones in both types of cheeses and then, a marked decrease was observed, which was more intense in H cheeses, reaching percentages of 6% at 90 days (data not shown). The minor incidence of acetoin in the volatile profile during ripening of cheeses was offset by the relevant role of diacetyl + 2-2-pentanone and 2-heptanone, whose percentages increased sharply in aged cheeses (data not shown).

Diacetyl and its reduction product, acetoin, derive from lactose and citrate metabolism whereas 2-pentanone is related to  $\beta$ -oxidation of free fatty acids (McSweeney & Sousa, 2000). Higher levels of diacetyl + 2-2-pentanone were found in H cheeses at 90 days, while no statistical

differences were detected earlier. The opposite behavior was observed for acetoin level which was higher in C cheeses and decreased in all samples during ripening.

Other ketones derived from free fatty acids catabolism are 2hexanone, 2-heptanone and 2-nonanone (Collins et al., 2003). They reached similar levels in both types of un-ripened cheeses but in the course of ripening a higher content in H than C cheeses was always recorded (p < 0.05). In agreement with our results, Cao, Fonseca, Schoenfuss, and Rankin (2014) reported higher concentrations of methylketones in blue cheeses made from homogenized milk at 7 MPa compared to the unhomogenized samples. On the other hand, the trend observed for each compound during ripening was different depending on the trial and type of cheese. As for propanone and butanone, they are normal constituents of raw milk and are thought to be derived from cow feed rather than from fatty acids metabolism (Toso, Procida, & Stefanon, 2002). According with the hypothesis about their origin, our results showed that the homogenization process does not promote their production. Moreover, the changes throughout ripening were erratic in all cheeses.

#### 3.4.5. Esters

Esters are common products derived of fatty acid catabolism (Collins et al., 2003). They were found in low proportions in all cheeses, ranging from 4% to 9%. Regardless of type of cheese, a slight increase in the percentages of esters was observed throughout ripening. This result was



Fig. 4. Percentage of chemical groups of compounds in cheeses made with homogenized milk (H) and unhomogenized milk (C) at each sampling time (3, 45 and 90 days). Values are average for two independent cheese making trials of each type of cheese, analyzed in duplicate.

expected, as esterification is one of the dominant chemical events during cheese ripening (Caporaso, Armento, & Sacchi, 2015). Within this group, mostly ethyl esters (ethyl acetate, ethyl butanoate, ethyl hexanoate and ethyl octanoate) were identified. Esters have a relevant role in the sensory characteristics of grana-type cheeses, providing fruity notes (Liu, Holland, & Crow, 2004). Ethyl acetate and ethyl butanoate showed higher levels (p < 0.05) in C than H cheeses for all ripening times, and ethyl hexanoate at 45 and 90 days in cheeses from trial A. Conversely, no differences in ethyl octanoate between cheeses from both trials were found. As for evolution during ripening, the three esters showed a general trend to increase. In cheeses made from homogenized milk a significantly increase was evidenced from 45 to 90 days whereas in C cheeses a progressive increase was detected throughout ripening (p < 0.05).

The formation of esters in cheeses is still largely unknown, and the reactions that could be involved are esterification and alcoholysis. Ethyl esters biosynthesis in cheeses is likely due to a combined effect of factors such as the presence of substrates (acids and alcohols), enzymes, and environment conditions. The role of alcohols, mainly ethanol, on ester biosynthesis in cheeses has been extensively investigated (Liu et al., 2004; Richoux, Maillard, Kerjean, Lortal, & Thierry, 2008). Data on the volatiles in different types of cheeses suggest that alcohol availability is the limiting factor of ester biosynthesis. In our study, the highest levels of the major ethyl esters (ethyl acetate, ethyl butanoate and ethyl hexanoate) observed in C cheeses could be attributed, at least partially, to a higher availability of ethanol in these cheeses as was above mentioned.

#### 4. Conclusions

The pre-treatment of cheese milk: cream homogenization followed by an incubation step, showed to have influence on lipolysis and production of volatile compounds in hard cooked cheeses. Indeed, fat accessibility seems to be a limiting factor in lipolysis, as an acceleration of this biochemical event was clearly evidenced in FFA production showing an early lipolysis. Some compounds derived from fatty acid catabolism such as hexanal, heptanal and nonanal and methylketones from  $C_5$  to  $C_9$  were preferentially formed in homogenized cheeses at different ripening times. Homogenization did not cause detrimental effects on cheese composition and proteolysis index.

Based on these results, the innovative treatment of milk low homogenization/incubation applied to hard cooked cheese technology has proved to be effective in lipolysis acceleration and FFA catabolism. It could be easily scaled up; however, further studies should be carried out in order to evaluate sensory and textural attributes of the products.

#### Acknowledgements

The authors acknowledge the support provided by Universidad Nacional del Litoral (CAI+D N° 50120110100082LI) and CONICET for María Ayelén Vélez fellowship. The authors gratefully thank Milkaut S.A. for milk supply.

#### References

- Addis, M., Pirisi, A., Di Salvo, R., Podda, F., & Piredda, G. (2005). The influence of the enzymatic composition of lamb rennet paste on some properties of experimentally produced PDO Fiore Sardo cheese. *International Dairy Journal*, 15(12), 1271–1278.
- ANMAT (2014). Administración Nacional de medicamentos, alimentos y tecnología médica. Código Alimentario Argentino. Cap. VIII: Alimentos lácteos. (http://www.anmat.gov. ar/alimentos/normativas\_alimentos\_caa.asp./Accessed 16.09.16).
- Ardö, Y. (2011). Blue mold cheese. In H. Roginski, J. W. Fiquay, & P. F. Fox (Eds.), Encyclopedia of dairy science (pp. 767–773). London, UK: Elsevier Academic Press.
- Barbieri, G., Bolzoni, L., Careri, M., Mangia, A., Parolari, G., Spagnoli, S., & Virgili, R. (1994). Study of the volatile fraction of Parmesan cheese. *Journal of Agricultural and Food Chemistry*, 42(5), 1170–1176.
- Battistotti, B., & Corradini, C. (1993). Italian cheese. Cheese: chemistry, physics and microbiologyVol. 2. (pp. 221–243), 221–243.
- Bradley, R. L., Arnold, E., Barbano, D. M., Semerad, R. G., Smith, D. E., & Vines, B. K. (1992). Chemical and physical methods. Standard methods for the examination of dairy products, 433–531.
- Brito, C., Manríquez, A. X., Molina, C. L. H., & Pinto, C. M. (2003). Maturation study of low fat Chanco cheese made with homogenized milk. *Archivos Latinoamericanos de Nutricion*, 53(3), 299–305.
- Candioti, M. C., Hynes, E., Quiberoni, A., Palma, S. B., Sabbag, N., & Zalazar, C. A. (2002). Reggianito Argentino cheese: Influence of *Lactobacillus helveticus* strains isolated from natural whey cultures on cheese making and ripening processes. *International Dairy Journal*, 12(11), 923–931.
- Cao, M., Fonseca, L. M., Schoenfuss, T. C., & Rankin, S. A. (2014). Homogenization and lipase treatment of milk and resulting methyl ketone generation in blue cheese. *Journal of Agricultural and Food Chemistry*, 62(25), 5726–5733.
- Caporaso, N., Armento, V., & Sacchi, R. (2015). Volatile profile of Conciato Romano cheese, a traditional Italian cheese, during ripening. *European Journal of Lipid Science and Technology*, 117(9), 1422–1431.
- Ceruti, R. J., Zorrilla, S. E., & Sihufe, G. A. (2016). Volatile profile evolution of Reggianito cheese during ripening under different temperature–time combinations. *European Food Research and Technology*, 242(8), 1369–1378.
- Collins, Y. F., McSweeney, P. L. H., & Wilkinson, M. G. (2003). Lipolysis and free fatty acid catabolism in cheese: A review of current knowledge. *International Dairy Journal*, 13(11), 841–866.
- Curioni, P. M. G., & Bosset, J. O. (2002). Key odorants in various cheese types as determined by gas chromatography-olfactometry. *International Dairy Journal*, 12(12), 959–984.
- Deegan, K. C., & McSweeney, P. L. H. (2013). Effects of low-pressure homogenisation pretreatment of cheesemilk on the ripening of cheddar cheese. *Dairy Science and Technology*, 93(6), 641–655.

Deegan, K. C., Heikintalo, N., Ritvanen, T., Putkonen, T., Rekonen, J., McSweeney, P. L. H., & Tuorila, H. (2013). Effects of low-pressure homogenisation on the sensory and chemical properties of Emmental cheese. *Innovative Food Science and Emerging Technologies*, 19, 104–114.

- Evers, J. M. (2004). The milkfat globule membrane Compositional and structural changes post secretion by the mammary secretory cell. *International Dairy Journal*, 14(8), 661–674.
- Gobetti, M., & DiCagno, R. (2003). Hard italian cheeses. In H. Roginsky, J. W. Fuquay, & P. F. Fox (Eds.), *Encyclopedia of dairy science. Vol. 1*. (pp. 378–385). Cornwall, UK: Academic Press.
- Hair, J. F., Anderson, R. E., Tatham, R. L., & Black, W. C. (1999). Análisis multivariante. Madrid: Prentice-Hall.
- Holland, R., Liu, S. Q., Crow, V. L., Delabre, M. L., Lubbers, M., Bennett, M., & Norris, G. (2005). Esterases of lactic acid bacteria and cheese flavour: Milk fat hydrolysis, alcoholysis and esterification. *International Dairy Journal*, 15(6–9), 711–718.
- ISO (2004). Cheese and processed cheese—determination of the total solids content. (reference method). ISO 5534-IDF 4. International organization for standardization.
- ISO (2008a). Cheese-determination of fat content-Van Gulik method. ISO 3433- IDF 222. International organization for standardization.
- ISO (2008b). Milk and milk products—guidance on sampling. ISO 707-IDF 50. International organization for standardization.
- ISO (2011). Milk and milk products—determination of nitrogen content—part 1: Kjeldahl principle and crude protein calculation. ISO 8968-IDF 20. International Organization for Standardization.
- Johnson, M. E. (2011). Preparation of cheese milk. In H. F. Roginski, & P. Fox (Eds.), Encyclopedia of dairy science (pp. 544–551). London, UK: Elsevier Academic Press.
- Kelly, A. L., Huppertz, T., & Sheehan, J. J. (2008). Pre-treatment of cheese milk: Principles and developments. [conference paper]. *Dairy Science and Technology*, 88(4–5), 549–572.
- Kondyli, E., Katsiari, M. C., Masouras, T., & Voutsinas, L. P. (2002). Free fatty acids and volatile compounds of low-fat feta-type cheese made with a commercial adjunct culture. *Food Chemistry*, 79(2), 199–205.
- Liu, S. Q., Holland, R., & Crow, V. L. (2004). Esters and their biosynthesis in fermented dairy products: A review. *International Dairy Journal*, 14(11), 923–945.
- MacGibbon, A., & Taylor, M. (2006). Composition and structure of bovine milk lipids. In P. F. Fox, & P. McSweeney (Eds.), Advanced dairy chemistry. Vol. 2. (pp. 1–35). New York, USA: Springler.
- Madadlou, A., Mousavi, M. E., Khosrowshahi asl, A., Emam-Djome, Z., & Zargaran, M. (2007). Effect of cream homogenization on textural characteristics of low-fat Iranian white cheese. *International Dairy Journal*, 17(5), 547–554.
- Malacarne, M., Summer, A., Franceschi, P., Formaggioni, P., Pecorari, M., Panari, G., & Mariani, P. (2009). Free fatty acid profile of Parmigiano-Reggiano cheese throughout ripening: Comparison between the inner and outer regions of the wheel. *International Dairy Journal*, 19(10), 637–641.
- McSweeney, P. L. H., & Sousa, M. J. (2000). Biochemical pathways for the production of flavour compounds in cheeses during ripening: A review. *Le Lait*, 80(3), 293–324.
- Metzger, L. E., & Mistry, V. V. (1994). A new approach using homogenization of cream in the manufacture of reduced fat cheddar cheese. 1. Manufacture, composition, and yield. *Journal of Dairy Science*, 77(12), 3506–3515.
- Michalski, M. C., Camier, B., Briard, V., Leconte, N., Gassi, J. Y., Goudédranche, H., & Fauquant, J. (2004). The size of native milk fat globules affects physico-chemical and functional properties of Emmental cheese. *Le Lait*, 84(4), 343–358.
- Nair, M. G., Mistry, V. V., & Oommen, B. S. (2000). Yield and functionality of cheddar cheese as influenced by homogenization of cream. *International Dairy Journal*, 10(9), 647–657.

- Olivecrona, T., Vilaro, S., & G. O. (2003). Lipases in milk. In P. F. Fox, & P. L. H. M (Eds.), Advanced in dairy chemistry. Vol. 1. (pp. 473–494). New York: Kluwer/Plenum.
- O'Mahony, J. A., Sheehan, E. M., Delahunty, C. M., & McSweeney, P. L. H. (2006). Lipolysis and sensory characteristics of cheddar cheeses ripened using different temperaturetime treatments. *Le Lait*, 86(1), 59–72.
- Ott, A., Germond, J. E., & Chaintreau, A. (2000). Origin of acetaldehyde during milk fermentation using 13C-labeled precursors. *Journal of Agricultural and Food Chemistry*, 48(5), 1512–1517.
- Perotti, M. C., Bernal, S. M., Meinardi, C. A., & Zalazar, C. A. (2005). Free fatty acid profiles of Reggianito Argentino cheese produced with different starters. *International Dairy Journal*, 15(11), 1150–1155.
- Richoux, R., Maillard, M. B., Kerjean, J. R., Lortal, S., & Thierry, A. (2008). Enhancement of ethyl ester and flavour formation in Swiss cheese by ethanol addition. *International Dairy Journal*, 18(12), 1140–1145.
- Rowney, M. K., Hickey, M. W., Roupas, P., & Everett, D. W. (2003). The effect of homogenization and milk fat fractions on the functionality of mozzarella cheese. *Journal of Dairy Science*, 86(3), 712–718.
- Svensson, I., Hernández, I., Virto, M., & De Renobales, M. (2006). Determination of lipase activity in cheese using trivalerin as substrate. *International Dairy Journal*, 16(5), 423–430.
- Toso, B., Procida, G., & Stefanon, B. (2002). Determination of volatile compounds in cows' milk using headspace GC-MS. *Journal of Dairy Research*, 69, 569–577.
- Vélez, M. A., Perotti, M. C., Wolf, I. V., Hynes, E. R., & Zalazar, C. A. (2010). Influence of milk pretreatment on production of free fatty acids and volatile compounds in hard cheeses: Heat treatment and mechanical agitation. *Journal of Dairy Science*, 93(10), 4545–4554.
- Vélez, M. A., Perotti, M. C., Rebechi, S. R., Meinardi, C. A., Hynes, E. R., & Zalazar, C. A. (2011). Effect of mechanical treatments applied to milk fat on fat retention and lipolysis in minicurds. *International Journal of Dairy Technology*, 64(2), 227–231.
- Voigt, D. D., Chevalier, F., Qian, M. C., & Kelly, A. L. (2010). Effect of high-pressure treatment on microbiology, proteolysis, lipolysis and levels of flavour compounds in mature blue-veined cheese. *Innovative Food Science and Emerging Technologies*, 11(1), 68–77.
- Walstra, P., Geurts, T. J., Noomen, A., Jellema, A., & Boekel, M. A. V. (1999a). Colloidal particles of milk. In O. R. Fennema, M. Karel, G. W. Sanderson, S. R. Tannenbaum, P. Walstra, & J. R. Whitaker (Eds.), *Dairy technology: Principles of milk properties and pro*cesses (pp. 107–149). New York: Marcel Dekker Inc.
- Walstra, P., Geurts, T. J., Noomen, A., Jellema, A., & van Boekel, M. A. (1999b). Cheese ripening and Properties. In O. R. Fennema, M. Karel, G. W. Sanderson, S. R. Tannenbaum, P. Walstra, & J. R. Whitaker (Eds.), *Dairy technology. Principles of milk properties and processes* (pp. 601–636). New York, USA: Marcel Dekker Inc.
- Wilkinson, M. G. (2007). Lipolysis and cheese flavour development. In B. C. W (Ed.), Improving the flavour of cheese (pp. 102–120). Cambridge, England: WoodHead Publishing Limited.
- Wolf, I. V., Perotti, M. C., Bernal, S. M., & Zalazar, C. A. (2010). Study of the chemical composition, proteolysis, lipolysis and volatile compounds profile of commercial Reggianito Argentino cheese: Characterization of Reggianito Argentino cheese. *Food Research International*, 43(4), 1204–1211.
- Wolf, I. V., Vénica, C. I., & Perotti, M. C. (2015). Effect of reduction of lactose in yogurts by addition of β-galactosidase enzyme on volatile compound profile and quality parameters. International Journal of Food Science and Technology, 50(5), 1076–1082.
- Zamora, A., Ferragut, V., Juan, B., Guamis, B., & Trujillo, A. J. (2011). Effect of ultra-high pressure homogenisation of milk on the texture and water-typology of a starterfree fresh cheese. *Innovative Food Science and Emerging Technologies*, 12(4), 484–490.