

# Characterization of Hg-phytochelatins complexes in vines (*Vitis vinifera* cv *Malbec*) as defense mechanism against metal stress

Adrian A. Spisso · Soledad Cerutti ·  
Fernanda Silva · Pablo H. Pacheco ·  
Luis D. Martinez

Received: 31 October 2013 / Accepted: 26 March 2014  
© Springer Science+Business Media New York 2014

**Abstract** An approach to understand vines (*Vitis vinifera*) defense mechanism against heavy metal stress by isolation and determination of Hg-phytochelatins (PCs) complexes was performed. PCs are important molecules involved in the control of metal concentration in plants. PCs complex toxic metals through –SH groups and stores them inside cells vacuole avoiding any toxic effect of free metals in the cytosol. The Hg-PCs identification was achieved by determination of Hg and S as hetero-tagged atoms. A method involving two-dimensional chromatographic analysis coupled to atomic spectrometry and confirmation by tandem mass spectrometry is proposed. An approach involving size exclusion chromatography coupled to inductively coupled plasma mass spectrometry on roots, stems, and leaves extracts describing Hg distribution according to molecular weight and sulfur associations is proposed for the first time.

Medium–low molecular weight Hg–S associations of 29–100 kDa were found, suggesting PCs presence. A second approach employing reversed-phase chromatography coupled to atomic fluorescence spectrometry analysis allowed the determination of Hg-PCs complexes within the mentioned fractions. Chromatograms showed Hg-PC<sub>2</sub>, Hg-PC<sub>3</sub> and Hg-PC<sub>4</sub> presence only in roots. Hg-PCs presence in roots was confirmed by ESI–MS/MS analysis.

**Keywords** *Vitis vinifera* · Mercury · Phytochelatins · Metal stress

## Introduction

Mercury (Hg) is one of the most toxic elements and has adverse impact on the human health and the environment. Mercury pollution is a ubiquitous problem resulting both from natural events and anthropogenic activities (Finkelman et al. 2002; Finkelman 2004; Wilcox et al. 2012; Dai et al. 2012). Natural sources and transport mechanisms include volcanic emissions, wind borne dust, geysers, thermal fluids, and sea-spray (Fergusson 1990; Marczenko and Lobinski 1996). Mercury removal from anthropogenic sources, such as chlorine plants, pesticides and paints factories, becomes very important because from these sources it can contaminate water reservoirs and irrigation networks, and thus it may be retained by plants and different agents at various levels of the food chain.

---

A. A. Spisso · S. Cerutti · P. H. Pacheco (✉) ·  
L. D. Martinez (✉)  
Instituto de Química de San Luis (INQUISAL–  
CONICET), Chacabuco y Pedernera, CP 5700 San Luis,  
Argentina  
e-mail: ppacheco@unsl.edu.ar

L. D. Martinez  
e-mail: ldm@unsl.edu.ar

F. Silva  
Instituto de Biología Agrícola de Mendoza (IBAM–  
CONICET), Alte.Brown 500, Chacras de Coria, CP 5505  
Mendoza, Argentina

Viticulture represents an important agricultural practice in many countries (Komárek et al. 2010). The quality of irrigation water is an important variable defining the transportation of metals and other elements necessary for vine growth from soil toward the plant (Fabani et al. 2009). Irrigation with contaminated waters can seriously increase the probability of toxic metals uptake into vine plants. Irrigation waters can be contaminated from different sources like industrial effluents (Saleem-Saif et al. 2005), wastewaters (Wu and Cao 2010), and geological processes (Domagalski et al. 2004).

The metal uptake capacity of *Vitis vinifera* has been reported elsewhere (Leita et al. 1998; Todić et al. 2006; Chopin et al. 2008). Elevated heavy metal concentrations were found in xylem saps of vines showing a relatively high mobility within the plants (Leita et al. 1998). Chopin et al. (2008) studies showed that the differences between elements intake resulted from vegetation uptake strategies and soil partitioning. In addition, it has been demonstrated that Hg content in vines cultivated in Hg contaminated soils is elevated compared with vines grown in soils with a normal Hg concentration (Todić et al. 2006; Spisso et al. 2013).

Plants respond to metal toxicity by initiating a wide range of cellular defense mechanisms (Thapa et al. 2012). These include immobilization, exclusion, and compartmentalization of metals; along with phytochelatin (PCs) synthesis (Sanità di Toppi and Gabrielli 1999). PCs are important molecules involved in the control of metal concentration in plants, fungus, and algae; being their main structure ( $\gamma$ -Glu-Cys) $_n$ -Gly with  $n = 2$ –11 (Kondo et al. 1985; Grill et al. 1987; Leopold and Gunther 1997; Gekeler et al. 1988). These peptides are produced in plants by  $\gamma$ -glutamyl-cysteine-dipeptidyl transpeptidase (phytochelatin synthase) as a consequence of exposure of plants to heavy metals, with glutathione (GSH) as a substrate. Several metal ions activate PCs synthesis, but Cd, Cu, Hg, and Zn are the most commonly studied because they induce a more intensive biological response in cells (Kondo et al. 1985). The role of PCs is toxic metals' coordination through –SH groups and metal–ion complex storing inside the cell's vacuole, avoiding any toxic effect of free metals in the cytosol (Zenk 1996). PCs accumulation has also been studied in different plants exposed to Hg contamination under controlled conditions (Sobrinho-Plata et al. 2009; Sobrinho-Plata et al. 2013). However, only a few

studies have been devoted to specific Hg-PCs complexes determination in plants (Chen et al. 2009; Carrasco-Gil et al. 2011).

In order to determine PCs, chromatographic techniques coupled to several detectors have been used. Reversed-phase chromatography (RPC) with UV detection was applied with pre- and post-column derivatization strategies (Habeed 1972; de Knecht et al. 1994; Hirata et al. 2001; Chen et al. 2009). Fluorescence detection was also employed by derivatization with monobromobimane (mBBBr), which forms fluorescent compounds with the thiol groups (mBBBr-PCs) (Newton et al. 1981; Ahner et al. 1994; Rijstenbil and Wijnholds 1996; Minocha et al. 2008). RPC coupled to amperometric detection has been also used (Dago et al. 2011). In addition, metal-PCs complexes can be determined by combining capillary electrophoresis or liquid chromatography with atomic and molecular detectors. Moreover, mass spectrometry techniques have also been used to confirm the presence of PCs (Vacchina et al. 1999; Chen et al. 2009; Wood and Feldmann 2012).

This research is an attempt to increase the knowledge of Hg metabolism in vines. For the first time, it is described the determination of PCs complexed with Hg in vines. To this end, vine plants were irrigated with Hg contaminated waters under greenhouse conditions. After Hg-PCs extraction, a two-dimensional separation was applied: a first separation employing size exclusion chromatography (SEC) coupled to inductively coupled plasma (ICP MS) followed by a second Hg-PCs complexes separation and determination by RPC coupled to atomic fluorescence spectrometry (AFS). Finally, Hg-PCs complexes presence was confirmed by tandem mass spectrometry.

## Materials and methods

### Chemicals

Ultrapure water ( $18 \text{ M}\Omega \text{ cm}^{-1}$ ) was obtained from EASY pure (RF Barnstedt, IA, USA). Bovine serum albumin (66 kDa), alcohol dehydrogenase (150 kDa),  $\beta$ -amylase (200 kDa), thyroglobulin (669 kDa) and apoferritin (443 kDa) from Sigma (St. Louis, MO, USA) were employed as standards for SEC. Phytochelatin standards (PC<sub>2</sub>, PC<sub>3</sub> and PC<sub>4</sub>) were obtained from Anaspec (USA). Nitric acid 65 % and trifluoroacetic acid (TFA)

were from Sigma-Aldrich. Mercury nitrate used in vine watering, sodium hydroxide, sodium monohydrogen phosphate, potassium dihydrogen phosphate and hydrogen peroxide were from Merck. Acetonitrile (ACN) and sodium dodecyl sulfate (SDS) were from Fisher Scientifics. Ethylenediaminetetraacetic acid (EDTA) (99 %) and tris (hydroxymethyl) aminomethane (Tris; 99, 85 %) were from Across organics. Potassium persulfate ( $K_2S_2O_8$ ) (99 %) was obtained from Fluka AG, (Switzerland). Sodium borohydride ( $NaBH_4$ ) was obtained from Riedel-de Haën. Mercury standards were from Perkin-Elmer. Ammonium acetate was from Biopack.

### Instrumentation

Microwave digestion was performed with a Milestone Start D microwave system (Sorisole, Italy), and with Milestone hermetically sealed 1 cm wall thickness polytetrafluoroethylene reactors (100 mL internal volume).

Mercury fluorescence measurements were carried out with an AFS, AI 3300, Aurora Instruments (Vancouver, BC, Canada). The apparatus was equipped with a two-channel peristaltic pump for the continuous fluorescence measurements. Hg hollow cathode lamp from Aurora Instruments (Vancouver, BC, Canada) was employed as Hg fluorescence excitation source. Samples and reagents involved in cold vapor generation (CV) were delivered by a Minipulse 3 peristaltic pump Gilson (Villiers-Le-Bell, France). Ultraviolet (UV) decomposition was achieved with a 400 W Hg vapor lamp (15 W G15T8 UV-C LONG LIFE high pressure Hg, PHILIPS) that ignited with a suitable starter and chock and surrounded by a 1.5 m PTFE tubing.

For extraction purposes, samples were dried in an Eppendorf Speed vac concentrator plus. A thermostatic ultrasonic washer (Cleanson) and Boeco Centrifuges U-320 R were also used.

ICP MS employed was a Perkin-Elmer Sciex, ELAN DRC-e (Thornhill, Canada). The argon (Ar) gas with a minimum purity of 99.996 % was supplied by Linde (Córdoba, Argentina). A high performance Teflon Nebulizer model PFA-ST, was coupled to a quartz cyclonic spray chamber with internal baffle and drain line cooled with the  $PC_3$  system from ESI (Omaha, NE, USA).

Separations were performed with a Series 200, Perkin-Elmer (Thornhill, Canada) binary pump. The columns used were: for SEC, TSK gel G3000SW

( $7.5 \times 300 \text{ mm} \times 10 \mu\text{M}$ ), Tosoh Biosep; and for RPC, Zorbax SB-Aq C18-RP ( $1.6 \times 150 \text{ mm}$ ,  $5 \mu\text{M}$ ) Agilent Technologies.

For confirmation of Hg-PCs structures, the mass spectrometry analyses were performed on a Quattro Premier<sup>TM</sup> XE Micromass MS Technologies triple quadrupole mass spectrometer with a ZSpray<sup>TM</sup> Electrospray ionization source (Waters, Milford, USA).

### Plants cultivation and supplementation

Plants were obtained from Estación Experimental Agropecuaria Mendoza, Instituto Nacional de Tecnología Agropecuaria (INTA). The experiment was carried out at Facultad de Ciencias Agrarias, Universidad Nacional de Cuyo, Mendoza, Argentina ( $33^\circ 0'S$ ,  $68^\circ 52'W$ ) at an altitude of 940 m.a.s.l. One-year-old plants of a selected clone of *V. vinifera* L. cv. *Malbec* were planted in 1.0 L plastic pots filled with 450 g of grape compost. Grape compost consists of 3 parts of pomace, 3 parts of loam and 2 parts of perlite (pH: 7.2; conductivity:  $18.3 \text{ m}\Omega \text{ cm}^{-1}$ ; organic matter 9 %). They were grown in a greenhouse at temperatures ranging from 23 to 27 °C (night and day). In order to reproduce a situation where vines are irrigated with Hg contaminated water, the short term supplementation procedure reported by Afton et al. (2009) was adapted with modifications. A total of six plants were divided into two groups: control and samples. Control plants were watered only with 300 mL tap water, while sample plants were watered daily for three days with 300 mL of a 100 mg/L Hg solution in the form of mercury nitrate. The health of these plants was visually indifferent to Hg supplementation. After one week, samples of leaves, stems and roots were collected. They were immersed in an ultrasonic bath for a complete soil removal, washed with ultrapure water, and lyophilized. These plant organs were stored at  $-5^\circ \text{C}$  to prevent deterioration and enzyme activity that could cause species interconversion. The samples were used within few hours of storing.

### Total Hg determinations

Determinations of total Hg in plant organs were performed by ICP MS after microwave assisted digestion. Thus 0.5 g of leaves were digested with 7 mL nitric acid ( $HNO_3$ ), and 1 mL hydrogen

peroxide ( $\text{H}_2\text{O}_2$ ); 0.3 g of stems with 8 mL of  $\text{HNO}_3$  and 2 mL of  $\text{H}_2\text{O}_2$  and 0.3 g of roots with 9 mL of  $\text{HNO}_3$  and 1 mL of hydrofluoric acid. The digestion method for roots and stems consisted of two steps: first, one ramp of 20 min to 180 °C; and second, hold step of 10 min at a temperature of 180 °C. Maximum power (1,000 watts) was applied in both steps. For leaves digestion, a two-step method of ramp and hold was also used, each one of 10 min at a temperature of 200 °C and a power of 1,000 watts. Previous to digestion, roots were washed with an EDTA solution to remove the adsorbed mercury.

#### Phytochelatins extraction and determination

Prior chromatographic separations, extractions were carried out from vine samples (leaves, stems, roots) as follows: an extraction stage involving liquid nitrogen and a mortar followed by the addition of 2 mL of a 2 % (w/v) SDS–30 mM Tris-HCl buffer solution (pH:7) was performed. The extraction was completed after 2 h sonication and centrifugation at 5,000 rpm at 4 °C. Supernatants were collected and filtered through 0.22  $\mu\text{m}$  filter (Osmonics®) prior injection (Mounicou et al. 2004). In order to evaluate the extraction efficiency, 15 and 3.5  $\mu\text{M}$  of Hg and each PCs; respectively, were added to the samples, extracted under the above mentioned procedure and analyzed by LC–AFS. Quantitative results were achieved.

Determination of Hg-PCs complexes was performed by a two-dimensional chromatographic procedure. First, SEC was performed coupling the chromatographer to ICP MS. A 50 mM buffer ammonium acetate was employed as eluent being adequate for coupling with ICP MS, since its volatility do not generate deposits onto the ICP cones. Bovine serum albumin (66 kDa), alcohol dehydrogenase (150 kDa),  $\beta$ -amilase (200 kDa), thyroglobulin (669 kDa) and apoferritin (443 kDa) were employed for calibration. In order to determine the Hg associated to different protein or peptide fractions, 200  $\mu\text{L}$  of each sample extracts (roots, stems and leaves) were injected and Hg and Sulfur (S) signals were monitored simultaneously. Sulfur determination was performed by means of its oxide detection, which is formed using oxygen as auxiliary gas. Thus sulfur acquires a mass/charge ratio of 48, being ICP MS adjusted for the simultaneous determination of Hg and  $\text{SO}^+$  (Spisso et al. 2013). The working conditions were: dwell time; 15 ms; sweeps/reading, 40; carrier gas used

was argon (Ar); 0.02 mL/min auxiliary gas used was oxygen ( $\text{O}_2$ ); the mobile phase used was 50 mM ammonium acetate -with 5 % of methanol (v/v); at a flow rate of 0.7 mL/min in isocratic elution mode; sample injection was 200  $\mu\text{L}$ . After the SEC approach, the fractions corresponding to Hg-PCs molecular weight were collected. These fractions were dried in Speed Vac and diluted to 1 mL with ultrapure water. The Hg-PCs complexes stability after the drying process was evaluated by standard addition, as described previously, with quantitative results. These fractions were then analyzed by RPC in a second dimension analysis. To this end, 100  $\mu\text{L}$  of the collected fractions were injected to LC coupled to AFS. A gradient elution mode was employed, the mobile phases were A: 0.1 % (v/v) TFA in  $\text{H}_2\text{O}$  and B: ACN (0–50 % B 28 min; hold 50 % B 2 min and return to 100 % A in 5 min); flow rate was 1 mL/min. The reagents for CV generation were 0.1 % (w/v)  $\text{K}_2\text{S}_2\text{O}_8$  in 30 % (v/v)  $\text{HNO}_3$  and 0.5 % (w/v)  $\text{NaBH}_4$  in 0.5 % (w/v) NaOH; flow rate were 1 and 2 mL/min respectively AFS conditions were: 253.7 nm Hg hollow cathode lamp wavelength; 300 V photo multiplier power (PMT); 35 mA primary current; 100 °C oven and 200 mL/min carrier gas (Ar).

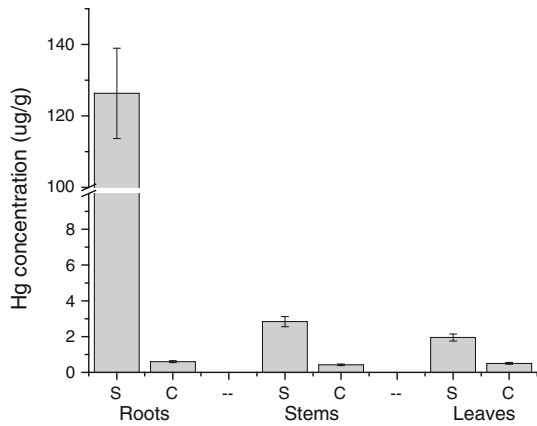
The choice of an AFS detector for reverse phase chromatography was mainly due to the fact that ICP MS does not support the introduction of large amounts of organic solvent, which is necessary for Hg-PCs complexes optimal separation.

In order to confirm the studied Hg-PCs complexes presence after RPC, mass analyses were performed on a triple quadrupole mass spectrometer configured with an electrospray ionization source operated in a positive (ES+) mode at a desolvation temperature 350 °C, with  $\text{N}_2$  as nebulizer. The source was kept at a temperature of 150 °C. The capillary voltage was maintained at 3.0 kV and the extractor voltage was set at 3.0 kV. Optimization and detection was performed in a full scan mode via direct infusion (syringe pump) into the mass spectrometer of the Hg-PCs complexes (1 mg/L standard solutions for optimization purposes).

## Results and discussion

### Total Hg determinations

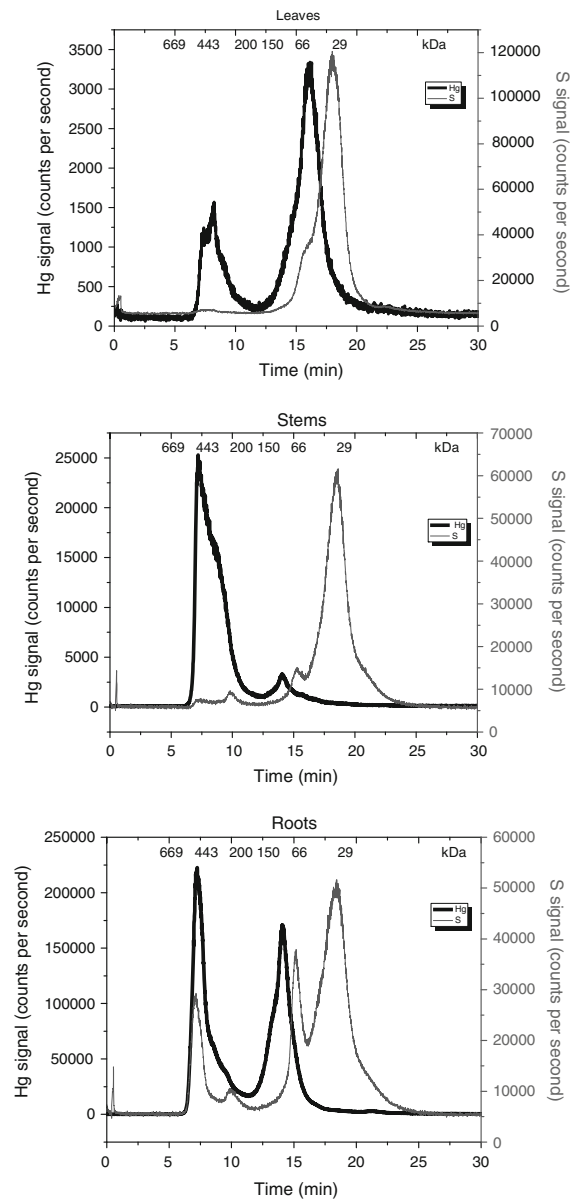
Total Hg determinations were performed in vine organs (roots, stems, and leaves) of control and



**Fig. 1** Total Hg concentration into the different vine compartments evaluated for 100 mg/L of Hg<sup>2+</sup> supplementation. Results are expressed as the mean ± standard deviation of 3 plants. S sample; C control

supplemented plants. In this work it was found that Hg concentrations in aerial organs were lower than roots in vines supplemented with Hg. However, these Hg levels are still higher than those found in control vines (0.42 ± 0.08 µg/g in stems, 0.5 ± 0.05 µg/g in leaves and 0.6 ± 0.05 µg/g in roots). It is frequent to find a certain degree of Hg accumulation in the leaves of control plants, as this metal is prone to volatilize and accumulate in above-ground organs in plants cultivated in closed controlled environments. Similar behaviour was observed in Arabidopsis plants grown in hydroponics in a growth chamber (Sobrino-Plata et al. 2014).

Herein obtained results are shown in Fig. 1. The maximum Hg concentrations were found in roots with a value of 126.31 ± 4.96 µg/g. In stems Hg values were of 2.84 ± 0.84 µg/g. Leaves showed Hg concentrations of 1.95 ± 0.2 µg/g. Results are expressed as the mean ± the standard deviation of three samples. From the results, it was found that Hg values were higher in roots compared with stems and leaves, coincident with Hg values reported in our previous work (Spisso et al. 2013). Mercury undergoes almost complete sequestration in roots of vine plants, which prevents Hg accumulation in the aerial organs (leaves and stems). This observation has been described in different plants, in which a greater Hg concentration in roots than in other organs was observed (Iglesia-Turiño et al. 2006; Rellán-Álvarez et al. 2006; Chen et al. 2009; Carrasco-Gil et al. 2011; Chen and Yang 2012).

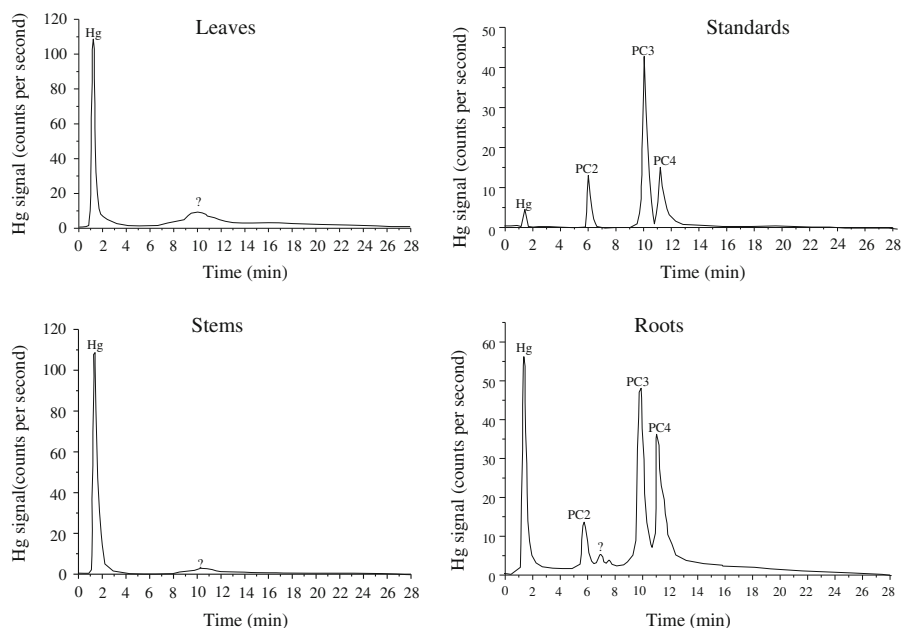


**Fig. 2** SEC–ICP MS chromatograms of plant compartments. Molecular weight markers can be observed in the upper side of graphics. Standards and samples injection: 200 µL

### Hg distribution and Hg associations with sulfur by SEC–ICP MS

The different Hg associations in plants were separated by SEC–ICP MS technique according to their molecular size. In this work, the column separation range used was wide: from 10 to 700 kDa. Hg and S signals were registered simultaneously by ICP MS. Sulfur was

**Fig. 3** Hg-PCs chromatograms by RPC-AFS analysis. Standards and samples injection: 100  $\mu$ L. Standards concentrations: 15 and 3.5  $\mu$ M for Hg and PCs respectively



monitored to evaluate possible Hg associations to proteins or peptides.

SEC chromatograms for roots, stems and leaves of vine plants are shown in Fig. 2. It can be observed that a Hg fraction appears in all vine organs in the range from 200 to 443 kDa. The estimated Hg concentrations in these fractions were 0.06 ; 0.984 and 6.133  $\mu$ g/mL in leaves, stems, and roots; respectively. Only in roots, a sulfur fraction appeared associated to this molecular size fraction. A second Hg fraction appeared in the 29–100 kDa range, coincident with the S fractions in all the studied plant organs, indicating possible Hg associations of medium–low molecular weight. The estimated mercury concentrations were 0.160 ; 0.115 and 6.084  $\mu$ g/mL in leaves, stems, and roots; respectively. These findings are in good agreement with separations reported in previous studies, which also described Hg distribution in two different molecular size fractions (Spisso et al. 2013).

After obtaining the SEC elution profile of Hg-S containing substances, medium–low molecular size fractions of the three vine organs were collected, as shown in Fig. 2. These fractions were chosen for collection considering parameters such as column separation capacity and molecular size that could be correspondent to Hg-PCs complexes. Fractions collected were from 12 to 19, 13 to 17, and from 11 to 17 min from leaves, stems, and roots chromatograms;

respectively. These fractions were freeze dried and then diluted with ultrapure water for further Hg-PCs complexes analysis.

#### Hg-PCs determination by RP-HPLC-AFS

Hg-PCs complexes determination in the different vine organs was performed using a C18 reverse phase column for separation and determination. To this end AFS was employed as detection system being Hg the hetero-tagged atom. The chromatographic conditions were specified above in the text. Separation was completed employing ACN and TFA as described in bibliography for PCs separation (Chen et al. 2009; Dago et al. 2011).

Figure 3 shows the chromatogram obtained from PCs standard analysis in root, stem, and leaves SEC collected fractions. Hg-PCs complexes were prepared in phosphate buffer, with a concentration of 15  $\mu$ M of Hg and 3.5  $\mu$ M of each PC. Two hours incubation at room temperature was also applied to ensure complex formation. This concentration relationship between Hg and PCs ensured optimal Hg-PCs formation, with minimum concentration of Hg<sup>2+</sup>, with a molar ratio of 2.14, 1.43 and 1.07 for PC<sub>2</sub>, PC<sub>3</sub>, and PC<sub>4</sub>; respectively. As observed in the chromatogram of Fig. 3, four peaks corresponding to Hg<sup>2+</sup> or unassociated Hg,



Hg-PC<sub>2</sub>, Hg-PC<sub>3</sub> and Hg-PC<sub>4</sub>, respectively, with baseline separation, were obtained.

Root chromatograms showed baseline separation between free Hg and different Hg-PCs complexes. PCs elution order was correspondent to those found in other published articles (de Knecht et al. 1994; Kawakami et al. 2006; Minocha et al. 2008). Elution times of Hg compounds were 77, 345, 593, and 660 s corresponding to free Hg<sup>2+</sup>, Hg-PC<sub>2</sub>, Hg-PC<sub>3</sub> and Hg-PC<sub>4</sub>; respectively. The estimated mercury concentration of each fraction was 0.914 µg/mL for Hg, 0.309 µg/mL for Hg-PC<sub>2</sub>, 1.272 µg/mL for Hg-PC<sub>3</sub>, and 1.11 µg/mL for Hg-PC<sub>4</sub>. Hg complexed with PCs corresponded to 1.43 % of the total Hg analyzed by SEC. Future studies will attempt to reveal other Hg associations since this time AFS was optimized for Hg-PCs determination. The above mentioned concentrations were calculated according to a calibration curve employing free Hg<sup>2+</sup> and Hg-PCs standards injected in the LC-UV-CV-AFS system and are referred to the volume employed for dilution after Speed Vac drying (1 mL).

The Hg-PCs chromatograms of leaves and stems presented similarities, showing both a peak at 76 s corresponding to free Hg<sup>2+</sup> and, also, a wide peak at 597 and 630 s in leaves and stems; respectively, with no correspondence with the investigated PCs.

From chromatograms analysis correspondent to the different studied vine organs, it was possible to conclude that Hg-PCs are present only in roots. This behavior can be explained considering that a higher Hg concentration favors a more effective Hg-PCs formation in roots. This behavior found in vines has also been observed in other plants where PCs were only identified in roots (Zeng et al. 2009).

#### Hg-PCs confirmation by ESI-MS/MS

After RPC-AFS analysis, fractions were collected as follows: one correspondent to the complex Hg-PC<sub>2</sub>, collected at 250–450 s and a second fraction corresponding to Hg-PC<sub>3</sub> and Hg-PC<sub>4</sub> complexes collected at 500–900 s. These last two complexes were collected in the same fraction since their retention times were close. Collection times were broad than elution peaks to assure complexes gathering. Both fractions were dried down and then reconstituted separately with water for ESI-MS/MS analysis.

**Table 1** Complex Hg-PCs

Hg-PC	Complex found	m/z
Hg-PC <sub>2</sub>	[HgPC <sub>2</sub> + H] <sup>+</sup>	740
	[HgPC <sub>2</sub> - H <sub>2</sub> O + H] <sup>+</sup>	722
Hg-PC <sub>3</sub>	[HgPC <sub>3</sub> + H] <sup>+</sup>	972
	[HgPC <sub>3</sub> - H <sub>2</sub> O + H] <sup>+</sup>	954
Hg-PC <sub>4</sub>	[HgPC <sub>4</sub> + H] <sup>+</sup>	1202
	[HgPC <sub>4</sub> - H <sub>2</sub> O + H] <sup>+</sup>	1184

The presence of the Hg-PCs complexes in roots was confirmed by determination of two ions for each studied complex, which are shown in Table 1. These Hg-PCs complexes are: m/z: 740 (Hg-PC<sub>2</sub>), 972 (Hg-PC<sub>3</sub>), 1202 (Hg-PC<sub>4</sub>), have been also reported by Chen et al. (2009). In this way Hg-PCs formation in vines under Hg stress was correspondent with Sobrino-Plata et al. (2013) research, where these complexes formation were determined in *Arabidopsis thaliana*.

#### Conclusion

Vine plants react to stress generated by heavy metals, producing different phytochelatins with metal complexing function. In this sense, the stress induced in vine plants through irrigation with Hg contaminated waters lead to the formation of Hg-PC<sub>2</sub>, Hg-PC<sub>3</sub> and Hg-PC<sub>4</sub> complexes.

As mentioned, higher Hg concentration levels in roots rather than in aerial organs (stems and leaves) were found. This could probably be explained by the way Hg mobilizes across the plant. This metal enters through the roots, where it is complexed by phytochelatins and thus cannot be mobilized to aerial organs. This sequestration process might be interesting considering it as an impediment for Hg to reach the edible vine organs.

A first approach by SEC-ICP MS analysis showed that Hg was distributed in plant organs in two different fractions, high and medium–low molecular weight; respectively, being this last one important since Hg-PCs compounds can be distributed within this fraction. In addition sulfur monitoring confirmed peptide presence in this medium–low molecular weight portion. A second strategy by RPC analysis showed a baseline separation of Hg-PC<sub>2</sub>, Hg-PC<sub>3</sub>, and Hg-PC<sub>4</sub> complexes in the medium–low molecular weight

fractions collected from SEC. Such complexes were only found in vine roots. The presence of these compounds was verified by ESI-MS/MS analysis, encompassing RPC results. Overall the proposed technique allowed a successful separation, isolation, and identification of Hg-PCs complexes.

Thus Hg-PCs complexes have been studied and determined in vine plants for the first time achieving a breakthrough in the elucidation of vine defense mechanism against heavy metals from irrigation water. This work acquires significance considering vines economic importance as grapes provider for wines production. Pollution of irrigation waters from different sources is a matter of great concern since affects the safety and quality of wines.

**Acknowledgments** This work was supported by Consejo Nacional de Investigaciones Científicas y Técnicas (CONICET), Agencia Nacional de Promoción Científica y Tecnológica (FONCYT) (PICTBID), Universidad Nacional de San Luis (Argentina), Universidad Nacional de Cuyo (Argentina), and Instituto Nacional de Técnicas Agropecuarias.

## References

- Afton SE, Catron B, Caruso JA (2009) Elucidating the selenium and arsenic metabolic pathways following exposure to the non-hyperaccumulating *Chlorophytum comosum*, spider plant. *J Exp Bot* 60:1289–1297
- Ahner BA, Price NM, Morel FMM (1994) Phytochelatin production by marine phytoplankton at low free metal ion concentrations: laboratory studies and field data from Massachusetts Bay. *Proc Natl Acad Sci USA* 91:8433–8436
- Carrasco-Gil S, Álvarez-Fernández A, Sobrino-Plata J, Millán R, Carpena-Ruiz RO, Leduc DL, Andrews JC, Abadía J, Hernández LE (2011) Complexation of Hg with phytochelatin is important for plant Hg tolerance. *Plant Cell Environ* 34:778–791
- Chen L, Yang L, Wang Q (2009) In vivo phytochelatin and Hg-phytochelatin complexes in Hg-stressed *Brassica chinensis* L. *Metallomics* 1:101–106
- Chen J, Yang ZM (2012) Mercury toxicity, molecular response and tolerance in higher plants. *Biometals* 25:847–857
- Chopin EIB, Marin B, Mkoungafoko R, Rigaux A, Hopgood MJ, Delannoy E, Cancès B, Laurain M (2008) Factors affecting distribution and mobility of trace elements (Cu, Pb, Zn) in a perennial grapevine (*Vitis vinifera* L.) in the Champagne region of France. *Environ Pollut* 156:1092–1098
- Dago A, González-García O, Ariño C, Díaz-Cruz JM, Esteban M (2011) Characterization of Hg(II) binding with different length phytochelatin using liquid chromatography and amperometric detection. *Anal Chim Acta* 695:51–57
- Dai S, Ren D, Chou CL, Finkelman RB, Seredin VV, Zhou Y (2012) Geochemistry of trace elements in Chinese coals: a review of abundances, genetic types, impacts on human health, and industrial utilization. *Int J Coal Geol* 94:3–21
- de Knecht JA, van Dillen M, Koevoets PLM, Schat H, Verkleij JAC, Ernst WHO (1994) Phytochelatin in cadmium-sensitive and cadmium-tolerant *Silene vulgaris*. Chain length distribution and sulfide incorporation. *Plant Physiol* 104:255–261
- Domagalski JL, Alpers CN, Slotton DG, Suchanek TH, Ayers SM (2004) Mercury and methylmercury concentrations and loads in the Cache Creek watershed, California. *Sci Total Environ* 327:215–237
- Fabani MP, Toro ME, Vázquez F, Díaz MP, Wunderlin DA (2009) Differential absorption of metals from soil to diverse vine varieties from the valley of tulum (Argentina): consequences to evaluate wine provenance. *J Agric Food Chem* 57:7409–7416
- Fergusson JE (1990) The heavy elements: chemistry environmental impact and health effects. Pergamon, Oxford
- Finkelman RB (2004) Potential health impacts of burning coal beds and waste banks. *Intern J Coal Geol* 59:19–24
- Finkelman RB, Orem W, Castranova V, Tatu CA, Belkin HE, Zheng B, Lerch HE, Maharaj SV, Bates AL (2002) Health impacts of coal and coal use: possible solutions. *Int J Coal Geol* 50:425–443
- Gekeler W, Grill E, Winnacker EL, Zenk MH (1988) Algae sequester heavy metals via synthesis of phytochelatin complexes. *Arch Microbiol* 150:197–202
- Grill E, Winnacker EL, Zenk MH (1987) Phytochelatin, a class of heavy-metal-binding peptides from plants, are functionally analogous to metallothioneins. *Proc Natl Acad Sci USA* 84:439–443
- Habeeb AF (1972) Reaction of protein sulfhydryl groups with Ellman's reagent. *Methods Enzymol* 25:457–464
- Hirata K, Tsujimoto Y, Namba T, Ohta T, Hirayanagi N, Miyasaka H, Zenk MH, Miyamoto K (2001) Strong induction of phytochelatin synthesis by zinc in marine green alga, *Dunaliella tertiolecta*. *J Biosci Bioeng* 92:24–29
- Iglesia-Turiño S, Febrero A, Jauregui O, Celdelas C, Araus JL, Bort J (2006) Detection and quantification of unbound phytochelatin 2 in plant extracts of *Brassica napus* grown with different levels of mercury. *Plant Physiol* 142:742–749
- Kawakami SK, Gledhill M, Achterberg EP (2006) Determination of phytochelatin and glutathione in phytoplankton from natural waters using HPLC with fluorescence detection. *TrAC, Trends Anal Chem* 25:133–142
- Komárek M, Čadková E, Chrástný V, Bordas F, Bollinger JC (2010) Contamination of vineyard soils with fungicides: a review of environmental and toxicological aspects. *Environ Int* 36:138–151
- Kondo N, Isobe M, Imai K, Goto T (1985) Synthesis of metallothionein-like peptides cadystin A and B occurring in a fission yeast, and their isomers. *Agric Biol Chem* 49:71–83
- Leita L, Mondini C, Nobili M, Simoni A, Sequi P (1998) Heavy metal content in xylem sap (*Vitis vinifera*) from mining and smelting areas. *Environ Monit Assess* 50:189–200
- Leopold I, Gunther D (1997) Investigation of the binding properties of heavy-metal-peptide complexes in plant cell cultures using HPLC-ICP-MS. *Fresenius J Anal Chem* 359:364–370
- Marczenko Z, Lobinski R (1996) Spectrochemical trace analysis for metals and metalloids. Elsevier, Amsterdam



- Minocha R, Thangavel P, Dhankher OP, Long S (2008) Separation and quantification of monothiols and phytochelatins from a wide variety of cell cultures and tissues of trees and other plants using high performance liquid chromatography. *J Chromatogr A* 1207:72–83
- Mounicou S, Meija J, Caruso J (2004) Preliminary studies on selenium-containing proteins in *Brassica juncea* by size exclusion chromatography and fast protein liquid chromatography coupled to ICP-MS. *Analyst* 129:116–123
- Newton GL, Dorian R, Fahey RC (1981) Analysis of biological thiols: derivatization with monobromobimane and separation by reverse-phase high-performance liquid chromatography. *Anal Biochem* 114:383–387
- Rellán-Álvarez R, Ortega-Villasante C, Álvarez-Fernández A, del Campo FFD, Hernández LE (2006) Stress responses of *Zea mays* to cadmium and mercury. *Plant Soil* 279:41–50
- Rijstenbil JW, Wijnholds JA (1996) HPLC analysis of non-protein thiols in planktonic diatoms: pool size, redox state and response to copper and cadmium exposure. *Mar Biol* 127:45–54
- Saleem-Saif M, Midrar-Ul-Haq RA, Memon KS (2005) Heavy metals contamination through industrial effluent to irrigation water and soil in Korangi area of Karachi (Pakistan). *Int J Agr Biol* 7:646–648
- Sanità di Toppi LP, Gabbriellini R (1999) Response to cadmium in higher plants. *Environ Exp Bot* 41:105–130
- Sobrinho-Plata J, Ortega-Villasante C, Flores-Caceres ML, Escobar C, Del Campo FF, Hernandez LE (2009) Differential alterations of antioxidant defenses as bioindicators of mercury and cadmium toxicity in alfalfa. *Chemosphere* 77:946–954
- Sobrinho-Plata J, Herrero J, Carrasco-Gil S, Pérez-Sanz A, Lobo C, Escobar C, Millán R, Hernández LE (2013) Specific stress responses to cadmium, arsenic and mercury appear in the metallophyte *Silene vulgaris* when grown hydroponically. *RSC Adv* 3:4736
- Sobrinho-Plata J, Carrasco-Gil S, Abadía J, Escobar C, Álvarez-Fernández A, Hernández LE (2014) The role of glutathione in mercury tolerance resembles its function under cadmium stress in *Arabidopsis*. *Metallomics* 6:356–366
- Spisso A, Pacheco PH, Gómez FJV, Silva MF, Martinez LD (2013) Risk assessment on irrigation of *vitis vinifera* L. cv malbec with Hg contaminated waters. *Environ Sci Technol* 47:6606–6613
- Thapa G, Sadhukhan A, Panda SK, Sahoo L (2012) Molecular mechanistic model of plant heavy metal tolerance. *Biomaterials* 25:489–505
- Todic S, Beslic Z, Lacic N, Tesic D (2006) Lead, mercury, and nickel in grapevine, *Vitis vinifera* L., in polluted and non-polluted regions. *Bull Environ Contam Toxicol* 77:665–670
- Vacchina V, Chassaigne H, Oven M, Zenk MH, Łobiński R (1999) Characterisation and determination of phytochelatin in plant extracts by electrospray tandem mass spectrometry. *Analyst* 124:1425–1430
- Wilcox J, Rupp E, Ying SC, Lim DH, Negreira AS, Kirchofer A, Feng F, Lee K (2012) Mercury adsorption and oxidation in coal combustion and gasification processes. *Int J Coal Geol* 90–91:4–20
- Wood BA, Feldmann J (2012) Quantification of phytochelatin and their metal (loid) complexes: critical assessment of current analytical methodology. *Anal Bioanal Chem* 402:3299–3309
- Wu GH, Cao SS (2010) Mercury and cadmium contamination of irrigation water, sediment, soil and shallow groundwater in a wastewater-irrigated field in Tianjin, China. *Bull Environ Contam Toxicol* 84:336–341
- Zeng X, Ma LQ, Qiu R, Tang Y (2009) Responses of non-protein thiols to Cd exposure in Cd hyperaccumulator *Arabis paniculata* Franch. *Environ Exp Bot* 66:242–248
- Zenk MH (1996) Heavy metal detoxification in higher plants—a review. *Gene* 179:21–30