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Application of Maillard reaction products on chia seed oil microcapsules with different core/wall ratios

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14 **Abstract**

15 This research studies the physical properties of microcapsules formulated with  
16 different concentrations of chia oil, using Maillard Reaction Products (MRPs) with  
17 different protein:carbohydrate ratio as encapsulants. Microcapsules were obtained  
18 from freeze-drying of O/W emulsions composed by non-heated/heated aqueous  
19 phases containing NaCas (10%wt) and lactose (10 or 20% wt/wt) blends. Chia oil  
20 (10, 15 or 20%wt/wt) constituted the oil phases. The moisture content of  
21 microcapsules was 0.31-2.23% d.b., while the water activity was ~0.500. The  
22 dispersibility and color were also studied. The microencapsulation efficiency varied  
23 between 41.43 and 83.95%. The bulk density was 323-551 kg/m<sup>3</sup> and 244-301 kg/m<sup>3</sup>  
24 for tapped and aerated density, respectively. All microcapsules exhibited an outer  
25 topography characterized by flakes and agglomerates without cracks or dents. The  
26 particle size distribution and D[3,2] of reconstituted emulsions were analyzed. The  
27 heat treatment improved the protection of chia oil against lipid oxidation in most  
28 samples, partially due to the antioxidant properties of the MRPs. Also, the oil content  
29 and the protein:carbohydrate ratio affected de oxidative stability. Thus, MRPs  
30 produced by heat treatment of NaCas-lactose mixture with different  
31 protein:carbohydrate ratios were effective for conferring microencapsulated chia oil  
32 additional oxidative stability.

33

34

35 **Keywords:** chia oil; omega-3 fatty acids; microencapsulation; physicochemical  
36 properties

37

## 38 1. Introduction

39 Chia (*Salvia hispanica* L., *Labiatae*) seed contains about 32-38% of oil by weight  
40 (Ayerza & Coates, 2005) and is a good source of polyunsaturated fatty acids  
41 (PUFAs), mainly  $\omega$ -3 ( $\alpha$ -linolenic ~ 60%), with a low content of saturated fatty acids  
42 (SFAs) and bioactive compounds (Ixtaina et al., 2011).

43 The consumption of  $\omega$ -3 PUFAs offers multiple health benefits, such as protection  
44 against the incidence of coronary diseases, inflammatory disorders, asthma, retina  
45 diseases, and helping brain function. Therefore, the incorporation of these  
46 compounds in human diet is desirable (O'Dwyer, O' Beirne, Ní Eidhin, O' Kennedy et  
47 2013; Kaushik, Dowling, Barrow, Adhikari, 2015). Health authorities of different  
48 countries have promoted the intake of foods containing high amounts of  $\omega$ -3 PUFAs,  
49 consequently a wide variety of commercial food products enriched with this type of  
50 fatty acids has recently been developed (Jacobsen et al., 2013).

51 Chia oil has a high nutritional value associated with its fatty acid profile. However, its  
52 high PUFAs content makes it very susceptible to the oxidation process (Ixtaina,  
53 Nolasco, Tomás, 2012). Thus, microencapsulation is a technology that can be used  
54 to protect this oil against oxidation during storage and/or processing. Thus, the  
55 microcapsules obtained with chia oil could be used as an ingredient to developing  $\omega$ -  
56 3 fortified foods.

57 The main purposes of this technique are to achieve high microencapsulation  
58 efficiency and to provide high oxidative stability of the core. These two objectives are  
59 closely related to the process employed for microencapsulation, the composition of  
60 the wall material and the core/wall ratio (Gharsallaoui, Roudaut, Chambin, Voilley,  
61 Saurel, 2007; Sanguansri & Augustin, 2007).

62 Spray-drying and freeze-drying are different processes applied for  
63 microencapsulation. Spray-drying is the most widely used process in the food  
64 industry since it is economical and flexible. Freeze-drying is a drying process carried  
65 out at low temperature, and it could be appropriate for microencapsulation of oils  
66 highly sensitive to the oxidation process, such as chia oil. Previous studies have  
67 shown the benefits of freeze-drying process to obtain microcapsules (Choi et al.  
68 2007; Chen, Zhong, Wen, McGillivray, Quek, 2013).

69 The different types of wall materials provide different extents of oxidative stability,  
70 depending primarily on their ability to inhibit oxygen transfer (Kaushik et al., 2015).  
71 Proteins and carbohydrates are commonly used for microencapsulation of oils with  
72 high  $\omega$ -3 content (Sanguansri & Augustin, 2007; Ixtaina, Julio, Wagner, Nolasco,  
73 Tomás, 2015). The proteins and carbohydrates blends are excellent for  
74 microencapsulation (Rosenberg & Sheu, 1996). The emulsification properties of  
75 proteins and particularly sodium caseinate (NaCas), seem to offer the functional and  
76 physical characteristics necessary to encapsulate lipid core materials (Hogan,  
77 McNamee, O'Riordan, O'Sullivan, 2001). The disaccharide lactose forms a  
78 continuous glass phase in which the protein chains are dispersed and improve the  
79 drying properties of the wall (Rosenberg & Sheu, 1996). Different researchers have  
80 studied the drying of O/W emulsions using NaCas and lactose (Calvo, Hernández,  
81 Lozano, González-Gómez, 2010; Ixtaina et al., 2015; Velasco, Marmesat,  
82 Dobarganes, Márquez-Ruiz, 2006). The Maillard reaction products (MRPs), formed  
83 when proteins and carbohydrates with reducing sugar groups are mixed under  
84 certain temperature and time conditions, can be used to enhance the oxidative  
85 stability of oils with high PUFAs content. The protein-carbohydrate conjugates  
86 formed as consequence of Maillard reaction have been shown to have emulsifying

87 and antioxidant capacity. Thus, they have been applied to microencapsulate different  
88 oils (Augustin, Sanguansri, Bode, 2006; Jacobsen, Sørensen, Nielsen, 2013; Rusli,  
89 Sanguansri, & Augustin, 2006).

90 Some studies regarding the microencapsulation of chia seed oil have been published  
91 (Rodea-González et al., 2012; Martínez et al., 2015; Ixtaina et al., 2015; Escalona-  
92 García et al., 2016; González, Martínez, Paredes, León, Ribotta, 2016). However,  
93 none of them reported the use of MRPs as wall material to encapsulate chia oil. This  
94 research was carried out to study whether MRPs produced by heat treatment of  
95 NaCas-lactose mixture with different protein:carbohydrate and core/wall ratios would  
96 be effective for conferring microencapsulated chia oil additional oxidative stability.

97 The aim of this research was to investigate the effects of the MRPs, the oil  
98 concentration and the protein:carbohydrate ratio in the wall on the physicochemical  
99 characteristics and oxidative stability of chia seed oil microencapsulated using  
100 NaCas and lactose by freeze-drying for the application as functional ingredient in  
101 foods.

102

## 103 **2. Materials and methods**

### 104 2.1. Materials

105 Commercial chia cold-pressed oil was provided by Nutracéutica Sturla S.R.L  
106 (Argentina) and stored for 3 days at  $4\pm 1$  °C without head space protected from light  
107 and oxygen.

108 Sodium caseinate was purchased from Sigma-Aldrich Company (St. Louis, MO,  
109 USA), D-lactose monohydrate from Cicarelli Laboratories Reagents S.A. (San  
110 Lorenzo, Argentina). All reagents were analytical grade.

111

## 112 2.2. Experimental design

113 A fully factorial design (3x2x2), with two replications, was applied to study the effects  
114 of three factors, including the MRPs -obtained by heat treatment at 60°C for 30 min-;  
115 the core/wall ratios; and the different concentrations of lactose. Twelve different  
116 emulsions were prepared (**Table 1**) and the microcapsules were produced from them  
117 as described in sections 2.3.1 and 2.3.3. The microcapsules were subjected to a  
118 storage trial during 30 days. About 15 g of each type of microcapsule was placed in  
119 an open Petri dish covered by foil with small holes and placed in desiccators at a  
120 relative humidity of 33% (using supersaturated solution of MgCl<sub>2</sub>) at room  
121 temperature.

122

## 123 2.3. Methods

### 124 2.3.1. Emulsion preparation

125 Chia oil-in-water (O/W) emulsions were composed of NaCas (10% in weight (wt)),  
126 different lactose concentrations (10 or 20% wt/wt), and 10, 15 or 20% (wt/wt) of chia  
127 oil (**Table 1**).

128 Prior to emulsification, the NaCas was dissolved in distilled water at 50°C using  
129 magnetic agitation. For emulsions containing lactose without heat treatment, the  
130 carbohydrate was incorporated in the aqueous phase at 25°C. In the case of  
131 emulsions with lactose and heat treatment, the protein-carbohydrate mixture was  
132 heated at 60°C in a water-bath and held for 30 min in order to promote the MRPs  
133 (Augustin et al., 2006). Nisine (0.0012g/100g) and potassium sorbate (0.1g/100g)  
134 were used to prevent microbial growth.

135 Preliminary homogenization was performed for 1 min at 9,500 rpm using an Ultra  
136 Turrax T-25 (IKA Labortechnik, Germany), equipped with a S25N-18G dispersing

137 tool . The resultant pre-emulsions were further subjected to a second stage of  
138 homogenization in a Panda 2K high pressure valve homogenizer (GEA Niro Soavi,  
139 Parma, Italy) at 600 bar, with four recirculation cycles.

140

### 141 2.3.2. Parent emulsion characterization

#### 142 2.3.2.1. Particle size distribution and mean diameter

143 The particle size distribution of the emulsions was determined by light scattering  
144 using a Mastersizer 2000 instrument equipped with a Hydro 2000MU as dispersion  
145 unit (Malvern Instruments Ltd., Worcestershire, UK) (Ixtaina et al., 2015). The pump  
146 speed was settled at 2,000 rpm. The refractive index of the disperse phase was  
147 1.47. The droplet size was reported as Sauter diameter ( $D [3, 2]$ ), which estimates  
148 the specific surface area of the emulsions (Ixtaina et al., 2015).

149 The Span value was calculated according to Eq. 1:

$$150 \text{ Span} = \frac{(d(v,90)) - (d(v,10))}{(d(v,50))} \quad (1)$$

151 where  $d(v,10)$ ,  $d(v,50)$ , and  $d(v,90)$  are diameters at 10%, 50%, and 90% cumulative  
152 volume calculated from the particle size distribution curves, respectively.

#### 153 2.3.3. Preparation of microcapsules by freeze-drying

154 First, the samples were frozen. For that, the emulsions (100g) were placed into  
155 plastic trays (12.5 cm × 16.0 cm), frozen at  $-20 \pm 1$  °C for 48h and then transferred to -  
156  $80 \pm 1$  °C for 24 h. Following, microcapsules were obtained from the frozen emulsions  
157 by freeze-drying in laboratory scale equipment for 48 h. The samples were ground  
158 using a manual mortar and sifted using a plastic mesh equivalent to ASTM No. 7  
159 sieve in order to standardize the powder size.



#### 160 2.3.4. Microcapsule characterization

##### 161 2.3.4.1. Moisture content

162 The moisture content of the chia oil powders (2 g) was measured gravimetrically by  
163 drying the microcapsules (24 h, 70°C, 29 in Hg) in a vacuum oven (Instrumentación  
164 Científica S.A., Buenos Aires, Argentina) (Baik et al., 2004).

##### 165 2.3.4.2. Water activity

166 This parameter was determined using an AquaLab Water Activity Meter CX2 model  
167 Decagon Devices Inc, USA, at 25±0.5°C.

##### 168 2.3.4.3. Essential fatty acid content

169 The 18:2 ( $\omega$ -6) and 18:3 ( $\omega$ -3) content was determined by  $^1\text{H}$  NMR spectroscopy.  
170 Approximately 300 mg of each sample was weighed and dissolved in 1.5 mL  
171 chloroform- $d_1$ . The mixture was ultrasonicated for 30 min and afterwards was shaken  
172 for 2 h. Then the mixture was centrifuged and 2 mL of dimethylsulfoxide- $d_6$  with  
173 tetramethylsilane (TMS) was added to the mixture.

174 A Bruker Avance III 500 MHz spectrometer (Bruker Biospin, Rheinstetten, Germany)  
175 with a BBFOPLUS SmartProbe probe equipped with a Bruker Automatic Sample  
176 Changer (B-ACS 120) was used to carry out the NMR measurements at ambient  
177 temperature.  $^1\text{H}$  NMR spectra were recorded using a standard 1D pulse sequence  
178 (PS) at a  $30^\circ$  flip angle with 512 scans, 131k time domain, 24.02 ppm spectral width,  
179 receiver gain of 90.5, and 5.45 s acquisition time. The data were recorded  
180 automatically by ICON-NMR (Bruker Biospin, Rheinstetten, Germany). All NMR  
181 spectra were manually phased, baseline-corrected and integrated by a Topspin 3.2  
182 (Bruker Biospin, Rheinstetten, Germany).

183 Specific NMR regions were used for quantification:  $\delta$  2.75-2.85 ppm (18:3) and  $\delta$   
184 2.69-2.75 ppm (18:2).

185 2.3.4.4. Microencapsulation efficiency of total oil,  $\omega$ -6 and  $\omega$ -3 PUFAs.

186 Microencapsulation efficiency of total oil (ME%) was performed according to  
 187 Augustin et al. (2015) with some modifications. About 1 g of microcapsules was  
 188 placed on filter paper (Whatman N° 4), washed three times with 10 mL of hexane,  
 189 collected on a flask and then evaporated under a nitrogen stream. The free oil  
 190 content was determined by weight difference. It was assumed that the total oil was  
 191 equal to the initial oil since previous study (Ixtaina et al., 2015) showed that all the  
 192 initial chia oil remained in the microcapsules. Microencapsulation efficiency was  
 193 calculated according to Eq (2):

$$194 \quad ME\% = \left( \frac{\text{Total Oil} - \text{Free Oil}}{\text{Total Oil}} \right) \times 100 \quad (2)$$

195  
 196 Microencapsulation efficiency of  $\omega$ -6 (ME% <sub>$\omega$ -6</sub>) and  $\omega$ -3 (ME% <sub>$\omega$ -3</sub>) PUFAs were  
 197 calculated by the data from the fatty acid analysis of encapsulated oil and total oil  
 198 determined by <sup>1</sup>H NMR spectroscopy according to Eqs. 3 and 4:

$$199 \quad ME\%_{\omega-6} = \frac{\omega-6 \text{ of microencapsulated oil}}{\omega-6 \text{ of total oil}} \quad (3)$$

$$201 \quad ME\%_{\omega-3} = \frac{\omega-3 \text{ of microencapsulated oil}}{\omega-3 \text{ of total oil}} \quad (4)$$

202

203 2.3.4.5. Powder bulk density and compressibility

204 In addition to the ME of microcapsules, other quality control parameters such as bulk  
 205 density, Carr Index and Hausner Ratio are used to evaluate the powder flRatio are  
 206 (Fitzpatrick, 2005).The bulk density can be defined as the mass of a powder divided

207 by the volume occupied by it. The bulk density can be classified as aerated and  
 208 tapped densities. The aerated bulk density ( $\rho_A$ ) was analyzed by allowing the  
 209 dispersed powder to settle in a container due to the gravity influence, whereas the  
 210 tapped bulk density ( $\rho_T$ ) was obtained by tapping the container holding the powder.  
 211 These densities were measured according to Holgado, Márquez-Ruiz, Dobarganes,  
 212 Velasco, 2013. For this propose a graduate cylinder (100 mL) with 25 g of powder  
 213 was used, and the respective densities were calculated according to Eqs. 5 and 6.

$$214 \quad \rho_A = \frac{m_0}{V_0} \quad (5)$$

215 where

216  $v_0$ : volume occupied by the powder ( $m^3$ )

217  $m_0$ : powder mass (kg)

$$218 \quad \rho_T = \frac{m_0}{V_T} \quad (6)$$

219 where

220  $v_T$ : volume occupied by the powder after tapping ( $m^3$ )

221 From the parameters previously described, the compressibility (C) was calculated  
 222 according to Eq (7):

$$223 \quad C = \frac{(\rho_T - \rho_A)}{\rho_T} \quad (7)$$

224

#### 225 2.3.4.6. Microstructure

226 The microcapsule morphology was study by scanning electron microscopy (SEM).  
 227 The microcapsules were fixed on a sample holder with graphite tape, and then  
 228 metalized with gold (SPI Supplies) Sputter. The samples were observed using a  
 229 FEI-Quanta 200 instrument in high vacuum mode operating at 20 Kv.

230

## 231 2.3.4.7. Color

232 Samples were homogeneously distributed in a glass Petri dish (diameter 95 mm)  
233 and the color of the microcapsule surface was measured using a Minolta colorimeter  
234 (CR-400, Konica Minolta Sensing Inc., Japan) calibrated with a white standard tile.  
235 Color was recorded using the  $L^*$  (lightness)  $a^*$  (red-green component) and  $b^*$   
236 (yellow-blue component) values of samples.

237

## 238 2.3.4.8. Particle size distribution and mean diameter of the reconstituted emulsion

239 The dispersion of the powder (solid content 10% wt/wt) was made by stirring the  
240 microcapsules in water at room temperature for 30 min. The measurements were  
241 carried out according to section 2.3.2.1.

242

## 243 2.3.4.9. Dispersibility

244 Dispersibility of the microcapsules was determined according to Klinkesorn,  
245 Sophanodora, Chinachoti, McClements, Decker (2005). Samples ~0.3 mg of  
246 powder/mL of distilled water were added within the stirring chamber (500 mL) of a  
247 laser diffraction instrument (Malvern Mastersizer Model 2000 E, Malvern  
248 Instruments, Worcestershire, UK) spinning at 2,000 rpm, measuring changes in  
249 mean particle diameter ( $D [3,2]$ ) and obscuration during 5 min.

250

## 251 2.3.4.10. Oxidative stability

252 An accelerated oxidation test of the bulk oil and the microcapsules was performed in  
253 a Rancimat (Metrohm 679, Switzerland) (AOCS Cd 12b-92, 2013) apparatus using

254 3.0 g of oil or 1.5 g of microcapsules at 98 °C with continuous bubbling of an air  
255 stream at 20 L/h. Stability was expressed as induction time ( $t_i$ ), in hours.

256

#### 257 2.3.4.11. Peroxide value

258 Peroxide value was evaluated spectrophotometrically according to the method of  
259 Díaz, Dunn, McClements, Decker, (2003). Briefly, the emulsions were reconstituted  
260 from the powders according to 2.3.4.7 section. The extraction of lipid hydroperoxides  
261 was made by mixing 300  $\mu$ l of the reconstituted emulsion with 1.5 mL of an iso-  
262 octane/isopropanol (3:1 v:v) mixture, vortexing 3 times for 10 s each. The phases  
263 were separated by centrifuging and the organic phase was used for analysis. The  
264 organic phase was added to 2.8 mL of a methano/butanol solution (2:1 v/v) followed  
265 by 15  $\mu$ L of 3.94 M thiocyanate solution and 15  $\mu$ L of 0.072 M acidic ferrous iron  
266 solution. After 20 min in the dark at room temperature, the absorbance was  
267 measured at 510 nm. Lipid hydroperoxide concentrations were determined using  
268 cumene hydroperoxide standard curve.

269

#### 270 2.3.5. Statistical analysis

271 Multifactorial ANOVA test was used to analyze the main effects of each factor and  
272 the interactions between them. Tukey's High Meaningful Difference test was  
273 performed ( $p \leq 0.05$ ) for mean multiple comparisons. Statgraphics Centurion  
274 software (Version XV.II for Windows, Manugistics Inc., USA) was used for the  
275 statistical analysis.

### 276 3. Results and discussion

#### 277 3.1. Parent emulsion characterization

##### 278 3.1.1. Particle size distribution and mean diameter

279 It is important to obtain emulsions with high physical stability due to the relatively  
280 long time required for freeze drying, during which possible losses of the material to  
281 be encapsulated could occur (Chen et al. 2013). In this sense, the particle size  
282 distribution and the mean diameter are relevant because these parameters are  
283 closely related to the physical stability of the emulsions.

284 **Figure 1** shows the particle size distribution curves for the parent emulsions  
285 prepared with different protein:carbohydrate ratios and oil concentrations. The  
286 particle diameters ranged from 0.1-10 and 0.1-239  $\mu\text{m}$  for emulsions with and  
287 without heat treatment, respectively. The particle size distribution profiles of the  
288 parent emulsions were bimodal, except for sample with 15% of chia oil, 20% of  
289 lactose without heat treatment which presented unimodal distribution. In the case of  
290 emulsions with a similar protein:carbohydrate ratio and the application of heat  
291 treatment, the particle size distribution was narrower than the other ones (**Fig. 1**). It  
292 can also be seen that the Span values of emulsions without heat treatment (1.1770-  
293 4.0900) were higher than those with thermal treatment (1.0260-2.5685), showing a  
294 lower polydispersibility level in these last systems. A similar result was obtained by  
295 Zhang et al. (2015), who observed that the emulsion with MRPs showed the smallest  
296 particle size and the narrowest size distribution. This behavior can be explained by  
297 the excellent emulsifying property of the protein-polysaccharide conjugates (Akhtar &  
298 Dickinson, 2007).

### 299 3.2 Microcapsule characterization

#### 300 3.2.1. Moisture content (MC)

301 **Table 2** shows that lactose concentration and heat treatment presented a very  
302 significant effect ( $p \leq 0.001$ ) on moisture content. Also, double and triple significant

303 interactions were found between factors, except lactose concentration x heat  
304 treatment.

305 The obtained values ranged between 0.31-2.23 % d.b. (**Table 3**), which are lower  
306 than those required to achieve chia oil microcapsules with a good stability during  
307 storage (3-4% d.b.) (Klaypradit & Huang, 2008).

308

### 309 3.2.2. Water activity ( $a_w$ )

310 There were no significant effects ( $p>0.05$ ) of oil load, lactose concentration or heat  
311 treatment on  $a_w$  (Tables 3 and 4). All samples showed values  $\sim 0.500$ , which were  
312 lower than 0.6 considered as the upper limit for a food to be microbiologically stable  
313 (Fazaeli, Emam-Djomeh, Kalbasi Ashtari, & Omid, 2012; Goyal et al., 2015). These  
314 values were higher than those reported by Ixtaina et al. (2015) for the microcapsules  
315 obtained by spray drying. Both  $a_w$  level and the moisture content of microcapsules  
316 obtained in this study would be appropriate for their incorporation in dehydrated food  
317 matrices. .

318

### 319 3.2.3. Microencapsulation efficiency of total oil (ME%), essential fatty acid content 320 and microencapsulation efficiency of $\omega$ -6 (ME% $\omega$ -6 ) and $\omega$ -3 (ME% $\omega$ -3 ) PUFAs

321 The ME% ranged between 41.4 and 83.9 % (**Table 4**), which were lower than those  
322 reported by Ixtaina et al. (2015) ( $\sim 95.0\%$ ) for microencapsulation of chia seed oil by  
323 spray-drying with NaCas and lactose. A lower ME% for the freeze-drying process in  
324 comparison with spray-drying was also found by Chen et al. (2013), who reported  
325 that this phenomenon could be produced by the dehydration of emulsifiers during the  
326 freezing of water phase, which promotes particle-particle interactions in emulsion

327 and reduces the emulsion stability. Thus, the encapsulated materials could be  
328 released from the core when ice crystals are removed during the drying stage.

329 The statistical analysis showed that the heat treatment was the most relevant factor  
330 with a negative impact, reducing the corresponding efficiency of the different  
331 microcapsules. A lesser influence was associated with the oil content and lactose  
332 concentration (**Table 2**). In this sense, a negative correlation was found between the  
333 total solid content and ME ( $r=-0.52$ ;  $p=0.0090$ ) and between oil content and ME ( $r=-$   
334  $0.43$ ;  $p=0.0374$ ). These results show the importance of having sufficient quantities of  
335 wall material for encapsulating chia oil. A significant interaction ( $p\leq 0.01$ ) between  
336 lactose concentration and heat treatment was found. Thus, for samples without heat  
337 treatment, no significant differences ( $p>0.05$ ) were detected for ME% between both  
338 of the lactose concentrations studied. However, in the case of samples with heat  
339 treatment, an increment in lactose concentration caused a decrease in ME%.

340 The essential fatty acid content of microencapsulated chia oil was 20.6-22.9 g  $\omega$ -  
341 6/100 g of oil and 58.2-65.9 g  $\omega$ -3/100 g of oil. These values were similar to those of  
342 bulk chia oil ( $\omega$ -6= 23.4 g/100g;  $\omega$ -3=60.0 g/100g), showing that microencapsulation  
343 process did not affect the essential fatty acid composition of chia oil. Thus,  
344 microcapsules could be used to fortified foods with this type of fatty acids.

345 Values of ME of  $\omega$ -6 and  $\omega$ -3 PUFAs calculated from the  $^1\text{H}$  NMR spectroscopy data  
346 presented a high correlation with those of ME of total oil obtained by the gravimetric  
347 analysis ( $r=0.97$ ;  $p=0.0000$ ). Thus, the highest ME of total oil,  $\omega$ -6 and  $\omega$ -3 PUFAs  
348 were found in non-heated samples with 10% of lactose and 10% of oil, while the  
349 lowest ones were recorded in heated samples with 20% of oil and 20% of lactose.

#### 350 3.2.4. Bulk Density



351 Regarding bulk density no effects ( $p>0.05$ ) of the different factors investigated in the  
352 experimental design were recorded) (**Table 2**). Bulk density varied between 323-551  
353  $\text{kg/m}^3$  and 244–301  $\text{kg/m}^3$  for tapped and aerated density, respectively (**Table 3**).  
354 These parameters depend on the particle size, distribution and characteristics of the  
355 material. Similar values were obtained for Quispe-Condori, Saldaña, Temelli (2011)  
356 for microcapsules with flax oil obtained by freeze drying.  
357 The compressibility in many powders is a measure of internal cohesion, flowability,  
358 and to some extent, deformability. A low compressibility indicates a less cohesive  
359 powder and a higher bulk density (Onwulata, Konstance, Holsinger, 1996).  
360 This property did not present significant differences ( $p>0.05$ ) between the  
361 experimental factors studied, but a significant oil load x heat treatment interaction  
362 ( $p\leq 0.01$ ) was recorded. This fact is important for the homogeneous character and  
363 reproducibility of the microcapsules to be subsequently included in food products  
364 (**Table 2**).

### 365 3.2.5. Microstructure

366 The scanning electronic micrographs (SEM) are shown in **Figure 2**. All the  
367 formulations exhibited an outer topography characterized by forming flakes and  
368 agglomerates with rough appearance without cracks or dents. The pores observed in  
369 cases 15OC10L, 15OC10LHT and 20OC10L were possibly formed by the cavities  
370 generated by the crystals of ice or bubbles of air retained during freezing. The  
371 existence of these pores would not affect the microencapsulation efficiency. Similar  
372 results were obtained by Gan, Cheng, Easa (2008), who worked with  
373 microencapsulated fish oil.

374 SEM micrographs of the microcapsules indicated that as the core:wall ratio  
375 increased, the flake size became larger and thicker.

376

### 377 3.2.6 Color

378 The color of the microcapsules is an important parameter because their  
379 incorporation as an ingredient in food products should not significantly alter the  
380 characteristics of the product.

381 The obtained results showed high  $L^*$  values (white and luminous), which decreased  
382 with storage (except for 15OC10L). Regarding  $a^*$  values, this parameters decreased  
383 as a function of storage time, whereas  $b^*$  values increased. These changes in color  
384 parameters showed yellower and a darker appearance at the end of the storage in  
385 comparison with the initial microcapsules (**Table 5**). Binsi et al (2017)

386 reported that the oxidation of triacylglycerols and free fatty acids can lead to changes  
387 in color, indicating the degree of deterioration of foods with high fat content. Thus,  
388 the color changes observed during storage would be associated with the oxidation of  
389 the surface oil of the microcapsules, which produced colored oxidation products.

### 390 3.2.7. Particle size distribution and mean diameter of the reconstituted emulsions

391 The mean diameter and droplet size distribution of reconstituted emulsions after  
392 freeze-drying were analyzed. The reconstitution of emulsions was made with  
393 distilled water (1 g solids/10 g emulsion) at  $\sim 25^\circ\text{C}$  for 30 min under stirring (Ixtaina et  
394 al., 2015).

395 **Figure 3** shows the particle size distribution curves of the reconstituted emulsions  
396 prepared with different wall protein:carbohydrate ratios and oil concentrations. All  
397 reconstituted emulsions showed a bimodal distribution, except the formulation  
398 20OC10L which presented three modes. It was observed that the influence of the

399 heat treatment improved the homogeneity of the systems studied. The same effect of  
400 the heat treatment on the width of the distribution had been recorded for the parent  
401 emulsions, shown that the good emulsifying property of the protein-carbohydrate  
402 conjugates produced from MRP was not affected by the microencapsulation process.  
403 In all cases the particle size distribution of the reconstituted emulsions was  
404 considerably wider (Span values: 3.1785-31.7865) than those of the parent  
405 emulsions. Similar results were reported during the microencapsulation of chia seed  
406 oil by spray drying (Ixtaina et al., 2015; Rodea-Gonzalez et al., 2012). The particle  
407 size increased with greater oil and lactose concentration and decreased with the  
408 application of heat treatment. This last case can be explained by the better  
409 emulsification obtained, which delays the flocculation. The curves with the highest  
410 homogeneity were associated with 10-15% of oil (**Fig. 3 A and B**). The statistical  
411 analysis indicated that the droplet size  $D[3,2]$  presented interactions between the  
412 factors, being the most important interactions oil load x heat treatment ( $p \leq 0.001$ ) and  
413 oil load x lactose concentration ( $p \leq 0.01$ ) (**Table 2**).

414

#### 415 3.2.8. Dispersibility

416 One of the most important properties of microcapsules is related to the speed and  
417 efficiency of powder to disperse in water (Klinkesorn et al., 2005). Therefore, the  
418 laser diffraction technique was used to obtain information about this parameter. The  
419 dispersibility of powdered emulsions was measured recording the obscuration and  
420 mean particle diameter  $D[3,2]$  changes as a function of stirring time.

421 At the initial time of the storage ( $t=0$ ), the obscuration increased sharply with the  
422 agitation up to approximately 1 min, and then remained constant after that time for  
423 most of the samples, except those with 20% of chia oil, 10% and 20% of lactose and

424 not heat treated, which continued to grow slightly (**Fig. 4 A**). Samples with heat  
425 treatment (**Fig. 4 B**) showed a similar behavior. The highest obscuration values were  
426 related to samples with 20% of chia oil and 10% of lactose (with and without heat  
427 treatment), whereas those with 20% of chia oil and 20% of lactose recorded the  
428 lowest values.

429 Additionally, the  $D [3,2]$  decreased quickly until 0.2 min, after which the particle size  
430 remained stable as a function of stirring time (**Fig. 5**).

431 The fast reduction in particle size and the increase in obscuration showed that most  
432 of the powder dissolved rapidly, giving a homogeneous suspension (Klinkesorn et  
433 al., 2005).

434 These parameters are important because they allow us to evaluate the rehydration  
435 of powder. In **Figure 5** it can be seen that the particle size significantly reduced in  
436 the first few seconds of stirring, which is very favorable for solubilization and  
437 subsequent application in instant foods.

438

### 439 3.2.9. Oxidative stability

440 The oxidative stability of microencapsulated chia oil was evaluated by Rancimat  
441 immediately after drying ( $t = 0$  d) and during storage ( $t = 30$  d). The oxidative stability  
442 of the chia oil was effectively enhanced by freeze-drying microencapsulation, since  
443 all systems presented higher induction times ( $t_i$ ) than those corresponding to bulk  
444 chia oil ( $t_i = 2.46 \pm 0.07$  h). At  $t = 0$  d, the highest induction time was found for the  
445 10OC20LHT sample (**Table 3**). At this time, the statistical analysis indicated that all  
446 factors affected the oxidative stability (**Table 2**). Also, double and triple interactions  
447 between factors were found, except lactose concentration x heat treatment. For  
448 samples with 10 and 15% of oil, the highest lactose concentration produced an

449 increase of the induction time. In contrast, samples with 20% of oil showed an  
450 inverse behavior. The heat treatment showed a positive effect on the induction time  
451 for systems with 20% chia oil, 10% lactose and 10% chia oil and 20% lactose. This  
452 can be explained by the formation of protein-lactose conjugates which reduce the  
453 amount of hydroperoxides in the powders. It can also be seen that in samples  
454 containing 10 or 15% of oil, a 1:2 NaCas:lactose ratio improved the oxidative stability  
455 in comparison with those with 1:1ratio. Thus, the protein:carbohydrate ratio is an  
456 important factor in the Maillard reaction since a greater amount of reducing sugars  
457 available to participate in the reaction increases its rate and extent . These results  
458 suggest that the conjugates obtained by the Maillard reaction in the wall material are  
459 appropriated to improve the oxidative stability of microencapsulated chia seed oil.  
460 Similar results were obtained by Zhang et al. (2015) in the microencapsulation of fish  
461 oil using caseinate and maltodextrin. Research studies have shown that the Maillard  
462 reaction antioxidant products are formed as a result of the interaction of sugars with  
463 amino acids whether these products are at the interface or in the continuous matrix  
464 of the powder (Lingnert, Vallentin, Erikssonsik, 1979; McGookin & Augustin, 1991).  
465 Because this reaction is very common in foods, especially those rich in heat-treated  
466 proteins, the heat treatment was applied in this study to promote antioxidant  
467 products that could protect the microencapsulated chia oil.

468 At the end of storage, the induction time decreased significantly for all the samples  
469 (data not shown). Only oil load x lactose interaction was found.

470 A similar trend was recorded for the influence of the heat treatment in terms of PV  
471 (data not shown). .

472

#### 473 **4. Conclusion**

474 Microcapsules of chia oil were investigated in order to evaluate the influence of  
475 MRPs, oil concentration and protein:carbohydrate ratio in the wall on the  
476 physicochemical characteristic and stability of chia oil. The oil load, the lactose  
477 concentration and the heat treatment of the aqueous phase influenced the  
478 microencapsulation efficiency of total oil,  $\omega$ -6 and  $\omega$ -3 PUFAs, the oxidative stability  
479 of microcapsules and the particle size of the reconstituted emulsions. Moisture and  
480 water activity levels were low and suitable for dried products. The essential fatty acid  
481 composition of microencapsulated chia oil was similar to that of bulk oil, recording  
482 high levels of essential fatty acids, mainly  $\omega$ -3 PUFAs. All formulations exhibited  
483 good and fast dispersibility which is important in order to the rehydration properties  
484 of powders. The application of the heat treatment was beneficial for most of the  
485 variables studied, except for microencapsulation efficiency. The obtained results  
486 showed that the MRPs produced by heat treatment of NaCas-lactose mixture with  
487 different protein:carbohydrate ratios were effective for conferring microencapsulated  
488 chia oil additional oxidative stability.

#### 489 **Conflict of interest**

490 The authors declare no conflict of interest.

491

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498

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### 629 Figure captions

630 **Fig. 1** Particle size distribution (% volume) of parent emulsions: A) without heat  
 631 treatment B) with heat treatment - . . . 10OC10L; ——— 15OC10L; — — 20OC10L;  
 632 - - - 10OC20L; ----- 15OC20L; ..... 20OC20L

633 **Fig. 2** Micrographs of chia oil microcapsules for different formulations: A) 10OC10L,  
 634 B) 15OC10L, C) 20OC10L, D) 10OC20L, E) 15OC20L, F) 20OC20L, G)  
 635 10OC10LHT, H) 15OC10LHT, I) 20OC10LHT, J) 10OC20LHT, K) 15OC20LHT, L)  
 636 20OC20LHT

637 **Fig. 3** Particle size distribution (% volume) of reconstituted emulsions: A) without  
 638 heat treatment B) with heat treatment - . . . 10OC10L; ——— 15OC10L;  
 639 — — 20OC10L; - - - 10OC20L; ----- 15OC20L; ..... 20OC20L

640 **Fig. 4** Influence of stirring time on obscuration of chia seed oil microcapsules:  
 641 A) without heat treatment B) with heat treatment - ◇ - 10OC10L —■— 15OC10L  
 642 - ▲ - 20OC10L —×— 10OC20L --\*-- 15OC20L ...⊙... 20OC20L

643 **Fig. 5** Influence of stirring time on mean diameter of chia seed oil microcapsules:

644 A) without heat treatment B) with heat treatment - ◇ - 100C10L - ■ - 150C10L

- ▲ - 200C10L - — - 100C2 - - \* - - 150C20 ... ⊙ ... 200C20L

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**Table 1**

Formulations for chia O/W emulsions previous to freeze-drying based on 3x2x2 full factorial design. Experimental parameters and samples codes

Chia oil concentration (%wt/wt)	Lactose concentration (% wt/wt)			
	10		20	
	Heat treatment		Heat treatment	
	Without	With	Without	With
10	10OC10L	10OC10LHT	10OC20L	10OC20LHT
15	15OC10L	15OC10LHT	15OC20L	15OC20LHT
20	20OC10L	20OC10LHT	20OC20L	20OC20LHT

**Table 3.** Physicochemical properties of microcapsules of chia seed oil at initial time (t=0 d)

Sample	Moisture content (% <sub>w</sub> , d.b.)	$a_w$ (25°C)	Oxidative stability ( $t_i$ , h)	Aerated bulk density (kg/m <sup>3</sup> )	Tapped bulk density (kg/m <sup>3</sup> )	Compressibility Index	Particle size of the reconstituted emulsion D[3,2] (µm)
10OC10L	1.34 <sup>bcd</sup>	0.515 <sup>a</sup>	11.91 <sup>ab</sup>	301 <sup>a</sup>	551 <sup>a</sup>	0.435 <sup>b</sup>	0.283 <sup>a</sup>
15OC10L	0.73 <sup>abc</sup>	0.483 <sup>a</sup>	21.31 <sup>cde</sup>	266 <sup>a</sup>	340 <sup>a</sup>	0.217 <sup>a</sup>	0.450 <sup>ab</sup>
20OC10L	1.02 <sup>abc</sup>	0.495 <sup>a</sup>	15.23 <sup>abc</sup>	244 <sup>a</sup>	323 <sup>a</sup>	0.244 <sup>a</sup>	0.957 <sup>ab</sup>
10OC20L	0.77 <sup>abc</sup>	0.500 <sup>a</sup>	25.13 <sup>de</sup>	266 <sup>a</sup>	450 <sup>a</sup>	0.322 <sup>ab</sup>	0.266 <sup>a</sup>
15OC20L	0.69 <sup>ab</sup>	0.520 <sup>a</sup>	34.87 <sup>f</sup>	270 <sup>a</sup>	402 <sup>a</sup>	0.265 <sup>a</sup>	0.280 <sup>a</sup>
20OC20L	0.31 <sup>a</sup>	0.508 <sup>a</sup>	11.74 <sup>ab</sup>	286 <sup>a</sup>	414 <sup>a</sup>	0.230 <sup>a</sup>	16.778 <sup>c</sup>
10OC10LHT	1.44 <sup>cd</sup>	0.481 <sup>a</sup>	9.62 <sup>a</sup>	260 <sup>a</sup>	351 <sup>a</sup>	0.217 <sup>a</sup>	0.292 <sup>a</sup>
15OC10LHT	2.23 <sup>e</sup>	0.488 <sup>a</sup>	20.29 <sup>bcd</sup>	261 <sup>a</sup>	359 <sup>a</sup>	0.249 <sup>a</sup>	0.390 <sup>a</sup>
20OC10LHT	0.76 <sup>abc</sup>	0.496 <sup>a</sup>	27.73 <sup>def</sup>	255 <sup>a</sup>	352 <sup>a</sup>	0.232 <sup>a</sup>	0.516 <sup>ab</sup>
10OC20LHT	0.74 <sup>abc</sup>	0.522 <sup>a</sup>	51.96 <sup>g</sup>	290 <sup>a</sup>	430 <sup>a</sup>	0.212 <sup>a</sup>	0.292 <sup>a</sup>
15OC20LHT	0.80 <sup>abc</sup>	0.497 <sup>a</sup>	29.33 <sup>ef</sup>	263 <sup>a</sup>	391 <sup>a</sup>	0.264 <sup>a</sup>	0.291 <sup>a</sup>
20OC20LHT	1.99 <sup>de</sup>	0.513 <sup>a</sup>	6.73 <sup>a</sup>	276 <sup>a</sup>	471 <sup>a</sup>	0.306 <sup>a</sup>	2.889 <sup>b</sup>

$a_w$  water activity at 25°C;  $t_i$ , induction time

Mean values (n=3). The coefficients of variation were lower than 10%. Different letters in each column indicate differences at  $p \leq 0.05$  between formulations, according to Tukey (HSD) test.

**Table 4.** Microencapsulation efficiency of total oil,  $\omega$ -6 and  $\omega$ -3 PUFAs of microcapsules of chia seed oil

Samples	ME (%)	ME $_{\omega-6}$ (%)	ME $_{\omega-3}$ (%)
10OC10L	83.9 <sup>b</sup>	91.6 <sup>f</sup>	81.4 <sup>f</sup>
15OC10L	74.7 <sup>b</sup>	81.2 <sup>de</sup>	78.4 <sup>ef</sup>
20OC10L	57.4 <sup>ab</sup>	64.3 <sup>b</sup>	58.4 <sup>bc</sup>
10OC20L	73.3 <sup>ab</sup>	79.9 <sup>de</sup>	74.3 <sup>ef</sup>
15OC20L	67.4 <sup>ab</sup>	75.0 <sup>cd</sup>	69.6 <sup>de</sup>
20OC20L	79.7 <sup>b</sup>	82.2 <sup>def</sup>	79.6 <sup>ef</sup>
10OC10LHT	77.2 <sup>b</sup>	86.0 <sup>ef</sup>	77.4 <sup>ef</sup>
15OC10LHT	72.6 <sup>ab</sup>	80.2 <sup>de</sup>	74.0 <sup>def</sup>
20OC10LHT	63.8 <sup>ab</sup>	68.2 <sup>bc</sup>	68.8 <sup>cde</sup>
10OC20LHT	61.7 <sup>ab</sup>	69.1 <sup>bc</sup>	63.1 <sup>bcd</sup>
15OC20LHT	55.4 <sup>ab</sup>	63.5 <sup>b</sup>	56.0 <sup>b</sup>
20OC20LHT	41.4 <sup>a</sup>	46.4 <sup>a</sup>	41.3 <sup>a</sup>

ME% microencapsulation efficiency of total oil; ME% $_{\omega-6}$  microencapsulation efficiency of  $\omega$ -6 PUFAs; ME% $_{\omega-3}$  microencapsulation efficiency of  $\omega$ -3-PUFAs

Mean values (n=3). The coefficients of variation were lower than 10%. Different letters in each column indicate differences at  $p \leq 0.05$  between formulations, according to Tukey (HSD) test.



**Table 2**

Multifactorial analysis of variance (ANOVA) for the physicochemical properties of microcapsules of chia seed oil

Main effects	D.F.	Sum of squares									Oxidative stability	
		MC	a <sub>w</sub>	ME	ME <sub>ω-6</sub>	ME <sub>ω-3</sub>	ρ <sub>a</sub>	ρ <sub>e</sub>	C	D[3,2]	Initial	Final
		Oil load (A)	2	0.036	0.000	737.95*	1083.15***	591.53***	1027.0	23172.3	0.011	195.168***
Lactose concentration (B)	1	0.814***	0.002	424.89*	512.24***	499.14***	693.4	13254.0	0.000	30.736**	480.078***	158.569
Heat treatment (C)	1	1.591***	0.000	705.94**	616.61***	623.32***	126.0	2604.2	0.009	58.101***	108.120*	78.156
AxB	2	1.205***	0.000	224.57	165.92***	112.08**	1348.0	13456.0	0.011	65.433**	1618.640***	729.329*
AxC	2	0.700**	0.000	90.12	105.03**	45.23	89.3	25106.3	0.047**	115.943***	242.242**	40.815
BxC	1	0.028	0.000	583.00**	513.84***	702.11***	287.0	5340.2	0.004	13.599	8.378	62.823
AxBxC	2	2.792***	0.002	454.96	347.21***	546.73***	2093.3	11825.3	0.006	28.000*	579.124***	48.160
Pure error	12	0.420	0.007	822.26	68.95	91.77	11583.5	43141.0	0.041	41.574	189.681	1049.950
Total	23	7.588	0.011	4043.70	3413.26	3211.91	17246.6	137899.0	0.130	548.554	3793.760	2499.510

D.F Degree of freedoms; MC moisture content; a<sub>w</sub> water activity at 25°C; ME microencapsulation efficiency of total oil; ME<sub>ω-6</sub> microencapsulation efficiency of ω-6 PUFAs; ME<sub>ω-3</sub> microencapsulation efficiency of ω-3-PUFAs ; ρ<sub>a</sub> aerated density; ρ<sub>e</sub> packed density; C compressibility index ; D[3,2] average oil droplet diameters of the reconstituted emulsions \*p ≤0.05. \*\*p≤0.01; \*\*\* p≤0.001

**Table 5.** Color of microcapsules of chia seed oil for different formulations during storage at 20±1°C

Samples	t=0 d			t=30 d		
	L*	a*	b*	L*	a*	b*
10OC10L	93.51±1.89 <sup>a</sup>	-1.01±0.18 <sup>cd</sup>	12.81±0.89 <sup>a</sup>	91.32±0.78 <sup>a</sup>	-1.37±0.00 <sup>a</sup>	17.84±3.00 <sup>a</sup>
15OC10L	92.21±0.27 <sup>a</sup>	-1.22±0.21 <sup>bc</sup>	14.06±0.16 <sup>a</sup>	94.68±5.27 <sup>a</sup>	-1.71±0.29 <sup>a</sup>	16.79±1.12 <sup>a</sup>
20OC10L	92.89±2.11 <sup>a</sup>	-1.68±0.04 <sup>ab</sup>	14.34±0.34 <sup>a</sup>	90.15±1.78 <sup>a</sup>	-2.22±0.04 <sup>a</sup>	15.59±0.29 <sup>a</sup>
10OC20L	91.94±0.44 <sup>a</sup>	-1.04±0.08 <sup>cd</sup>	13.58±0.28 <sup>a</sup>	89.71±1.77 <sup>a</sup>	-1.66±0.24 <sup>a</sup>	16.38±0.19 <sup>a</sup>
15OC20L	91.49±0.07 <sup>a</sup>	-0.96±0.10 <sup>cd</sup>	14.29±0.34 <sup>a</sup>	90.61±1.05 <sup>a</sup>	-1.58±0.16 <sup>a</sup>	15.87±1.11 <sup>a</sup>
20OC20L	92.70±1.24 <sup>a</sup>	-1.83±0.01 <sup>a</sup>	15.04±0.05 <sup>ab</sup>	89.26±3.37 <sup>a</sup>	-2.41±0.00 <sup>a</sup>	17.92±0.00 <sup>a</sup>
10OC10LHT	91.57±0.38 <sup>a</sup>	-1.01±0.18 <sup>cd</sup>	15.05±1.43 <sup>ab</sup>	88.67±1.34 <sup>a</sup>	-1.50±0.08 <sup>a</sup>	15.48±0.00 <sup>a</sup>
15OC10LHT	93.42±1.58 <sup>a</sup>	-0.98±0.0 <sup>cd</sup>	14.02±1.27 <sup>a</sup>	91.35±0.47 <sup>a</sup>	-1.59±0.24 <sup>a</sup>	19.89±3.75 <sup>a</sup>
20OC10LHT	92.06±0.43 <sup>a</sup>	-1.17±0.17 <sup>bc</sup>	14.775±0.05 <sup>a</sup>	89.85±0.66 <sup>a</sup>	-1.62±0.09 <sup>a</sup>	16.35±0.34 <sup>a</sup>
10OC20LHT	91.47±0.81 <sup>a</sup>	-0.54±0.09 <sup>d</sup>	13.625±0.35 <sup>a</sup>	89.95±0.89 <sup>a</sup>	-0.89±0.14 <sup>a</sup>	15.70±0.65 <sup>a</sup>
15OC20LHT	92.85±1.63 <sup>a</sup>	-0.53±0.03 <sup>d</sup>	13.57±1.00 <sup>a</sup>	91.31±2.55 <sup>a</sup>	-0.85±0.01 <sup>a</sup>	14.54±0.55 <sup>a</sup>
20OC20LHT	90.00±0.38 <sup>a</sup>	-1.27±0.23 <sup>bc</sup>	17.64±0.21 <sup>b</sup>	88.14±2.52 <sup>a</sup>	-1.81±0.03 <sup>a</sup>	18.47±0.20 <sup>a</sup>

Different letters in each column indicate differences at  $p \leq 0.05$  between formulations, according to Tukey (HSD) test

FIGURE 1

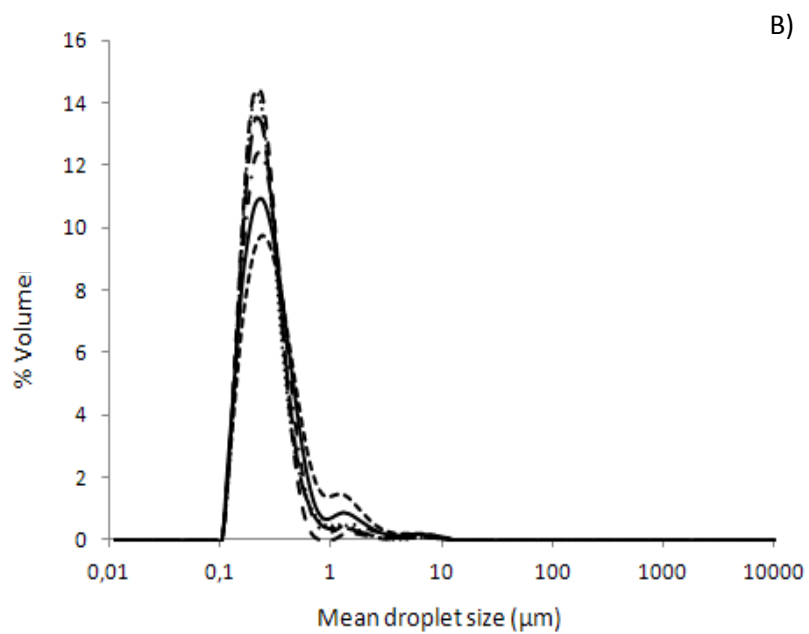
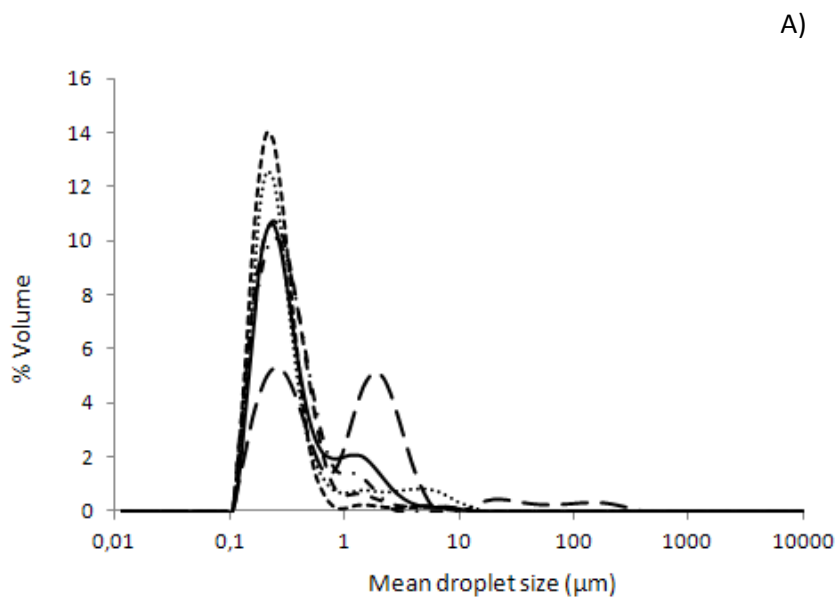


Figure 2.

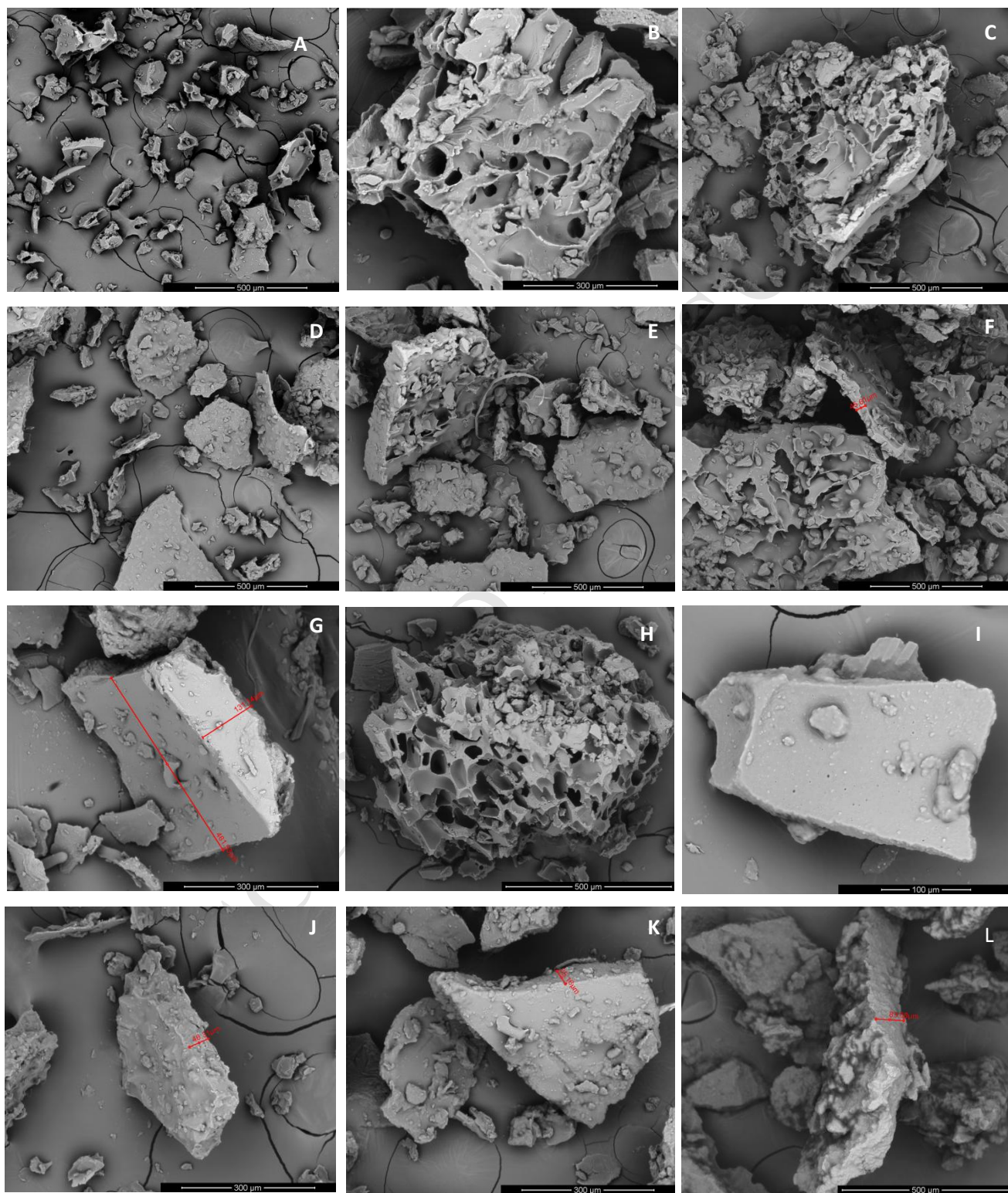


Figure 5

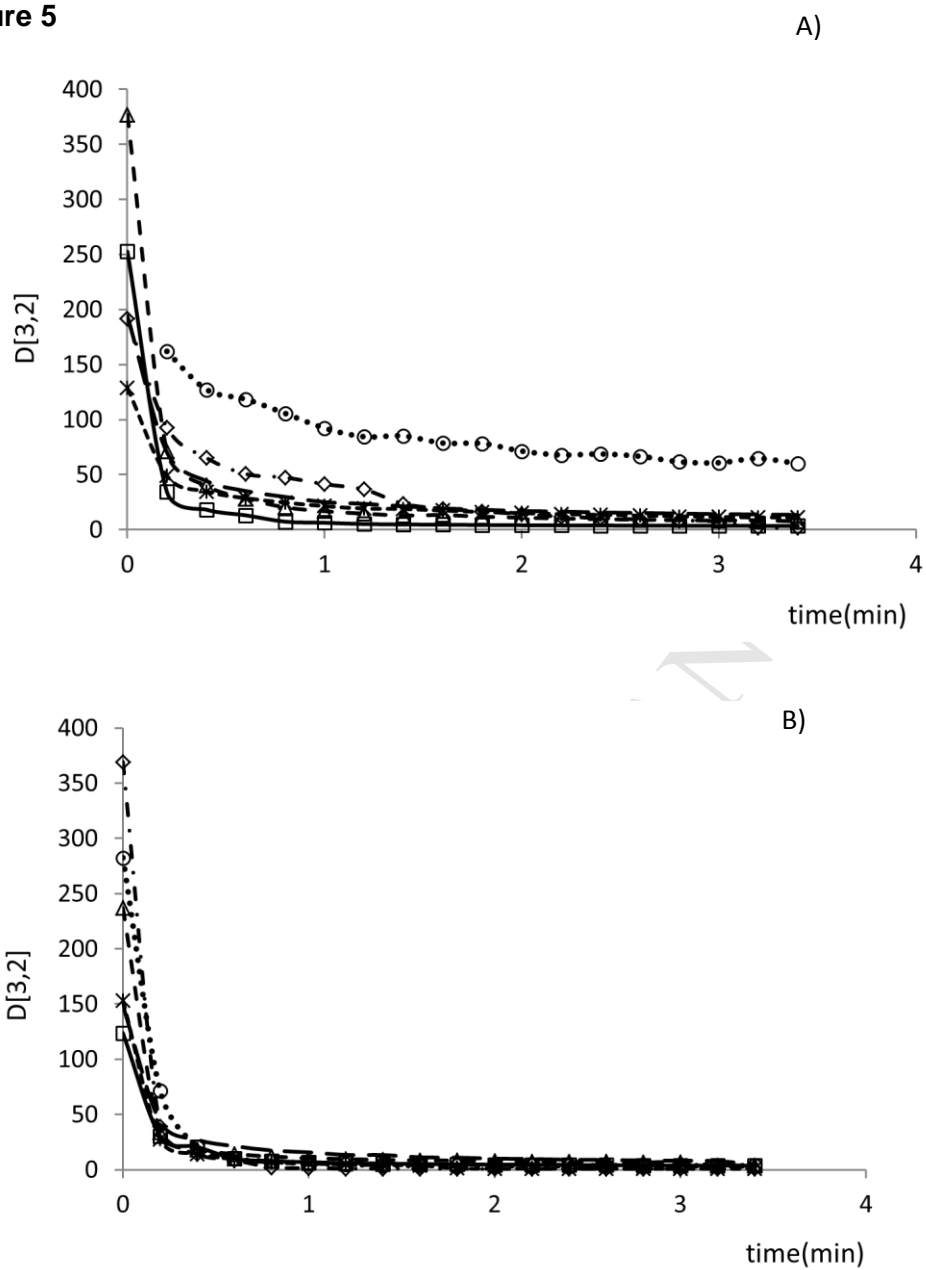


Figure 4

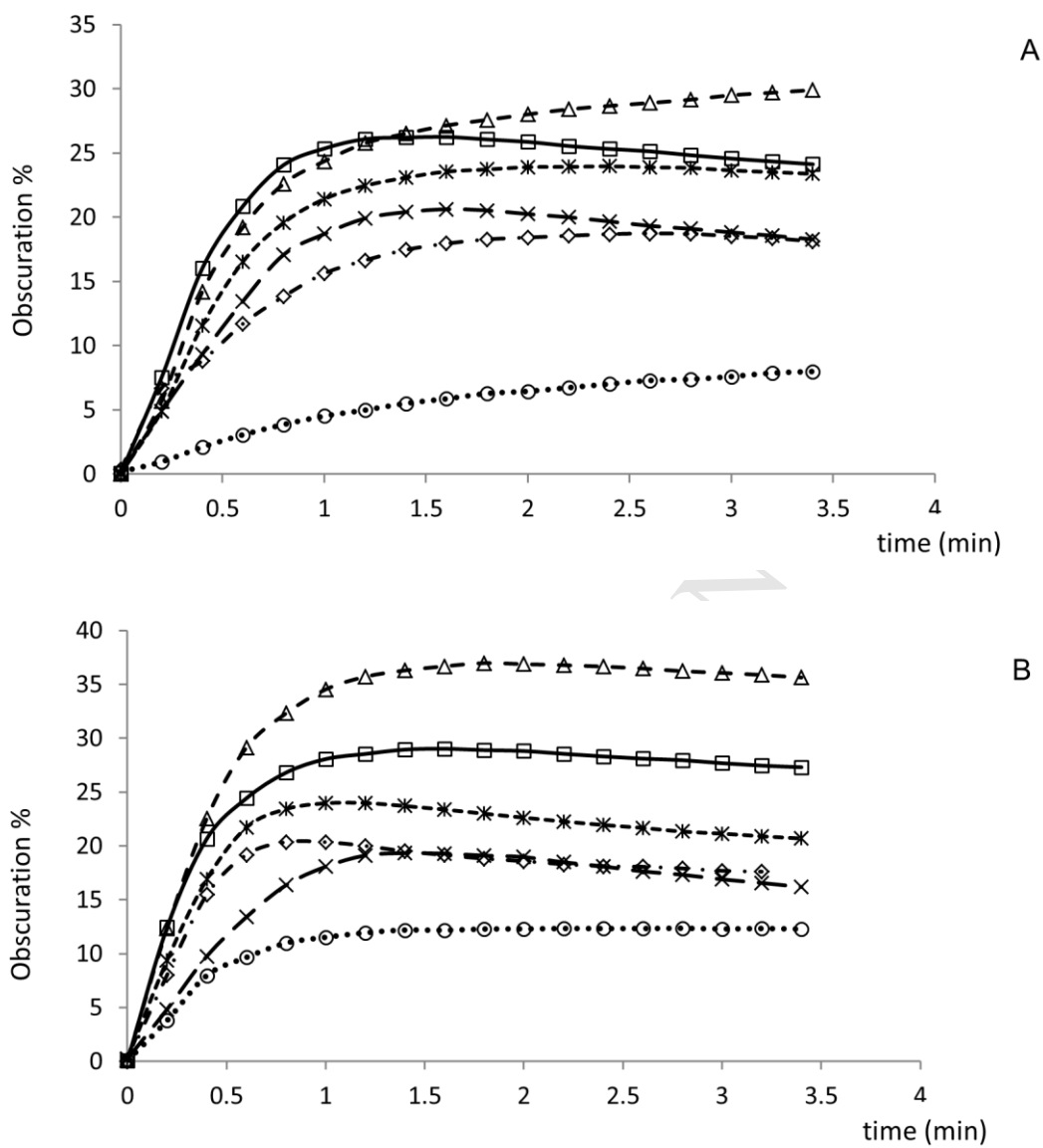
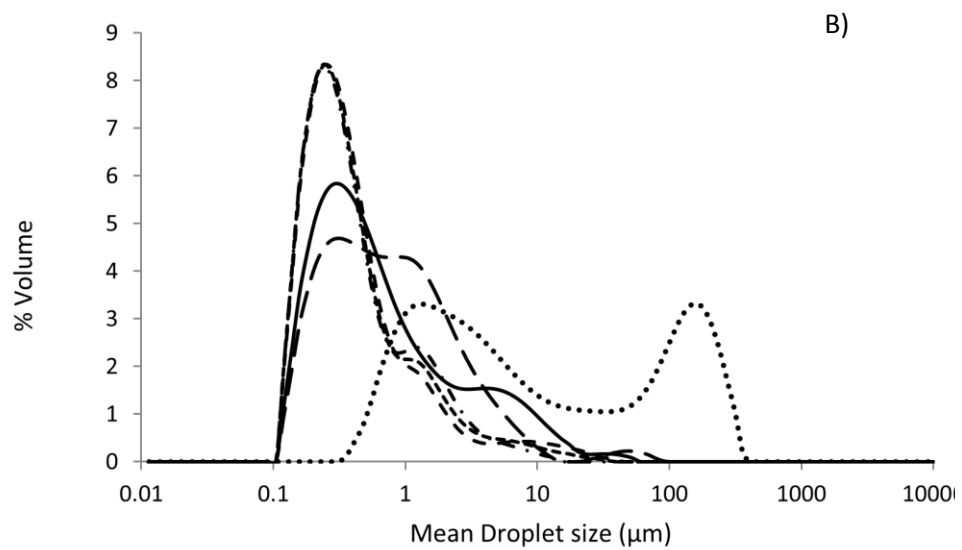
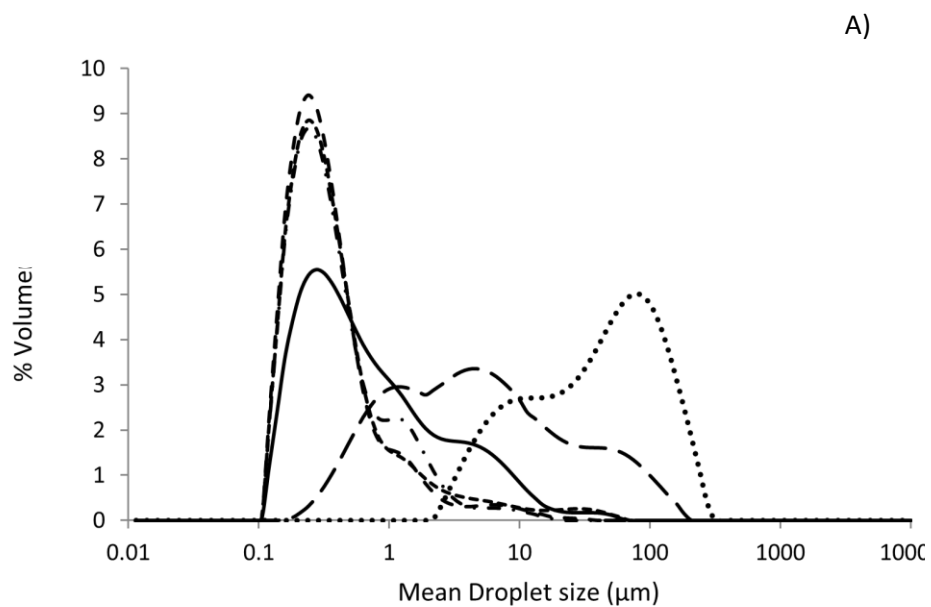


Figure 3



**Highlights**

- \* Oxidative protection of chia oil by microencapsulation with diverse core/wall ratio
- \* Oxidative protection of microencapsulated chia oil by the Maillard Reaction Products
- \* Highest oxidative stability in systems with 10% oil, 20% lactose and heat treatment
- \* Highest microencapsulation efficiency in 10% oil-10% lactose-systems (ME> 80%)
- \* Flake like powder particles with good physicochemical properties and dispersibility