

Application of an edible coating developed with Andean potato starch and carboxymethyl-cellulose for lipid reduction during frying

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SUMMARY: This work aimed to search for technological alternatives for the use of different varieties of Andean potatoes (*Solanum tuberosum* ssp andigenum) reintroduced in north-western Argentina. Coatings made with starch (S) extracted from the Runa variety, carboxymethyl-cellulose (CMC), and their combination (S/CMC) were studied. These coatings were applied to chips of Waycha variety potatoes before being fried; the effect of prior bleaching of the chips with different media was also evaluated: water, calcium chloride solution, and ascorbic acid. The results showed that the type of oil used did not affect their absorption by the chips. The bleaching treatments with calcium chloride and S/CMC coating significantly reduced the oil absorption ($39.5 \% \pm 0.7$), delayed its oxidation, and decreased the loss of tocopherols during the frying process. It also contributed to the physical and sensory characteristics of the final product, which presented high acceptability by consumers.

KEYWORDS: *Andean potato starch, Andean potato chips, coating, frying, oil absorption*

RESUMEN: Este trabajo tuvo como objetivo buscar alternativas tecnológicas para el aprovechamiento de diferentes variedades de papa andina (*Solanum tuberosum* ssp andigenum) reintroducidas en el noroeste argentino. Se estudiaron recubrimientos elaborados con almidón (S) extraído de la variedad Runa, carboximetilcelulosa (CMC) y su combinación (S/CMC). Estos recubrimientos se aplicaron a chips de papa variedad Waycha antes de ser fritos; también se evaluó el efecto del blanqueo previo de los chips con diferentes medios: agua, solución de cloruro de calcio y ácido ascórbico. Los resultados mostraron que el tipo de aceite utilizado no afectó su absorción por los chips. Los tratamientos de blanqueo con cloruro de calcio y recubrimiento S/CMC redujeron significativamente la absorción de aceite ($39,5\% \pm 0,7$), retrasaron su oxidación y disminuyeron la pérdida de tocoferoles durante el proceso de fritura. También contribuyeron a las características físicas y sensoriales del producto final, que presentó alta aceptabilidad por parte de los consumidores.

PALABRAS CLAVE: *almidón de papa Andina, hojuelas de papa andinas, recubrimiento, fritura, absorción de aceite*

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45 1. INTRODUCTION

46 Many strategies have been proposed to reduce oil absorption in food during frying (Crosa *et*
47 *al.*, 2014). The application of hydrocolloid coatings has been one of the most promising
48 methods (Naghavi *et al.*, 2018). These modify the food surface, decreasing its permeability.
49 Among the materials used to make coatings are methylcellulose (MC), hydroxypropyl-cellulose
50 (HPC), carboxymethyl-cellulose (CMC) and hydroxypropylmethyl-cellulose (HPMC) (Mai *et*
51 *al.*, 2007). The use of natural polymers in packaging and food additive applications is gaining
52 popularity, due to the prevention of environmental problems. Edible coatings must contain
53 substances that comply with food standards and must also be economical, easy to apply and
54 respectful of the environment. Cellulose is a very abundant natural polysaccharide, it is the
55 structural component of the cell wall of plants, therefore, it is obtained from various natural
56 sources, from food waste, cereal bran and fruit peels. The main reasons for the common use of
57 CMC are its viscosity, flocculant property, excellent oil resistance, transparency, non-toxicity,
58 and low price. It has GRAS status from the FDA (Yalsin *et al.*, 2022). Starch, as one of the most
59 abundant, sustainable, and low-cost commercial biopolymers, has versatile applications in many
60 industries (e.g., food, paper, packaging, etc.). Potato varieties to produce fried products at the
61 industrial level must meet external and internal quality characteristics. Environmental and
62 genetic effects determine external qualities. The internal ones refer to the chemical composition
63 of tubers that includes the content of sugars, dry matter, and starch, among others. Reducing
64 sugars (fructose and glucose) have a critical role in the industrial process, so the legislation
65 establishes contents lower than 0.25% for French fries' elaboration and less than 0.2% (Colman
66 *et al.*, 2009) for potatoes in flakes. Frying is an operation widely used in the food industry. It is
67 based on the transfer of heat from hot oil to foods, causing water removal and oil absorption.
68 The high consumption of fried products is a risk factor for health due to the high energy density
69 and the possible formation of toxic compounds. The origin and oil composition can influence
70 the food oil absorption during frying process, temperature and frying time, type of food,
71 porosity, and pretreatments applied (Alvis *et al.*, 2015). During the frying occur, reactions

72 responsible for taste, color, and texture and also unhealthy components formation such as polar
73 compounds (Jimenez *et al.*, 2017). During repeated use of the frying oil, it degrades and
74 changes its composition, generating a mixture of polar compounds that act as wetting agents
75 that reduce the surface tension between water and oil causing an increase in oil absorption.

76 This paper proposed to search technological alternatives for use of Andean potatoes
77 (*solanum tuberosum ssp. andigenum*) genotypes reintroduced in northwestern Argentina to
78 revalue these relegated foods. For this purpose, the objective was to study the application of
79 films elaborated with hydrocolloids and starches extracted from Andean potatoes and to study
80 their application in deep frying of chips in order to diminish the oil absorption.

81

82 **2. MATERIALS AND METHODS**

83

84 **a. Materials**

85 Starch was extracted from Runa Andean potato variety (30% amylose content) (Calliope *et*
86 *al.*, 2019), carboxymethyl-cellulose (CMC) (food grade, Rettenmaier & Sohne GMBH + Co),
87 calcium chloride and ascorbic acid (food grade), were used for coatings formulation. Waycha
88 Andean variety and Spunta high -consumption commercial potato- were used for frying assays.
89 Commercial potato chips were used as control. Sunflower oil and blend oil (sunflower/soybean)
90 were used for frying.

91 **b. A) Formulation and properties of films**

92

93 *Films formulation*

94 The components were used as aqueous solutions in the following concentrations: S (1%);
95 S/CMC (1:05; 1.5%) and CMC (1%); 13 g of each formulation were poured onto 10 cm

96 diameter polystyrene plates and allowed to dry 6 h in forced flow oven at 35 °C and then
97 maintained 15 h at 53% of relative humidity and room temperature.

98 *Thickness*

99 The film thickness was measured with an analog micrometer (Digimes, Argentina,
100 sensibility 0.0001mm); the measurement was evaluated with the average of 5 different points of
101 the film.

102 *Water Vapour Permeability (WVP)*

103 The WVP was determined gravimetrically according to the standard method ASTM E96
104 (2000). Films were conditioned 48 h in a desiccator at 25 °C and 53% relative humidity (RH)
105 using Mg (NO₃)₂ supersaturated solution. The measurements of WPV (quadruplicate) were
106 performed according to Slavutsky and Bertuzzi (2015).

107 The water vapour transmission rate (WVTR) was calculated from equations 1 and 2

$$108 \quad WVTR = \frac{G}{A} \quad (1) \qquad P = cte. \frac{WVTR \cdot l}{P_{w0} - P_{w2}} \quad (2)$$

109 Where: G: slope of linear regression; A: area of the exposed film; P: water vapour permeability;
110 l: film thickness; p_{w0}: partial pressure of water vapour in the air on the surface of distilled
111 water; p_{w2}: partial pressure of water vapour on the surface of the film outside the cup; cte:
112 constant to satisfy unit conversion.

113 *Solubility*

114 It was measured as a percentage of dry matter of the film solubilized in water for 24 h
115 immersion. The samples, previously dried in an oven at 105 °C, were weighed (1 g) and placed
116 in a beaker with 50 mL of distilled water at 30 °C, with constant stirring. The non-solubilized
117 material was then separated by centrifugation (Sigma 4K10, Germany) and dried to determine
118 the weight of the dry matter. The tests were performed in triplicate and the solubility was
119 calculated as follows:

$$120 \quad \text{solubility} = \frac{\text{Initial dry weight} - \text{final dry weight}}{\text{Initial dry weight}} \times 100 \quad (3)$$

121 *Sorption Isotherms*

122 The films were cut in pieces of approximately 2 cm² and placed 48 h in a desiccator with
123 P2O5. Then, they were placed in containers with controlled relative humidity using different
124 supersaturated saline solutions (range of aw: 0.10 to 0.90) (Spiess and Wolf, 1983). The weight
125 of the samples was recorded until the difference between the two consecutive weighing was less
126 than 1 mg. Absorption tests were performed in triplicate at each aw. The data obtained were
127 adjusted by the sorption model of BET (Eq. 4).

$$128 \quad w_e = \frac{w_0 \cdot C \cdot a_w}{(1 - a_w) \cdot (1 + (C - 1) a_w)} \quad (4)$$

129 Where w_e is the equilibrium moisture content (g water/100g dry film), w_0 is the moisture
130 content in monolayer (g water/100g dry film) and C is adsorption constant of the first layer
131 dependent on temperature. The quality of the fit was assessed through R².

132 *Colour*

133 The films colour was determined with a colorimeter (Colorquest XE Hunter Lab, USA)
134 versus a standard film (L=94; a=-0.11 and b=3.2). All measurements were performed in
135 triplicate. Total colour difference (ΔE), was calculated as:

$$136 \quad \Delta E = ((L_{\text{standar}} - L_{\text{sample}})^2 + (a_{\text{standar}} - a_{\text{sample}})^2 + (b_{\text{standar}} - b_{\text{sample}})^2)^{1/2} \quad (5)$$

137 *Contact angle*

138 To evaluate the wettability of the oils to the different surfaces, the contact angles were
139 measured using a goniometer (Standard Goniometer with DROP image model 200-00, Ramé-
140 Hart Instrument Co., USA). The oil (10 μ L) was dropped on the surface of the film, using a
141 micro-syringe. The contact angle was measured in 5 points on each film (Zdanowicz and
142 Johansson, 2016). Each analysis was performed in sextuplicate at 25 °C.

143 **c. B) Potato Chips**

144 *Analysis of the raw material*

145 Reducing sugar content was determined by the spectrophotometric method of 3.5
146 dinitrosalicylic acid; absorbance at 540 nm was measured (Mapada, model UV 6300 PC).
147 Glucose p.a. (Merck) for the standard curve was used.

148 *Blanching treatments*

149 The potatoes were washed and cut without peeling in the form of chips (2.5 mm thick);
150 then three scalding processes were applied: 1) water boiling for 5 min, 2) aqueous solution of
151 calcium chloride (0.5%), and 3) ascorbic acid solution (1%), with the same temperature/time
152 conditions.

153 *Coating application*

154 After blanching the potatoes were drained on absorbent paper and immediately immersed
155 in solutions described in section 2.2.1., at 25 °C for 2 min, then the potatoes were drained and
156 the surface moisture was removed in a convection oven at 40 °C for 20 min. An uncoated
157 sample was used as control.

158 *Frying Process*

159 The potatoes (150 ± 5 g) were fried in 3 L of oil in a domestic fryer without reposition. The
160 temperature/time conditions were 180 ± 10 °C/3 min. The fried chips were drained 2 min in the
161 fryer basket and stored 24 h until analysis. The oil absorption of chips with different blanching
162 and coatings was studied in the first frying cycle. To determine the behaviour of the oils
163 concerning the tested coatings, 40 frying cycles were performed in each type of oil; 10 g of oil
164 was taken in cycles 1, 20, 40, and stored at -20 °C until analysis.

165 *Chip evaluation*

166 In the first frying cycle, the lipid content (AOAC 920.39) and moisture/solid matter
167 (AOAC 925.09) were determined.

168 To calculate the reduction in oil absorption (% ROA, Equation 6), a commercial potato
169 chip was taken as a reference.

170
$$\% RAA = 100 - \left(\frac{\%LChA \times 100}{\%LChC} \right) \quad (6)$$

171 % LChA: percentage of lipids of Andean potato chips, scalded with coatings

172 % LChC: percentage of lipids of commercial potato chips, label value: 30.4 g/100 g
173 potato.

174 *Colour*

175 It was measured by a Colorimeter (Colorquest XE Hunter Lab, USA). The average of 5
176 readings was calculated. A chip without frying was taken as reference. The measure of colour
177 change was evaluated according to Equation 5. Where $L^*a^*b^*$ standard were the values for fresh
178 potato and $L^*a^*b^*$ sample the values of fried chips.

179 *Sensory evaluation*

180 Chips samples from the first frying cycle in the two types of oil were used to carry out the
181 sensory analysis with 48 untrained consumers. Six sensory attributes were evaluated using a 5-
182 point hedonic scale, where 5 corresponded to the maximum score.

183 **d. C) Analysis of fresh and used oils**

184 *Fatty acid composition*

185 Fatty acid methyl esters (FAMES) were prepared according to IUPAC 2.301. The FA were
186 quantified in a gas chromatograph model 2014 (Shimadzu, Japan) equipped with column SP
187 2560 (100 mm x 0.25 mm). A mixture of FAME (Supelco FAME Mix C4-C24 18919) was
188 employed as standard.

189 *Calculated oxidizability (Cox)*

190 The Cox value of oils was calculated by the percentage of unsaturated C18 fatty acids,
191 applying the equation 7, proposed by Rossi *et al.* (2013):

$$192 \text{ Cox} = [1(\text{oleic acid \%}) + 10.3(\text{linoleic acid \%}) + 21.6(\text{linolenic acid \%})]/100 \quad (7)$$

193 *Tocopherols*

194 They were determined by AOCS Method Ce8-89. A chromatograph (Shimadzu model 20,
195 Japan) was used, with a fluorescence detector, a Phenomenex C18 silica column (250 × 4.6 mm,
196 5.0 μm); the mobile phase was acetonitrile, methanol, water with phosphoric acid and
197 isopropanol (the flow rate was kept constant at 1.0 mL/min). Tocopherol isomers (α-, β-, γ-, δ-)
198 were identified using standards (Sigma Aldrich). Isopropanol (1 mL) was added to the oil
199 sample (30 mL) and then injected into the HPLC equipment.

200 *Polar Compounds*

201 They were determined by adsorption chromatography (IUPAC 2.507). Stationary phase
202 Silica gel (Merck) particle size 0.063-0.200 mm was used, as the mobile phase for non-polar
203 compounds ethyl ether/petroleum ether 10:90 v/v, and diethyl ether for the polar fraction. Polar
204 compounds were quantified, such as the difference between the initial mass of oil and the eluted
205 non-polar fraction.

206 e. D) Statistical Analysis

207 The means were analysed by analysis of Variance. Differences between samples were
208 analysed applying Tukey test. Differences between treatments ($p < 0.05$) were considered
209 significant. To determine the influence of the scalding solution, coating addition, and oil
210 absorption during the frying, the 3-way interaction method was applied. Software Infostat 2017
211 and Graph pad prism version 5.01 were used.

212

213 3. RESULTS AND DISCUSSION

214

215 a. Films

216

217 The average results of the measured parameters are shown in Table 1. The water vapor
218 permeability (WVP) of edible films should be as low as possible to control the transfer of
219 moisture between the food and the surrounding atmosphere. S/CMC films showed a
220 significantly lower WVP value than CMC films which would be related to the thickness of the
221 film and the intrinsic characteristic of each material (Basiak *et al.*, 2017). The permeation
222 phenomenon depends on three stages: adsorption, diffusion, and desorption. The diffusion stage
223 depends on the thickness of the film, while the other two are independent of it. Starch has a
224 semi-crystalline structure; the amylose is capable of forming a tortuous path that decreases the

225 diffusion of water through the film. The different behavior of starch and CMC films may be due
226 to this. No significant difference was observed between S/CMC and S films. This indicated that
227 in the composite films, permeability was related to the presence of starch. Almasi *et al.* (2010)
228 postulated that starch forms hydrogen bonds with the hydroxyl groups of the CMC, and this
229 strong structure could reduce the diffusion of water in the material. Therefore, the addition of
230 CMC would improve the water-resistance of the starch matrix. The water solubility of the S
231 films was high and higher than that reported by Basiak *et al.* (2017). This would be related to
232 the higher amylose content of the starch of potato var. Runa used. The combination with CMC
233 produced a decrease in solubility of approximately 50%. This behavior was also observed by
234 Ghanbarzadeh *et al.* (2010). The solubility of the S/CMC composite films was related to the
235 behavior observed in the permeability study.

236 Figure 1 shows the absorption isotherms obtained at 25 °C for the three formulations.
237 Curves were typical of polymers with affinity for water. The curves showed a slight relative
238 slope at low a_w values, while they were exponential at a_w values greater than 0.60. Other
239 authors reported similar behavior in starch-based films (Slavutsky *et al.*, 2015). Experimental
240 data indicated that the CMC film had the lowest water absorption, while the S/CMC film had
241 similar behavior to the S film. The adjustment parameters obtained with the BET model
242 (Equation 4) of each film showed that the constant c influences the sigmoidal shape of the
243 isotherms, particularly in the low range of a_w . The values of c would indicate that the moisture
244 absorption of the matrix studied could occur more easily in the upper layers than in the
245 monolayer. These results showed that the stage that controls the water permeability of the S and
246 S/CMC films was diffusion and not the adsorption / desorption of water, so the influence of
247 thickness would be less significant. This confirmed that the composite films have lower
248 solubility and water vapor permeability.

249 Starch films showed greater opalescence, which could be explained by their greater
250 thickness, which was probably due to the higher amylose content (Basiak *et al.*, 2017).
251 Regarding color, parameter L showed significant differences being lower for the S film. In

252 parameter a, all the films had a greenish hue being higher for the CMC and in b the films hue
253 was yellow with no significant differences among them. In general, the optical values of the
254 films presented good transparency. Ghanbarzadeh *et al.* (2010) observed in a study conducted
255 with S/CMC composite films that the CMC aggregate produced clearer films. These changes
256 can be further described with the ΔE function which had a significant decrease ($p < 0.05$) when
257 CMC was added.

258 The contact angle was used as an indicator of the degree of interaction between the oils and
259 the surface of the films. The three films had surfaces with moisturizing properties confirming
260 their hydrophilicity. An increase in the contact angle between the oil and the film indicates a
261 lower affinity between both materials. The highest value (20.6°) was obtained for S/CMC films
262 in blend oil. This is according to the frying experiments (Table 2 and Fig. 2), in which the chips
263 potatoes with the lowest oil content were those covered by S/CMC. The contact angles of the
264 CMC and S films were smaller indicating surfaces with greater affinity to oils.

265 **b. Chips**

266 Andean potato var. Waycha, and Spunta contained 26.38 ± 0.31 and $15.00 \pm 0.20\%$ of dry
267 matter (DM) and 0.18 ± 0.06 and $0.28 \pm 0.08\%$ of reducing sugars (RS) respectively. The
268 appropriate value for the production of potato chips is 25% DM and 0.2% RS (Colman *et al.*,
269 2009). The Waycha potato met both characteristics. The DM content decreased with all the
270 bleaching treatments. Significant differences in DM content were found between chips scalded
271 with water (observed decrease from 26.4 to 23.8%), with respect to treatments with ascorbic
272 acid (22.6%) and calcium chloride (22.1%). This could be due to different migrations of soluble
273 potato compounds to the bleaching medium. The oil content was significantly affected ($p > 0.05$)
274 by kind of scalding and the type of coating used (Table 2). The oils employed did not
275 significantly influence absorption. Figure 2 a and b, shows the percentages of oil reduction
276 according to the treatments applied taking as reference a commercial potato (label value 30.4 g
277 oil/100g). Control samples scalded in water and without coating showed a slight reduction in oil
278 content (16.1-18.5%), compared to those scalded in calcium chloride with S/CMC coating (39.1-

279 40.1%). These results indicated that calcium chloride stabilized the structure of the tissue during
280 the frying process. The texture of the potato depends on the presence of pectinic substances,
281 which are part of the intercellular material. Pectinolytic enzymes produce free carboxylic
282 groups, which can react with divalent ions such as calcium and magnesium, creating more rigid
283 structures and increasing firmness. The formation of these calcium/pectin complexes causes the
284 reaffirmation of the cell wall and increase the stiffness of the medium of the lamina-cell wall
285 (Hernandez *et al.*, 2014), and therefore its structure better resists the frying process. Table 2 and
286 Figure 2 a) and b), shows that the coatings influenced the decrease in oil absorption. The results
287 showed significant differences ($p > 0.05$) between treatments. The one with the greatest effect
288 was the coating of S/CMC ($39.5\% \pm 0.7$) combined with scalding in calcium chloride, while the
289 one with the least effect was the control without coating and scalding in water ($17.3\% \pm 1.7$).
290 This reduction in oil absorption could be attributed to the fact that the starch undergoes
291 structural changes in which the crystals of amylose and amylopectin are reorganized and
292 promote the formation of a gel that functions as a barrier to the entry of oil (Hasbún *et al.*,
293 2009). Varela and Fiszman (2011) and Freitas *et al.* (2009) postulated that CMC increases water
294 retention capacity and, consequently, prevents the replacement of water with oil. In addition,
295 since the polymer is hydrophilic, it forms a thin layer on the surface of the food that acts as a
296 barrier to the incorporation of oil. Ali *et al.* (2012) observed that CMC increases surface
297 tension, which facilitates the draining of surface oil. Likewise, calcium chloride is a cross-
298 linking agent that forms a fine network, which prevents the migration of oil to the potato during
299 the frying process (Hasbún *et al.*, 2009).

300 The results showed that the type of oil had no significant influence on absorption ($p > 0.05$).

301 Table 3 shows colour change during 40 frying cycles. In the parameters L^*a^* and b^* within the
302 frying cycles with coatings, there were significant differences in L. It was observed with L value
303 for chips with film, as the cycles pass, the product is darkening. For parameter a, the potatoes
304 with coatings in both oils showed significant differences between the first and the last frying,
305 with an increase in redness in last cycles, possibly due to the effect of the coating. While in

306 parameter b, there was greater variation, with the exception of chips without coating fried in
307 blend oil. In the other treatments, the intensity of the yellow colour varied, not finding a defined
308 pattern. In all cases, the yellow/gold colour, typical of fried products, was characteristic. The
309 parameter ΔE was used to evaluate the colour change between the different processes tested; no
310 statistical differences were observed between the treatments since the observed variations were
311 very wide. Table 3 shows the average hedonic classification of sensory attributes. Scalded
312 samples with coatings for both types of oils obtained high scores on almost all attributes. The
313 texture and overall acceptability of these samples received higher scores compared to the
314 uncoated sample fried in sunflower oil. The uncoated sample fried in sunflower oil had the
315 lowest score for the texture. While the uncoated sample fried in blend oil obtained the lowest
316 taste score. The flavours described by consumers were "bitter, burned and taste oily." These
317 indicated that the coating contributed to eliminating these perceptions. Garcia *et al.* (2002)
318 reported that edible coatings with CMC affected the colour of potato chips samples but did not
319 change the texture characteristic. In this study, the coating with S/CMC did not affect the colour
320 and gave a better "crunchy" texture to the fried chips in both types of oils used.

321 The sample with coating and fried in sunflower had a significantly greater acceptance, the
322 texture being notable for its high crunchiness.

323 During the frying, the oil is exposed to high temperatures in presence of air and humidity,
324 which generates oxidation, hydrolysis and polymerization reactions (Rimac-Brnčić *et al.*,
325 2004). This is why the oil usage times are reflected in the colour and changes in its composition
326 (Navas *et al.*, 2007).

327 **c. Analysis of fresh and used oils**

328 Figure 3 shows the fatty acid fractions: saturated (SFA), monounsaturated (MFA),
329 polyunsaturated (PFA), and Trans (TFA) of fresh oils and after being used in 40 frying cycles.
330 The SO had a lower content of polyunsaturated fatty acids (5.6%) than the OB (7.03); so, its
331 calculated oxidability was different too. There was an increase in the SFA content in OB with
332 increasing frying cycles, and in TFA when potato chips were uncoated. In the SO the MFA

333 fraction increased and TFA were generated when uncoated chips were fried. This probably is
334 related to the temperature/time of oil use. Cis to Trans isomerization begins when the frying
335 temperature is higher than 150 °C (Bhardwaj *et al.*, 2016). However, the TFA content of oils
336 with 40 frying cycles was less than the maximum established (5%) by Argentine legislation.

337 Tocopherols are natural components of oils with a protective effect against oil oxidation.
338 During the processing, storage, and use of oils there are partial losses of these components.
339 Table 3 shows the content of the different types of tocopherols of SO and OB fresh and after
340 being used in 40 frying cycles. The tocopherols content was significantly higher in the OB than
341 in SO, both fresh. When both were used in 40 frying the total tocopherols content decreased
342 significantly; this was less noticeable when covered chips were frying. Rossi *et al.* (2017)
343 reported that frying potato chips in eight different types of vegetable oils, including sunflower
344 pure and mixture, tocopherols decreased rapidly after the third hour of continuous use of the
345 oils. In this study, the frying cycles exceeded 3 hours of oil use. The mechanism of the reaction
346 to eliminate radicals of tocopherols requires that they lose their mobile hydrogen atom in the
347 hydroxyl group, forming more stable free radicals than fatty acids, it follows that rapid
348 oxidation of tocopherols corresponds to greater antioxidant power. However, if all other
349 fractions are taken into account, it could be assumed that, in the case of oils containing higher
350 levels of PFA, the double bond that determines unsaturation competes with tocopherols as
351 substrates for oxidation, determining a less rapid decrease of these antioxidants. In contrast, in
352 the case of low polyunsaturated oils, tocopherols would constitute the substrates that react more
353 easily with oxygen. In fact, it was reported in the literature that, in the propagation phase of the
354 reaction, peroxy fatty acid-free radicals preferentially react with the phenolic hydrogen of the
355 tocopherol molecule (Rossi *et al.*, 2017). The PC concentrations of fresh oils were for OB $5.9 \pm$
356 3.7% and $5.3 \pm 2.5\%$ for SO, which are within the reported values (Ramírez Botero *et al.*,
357 2012). After 40 frying, the PC content increased to 13.0 ± 0.7 and 10.5 ± 1.4 in SO and to 14.5
358 ± 1.4 and $11.9 \pm 0.5\%$ in OB when chips with and without coating were fried, respectively. The
359 higher PC content in the frying 40 in OB with film ($p>0.5$), is probably due to the oil

360 composition since SO has greater stability against oxidation due to its lower degree of
361 unsaturation.

362 These results indicated that under the conditions used in this study, CP content was not
363 generated in concentrations higher than the limits established by countries that have their
364 content legislated. For example, Spain, France, Italy, and Chile accept as maximum value 25%
365 of polar compounds content, while Germany accepts 24% and Austria and Switzerland up to
366 27% (Suaterna Hurtado, 2009).

367 **4. CONCLUSIONS**

368 The Waicha variety was suitable to obtain potato chips because dry matter and reducing
369 sugars content meet the conditions established for that use.

370 The coating formulated with starch extracted of Runa variety combined with CMC, applied
371 to chips scalded in calcium chloride contributed to decreasing the oil absorption during the
372 frying process. This formulation delayed oil oxidation and decreased loss of tocopherols during
373 the frying process. It also contributed to improving the physical and sensory characteristics of
374 the final product, which had high acceptability by consumers.

375 These results confirm two technological applications for Andean potatoes that can be used to
376 contribute to a healthy diet. Additionally, both materials used in the preparation of edible
377 coatings are ecological and respectful of the environment since they will contribute to the
378 recycling of waste from food and other industries.

379

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383

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Table 1. Parameters measured in the developed films

Parameters		S	CMC	S/CMC
Thickness (μm) 10^{-5}		6.42 ± 0.20^c	4.86 ± 0.07^a	5.63 ± 0.08^b
Barrier properties water vapour permeability ($10^{-10} \text{ g m}^{-1} \text{ s}^{-1} \text{ Pa}^{-1}$)		2.12 ± 0.23^{ab}	2.73 ± 0.21^b	1.94 ± 0.29^a
Solubility in water (%)		26.24 ± 6.66^b	9.25 ± 2.23^a	13.06 ± 1.34^a
BET	w_0	3.09 ± 0.13^b	2.00 ± 0.10^a	3.23 ± 0.15^b
	c	15.87 ± 2.58^b	5.01 ± 4.50^a	6.15 ± 5.36^a
	R^2	0.9292	0.9388	0.9240
Colour	L	92.98 ± 0.09^a	93.32 ± 0.06^b	93.24 ± 0.17^b
	a	-0.33 ± 0.03^c	-0.57 ± 0.02^a	-0.52 ± 0.02^b
	b	3.60 ± 0.19^a	3.90 ± 0.27^a	3.47 ± 0.87^a
	AE	1.11 ± 0.15^b	1.08 ± 0.22^{ab}	0.91 ± 0.15^a
Surface properties	θ (OB)	15.80 ± 1.20^a	18.90 ± 3.90^{ab}	20.60 ± 3.90^b
	θ (SO)	18.80 ± 4.10^a	16.80 ± 1.80^a	19.10 ± 3.10^a

486 w_0 : monolayer moisture content, c: constant related to the heat sorption for monolayer,
487 R^2 : coefficient of determination. θ (OB): contact angle of blend oil; θ (SO): contact
488 angle of sunflower oil; colour parameters (L, a, b). Values having the same letter for a
489 parameter within the same row are not significantly different at p level > 0.05.

490 S: starch; CMC: carboxymethyl-cellulose; S/CMC: combination starch with carboxymethyl-
491 cellulose

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