Asian Jr. of Microbiol. Biotech. Env. Sc. Vol. 25, No. (1): 2023: 1-8 © Global Science Publications

ISSN-0972-3005

NEW ANTIFUNGAL AGENTS: SYNTHESIS, CHARACTERIZATION AND BIOLOGICAL ACTIVITY OF SOME MANNICH BASES DERIVED FROM 2-MERCAPTOBENZOTHIAZOLE

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(Received 1 March, 2022; Accepted 16 September, 2022)

Key words: Mannich reaction, 2-mercaptobenzothiazole, Antifungal activity, Agar well diffusion

Abstract– A series of new *3-[(amino)methyl]-1,3-benzothiazole-2-thione* **4(a-1)** were synthesized by Mannich reaction from formaldehyde, aliphatic or aromatic amines and 2-mercaptobenzothiazole. All synthesized compounds were in good agreement with elemental and spectral data (¹H NMR and mass spectroscopy) and were evaluated for *in vitro* antifungal activity by agar well diffusion method. Fluconazole was included in the assays as a commercially available reference compound. Four fungal strains were used to perform the study: *Cladosporium cladosporioides, Aspergillus niger* ATCC 16404, *Botrytis sp.* and *Rhizopus sp.* Compounds **4a-c, 4e-g, 4i** and **4l** were found to be active against the four moulds mentioned, due to the presence of inhibitory halo. **4h** was active in case of *C. cladosporioides*. Compounds **4d, 4j** and **4k** did not show inhibition against fungi used for the test. Fluconazole showed biological activity against *Aspergillus niger* ATCC 16404 and *Botrytis sp.* Therefore, 8 of the 12 synthesized substances would be considered as promising products to the treatment of fungal diseases.

INTRODUCTION

Antimicrobial resistance (AMR) represents one of the greatest threats to global public health. Resistance becomes a public health problem when resistant strains compromise the effectiveness of prescribed antibiotic therapy. The resistant microorganisms make up a large reservoir of genes that can potentially transfer resistance to human, animal and environmental pathogens. Using the whole genome sequencing, experts can identify resistant genes in bacteria, rather than current phenotypic methods that test bacteria for resistance to specific antibiotics. This, does not only have the potential to predict AMR more efficiently, it also generates a wealth of data that can be used for other epidemiological studies and analyzes. Also, from

genomics it is feasible to identify new pharmacological targets and contribute to the design of new antimicrobial agents with a defined structure and a more efficient mode of action (Walsh *et al.*, 2000; Ritter and Wong, 2001).

As the incidence and prevalence of invasive fungal infections has increased dramatically in the last 20 years, the development of new, more selective and less toxic antifungal drugs has become imperative (Maertens and Boogaerts, 2005; Datry and Bart-Delabesse, 2006). In this direction, the scientific community has explored numerous selection methods to evaluate different natural or synthetic drugs. In this sense, 2-Mercaptobenzothiazole (2-MBT) constitutes an important pharmacophore, possessing several pharmacological functions. 2-MBT and its

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derivatives have shown to have biological activity as potent antimicrobials and antifungals agents (Quiroga et al., 2002; Kok et al., 2007; Grover and Moore, 1962), neurotransmission inhibitors (Benavides et al., 1985; Mizoule et al., 1985) and as antitumor agents (Leong et al., 2004; Yldiz-Oren et al., 2004; Lockhart et al., 2005). The molecule of 2-MBT contain extended π -delocalized systems which are capable of binding to complex molecules such as Deoxyribonucleic Acid (DNA) via π – π interactions and consequently exhibit interesting biological properties, for example, for the treatments of cancer and infectious diseases caused by certain microorganisms such as bacteria, fungi and parasites (Triphati and Mishra, 2007; Pavlovic et al., 2007; Susithra et al., 2022).

The present study was undertaken in order to synthesize and to explore the biological activity of some new compounds having this heterocyclic ring. Therefore, the present work reports the synthesis of some Mannich's bases of 2-MBT and their *in vitro* antifungal activity as a part of our program aimed at the development of new heterocyclic compounds with potential antifungal activities.

MATERIALS AND METHODS

General materials

All the chemicals and reagents used in synthetic procedures were of reagent grade and used without further puriûcation. These were used as received from Sigma-Aldrich Co., Inc., Saint Louis, USA. The purity of the synthesized compounds was ascertained by Thin Layer Chromatography (TLC) on silica gel GF254 with mixtures of hexane/ethyl acetate (3:1) as solvent systems and using panisaldehyde as detecting agent. Melting points were determined on a Büchi 510 micro melting point determination apparatus in open capillary tubes and are uncorrected. High-resolution mass spectrometry (HRMS) data were obtained on a Bruker micro QTOF-Q11 mass spectrometer equipped with an electrospray ionization (ESI). Proton ¹H NMR spectra were recorded on Bruker

Avance Ultra Shield Spectrometer (Bruker, 300 MHz, Russ, Germany) using Chloroform-*d* (CDCl₃) and Dimethyl sulfoxide-*d6* (DMSO-*d6*) as a solvents and tetramethyl silane (TMS) as an internal standard. Chemical shift value is expressed in delta (δ) parts per million (ppm). Antimicrobial tests were done using potato dextrose agar (PDA) and these were purchased from Britania Lab, Argentina. Microbial strains employed were *Cladosporium cladosporioides*, *Aspergillus niger* ATCC 16404, *Botrytis sp.* and *Rhizopus sp.*, all from the collection of the laboratory of IDIC-UCP Goya, Corrientes, Argentina. Fluconazole (Vannier) was used as a commercially available reference compound.

Synthetic procedures

General procedure for the synthesis of 3-[(amino)methyl]-1,3-benzothiazole-2-thione **compounds 4(a-1):** In a round-bottomed flask, aqueous formaldehyde (1) (5 ml of 37% solution; 0.05 mole) was added dropwise to aliphatic or aromatic amines 2(a-1) (0.05 mole) under stirring at 30 °C (to aliphatic amines at 5 °C). After formation of a solid white precipitate and with vigorous mixing 2-MBT (3) (8.2 g; 0.05 mole) diluted in acetone (2 mL) was added in small amounts under continuous stirring. Behind 10 minutes, the mixture was carefully heated at 55 °C whereas obtained a vellow solution. After standing for 10 min the solution was cooled to 5 °C and diluted with water (60 ml) until obtain needles of product crystallized. Finally, the crystals were filtered and were purified by crystallization from ethanol (Figure 1).

By adopting this procedure, 12 compounds were synthesized. Synthetic pathway for preparation of title compounds is shown in Figure 1. The aliphatic and aromatic amines **2(a-1)**, *pka* value of amines and yields for **4(a-1)** products are summarized in Table 1.

Spectral data of synthesized compounds

3-(morpholin-4-ylmethyl)-1,3-benzothiazole-2-thione (4a): White solid (25%). Mp = 148-149 °C (Lit. 149-150). ¹H NMR [CDCl₃, 300MHz]: 5 = 2.63 (t, 4H, CH₂), 3.69 (t, 4H, CH₂), 4.91 (s, 2H, CH₂), 7.33-7.42

Fig. 1. Synthesis of 3-[(amino)methyl]-1,3-benzothiazole-2-thione 4(a-1)

Table 1. Amines, *pKa* values of amines **2(a-1)** and yields for **4(a-1)** compounds

Entry	Amines	рКа ^а	Products	Yields (%) ^b
1	HN O	8.49	S S N CH_2 N O Aa	25
2	HN 2b	2.90	$S \longrightarrow S$ $N-CH_2 N$ 4b	50
3	HN_NH	9.73	$S \longrightarrow S$ $N-CH_2 N$ NH	80
4	H_2N \longrightarrow NO_2	14.30	$\begin{array}{c c} S & H \\ N-CH_2 & N-CH_2 & N-CH_2 \end{array}$	87
5	H_2N — CF_3 CI $\mathbf{2e}$	11.50	$\begin{array}{c c} S & S & H & Cl \\ N-CH_2 & N & CI \end{array}$ $\begin{array}{c} CI & \\ \mathbf{4e} & \\ \end{array}$	30
6	$H_2N \longrightarrow F$ F $\mathbf{2f}$	11.37	$\begin{array}{c c} S & F \\ \hline & N-CH_2 \\ \hline & F \\ \end{array}$	61
7	H_2N F F F F F F	14.28	$ \begin{array}{c c} \mathbf{4f} \\ \hline S \\ S \\ N-CH_2 \\ N \end{array} $ $ \begin{array}{c c} F \\ F \\ F \end{array} $ $ \begin{array}{c c} F \\ F \\ F \end{array} $	84
8	H_2N Br $2h$	11.50	$\begin{array}{c c} S & S & H \\ N-CH_2 & N \end{array}$ $\begin{array}{c c} Br \end{array}$	96
9	H_2N O- CH_3	8.70	$\begin{array}{c c} S & H \\ \hline N-CH_2 & N- \\ \end{array} \\ \begin{array}{c} O-CH_3 \\ \end{array}$	74
10	H_2N H_2N $2j$	9.54	4i S S S H N-CH ₂ N N-CH ₂ N H	41

Table 1. Continued ...

Entry	Amines	рКа ^а	Products Yields (%) ¹	,
11	$\begin{array}{c} O \\ H_2N \longrightarrow \begin{array}{c} O \\ \vdots \\ S - OH \\ O \end{array} \\ \mathbf{2k} \end{array}$	10.36	4j S S S H O S O S O O O O O O O O	51
12	H_2N H_3C CH_3 NH_2 $2I$	11.38	S S H CH_3 H_3C NH_2 AI	68

a. pKa values of aliphatic and aromatic amines **2(a-1)** at 25 °C b. Isolated yields.

(m, 2H, CH), 7.46-7.74 (m, 2H, CH). 13 C NMR [75 MHz]: \ddot{a} = 52.12 (CH₂), 66.73 (CH₂), 71.39 (CH₂), 113.84 (CH), 123.65 (CH), 124.61 (CH), 125.85 (C), 127.16 (CH), 142.36 (C), 189.17 (C=S). HRMS (ESI): Anal. Calcd. for [C₁₂H₁₄N₂OS₂]: 266.0548; found: 266.0537.

3-(piperidin-1-ylmethyl)-1,3-benzothiazole-2-thione (**4b**): White solid. Mp = 159-161 °C. ¹H NMR [CDCl₃, 300MHz]: 5 = 1.18-1.22 (m, 2H, CH₂), 1.59 (q, 4H, CH₂), 2.63 (q, 4H, CH₂), 4.91 (s, 2H, CH₂), 7.32-7.43 (m, 2H, CH), 7.49-7.74 (m, 2H, CH). ¹³C NMR [75 MHz]: $\ddot{a} = 24.61$ (CH₂), 26.15 (CH₂), 53.34 (CH₂), 71.01 (CH₂), 113.80 (CH), 123.62 (CH), 124.59 (CH), 125.35 (C), 127.11 (CH), 142.38 (C), 189.11 (C=S). HRMS (ESI¹): Anal. Calcd. for [C₁₃H₁₆N₂S₂]⁻: 264.0755; found: 264.0741.

3-(piperazin-1-ylmethyl)-1,3-benzothiazole-2-thione (**4c**): White solid. Mp = nd (not determined, decomposes above 221 $^{\circ}$ C). 1 H NMR [CDCl₃, 300MHz]: 5 = 2.02-2.06 (m, 1H, NH), 2.69-2.74 (m, 8H, CH₂), 4.91 (s, 2H, CH₂), 7.32-7.43 (m, 2H, CH), 7.49-7.74 (m, 2H, CH). 13 C NMR [75 MHz]: δ = 45.29 (CH₂), 51.53 (CH₂), 70.69 (CH₂), 113.87 (CH), 123.69 (CH), 124.58 (CH), 125.81 (C), 127.18 (CH), 142.39 (C), 189.15 (C=S). HRMS (ESI⁻): Anal. Calcd. for [C₁₂H₁₆N₃S₂]⁻: 265.0707; found: 265.0711.

3-[(4-nitroanilino)methyl]-1,3-benzothiazole-2-thione (4d): White solid (87%). Mp = 197-199 °C. ¹H NMR

[DMSO-d6, 300MHz]: 5 = 5.05 (s, 2H, CH₂), 7.06-7.11 (m, 1H, NH), 7.32-7.39 (m, 2H, CH), 7.51-7.53 (m, 2H, CH), 7.81-7.83 (m, 4H, CH). 13 C NMR [75 MHz]: δ = 60.75 (CH₂), 112.93 (CH), 116.71 (CH), 122.74 (CH), 123.98 (CH), 124.58 (CH), 125.25 (C), 127.18 (CH), 143.31 (C), 145.39 (C), 189.92 (C=S). HRMS (ESI-): Anal.Calcd. for [C₁₄H₁₁N₃O₂S₂]-: 317.0293; found: 317.0301.

3-{[2,6-dichloro-4-(trifluoromethyl)anilino]methyl]-1,3-benzothiazole-2-thione (4e): White solid (30%). Mp = 186-187 °C. ¹H NMR [DMSO-d6, 300MHz]: 5 = 5.05 (s, 2H, CH₂), 7.38-7.41 (m, 2H, CH), 7.48-7.56 (m, 2H, CH), 7.55-7.57 (m, 1H, NH), 7.83-7.87 (m, 2H, CH). ¹9F NMR [300 MHz] δ = -63.22 (s, 2F). ¹³C NMR [75 MHz]: δ = 60.73 (CH₂), 75.4 (C), 113.4 (C), (116.5, 120.1, 123.7, 127.3, q, ¹J = 273.58Hz -CF₃), 126.2 (CH), (133.8, 134.2, 134.7, 134.9, q, ²J = 34.59Hz -CF₃), 136.7(C), 142.9 (CH), 151.4(C), 189.88 (C=S). HRMS (ESI⁻): Anal.Calcd. for [C₁₅H₉Cl₂F₃N₂S₂]⁻: 407.9536; found: 407.9521.

3-[(3,4,5-trifluoroanilino)methyl]-1,3-benzothiazole-2-thione (**4f**): White solid (61%). Mp = 187-188 °C. ¹H NMR [DMSO-d6, 300MHz]: 5 = 5.05 (s, 2H, CH₂), 6.38-6.41 (m, 2H, CH), 7.50-7.55 (m, 2H, CH), 7.53-7.57 (m, 1H, NH), 7.84-7.88 (m, 2H, CH). ¹9F NMR [300 MHz] δ = -62.30 (s, 3F). ¹3C NMR [75 MHz]: δ = 60.70 (CH₂), 77.2 (C), 113.6 (C), 121.6 - 125.2 (q, ¹J = 272.58Hz, CF₃; it was not possible to assign the other half of this quartet), 123.9 (CH), 127.2 (CH), 130.4 -

130.8 (q, 2J = 32.0Hz), 140.0 (C), 141.9 (C), 150.0(C), 189.91 (C=S). HRMS (ESI-): Anal .Calcd. for $[C_{15}H_9Cl_2F_3N_2S_2]$: 326.0159; found: 326.0142.

3-[(2,3,4,5,6-pentafluoroanilino)methyl]-1,3-benzothiazole-2-thione (**4g**): White solid (84%). Mp = 169.2 °C. ¹H NMR [DMSO-d6, 300MHz]: 5 = 5.05 (s, 2H, CH₂), 7.42-7.48 (m, 2H, CH), 7.52-7.55 (m, 1H, NH), 7.85-7.88 (m, 2H, CH). ¹°F NMR [300 MHz] δ = -158.95 (t, 2F), -148.58 (t, 1F), -143.24 (d, 2F). ¹³C NMR [75 MHz]: δ = 61.01 (CH₂), 76.4 (C), 113.1 (C), 143.4 (CH), 152.1 (C), 189.90 (C=S). It was not possible to assign the other carbons corresponding to this molecule. HRMS (ESI¹): Anal.Calcd. for [C₁,H₂F₂,N₃S₂]: 361.9971; found: 361.9959.

3-[(2-bromoanilino)methyl]-1,3-benzothiazole-2-thione (**4h**): White solid (96%). Mp = 156-158 °C. ¹H NMR [DMSO-d6, 300MHz]: 5 = 5.05 (s, 2H, CH₂), 6.73-6.77 (m, 1H, CH), 7.01-7.03 (m, 1H, CH), 7.17-7.21 (m, 1H, CH), 7.42-7.48 (m, 2H, CH), 7.52-7.54 (m, 1H, NH), 7.56-7.58 (m, 1H, CH), 7.83-7.87 (m, 2H, CH). ¹³C NMR [75 MHz]: δ = 60.73 (CH₂), 112.91 (CH), 116.73 (CH), 122.76 (CH), 123.95 (CH), 124.56 (CH), 125.23 (C), 127.20 (CH), 143.30 (C), 145.41 (C), 190.01 (C=S). HRMS (ESI-): Anal.Calcd. for [C₁₄H₁₁BrN₂S₂]-: 351.9526; found: 351.9539.

3-[(4-methoxyanilino)methyl]-1,3-benzothiazole-2-thione (4i): White solid (74%). Mp = 182-183 °C. ¹H NMR [DMSO-d6, 300MHz]: 5 = 3.76 (s, 3H, O-CH₃), 5.05 (s, 2H, CH₂), 6.79-6.81 (m, 2H, CH), 7.13-7.15 (m, 1H, NH), 7.22-7.24 (m, 2H, CH), 7.36-7.38 (m, 2H, CH), 7.80-7.82 (m, 2H, CH). ¹³C NMR [75 MHz]: δ = 55.33 (CH₃), 60.49 (CH₂), 112.69 (CH), 116.70 (CH), 122.71 (CH), 123.90 (CH), 124.55 (CH), 125.25 (C), 127.16 (CH), 143.33 (C), 145.37 (C), 189.04 (C=S). HRMS (ESI-): Anal.Calcd. for [C₁₅H₁₄N₂OS₂]-: 302.0548; found: 302.0531.

3-[(2-aminoanilino)methyl]-1,3-benzothiazole-2-thione (**4j**): White solid (41%). Mp = nd (not determined, decomposes above 210 9 C). 1 H NMR [DMSO-d6, 300MHz]: 5 = 4.02 (s, 2H, NH), 5.05 (s, 4H, CH₂), 6.73-6.77 (m, 3H, CH), 7.01-7.03 (m, 1H, CH), 7.38-7.41 (m, 4H, CH), 7.55-7.57 (m, 2H, NH), 7.83-7.87 (m, 4H, CH). 13 C NMR [75 MHz]: δ = 60.31 (CH₂), 112.55 (CH), 116.65 (CH), 122.76 (CH), 123.10 (CH), 124.50 (CH), 125.13 (C), 127.17 (CH), 143.31 (C), 145.39 (C), 189.14 (C=S). HRMS (ESI-): Anal.Calcd. for [C₁₂H₁₈N₄S₄]: 466.0414; found: 466.0422.

 $3-[(1-naphthalene\ sulphonic\ acid-4-amino)methyl]-1,3-benzothiazole-2-thione\ (4k):$ White solid (51%). Mp =

nd (not determined, decomposes above 210 °C). ¹H NMR [DMSO-*d6*, 200MHz]: 5 = 5.05 (s, 2H, CH₂), 7.25-7.34 (m, 3H, CH), 7.39-7.45 (m, 2H, CH), 7.77-7.84 (m, 2H, CH), 7.90-7.92 (m, 1H, CH), 8.08-8.12 (m, 1H, CH), 8.20-8.22 (m, 1H, NH), 8.40-8.42 (m, 1H, CH), 9.26 (s, 1H, OH). ¹³C NMR [75 MHz]: $\delta = 77.05$ (CH₂), 107.71 (CH), 112.55 (CH), 116.65 (CH), 122.41 (CH), 122.76 (CH), 123.10 (CH), 124.50 (CH), 125.32 (C), 125.82 (CH), 126.13 (C), 127.17 (CH), 129.41 (C), 129.91 (C), 130.21 (CH), 143.31 (C), 145.39 (C), 189.14 (C=S). HRMS (ESI-): Anal.Calcd. for [C₁₂H₁₈N₄S₄]: 402.0167; found: 402.0154.

3-[(3,3´-dimethylnaphthidine)methyl]-1,3-benzothiazole-2-thione (41): White solid (68%). Mp = 187-188 °C. ¹H NMR [DMSO-d6, 200MHz]: 5 = 2.27 (s, 3H, CH₃), 2.34 (s, 3H, CH₃), 4.86-4.93 (m, 2H, NH₂), 5.05 (s, 2H, CH₂), 7.22-7.24 (s, 1H, CH), 7.26-7.28 (m, 1H, NH), 7.40-7.46 (m, 6H, CH), 7.80-7.86 (m, 3H, CH), 7.88-7.94 (m, 3H, CH), 8.06-8.08 (m, 1H, CH). 13 C NMR [75 MHz]: δ = 18.12 (CH₃), 18.31 (CH₃), 75.15 (CH₂), 118.71 (C), 116.65 (CH), 122.41 (CH), 122.76 (CH), 123.10 (CH), 124.50 (CH), 125.32 (C), 125.82 (CH), 126.61 (C), 128.55 (CH), 127.17 (CH), 129.41 (C), 129.91 (C), 131.21 (C), 143.31 (C), 139.39 (C), 191.33 (C=S). HRMS (ESI-): Anal.Calcd. for [C₃₀H₃₂N₃S₄]: 491.1490; found: 491.1479.

Antimicrobial evaluation

Agar well diffusion tests: With the porpoise of determine the existence of antimicrobial activity of synthesized compounds 4(a-1), agar well diffusion test was applied (Balouiri et al., 2016, Valgas et al., 2007, Magaldi, et al., 2004, Devillers et al., 1989). Stock solution of every compound had a concentration of 1500 µg/mL, and were prepared with sterile distillated water, dimethyl sulfoxide at 5-20% depending on their solubility and tween 20 at 1%. Seven-day-old cultures of *Cladosporium* cladosporioides, Aspergillus niger ATCC 16404, Botrytis sp. and Rhizopus sp. (PDA, 27 ± 1°C) were used to prepare inoculums at concentration between 0,4 x 10⁴ - 5 x 10⁴ CFU/mL to ensure reproducible values (NCCLS, 2002). Each conidial suspension was quantified with a Neubauer chamber. Inoculum's concentrations were verified by spread plate method, using PDA. These plates were incubated for 48 to 72 hours, at 27°C ± 1°C, until colonies could be count. For the assay, 20 mL of PDA at 45 °C was mixed with 1 mL of inoculum and poured in a 90 mm Petri dish. 5 wells of 8 mm were cut in each plate and filled with 100 µL of the 2-MBT derivatives

stock solutions (150 µg of pure compound). Wells of a control plate were filled with water and water + DMSO at 20% with Tween 20 at 1% (control A and B, respectively). All plates were done duplicated and were incubated for 48 hours at 27 \pm 1°C and inhibitory halos formation were controlled. When an inhibitory halo was found, the compound was reported as positive (+), but when it was not present, it was reported as negative (-). The appearance of inhibition halos indicated that tested compounds would be active against the evaluated microorganism. Fluconazole (1500 µg/mL, watersoluble antifungal agent) was used as a reference drug.

RESULTS AND DISCUSSION

Synthesis

Twelve benzothiazole derivatives were prepared with affords moderate to good yields from formaldehyde, amines and 2-MBT. The synthesis of 3-[(amino)methyl]-1,3-benzothiazole-2-thione **4(a-1)** by Mannich reaction was accomplished as presented in Figure 1. In the first step it involves the nucleophilic addition of aliphatic or aromatic amines **2(a-1)** to formaldehyde (1) followed by dehydration to form the Schiff base. The Schiff base is an electrophile which reacts in the second step through an electrophilic addition on iminothiol tautomer of 2-MBT (3) containing an acidic proton. Finally, the compounds 4(a-1) were cooled in water, filtered, and washed with ethanol. Synthesized compounds were obtained as pure crystals and were characterized by the basic analysis of the spectroscopic data obtained (HRMS, ¹H-NMR, ¹³C-NMR, ¹⁹F-NMR experiments). In particular, 2-MBT contain an extended ðdelocalized systems and present an iminothiolthioamide tautomerism (Balestrero et al., 1986; Hassan and Khan, 2021), shown below (Figure 2) Following the procedure described by Holbová et al. (1976), the molecule of 2-MBT was substituted in the

Fig. 2. Prototropic equilibria iminothiol-thioamide of 2-MBT

thioamide

position 3 of iminothiol tautomer. These results confirm that the iminothiol tautomer is the predominant tautomer for Mannich reaction and pKa values of amines employed promote only the formation of monoderivative products, substituted in the position 3 of 2-MBT ring. All compounds were characterized by ¹H-RMN, ¹³C-RMN and HRMS spectroscopic data, respectively. For aliphatic amines the ¹H-RMN spectrum (CDCl₂) showed a singlet at δ 4.91ppm conrmed signal for methylene protons N-<u>CH</u>,-NH-R, multiplets ranging from δ 7.80–7.30 ppm conrmed signal for aromatic protons, respectively. For aromatic amines the ¹H-RMN (DMSO-d6) spectrum showed a singlet at δ 5.05 ppm conrmed signal for methylene protons N-CH2-NH-R, multiplets ranging from δ 8.12–6.86 conrmed aromatic protons and singlet ranges at δ 7.17-6.10 conrmed the presence of NH-R, respectively. Spectral data of ¹³C and ¹⁹F confirm the final structures for compounds 4(a-1).

Antimicrobial evaluation

The synthesized compounds were evaluated for their antifungal activity against Cladosporium cladosporioides, Aspergillus niger ATCC 16404, Botrytis sp. and Rhizopus sp. Agar well diffusion test was applied. A general view of the obtained results is exposed in Table 2. Compounds 4a-c, 4e-g, 4i and 4l showed antifungal activity against the four mentioned fungi, due to the presence of an inhibitory halo around the well with the corresponding compound. 4h only presented an inhibitory zone in case of *C. cladosporioides*; so its activity is considered moderate to low. Compounds 4d, 4j and 4k did not show activity against the molds used for the test. Fluconazole showed biological activity against Aspergillus niger ATCC 16404 and Botrytis sp.

Taking into account the results achieved, 8 of the 12 synthesized products have shown antifungal activity against filamentous fungi. Such promising substances should be studied as potential drugs to treat patients with pathogens caused diseases.

CONCLUSION

Twelve compounds **4(a-1)** were synthesized by Mannich reaction from formaldehyde, aliphatic and aromatic amines and 2-mercaptobenzothiazole. The yields obtained were satisfactory and the purity obtained for each product was >99 %. All compounds were characterized by nuclear magnetic resonance (¹H-RMN, ¹³C-RMN, ¹⁹F-RMN) and high-resolution mass spectrometry (HRMS) spectroscopic data. Synthesized products were evaluated against

Table 2. Antimicrobial activity of synthesized compounds 4(a-1) by agar well diffusion method

Compounds	Cladosporium	Aspergillus niger cladosporioides	Botrytis sp. ATCC 16404	Rhizopus sp.
4a	+	+	+	+
4b	+	+	+	+
4c	+	+	+	+
4d	-	-	-	-
4e	+	+	+	+
4f	+	+	+	+
4g	+	+	+	+
4h	+	-	-	-
4i	+	+	+	+
4j	-	-	-	-
4k	-	-	-	-
41	+	+	+	+
Fluconazole	-	+	+	-
Control A	-	-	-	-
Control B	-	-	-	-

Positive (+): an inhibitory halo was found. Negative (-): inhibitory halo not present.

Compounds concentration: 1500 µg/mL. Fluconazole concentration: 1500 µg/mL.

Reading done at 48 h of incubation at $27^{\circ}\text{C} \pm 1^{\circ}\text{C}$

Culture media: potato dextrose agar (PDA)

Wells size: 8 mm

four different fungal species by agar diffusion test. The antifungal activity displaying different degree of antimicrobial activity, i.e. compounds **4a-c**, **4e-g**, **4i** and **4l** showed antifungal activity against *Cladosporium cladosporioides*, *Aspergillus niger* ATCC 16404, *Botrytis sp.* and *Rhizopus sp.*, while **4d**, **4j** and **4k** did not show activity against the molds used for the test. These synthesized products are promising and will continue to be studied as a part of our program aimed at the development of new heterocyclic compounds with potential antifungal activities.

ACKNOWLEDGEMENTS

This work was supported by Agencia Nacional de Promoción Científica y Tecnológica (ANPCyT) (Project code: PICT 2016-4642), Consejo Nacional de Investigaciones Científicas y Técnicas (CONICET) (Project code: PIP 2015-024) and Universidad Nacional del Litoral (UNL) (Project code: CAID 2020-166). Authors thank for their financial support and contribution to the development of this work. We are thankful to Universidad de la Cuenca del Plata and INTA-Bella Vista, Corrientes, Argentina, for giving the facilities to carry out the antimicrobial activity.

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