



Metal coordination and peripheral substitution modulate the activity of cyclic tetrapyrroles on α S aggregation: a structural and cell-based study

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Abstract

The discovery of aggregation inhibitors and the elucidation of their mechanism of action are key in the quest to mitigate the toxic consequences of amyloid formation. We have previously characterized the anti-amyloidogenic mechanism of action of sodium phthalocyanine tetrasulfonate ($[\text{Na}_4(\text{H}_2\text{PcTS})]$) on α -Synuclein (α S), demonstrating that specific aromatic interactions are fundamental for the inhibition of amyloid assembly. Here we studied the influence that metal preferential affinity and peripheral substituents may have on the activity of tetrapyrrolic compounds on α S aggregation. For the first time, our laboratory has extended the studies in the field of the bioinorganic chemistry and biophysics to cellular biology, using a well-established cell-based model to study α S aggregation. The interaction scenario described in our work revealed that both N- and C-terminal regions of α S represent binding interfaces for the studied compounds, a behavior that is mainly driven by the presence of negatively or positively charged substituents located at the periphery of the macrocycle. Binding modes of the tetrapyrrole ligands to α S are determined by the planarity and hydrophobicity of the aromatic ring system in the tetrapyrrolic molecule and/or the preferential affinity of the metal ion conjugated at the center of the macrocyclic ring. The different capability of phthalocyanines and meso-tetra (*N*-methyl-4-pyridyl) porphine tetrachloride ($[\text{H}_2\text{PrTPCl}_4]$) to modulate α S aggregation in vitro was reproduced in cell-based models of α S aggregation, demonstrating unequivocally that the modulation exerted by these compounds on amyloid assembly is a direct consequence of their interaction with the target protein.

Keywords Misfolding · Amyloid · Neurodegeneration · Inhibitors

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Introduction

Parkinson's disease (PD) is the second most common age-related neurodegenerative disorder after Alzheimer's disease, affecting millions of people worldwide [1]. Most of the PD patients receive their diagnosis at the age of 50–60, with clinical symptoms comprising motor impairments including bradykinesia, tremor, akinesia, rigidity and other non-motor symptoms [2]. The motor manifestations are attributed to the irreversible loss of dopaminergic neurons of the *substantia nigra* (*SN*) *pars compacta*, whereas the formation of proteinaceous inclusions known as Lewy bodies is recognized as the pathological hallmark of the disease [3]. Two decades ago, immunostaining of SN sections of PD patients indicated that the protein α -synuclein (α S) is the major component of Lewy bodies, suggesting a role of α S in this pathology [4–6]. Lewy bodies are not restricted to PD but also found in other neurodegenerative diseases such

multiple system atrophy and dementia with Lewy bodies, collectively known as synucleinopathies [7–9]. Accordingly, over the last 20 years effort of numerous laboratories led to the development of different *in vitro* and *in vivo* systems, which were targeted to characterize α S, its aggregation, and the formation of Lewy bodies to shed light on the mechanism of pathogenesis in PD [10–19].

α -Synuclein (14 kDa) is a highly soluble, intrinsically disordered protein that is abundantly expressed in the brain and predominantly localizes to presynaptic terminals, in close proximity to synaptic vesicles [20–24]. Structurally, monomeric α S is characterized by an amphipathic N-terminal region (residues 1–60), involved in lipid binding; a central, highly hydrophobic self-aggregating sequence known as non-amyloid β component (NAC; residues 61–95), presumed to initiate fibrillation; and the acidic C-terminal region (residues 96–140), that is critical for blocking rapid α S filament formation [25–28] (Fig. 1a). Since misfolding and amyloid aggregation of α S is thought to play a critical role in PD [29], the aggregation pathway of the protein represents then an obvious target for therapeutic intervention in this disorder [16–19, 30, 31]. Therefore, understanding the mechanistic basis behind α S amyloid assembly and its inhibition is of high clinical importance. In this context, the discovery of small molecules targeting disease-associated protein aggregation is considered one of the most active therapeutic approaches toward neurodegenerative disorders [17, 19, 32]. Not only small molecules have been shown to modulate the aggregation of amyloid proteins both *in vitro* and *in vivo*, but have been also used as molecular and structural probes to gain insight into the amyloid aggregation process [33–36]. From the screening of large libraries of small molecules, potential candidates with different chemical structures were found to modulate the aggregation of distinct amyloid proteins [17, 37–55]. Notably, poly-aromatic

scaffolds belonging to flavonoids, polyphenols, porphyrins, and phthalocyanines were predominantly identified by these screenings. Indeed, distinct variants of phthalocyanines and porphyrins have been tested for their ability to impair amyloid assembly of proteins linked to neurodegeneration [36, 37, 56–59]. These molecules are cyclic tetrapyrroles, a class of compounds whose distinguished characteristic is the planarity and hydrophobicity of its aromatic ring system. The phthalocyanine tetrasulfonate (PcTS) compounds are among the most widely investigated tetrapyrroles [56]. The structure of PcTS contains four sulfonic acid groups at the periphery of the aromatic rings, whereas the central cavity of the molecule can remain ligand-free or conjugated to metal ions of various valences (Fig. 1b).

Particularly, $[\text{Na}_4(\text{H}_2\text{PcTS})]$ was shown to exhibit anti-scrapie activity [41, 43, 60–62] and inhibition of α S [35, 37, 57] and tau amyloid assembly [58]. However, this compound was not able to impair the amyloid fibril formation of A β 40 [36]. These results argued against the hypothesis that cyclic tetrapyrroles might have a common mechanism of action in slowing the formation of a variety of pathological aggregates. From a mechanistic perspective, and to fully understand the way these compounds might modulate protein aggregation, it is not only of paramount importance to decipher the structural basis of the implied protein-ligand interactions but also to investigate the molecular requirements of the small molecules that are critical for efficient and specific anti-amyloid activity. In that direction, in this work we studied how the nature of the conjugated metal ion and peripheral substituents in tetrapyrrolic compounds may influence or modulate the activity of these compounds on the amyloid fibril assembly of the protein α S. Our results demonstrated that both the nature of the metal ions conjugated to the central nitrogens and the properties of the functional groups at the periphery of the aromatic macrocycle

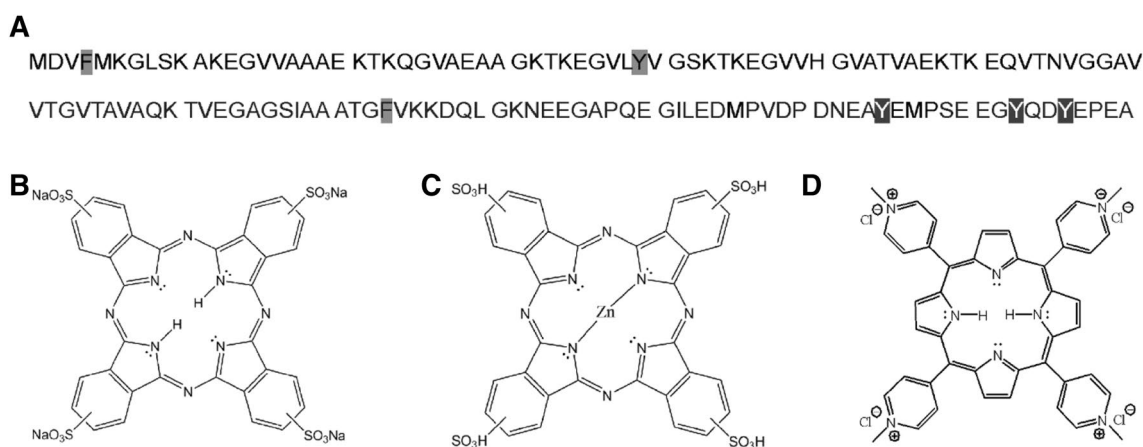


Fig. 1 **a** Primary sequence of the full-length protein α S. Aromatic residues are shaded in gray at the N-terminal and NAC regions, and in black at the C terminus. Chemical structure of phthalocyanine and porphine compounds: **b** $[\text{Na}_4(\text{H}_2\text{PcTS})]$, **c** $[\text{Zn}(\text{H}_4\text{PcTS})]$ and **d** $[\text{H}_2\text{PrTPC}_{14}]$

in phthalocyanines and porphyrins are factors that influence the location of these compounds at well-defined regions of α S and its interaction with specific residues in the protein sequence. Interestingly, we found that the interaction profile of these molecules with α S correlates directly with their modulatory (inhibition/acceleration) effects on the amyloid fibril assembly of the protein. Most importantly, the effects observed in vitro were reproduced on a well-established cell-based model for the study of α S aggregation. Overall, the results reported in this work provide a framework in which bioinorganic chemistry, biophysics and cell biology can be combined to understand the structural rules directing the interaction of aggregation inhibitors with target proteins and their implications on its amyloid fibril assembly.

Materials and methods

Protein and reagents

Unlabeled and ^{15}N isotopically enriched N-terminally acetylated α S was obtained by co-transforming *E. coli* BL21 with the plasmid harboring the wild type α S gene and a second one that encodes for the components of yeast NatB acetylase complex [63]. Both plasmids carried different antibiotic resistance, namely Ampicillin and Chloramphenicol to select the doubly transformed *E. coli* colonies. Purification was carried out as previously reported [64] with the exception that both antibiotics were included in the growth flasks to avoid plasmid purge during growth and expression. The final purity of the α S samples was determined by SDS-PAGE. Purified protein samples were dissolved in 20 mM MES buffer supplemented with 100 mM NaCl at pH 6.5 (Buffer A). Protein concentrations were determined spectrophotometrically by measuring absorption at 274 nm and using an epsilon value of $5600\text{ M}^{-1}\text{ cm}^{-1}$. Sodium phthalocyanine tetrasulfonate ($[\text{Na}_4(\text{H}_2\text{PcTS})]$) was purchased from MP Bio-medicals (Solon, Ohio); aluminium phthalocyanine chloride tetrasulfonic acid ($[\text{Al}(\text{H}_4\text{PcTS})\text{Cl}]$), zinc phthalocyanine tetrasulfonic acid ($[\text{Zn}(\text{H}_4\text{PcTS})]$) and *meso-tetra* (*N-methyl-4-pyridyl*) porphine tetrachloride ($[\text{H}_2\text{PrTPCl}_4]$) were from Frontier Scientific Inc. (Logan, Utah).

NMR experiments

NMR spectra were registered on a Bruker 600 MHz HD Avance III spectrometer, equipped with a cryogenically cooled triple resonance ^1H ($^{13}\text{C}/^{15}\text{N}$) TCI probe. Two-dimensional 2D ^1H - ^{15}N heteronuclear single quantum correlation (HSQC) experiments were performed with pulsed-field gradient enhanced pulse sequences [65] on 50–100 μM ^{15}N -labeled protein samples dissolved in buffer A, at 15 °C. One-dimensional 1D ^1H -NMR experiments were acquired

at 15 °C on 50–100 μM unlabeled α S samples dissolved in buffer A. Aggregation did not occur under these low-temperature conditions and absence of stirring. For the mapping experiments, ^1H - ^{15}N HSQC amide cross-peaks affected during titrations with the ligand molecules were identified by comparing their intensities (I) with those of the same cross-peaks in the data set of free protein (I_0) [35]. The I/I_0 ratios of 90–100 non-overlapping cross-peaks were plotted as a function of the protein sequence to obtain the intensity profiles. Acquisition and processing of NMR spectra were performed using TOPSPIN 7.0 (Bruker Biospin). 2D spectra analysis and visualization were performed with CCPN.

Electron microscopy

10- μl aliquots withdrawn from aggregation reactions of α S in the absence or presence of phthalocyanines molecules and $[\text{H}_2\text{PrTPCl}_4]$ were adsorbed onto Formvar/carbon-coated copper grids (Pella Inc.) and negative stained with 2% (w/v) uranyl acetate. Images were obtained at various magnifications (1000–90,000 \times) using a Philips CM120 transmission electron microscope, at the Facility for Transmission Electron Microscopy of the Max Planck Institute for Biophysical Chemistry (Göttingen).

Cell culture and transfection

For the studies with cell-based models of α S aggregation, H4 neuroglioma cells (ATCC[®] HTB-148TM) were maintained at 5% CO_2 and 37 °C in Dulbecco's Modified Eagle's Medium (DMEM) (Gibco) supplemented with 10% fetal bovine serum (Gibco) and 1% Penicillin Streptomycin (Gibco). 1 day before transfection cells were seeded in 12-well plates. Cells were co-transfected with plasmids encoding a C-terminally modified α S (SynT construct) and synphilin-1 [15] using FuGene (Promega), according to the manufacturer's instructions. After 24 h of transfection, phthalocyanines and $[\text{H}_2\text{PrTPCl}_4]$ were added to the cells in concentrations of 10 μM . Cells were treated with buffer A to exclude vehicle effects. Inclusion formation was assessed 48 h post-transfection.

Immunocytochemistry

Cells were fixed with 4% paraformaldehyde in PBS 48 h after transfection and start of phthalocyanines and $[\text{H}_2\text{PrTPCl}_4]$ treatment. For permeabilization, cells were subsequently treated with 0.1% Triton X-100 and blocked with 1.5% bovine serum albumin (BSA) in PBS. Cells were then incubated with anti- α S primary antibody (BD 610787, 1:1000) and anti-V5 tag (Abcam ab9116, 1:100), either 4 h at room temperature or overnight at 4 °C, and 1 h in Alexa Fluor 488 donkey anti-rabbit and Alexa Fluor 555 donkey anti-mouse

as secondary antibodies (Invitrogen A21202 and A31570, respectively, 1:1000). Cells were further stained with Hoechst (Molecular Probes 33258) for 5–10 min. Images were captured using a Nikon C2 Plus confocal microscope.

Quantification of α S intracellular inclusions

Transfected cells were detected and scored based on the absence and presence of α S intracellular inclusions. We also quantify the pattern of α S intracellular inclusions, by classifying them in different groups: 1–4 inclusions, 5–9 inclusions and equal to/more than 10 inclusions. Results were expressed as the percentage of the total number of transfected cells obtained from three independent experiments [15]. At least 50–100 cells were counted per condition.

Results and discussion

Influence of metal coordination and peripheral substituents on the interaction of tetrapyrroles with α S

We investigated the details of the binding of $[\text{Na}_4(\text{H}_2\text{PcTS})]$ (Fig. 1b), its Zn-loaded form $[\text{Zn}(\text{H}_4\text{PcTS})]$ (Fig. 1c) and the $[\text{H}_2\text{PrTPCl}_4]$ (Fig. 1d) to the protein α S by NMR spectroscopy. To analyze these interactions we used ^1H – ^{15}N HSQC spectroscopy. The ^1H – ^{15}N HSQC spectrum of a 100 μM sample of uniformly ^{15}N -labeled α S recorded in buffer A at 15 °C is shown in Fig. 2a. The resonances were well resolved, with a limited dispersion of chemical shifts, reflecting the disordered state and the high degree of flexibility of the backbone. As previously reported, the interaction of α S with $[\text{Na}_4(\text{H}_2\text{PcTS})]$ caused significant broadening and chemical shift changes in a number of residues at the N-terminal region of the protein [35] (Fig. 2b). No perturbations were observed for the amide groups of residues located in the NAC region or at the C terminus. Close analysis of the signals exhibiting severe changes on α S upon addition of $[\text{Na}_4(\text{H}_2\text{PcTS})]$ shows that the interaction is centered at Phe-4 and Tyr-39 residues. A small broadening and shift perturbation become also evident at Phe-94. The perturbations caused by the $[\text{Zn}(\text{H}_4\text{PcTS})]$ were also located at the N-terminal region of α S; however, the most affected residue was the histidine at position 50 (Fig. 2c). Minor effects were also detected at Phe-4 and Tyr-39, together with a lack of binding to Phe-94 (Fig. 2c). On the other hand, no broadening or chemical shift perturbations were detected when the interaction of α S with $[\text{Al}(\text{H}_4\text{PcTS})\text{Cl}]$ (Fig. S1) [37].

To evaluate the influence that charged substituents located at the periphery of the macrocycle might exert on the molecular interactions of these compounds with target proteins, we studied the binding features of the $[\text{H}_2\text{PrTPCl}_4]$ molecule

to the protein α S. Contrasting with the studied phthalocyanines, the substituents groups located at the periphery of the macrocycle in $[\text{H}_2\text{PrTPCl}_4]$ are positively charged (*N*-methyl-4-pyridyl). As observed in Fig. 2d, the titration of α S with $[\text{H}_2\text{PrTPCl}_4]$ caused broadening and chemical shift changes of amide resonances in the segment Q120–A140, located at the highly acidic C-terminal region of the protein. The residues exhibiting the strongest perturbations in the amide resonances were Tyr-125, Tyr-133 and Tyr-136 (Figs. 2d and S2). Thus, the substitution of the negatively charged sulfonate groups at peripheral positions of cyclic tetrapyrroles by positively charged *N*-methyl-pyridyl groups influences markedly the binding features of these molecules, by redirecting the interaction from the N-terminal region to the C-terminus of α S.

The interaction of the tetrapyrrolic compounds with aromatic moieties in the protein was then analyzed by NMR experiments aimed at detecting directly the selective perturbation of resonances of aromatic side chains in α S upon ligand binding (Fig. S3). The distribution of aromatic residues throughout the α S sequence provides excellent probes to investigate the interaction of cyclic tetrapyrroles with α S. The binding features observed for the $[\text{Na}_4(\text{H}_2\text{PcTS})]$, $[\text{Zn}(\text{H}_4\text{PcTS})]$ and $[\text{H}_2\text{PrTPCl}_4]$ complexes confirm the role as anchoring moieties played by the aromatic side chains of Phe-4, Tyr-39, and His-50 at the N-terminus and by Tyr-125, Tyr-133 and Tyr-136 at the C-terminal region of α S (Fig. S3). In particular, these spectra allowed us to conclude that Phe-4 and Tyr-39 are the main anchoring residues for $[\text{Na}_4(\text{H}_2\text{PcTS})]$ binding to the N-terminal region of α S, highlighting the role of specific π – π interactions between the aromatic ring system of these molecules and aromatic residues in the amyloidogenic proteins in the formation of the α S– $[\text{Na}_4(\text{H}_2\text{PcTS})]$ complex. Interestingly, in the case of $[\text{Zn}(\text{H}_4\text{PcTS})]$, whereas the divalent cation Zn(II) was shown to bind preferentially to the C-terminus of α S [66], electrostatic repulsions between the peripheral sulfonate groups in the ring system and the highly negatively charged C-terminal region disfavor the location of the $[\text{Zn}(\text{H}_4\text{PcTS})]$ complex at that region, redirecting its binding toward the N-terminal interface. In that region, the His-50 residue acts as the main anchoring residue for $[\text{Zn}(\text{H}_4\text{PcTS})]$ binding, indicating that the preferential affinity of the Zn(II) ion coordinated to the center of the phthalocyanine modulates the binding features of this compound by targeting other binding site in the α S protein. Although more work is needed to confirm this hypothesis, this interpretation is in agreement with the more pronounced propensity of Zn(II) ions to form flexible and open coordination geometries, as we demonstrated recently for the interaction of $[\text{Zn}(\text{H}_4\text{PcTS})]$ with the amyloid β -peptide [36].

Clearly, the presence of positively charged substituents at the periphery of the macrocycle in the $[\text{H}_2\text{PrTPCl}_4]$

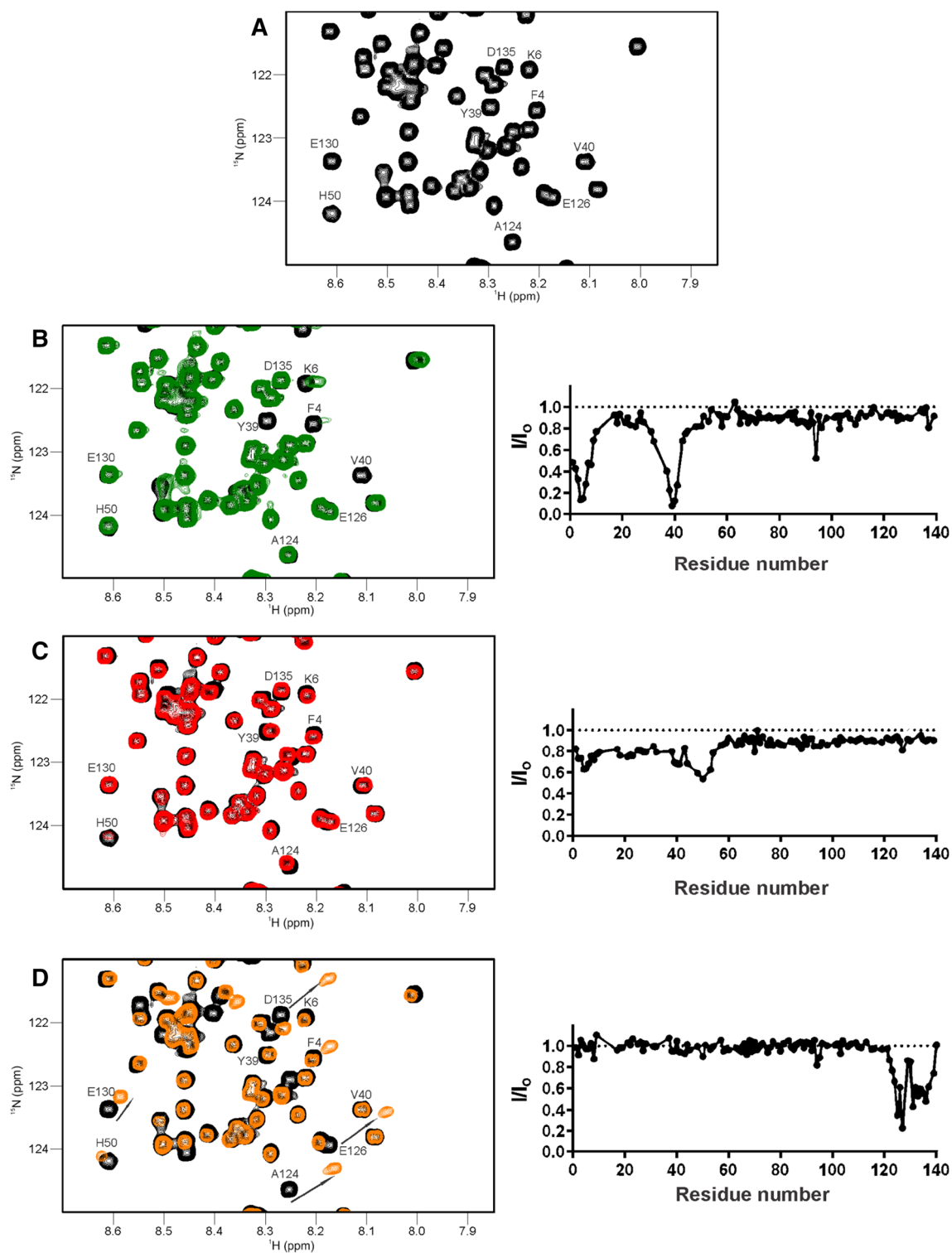


Fig. 2 Structural characterization of the interaction between phthalocyanines and $[\text{H}_2\text{PrTPC}]_4$ with αS monitored by NMR spectroscopy. **a** ^1H - ^{15}N HSQC spectra of 100 μM αS . **b** Overlaid contour plots of ^1H - ^{15}N HSQC spectra and I/I_0 profile of the backbone amide groups of 100 μM αS in the absence (black) or presence (green) of 50 μM $[\text{Na}_4(\text{H}_2\text{PcTS})]$. **c** Overlaid contour plots of ^1H - ^{15}N HSQC spectra and I/I_0 profile of the backbone amide groups of 100 μM αS in the

absence (black) or presence (red) of 50 μM $[\text{Zn}(\text{H}_4\text{PcTS})]$. **d** Overlaid contour plots of ^1H - ^{15}N HSQC spectra and I/I_0 profile of the backbone amide groups of 100 μM αS in the absence (black) and presence (orange) of 50 μM $[\text{H}_2\text{PrTPC}]_4$. Amino acid residues of αS affected by the interaction with the tetrapyrrole compounds are identified. ^1H - ^{15}N HSQC spectra were recorded at 15 $^\circ\text{C}$ using ^{15}N isotopically enriched αS (100 μM) samples

molecule allows this molecule to interact electrostatically with negative patches at the C-terminal region of α S, directing the location of the $[\text{H}_2\text{PrTPCl}_4]$ molecule to that region of the protein structure. There, the cluster of tyrosines (Y-125, Y-133 and Y-136) at the C-terminus constitute the target site for $[\text{H}_2\text{PrTPCl}_4]$ binding to α S, as shown in Figs. 2D and S2. As observed with the $[\text{Na}_4(\text{H}_2\text{PcTS})]$ molecule at the N-terminus binding interface, aromatic interactions between the macrocycle ring of $[\text{H}_2\text{PrTPCl}_4]$ and the Tyr cluster at the C-terminus contribute strongly to the binding process.

Finally, from the analysis of panels b–d of Fig. 2 we also found indications of significant differences in the degree of broadening induced by $[\text{Na}_4(\text{H}_2\text{PcTS})]$, its Zn(II)-loaded form and $[\text{H}_2\text{PrTPCl}_4]$ on the amide resonances of α S. We demonstrated previously that the degree of phthalocyanines self-association correlates well with their capabilities to bind α S, with low order stacked aggregates of the cyclic tetrapyrroles being identified as the active interacting species [37]. The planarity and aromaticity of the central heterocycle appear to be determining factors for such self-association reactions, which, however, can be perturbed or heavily impaired by the nature of the conjugated metal ion. Indeed, the smaller effects induced by $[\text{Zn}(\text{H}_4\text{PcTS})]$ binding to α S reflect the lower tendency of this phthalocyanine derivative to self-associate [37]. As expected, this behavior becomes more evident in the Al(III) derivative, which was shown to remain mostly as a monomer in solution and thus, showed a complete lack of interaction with the protein (Fig. S1) [37]. In this compound, axial coordination of Al(III) with chloride anions give rises to distorted structures that impair the stacking of the $[\text{Al}(\text{H}_4\text{PcTS})\text{Cl}]$ molecules.

Effects of tetrapyrrolic compounds on the aggregation of α S in vitro and in cell-based model

Ultrastructural visualization of the final product of aggregation of α S using TEM shows that normal amyloid fibrils are formed in the absence of tetrapyrrolic compounds (Fig. 3a). We found that $[\text{Na}_4(\text{H}_2\text{PcTS})]$ and $[\text{Zn}(\text{H}_4\text{PcTS})]$ produced a variety of small, amorphous, non-fibrillar aggregates (Fig. 3b), as well as short, flat fibrillar α S aggregates, respectively (Fig. 3c). Conversely, the α S aggregates in $[\text{Al}(\text{H}_4\text{PcTS})\text{Cl}]$ -treated samples showed an amyloid morphology comparable with that of untreated protein samples (Figure S4). On the other hand, Fig. 3d shows that α S amyloid assembly through $[\text{H}_2\text{PrTPCl}_4]$ mediation are made of large and abundant aggregates, comprising clusters of fibrils.

Finally, we tested the potential toxicity of phthalocyanines and $[\text{H}_2\text{PrTPCl}_4]$ in human neuroglioma (H4) cells. For this cell line, the compounds were innocuous at concentrations as high as 20 μM (data not shown). Then, we used a well-established cell model to assess α S inclusion formation. H4 cells were transiently transfected with C-terminally modified α S (SynT) and synphilin-1, which results in the formation of LB-like inclusions, as previously described [15]. The formation of α S inclusions was assessed 48 h after treatment (Fig. 4a). Upon treatment with 10 μM $[\text{Na}_4(\text{H}_2\text{PcTS})]$ and $[\text{Zn}(\text{H}_4\text{PcTS})]$ we observed a significant increase in the number of transfected cells devoiding α S inclusions relative to untreated samples (Fig. 4b). Analysis of α S inclusion formation in the $[\text{Al}(\text{H}_4\text{PcTS})\text{Cl}]$ -treated cells showed a pattern comparable with that of untreated samples (Figs. 4a, b and S5). By contrast, $[\text{H}_2\text{PrTPCl}_4]$ treatment promoted an increase in the number of transfected cells displaying α S aggregates (Figs. 4a, b and S5). Overall, our results

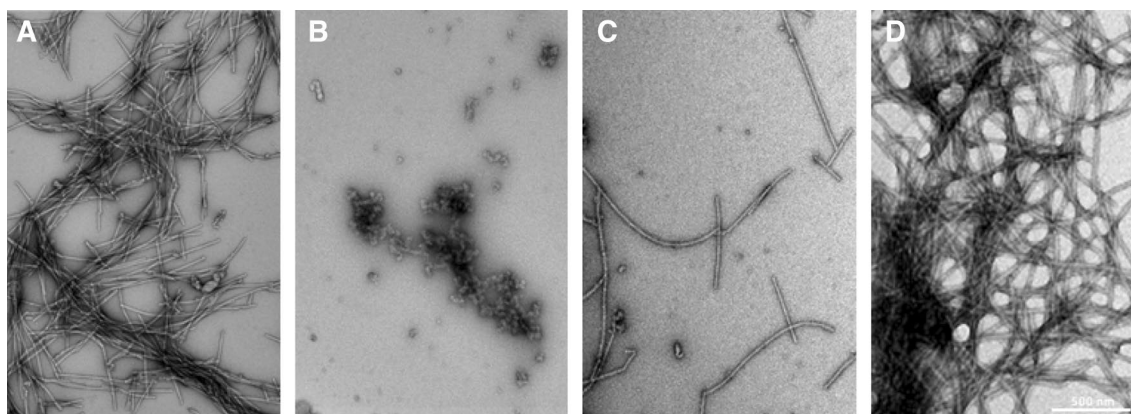


Fig. 3 Modulation of α S aggregation by phthalocyanines and $[\text{H}_2\text{PrTPCl}_4]$. Representative negative stain EM images of α S aggregates (100 μM samples) generated in **a** the absence and presence

of 100 μM , **b** $[\text{Na}_4(\text{H}_2\text{PcTS})]$, **c** $[\text{Zn}(\text{H}_4\text{PcTS})]$ and **d** $[\text{H}_2\text{PrTPCl}_4]$. Fibrils of identical morphologies to that shown in panel A were obtained for α S in the presence of $[\text{Al}(\text{H}_4\text{PcTS})\text{Cl}]$ (Fig. S4)

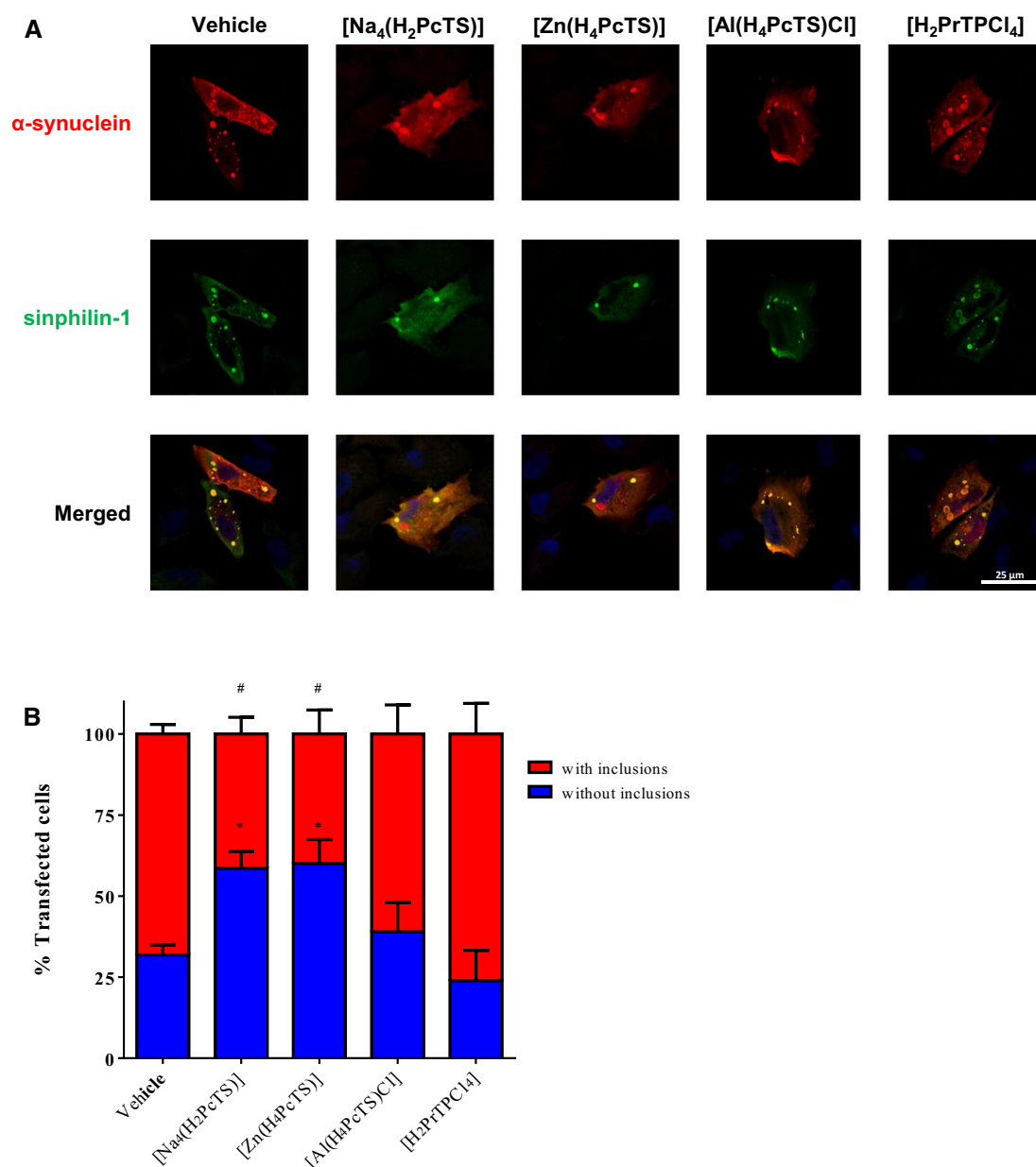


Fig. 4 Modulation of α S inclusion formation in cultured cells. **a** Representative images of intracellular α S inclusion formation in human cultured cells in the presence of phthalocyanines and [H₂PrTPC1₄]. Human H4 neuroglioma cells were co-transfected with plasmids encoding SynT and Synphilin-1, and inclusion formation was assessed 48 h post-transfection. Cells were incubated in the absence (vehicle) and the presence of the studied tetracyclic compounds (10 μ M). **b** Quantification of α S inclusions. Transfected cells were

detected and scored based on the absence or presence of α S inclusions. Results were expressed as the percentage of the total number of transfected cells obtained from three independent experiments. At least 50 cells were scored per experiment ($n=3$). A detailed quantification of the pattern of α S inclusion formation, in the absence or presence of the studied compounds is shown in Fig. S5. * $p < 0.05$ cells without inclusions vs. vehicle treatment; # $p < 0.05$ cells with inclusions vs. vehicle treatment

demonstrated that the effects induced by phthalocyanines and [H₂PrTPC1₄] on α S fibril formation in vitro are reproduced well on the H4 cell-based model of α S aggregation.

Conclusions

The results reported in this work provide a firm basis to understand the structural rules directing the binding of tetrapyrrolic compounds to α S, and their implications for the amyloid aggregation of the protein. Because the structural basis for the activity of these molecules as modulators of

amyloid aggregation is starting to emerge, combined efforts from the fields of chemistry, structural and cell biology are needed at this stage to elucidate their binding modes with the target proteins, the precise molecular mechanism(s) of action of these molecules and its biological activity on cell models of the disease. The interaction scenario described in our work revealed that both N- and C-terminal regions of α S represent binding interfaces for the studied compounds, a behavior that is mainly driven by the presence of negatively or positively charged substituents located at the periphery of the macrocycle, that can mediate electrostatic interaction with positive or negative patches on the proteins structure. With the polyaromatic compounds located at the N- or C-terminus, the interaction mode is determined by (1) the planarity and hydrophobicity of the aromatic ring system in the tetrapyrrolic molecule and/or (2) the preferential affinity of the metal ion conjugated at the center of the macrocyclic ring.

The differences observed between the binding preferences and interaction profiles of $[\text{H}_2\text{PrTPCl}_4]$ and phthalocyanines provide a structural explanation for the promotion of α S amyloid fibril formation by $[\text{H}_2\text{PrTPCl}_4]$ molecules against the inhibitory effect exerted by $[\text{Na}_4(\text{H}_2\text{PcTS})]$. Regarding to the mechanistic basis behind the inhibitory effects of $[\text{Na}_4(\text{H}_2\text{PcTS})]$ and $[\text{Zn}(\text{H}_4\text{PcTS})]$ on α S amyloid fibril assembly, the amyloid blocking effects of these phthalocyanines might be related to a loss of function of aromatic side chains involved in the filament-assembly mechanism, likely comprising early intra- and intermolecular interactions necessary for amyloid structural transition and implying to the affected residues. On the other hand, we propose that formation of the α S- $[\text{H}_2\text{PrTPCl}_4]$ complex reduces and/or interferes with the attractive intramolecular interactions of the C-terminus with the N-terminal and NAC regions in the protein, decreasing the stability of monomeric α S and promoting its aggregation.

Interestingly, the different capability of phthalocyanines and $[\text{H}_2\text{PrTPCl}_4]$ to modulate α S aggregation in vitro was also observed in cell-based models of PD, demonstrating unequivocally that the inhibitory mechanism exerted by phthalocyanines on amyloid assembly is a direct consequence of their interaction with target proteins. Added to that, the versatility of these compounds to condone the coordination of different metal ions and a wide range of peripheral substituents, which in turn can modulate physicochemical properties such as their solubility and distribution in biological fluids, emphasize the potential of this scaffold for further optimizations in terms of pharmacokinetics and delivery to the brain.

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