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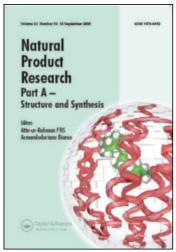
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Natural Product Research

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713398545

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Online Publication Date: 01 January 2009

To cite this Article Casado, Mariela, Ortega, María Gabriela, Peralta, Mariana, Agnese, Alicia Mariel and Cabrera, José Luis(2009)'Two new alkamides from roots of Acmella decumbens', Natural Product Research, 23:14,1298 — 1303

To link to this Article: DOI: 10.1080/14786410802518201 URL: http://dx.doi.org/10.1080/14786410802518201

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Two new alkamides from roots of Acmella decumbens

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(Received 18 April 2007; final version received 29 September 2008)

From the roots of *Acmella decumbens* (Sm) R.K. Jansen three compounds belonging to the alkamide family were isolated: (2*E*, 4*E*)-*N*-hydroxiphenylethyl-2,4-decadien-9-inamide (1); (4*E*, 6*E*)-*N*-isobutyl-4,6-undecen-10-inamide; (2) and (2*Z*)-*N*-phenylethyl-2-nonane-6,8-diinamide (3). The structures were determined by means of IR, MS, ¹H NMR, ¹³C NMR, DEPT 135 and COSY. Compounds 1 and 2 are reported for the first time, while 3 was previously isolated from *Spilanthes acmella* L.

Keywords: Acmella decumbens; alkamides; acmelline; decumbine

1. Introduction

Acmella decumbens (Sm) R.K. Jansen (= Spilanthes decumbens (Smith) A.H. Moore) (Asteraceae) is a species whose habitat extends from southeastern Uruguay to Río Grande do Sul in Brazil. In Argentina, its distribution corresponds to the northeastern and central regions of the country. This plant grows in very different soils, varying in density and dimensions of foliage (Cabrera, 1971, 1974, 1978; Jansen, 1985; Zuloaga & Morrone, 1999). Its roots produce a pungent effect accompanied by a sleepy sensation of the tongue when they are bitten. For this reason, in folk medicine analgaesic properties are attributed to this plant, and it is used as an odontalgic agent (Cabrera, 1978). The species belongs to a genera that is characterised by the presence of alkamides (Greger, 1984). These compounds have been biologically studied, showing that several of them have insecticidal properties; others are immunostimulatory or antiviral, develop inhibitory activity of the synthesis of RNA, or cause inhibitory activity of the arachidonic acid metabolism. Alkamides in general are characterised by a pungent effect that they produce when they are bitten, as well as a sialagogue effect (Bauer & Reminger, 1989; Greger, 1984, 1988; Nagashima & Nakatani, 1992; Ramsewak, Erickson, & Nair, 1999).

In the present work, we report the isolation and identification of two new alkamides from the hexane root extract of A. decumbens, (2E, 4E)-N-hydroxiphenylethyl-2,4-decadien-9-inamide (acmelline, 1), (2E, 4E)-N-isobutyl-4,6-undecen-10-inamide (decumbine, 2), as well as a third already known compound (2Z)-N-phenylethyl-2-nonane-6,

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8-diinamide (3). The methodology used for isolation and purification is described, together with the spectroscopic data obtained and a discussion on the structure elucidation.

2. Results and discussion

Three compounds (1–3) (Figure 1) belonging to the alkamide family were obtained by purification of the hexane root extract of *A. decumbens*. Compounds 1 and 2 are described for the first time in the present work and their structures were determined by IR, MS, ¹H RMN, COSY H–H, ¹³C NMR, DEPT and elemental analysis, while 3 has been isolated previously from *Spilanthes acmella* L. (Nagashima & Nakatani, 1992).

Compound 1 was isolated as yellow oil. The IR spectra showed absorption bands at 3304, 1632 and $1532\,\mathrm{cm}^{-1}$, the characteristics of a secondary amide, at $2221\,\mathrm{cm}^{-1}$ an absorption band due to a triple bond presence in the structure and at $2982\,\mathrm{cm}^{-1}$ a band that corresponds to aromatic hydrogens (Nagashima & Nakatani, 1992; Martin & Becker, 1984; Saadali, Boriky, Blaghen, Vanhaelen, & Talbi, 2001). The MS presents an ion at m/z 284 that corresponds to $[M+1]^+$, as well as the following ions that correspond to other breakdown fragments of the compound at m/z 255 ($C_{16}H_{17}NO_2$), 164 ($C_9H_{10}NO_2$), 93 (C_6H_5O) and 58 (C_2H NO) (Saadali et al., 2001). The 1H NMR spectrum presented a wide singlet at δ_H 5.49 (1H), which corresponded to the H of an amide group, and a multiplet at δ_H 7.21–7.35 (4H, H 4', 5', 7', 8'), attributable to four aromatic protons. The ^{13}C NMR spectrum showed the signals of aromatic ring carbons at δ_C 138.42, 126.56, 128.75, 146.83, 128.75 and 126.56. The triple bond of the structure was confirmed by the

Compound 1

Compound 2

Compound 3

Figure 1. Alkamides from *A. decumbens*. Compound 1: (2*E*, 4*E*)-*N*-hydroxiphenylethyl-2,4-decadien-9-inamide. Compound 2: (4*E*, 6*E*)-*N*-isobutyl-4,6-undecen-10-inamide. Compound 3: (2*Z*)-*N*-phenylethyl-2-nonane-6,8-diinamide.

presence of a wide singlet at $\delta_{\rm H}$ 1.95 (1H, H10) in the ¹H NMR spectrum, and by two signals in the ¹³C NMR spectrum assignable to the sp carbons at $\delta_{\rm C}$ 77.8 and 64.8. Four signals were observed in the ¹H NMR spectrum at 5.79 (1H; H4; *J*:15), 5.82 (1H; H2; *J*:15,7), 6.23 (1H; H5; *J*:15,7) and 7.24 (1H; H3; *J*:15,7) that were assigned to four olefinic protons (Nagashima & Nakatani, 1992; Martin & Becker, 1984). In the ¹³C NMR spectrum, four signals that correspond to the carbons involved in the double bonds at $\delta_{\rm C}$ 123.83, 142.14, 124.85 and 143.92 were observed. Double bond *trans*-configuration (*E*) of the molecule was determined by the coupling constant (*J*) value of 15 Hz (Greger & Hofer, 1989). By means of the homonuclear correlation experiment COSY, it was possible to establish all the proton–proton correlations of the molecule. Methylene and methine groups present in the structure were identified by means of the DEPT experiment. From all of the above data, compound 1 was identified as (2*E*, 4*E*)-*N*-hydroxiphenylethyl-2,4-decadien-9-inamide (acmelline) (Table 1).

Compound **2** was isolated as yellow oil. The MS presented a molecular ion [M]⁺ at m/z 233, consistent with the molecular formula (C₁₅H₂₃NO), and another ion at m/z 57 that corresponded to a terminal isobutyl fragment (Greger & Hofer, 1989; Nagashima & Nakatani, 1992; Martin & Becker, 1984). The IR spectrum revealed a secondary amide group presence (3310,1633 and 1531 cm⁻¹) and a triple bond in the structure at 2221 cm⁻¹. The ¹H NMR spectrum presented a wide singlet at $\delta_{\rm H}$ 1.96 (1H; H11) that was attributable to the triple bond, confirmed by two signals in the ¹³C NMR spectrum corresponding to the sp carbons at $\delta_{\rm C}$ 77.7 and 64.8, respectively. The ¹H NMR spectrum presented a wide singlet at $\delta_{\rm H}$ 5.42 (1H; NH) that corresponded to the H of the amide group. The presence of a terminal isobutyl group was attributed to a quadruplet at $\delta_{\rm H}$ 3.14 (2H; H1'; J:7), a multiplet at $\delta_{\rm H}$ 1.77 (1H; H2') and a doublet at $\delta_{\rm H}$ 0.93 (6H; H3', 4'; J:7),

Table 1. Compound 1: ¹H NRM, COSY (200.13 MHz, CDCl₃), ¹³C NRM, DEPT (50.03 MHz, CDCl₃).

H no.	$\delta_{\rm H}$ ppm (J Hz)	COSY	$\delta_{\rm C}$ ppm	DEPT
1			165.59	
2	5.82 dt (15,7)	H_3	123.83	CH
3	7.24 d (15)	H_2H_4	142.14	CH
4	5.79 d (15)	H_{3}, H_{5}	124.85	CH
5	6.23 dd (15,7)	H_6, H_4	143.92	CH
6	2.85 dd (7,7)	H_5, H_7	30.39	CH_2
7	2.41 d (7)	H_6,H_8	20.08	CH_2
8	1.49 t	H_7	26.94	CH_2
9	_	_	77.83	_
10	1.95 s	_	64.89	CH
NH	5.49 s	_	_	_
1'	3.58 m	$H_{2'}$	40.38	CH_2
2'	2.85 dd (7,7)	$\overline{\mathrm{H}}_{1'}^-$	35.63	CH_2
3'	_	_	138.42	_
4′	7.21–7.35 m	$H_{5',7',8'}$	126.56	CH
5'	7.21-7.35 m	$H_{4',7',8}$	128.75	CH
6'	_	_	146.83	_
7'	7.21–7.35 m	$H_{4',5',8}$	128.75	CH
8'	7.21–7.35 m	$H_{4',5',7'}$	126.56	CH
OH	9.27 s, b	,- ,-	-	_

and three signals in the 13 C NMR spectrum at 46.8, 28.5 and 18.9, respectively (Nagashima & Nakatani, 1992). In the 1 H NMR spectrum, three double triplets at $\delta_{\rm H}$ 5.81 (2H; H5,7; J:15,7), 6.08 (1H; H4; J:15,7) and 6.83 (1H; H6; J:15,7) were assigned to the olefinic protons confirmed by the 13 C NMR spectrum, which were four signals corresponding to the carbons involved in the double bonds at $\delta_{\rm C}$ 124.3, 143.5, 124.0 and 137.1 (Greger & Hofer, 1989; Nagashima & Nakatani, 1992; Martin & Becker, 1984; Saadali et al., 2001). The *trans*-configuration (E) of the double bonds was determined by the coupling constant (J) value of 15 Hz (Greger & Hofer, 1989). By means of the homonuclear correlation experiment COSY, all the proton–proton correlations of the molecule were established. Primary, secondary and tertiary carbon assignation (groups CH₃, CH₂ and CH) was corroborated by means of DEPT. According to all these antecedents, compound 2 was identified as (2E, 4E)-N-isobutyl-4,6-undecen-10-inamide (decumbine) (Table 2).

Compound 3 was obtained as pale yellow oil. It was previously isolated from *S. acmella* L.; its structure was assigned by comparison to the spectroscopic data of IR, MS and ¹H NMR cited in the literature (Nagashima & Nakatani, 1992).

3. Experimental

3.1. General

¹H NMR and COSY spectra were measured on a Bruker AC spectrometer at 200.13 MHz. ¹³C NMR and DEPT were measured on the same equipment at 50.03 MHz, using CDCl₃ as solvent. IR spectra were obtained with a Nicolet 5 SXC FTIR infrared spectrophotometer. EIMS were obtained on a Variant Mat CH 7-A at 70 eV. Microanalysis of CHNS was recorded on a CARLO ERBA EA 1108. Column chromatography (CC) was performed on silica gel (Merck 70–230 mesh). Thin layer chromatography (TLC) was performed on precoated silica gel plates (Merck Kiesselgel 60 F254). Spots were revealed

Table 2. Compound **2**: ¹H NMR, COSY (200.13 MHz, CDCl₃), ¹³C NMR, DEPT (50.03 MHz, CDCl₃).

H no.	$\delta_{\rm H}~({\rm ppm})$	COSY	$\delta_{\rm C}$ ppm	DEPT
1			165.0	
2	2.34 dt (7.7)	H_3	26.9	CH_2
3	2.89 dd (15.7)	H_2, H_4	27.4	CH_2
4	6.08 dt (15.7)	H_3, H_5	124.3	CH
5	5.81 dt (15.7)	H_6, H_4	143.5	CH
6	6.86 dt (15.7)	H_5,H_7	124.0	CH
7	5.81 dt (15.7)	H_6,H_8	137.1	CH
8	1.81 dd (15.7)	H_7,H_9	30.9	CH_2
9	2.34 dt (7.7)	H_8	46.6	CH_2
10	_	_	77.7	
11	1.96 s	_	64.8	CH
NH	5.42 s	_	_	
1'	3.14 c (7)	$H_{2'}$	46.8	CH_2
2'	1.77 m	$H_{1',3',4'}$	28.5	CH_2
3'4'	0.93 d (7)	H_2	18.9	CH_3

under UV light by spraying the plates with Dragendorff reactive. Preparative TLC was performed on plates prepared with silica gel 60 GF254.

3.2. Plant material

Roots of *A. decumbens* were collected in Embalse de Río Tercero, Los Cocos, Athos Pampa (Córdoba Province, Argentina) between the end of December 2002 and the beginning of January 2003. Plant material was identified by Prof. Dr Luis Ariza Espinar (Instituto Multidisciplinario de Biología Vegetal, CONICET-Universidad Nacional de Córdoba). A voucher specimen has been deposited at the Museo Botánico de Córdoba (UNC) as CORD 775.

3.3. Extraction and isolation

Plant material was dried at room temperature. Fragmented *A. decumbens* roots (1180 g) were extracted for 72 h using *n*-hexane in a Sohxlet apparatus. The hexane extract (12 g) was submitted to CC eluted with trichloroetilene: EtOAc (9:1) in increasing proportions of EtOAc, up to 100%. All the fractions were monitored by TLC, using toluene: EtOAc (7:3). Two main fractions were obtained and both were posteriorly purified by CC with petrol: EtOAc (8:2) as mobile phase. The eluents were monitored by TLC with *n*-hexane: acetone (3:1) as mobile phase. Final separation and purification of 1 (7.4 mg), 2 (9.3 mg) and 3 (2.5 mg) were achieved by several successive preparative TLC with the same last mobile phase.

Acmelline (2E, 5E)-N-hydroxiphenylethyl-2,5-decadien-9-inamide (1): Yellow oil. IR (KBr) $\nu_{\rm max}$ cm⁻¹: 3304, 2982, 2952, 2221, 1668, 1632, 1532, 1300. MS: 284 [M+1]⁺ (C₁₈H₂₂NO₂), 255 (C₁₆H₁₇NO₂), 164 (C₉H₁₀NO₂), 107 (C₇H₇O), 93 (C₆H₅O), 58 (C₂H₄NO). Elemental analysis Calcd for C₁₈H₂₁NO₂: C 76.33%; H 7.42%, N 4.94%, found C 75.09%, H 7.01%, N 4.20%. ¹H NMR, COSY, DEPT and ¹³C NMR (Table 1).

Decumbine (4*E*, 6*E*)-*N*-isobutyl-4,6-undecen-10-inamide (2): Yellow oil. IR $\nu_{\rm max}$ cm⁻¹: 3310, 2966, 2926, 2850, 2221, 1682, 1633, 1531, 1373. MS: 233 ($C_{15}H_{23}NO$), 219 ($C_{14}H_{21}NO$), 131 ($C_{10}H_{11}$), 77 (C_6H_5), 57 (C_4H_9). Elemental analysis Calcd for $C_{15}H_{23}NO$: C 77.25%, H 9.87%, N 6.0%, found C 75.92%, H 10.07%, N 6.71%. ¹H NMR, COSY, DEPT and ¹³C NMR (Table 2).

(2Z)-N-phenylethyl-2-nonane-6,8-diinamide (3): IR $\nu_{\rm max}$ cm⁻¹: 3320, 3021, 2930, 2904, 2250, 1665, 1634, 1550, 1330. MS: 251 ($C_{17}H_{17}NO$), 131($C_{9}H_{7}O$), 105 ($C_{8}H_{9}$), 91 ($C_{7}H_{7}$), 77 ($C_{6}H_{5}$). All the NMR spectroscopic data are coincident with those reported in the literature (Nagashima & Nakatani, 1992).

Acknowledgements

The authors are thankful to Prof. Dr Luis Ariza Espinar (Museo Botánico de Córdoba, Universidad Nacional de Córdoba) for the identification of the species. This study was supported by PICT 1728 N° 2398 and SeCyT (UNC) grants.

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