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# Bulk-sensitive Mo 4d electronic structure of Sr<sub>2</sub>FeMoO<sub>6</sub> probed by high-energy Mo L<sub>3</sub> resonant photoemission

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**Abstract** – We studied the Mo 4d electronic structure of Sr<sub>2</sub>FeMoO<sub>6</sub> using high-energy Mo L<sub>3</sub> resonant photoemission. The experimental spectra are in good agreement with modified Becke-Johnson (mBJ) band structure calculations. The energy dependence of the spectra can be explained by the changes in the photoemission cross-sections. The Mo L<sub>3</sub> resonant spectrum shows Mo 4d character below the Fermi level and mixed Mo 4d-O 2p character around 8.0 eV. The Mo 4d weight in the resonant spectrum is in good agreement with the calculated interference between the direct and decay terms. The high photon energy used in this study (about 2520 eV) provides a bulk-sensitive determination of the Mo 4d electronic structure.

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**Introduction.** – The 4d transition metal compounds and alloys present a broad range of interesting physical properties. For instance, ferroelectricity in KNbO<sub>3</sub>, magnetoresistance in Sr<sub>2</sub>FeMoO<sub>6</sub>, 1D character in K<sub>0.3</sub>MoO<sub>3</sub>, ferromagnetism in SrRuO<sub>3</sub>, giant moments in Fe-Pd alloys, etc. These materials are used in high-*k* capacitors, optical devices, magnetic sensors, electrical contacts, etc. In addition, Rh and Pd alloys and compounds are very important in many catalytic processes. The physical properties of these compounds are directly related to the corresponding electronic structure. For this reason, there is a need to develop better techniques to study the transition metal 4d states.

Photoemission is a very powerful method to study the electronic structure of transition metal compounds. In particular, the Cooper minimum [1] in the 4d photoemission cross-section was used to investigate some of these compounds [2–7]. In this method, a reference spectrum is contrasted with one taken at the Cooper minimum (80–120 eV), so the differences in the spectra can be qualitatively attributed to features with 4d character. However, this relatively simple and widely used technique

has two drawbacks: first, the energy difference between the reference and the Cooper minimum spectrum is relatively large, so the photoemission cross-section of the other electronic levels involved also changes considerably [8]. Secondly, the results are surface sensitive because the photoelectron escape depth at the Cooper minimum is about 5 Å [9].

In this work, we study the electronic structure of Sr<sub>2</sub>FeMoO<sub>6</sub> using high-energy Mo L<sub>3</sub> resonant photoemission (RPES). The difference between the *on*- (2527 eV) and *off*-resonance (2515 eV) spectra is directly related to the Mo 4d contribution. The other electronic levels do not change much because the energy difference is relatively small, and the results are more bulk-sensitive because the photoelectron escape depth at these energies, about 35 Å [9], is much larger. For these reasons, the present method is a particularly powerful technique to obtain the Mo 4d contribution. We show below that the experimental Mo 4d bands are in good agreement with band structure calculations. The RPES spectra show a minority Mo 4d band at the Fermi level, and mixed Mo 4d-O 2p states at higher energies. These results are important

to understand the magnetic ordering and the electronic transport in this material.

The  $\text{Sr}_2\text{FeMoO}_6$  compound is an ordered double perovskite with a tetragonal structure. This material is a half-metallic ferromagnet with a relatively high  $T_C$  of about 420 K [10]. Single crystals of this material present a relatively small magnetoresistance, whereas polycrystalline samples exhibit a much larger (30%) magnetoresistance [11]. In view of this, the observed magnetoresistance is usually attributed to tunneling effects across grain boundaries [11,12]. The magnetism here is due to an antiferromagnetic alignment of the localized Fe 3d and the itinerant Mo 4d moments [13]. The magnetic moment of the samples, about  $3.2 \mu_B$ , is very sensitive to the concentration of Fe-Mo anti-sites [14]. The electronic structure of this compound was investigated by photoemission as well as theoretical calculations [15–18].

**Experimental details.** – The  $\text{Sr}_2\text{FeMoO}_6$  sample was synthesized using the conventional solid-state reaction method. The proper amount of the reagents were repeatedly mixed and calcined in air at  $950^\circ\text{C}$  for 24 h. The resulting powder was reduced in a flowing mixture of 1%  $\text{H}_2$ -Ar gas at  $1050^\circ\text{C}$  for 1 h. The material was ground, pressed into pellets, and sintered in a vacuum of  $2 \times 10^{-8}$  bar at  $1200^\circ\text{C}$  for 12 h. The powder XRD analysis confirmed the presence of a single-phase material. The Rietveld refinement indicated a tetragonal  $I4/mmm$  structure with  $a = 5.5761 \text{ \AA}$  and  $c = 7.9078 \text{ \AA}$ . The relative intensity of the (101) reflection showed that the concentration of anti-sites was less than 3% [14].

The photoemission experiment was performed at the Laboratório Nacional de Luz Síncrotron (LNLS). The beamline is equipped with a focusing mirror, a double-crystal monochromator and an UHV experimental chamber [19]. The energy scale of the monochromator was calibrated using the Si  $K$  and the Mo  $L_3$  absorption edges. The photoemission spectra were acquired using a SPECS Phoibos 150 electron energy analyzer. The energy scale of the analyzer was calibrated using the Fermi level of a clean gold foil. The resonant photoemission spectra were taken around 2520 eV with an energy resolution of about 0.4 eV. The samples were repeatedly scraped with a diamond file to remove the surface contamination.

#### Calculation details. –

**Band structure calculation.** The band structure of  $\text{Sr}_2\text{FeMoO}_6$  was calculated using the WIEN2k program [20]. The calculation was performed using the modified Becke-Johnson (mBJ) exchange potential [21]. This method is very efficient and produces good results for transition metal oxides [22]. Both the space group and the lattice parameters were extracted from the XRD measurements. The atomic positions and muffin-tin radii are given in table 1.

The grid of  $\mathbf{k}$ -points was set to  $11 \times 11 \times 11$ , and the plane-wave basis cutoff was set to  $RK_{\text{max}} = 7.0$ . The total

Table 1: Atomic positions and muffin-tin radii used in the band structure calculations of  $\text{Sr}_2\text{FeMoO}_6$ .

Atom	Atomic positions			$R_{\text{MT}}$ (a.u.)
	$x$	$y$	$z$	
Sr	0.500	0.000	0.250	2.50
Fe	0.000	0.000	0.000	1.98
Mo	0.000	0.000	0.500	1.92
O <sub>1</sub>	0.000	0.000	0.254	1.71
O <sub>2</sub>	0.249	0.249	0.000	1.71

energy convergence criterion for the self-consistent calculation was set to  $10^{-5}$  eV.

**Resonant photoemission.** The resonant photoemission spectrum was analyzed using the following Hamiltonian:

$$H = H_0 + V. \quad (1)$$

The  $H_0$  part of the Hamiltonian describes the active electron energy levels:

$$H_0 = \sum_i E_{2p} c_i^\dagger c_i + \sum_i E_{4d} d_i^\dagger d_i + \sum_i E_{\epsilon f} e_i^\dagger e_i, \quad (2)$$

where the first term corresponds to the core Mo 2p electrons with energy  $E_{2p}$ , the second term describes the valence Mo 4d electrons with energy  $E_{4d}$ , and the third term relates to the continuum  $\epsilon f$  electrons with energy  $E_{\epsilon f}$ . The Mott-Hubbard repulsion between the Mo 4d electrons was not considered because  $\text{Sr}_2\text{FeMoO}_6$  is a  $4d^1$  compound [23,24]. The Fe 3d and O 2p electrons were not considered because the resonant photoemission involves a Mo 2p core level and it is, thus, a local process. Further, the spectral weight of the Mo 4d electrons was obtained from the band structure calculations, which already include the interactions of the Mo 4d states with the O 2p and Fe 3d electrons. Finally, the energy and wave function of the Mo 2p orbital were obtained from the atomic calculation program by Cowan [25].

The  $V$  part of the Hamiltonian describes the corresponding transitions:

$$V = \sum_{i,j} V_{ph} e_i^\dagger d_j + \sum_{i,j} V_{ab} d_i^\dagger c_j + \sum_{i,j,k,l} V_{Aug} c_i^\dagger e_j^\dagger d_k d_l + \text{h.c.}, \quad (3)$$

where the first term corresponds to the  $4d$ - $\epsilon f$  photoemission process, the second term describes the  $2p$ - $4d$  absorption transition, and the third term relates to the  $L_3 N_{4,5} N_{4,5}$  Auger decay. The  $V_{ph}$  parameter is given in terms of the  $\langle \epsilon f || r || 4d \rangle$  matrix element, the  $V_{ab}$  parameter is written in terms of the  $\langle 4d || r || 2p \rangle$  matrix element, and the  $V_{Aug}$  parameter is related to the  $R^1(2p, \epsilon f; 4d, 4d)$  and  $R^3(2p, \epsilon f; 4d, 4d)$  integrals. Photoemission to the  $\epsilon p$  channel was neglected because the  $\langle \epsilon p || r || 4d \rangle$  matrix element

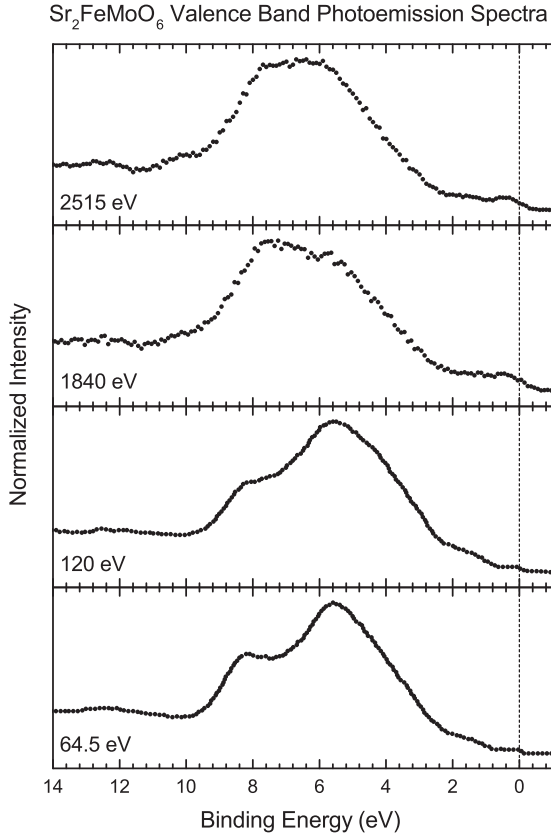


Fig. 1: High energy-photoemission spectra of Sr<sub>2</sub>FeMoO<sub>6</sub> obtained at 2515 and 1840 eV compared to previous low-energy photoemission spectra taken at 120 and 64.5 eV (taken from ref. [15]).

is much smaller. Likewise, Auger decay into the  $\epsilon p$  channel was also neglected because the  $R^1(2p, \epsilon p; 4d, 4d)$  and  $R^3(2p, \epsilon p; 4d, 4d)$  integrals are much smaller. The dipole matrix elements as well as the radial integrals were calculated using the code by Cowan [25]. The quadrupole contribution to the cross-section can be neglected, because the calculation shows that it is 15 times smaller than the dipole part. We note that this approach does not describe all the many-body effects of the resonant photoemission process. However, the advantage is that all the ingredients of the model are obtained using *ab initio* calculations.

### Results and discussion. –

**Photoemission spectra.** The high-energy photoemission spectra of the Sr<sub>2</sub>FeMoO<sub>6</sub> compound taken at 2515 and 1840 eV are presented in fig. 1. These are compared to previous low-energy photoemission spectra measured at 120 and 64.5 eV (taken from ref. [15]). The structure around 8 eV is mainly attributed to Fe 3d-O 2p and Mo 4d-O 2p bonding states. The feature about 6 eV is mostly assigned to Fe 3d-O 2p and O 2p non-bonding states. Finally, the intensity close to the Fermi level, from 2 to 0 eV, corresponds to Fe 3d and Mo 4d bands.

The relative intensity of the features in the spectra changes strongly with the photon energy. The most

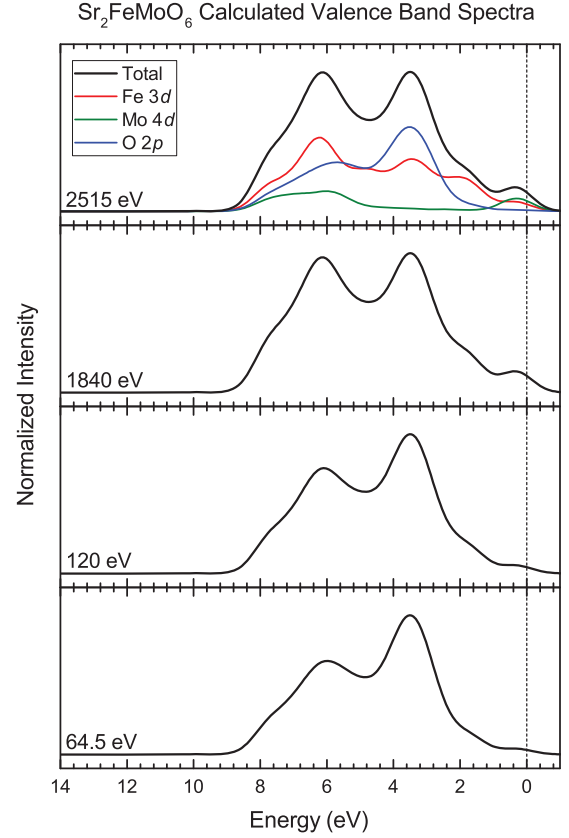


Fig. 2: (Colour online) Calculated photoemission spectra of Sr<sub>2</sub>FeMoO<sub>6</sub> as a function of the photon energy. The spectra are composed by the Fe 3d, Mo 4d and O 2p contributions. These were weighted according to the corresponding photoemission cross-sections.

dramatic effect is the decrease of the non-bonding O 2p intensity at higher energies. There are also lesser changes in the relative intensity of the other structures in the spectra. A similar photon energy dependence was already reported in the spectra of the VO<sub>2</sub> [26] and LaVO<sub>3</sub> [27] oxides. We show below that these changes can be explained by the energy dependence of the photoemission cross-sections.

Figure 2 shows the calculated photoemission spectra of Sr<sub>2</sub>FeMoO<sub>6</sub> as a function of the photon energy. The calculated spectra are the sum of the individual Fe 3d, Mo 4d and O 2p contributions (the Sr 4d contribution to the valence band is much smaller and can be neglected). These components were weighted according to their corresponding photoemission cross-sections. The photon energy of the cross-sections were obtained by interpolation of the values reported in the literature [8]. First, we note that the calculated spectra are in reasonably good agreement with the experimental results. Second, the band structure calculations support the assignments of the spectra made above. Finally, the changes in the calculated spectra with the photon energy resemble those observed in the experiment.

**Resonant photoemission.** Figure 3 shows the photoemission spectra of Sr<sub>2</sub>FeMoO<sub>6</sub> taken at 2515 eV

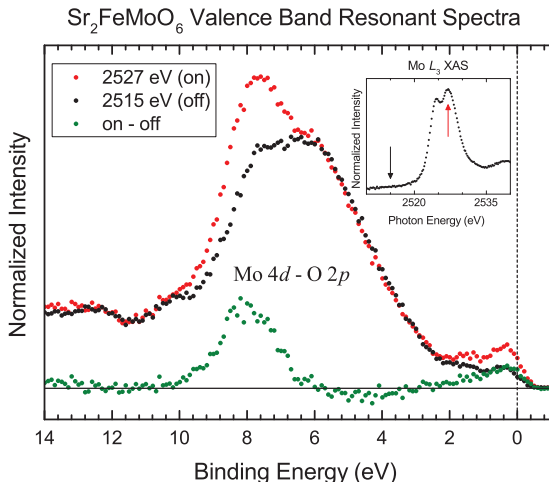


Fig. 3: (Colour online)  $L_3$  resonant photoemission spectra of the  $\text{Sr}_2\text{FeMoO}_6$  compound. The *off*-resonance spectrum was taken at 2515 eV, whereas the *on*-resonance spectrum was obtained at 2527 eV. The inset shows the Mo  $L_3$  x-ray absorption spectrum indicating the photon energies used in the above spectra. The lower part presents the difference between the *on* and *off* photoemission spectra.

(*off*-resonance) and 2527 eV (*on*-resonance). The inset shows the Mo  $L_3$  x-ray absorption spectrum indicating the photon energies used in the spectra. The *off*-resonance photoemission spectrum is only due to the direct photoemission process ( $2p^6 4d^1 + h\nu \rightarrow 2p^6 4d^0 + e$ ). In the *on*-resonance spectrum, the indirect decay channel ( $2p^6 4d^1 + h\nu \rightarrow 2p^5 4d^2 \rightarrow 2p^6 4d^0 + e$ ) also contributes. The constructive interference between these two processes enhances the intensity of the Mo  $4d$  states [28–30].

The lower part of fig. 3 presents the difference between the *on* and *off* photoemission spectra. This difference is directly related to the Mo  $4d$  contribution to the valence band. The enhancement in the difference spectrum below the Fermi level, around 0.5 eV, is attributed to the Mo  $4d$  band. The increased intensity around 8.0 eV is assigned Mo  $4d$  states mixed with O  $2p$  character. This increase is relatively large indicating that the covalent Mo  $4d$ -O  $2p$  mixing is strong.

The intensity  $I(\nu)$  of the resonant photoemission spectrum is given by [30]

$$I(\nu) = \left| \langle f | V_{ph} | i \rangle + \sum_m \frac{\langle f | V_{Aug} | m \rangle \langle m | V_{ab} | i \rangle}{h\nu - E_{2p} + i\Gamma_m} \right|^2, \quad (4)$$

where  $i$  denotes the initial  $2p^6 4d^1$  state,  $f$  stands for the final  $2p^6 4d^0$  state,  $m$  denotes an intermediate  $2p^5 4d^2$  state, the lifetime broadening is  $\Gamma_m = \pi |\langle f | V_{Aug} | m \rangle|^2$ , and the sum runs over all the possible intermediate states. The first term in the equation corresponds to the direct process, whereas the second term describes the intermediate decay channel. We note that the  $V_{ph}$  and  $V_{ab}$  terms were treated in the first order, whereas the  $V_{Aug}$  decay was evaluated to all orders [30].

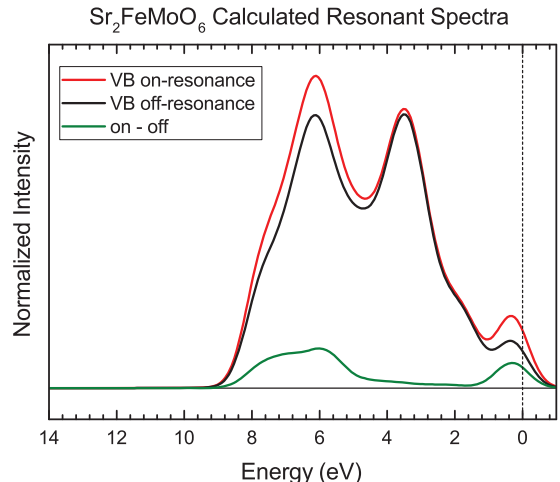


Fig. 4: (Colour online) Calculated  $L_3$  resonant photoemission spectra of the  $\text{Sr}_2\text{FeMoO}_6$  compound. These include the *off*- and *on*-resonance spectra, as well as the difference spectrum.

Table 2: Parameters used in the calculation of the resonant photoemission spectra.

$\langle \epsilon f    r    4d \rangle$	$-0.0309 \text{ a.u.}/\text{eV}^{1/2}$
$\langle 4d    r    2p \rangle$	$-0.0397 \text{ a.u.}$
$R^1(2p, \epsilon f; 4d, 4d)$	$0.0312 \text{ eV}^{1/2}$
$R^3(2p, \epsilon f; 4d, 4d)$	$0.0195 \text{ eV}^{1/2}$

The values of the parameters used in the calculation of the intensity of the resonant photoemission spectrum are given in table 2.

The calculated *off*- and *on*-resonance spectra, as well as the difference spectrum of  $\text{Sr}_2\text{FeMoO}_6$ , are presented in fig. 4. The *off*-resonance spectrum corresponds to the sum of the Fe  $3d$ , Mo  $4d$  and O  $2p$  components, weighted according to their relative photoemission cross-sections. In the *on*-resonance spectrum, the Fe  $3d$  and O  $2p$  character remains the same, whereas the Mo  $4d$  contribution is normalized according to eq. (4). We note that the calculation is in relatively good agreement with the experimental results. It describes the main features in the spectra as well as the changes observed between the *off*- and *on*-resonance.

*Comparison with previous work.* The Mo  $4d$  states in the  $\text{Sr}_2\text{FeMoO}_6$  compound were also studied using the Cooper minimum method [15–18]. The photon energy difference in this method is rather large, about 40–50%, whereas in the present work it is relatively small, around 0.5%. Thus, the variation of the photoemission cross-section of the other electronic levels can be safely neglected [8]. Furthermore, the electron escape depth at the Cooper minimum is very small, about 5 Å, whereas at the Mo  $L_3$  edge is much larger, around 35 Å [9]. Therefore, the present results are more bulk-sensitive than those obtained using the Cooper minimum method. The Mo  $4d$

electronic structure in Sr<sub>2</sub>FeMoO<sub>6</sub> was also studied using Mo M x-ray emission spectroscopy, but these results present only a broad structure about 6.0 eV and do not reveal the states at the Fermi level [31].

The present results reveal the contribution of the Mo 4d states to the bulk electronic structure of the Sr<sub>2</sub>FeMoO<sub>6</sub> double perovskite. They indicate the presence of Mo 4d states at the Fermi level and mixed Mo 4d-O 2p states around 8.0 eV. These findings are crucial to understand the indirect Mo 4d-O 2p-Fe 3d hopping, which is directly related to the electronic transport mechanism in this compound [10]. Further, the relatively large Mo 4d-O 2p hybridization is also necessary to explain the antiferromagnetic coupling between the Fe 3d and Mo 4d spins [13].

The electronic structure of the NiO and Fe<sub>2</sub>O<sub>3</sub> compounds were studied using hard x-ray resonant photoemission [32]. This study was performed at the Ni K and Fe K x-ray absorption edges and probed the secondary Ni 3d and Fe 4p states [32]. The electronic structure of CeRh<sub>3</sub> was also studied using hard x-ray resonant photoelectron spectra [33]. In this case, the spectra were taken at the Ce L<sub>3</sub> x-ray absorption edge and probed the Ce 3d core levels [33]. The present study was carried out at the Mo L<sub>3</sub> edge and probed directly the active Mo 4d states in the valence band.

**Summary and conclusions.** – In summary, We studied the Mo 4d electronic structure of Sr<sub>2</sub>FeMoO<sub>6</sub> using high-energy Mo L<sub>3</sub> resonant photoemission. The Mo 4d character obtained from the spectra is in good agreement with band structure calculations. The resonant spectrum shows Mo 4d character at the Fermi level and mixed Mo 4d-O 2p character around 8.0 eV. The present findings are important to understand the electrical and magnetic properties of the Sr<sub>2</sub>FeMoO<sub>6</sub> compound.

To conclude, the high-energy resonant photoemission technique is very powerful and provides bulk-sensitive results for Sr<sub>2</sub>FeMoO<sub>6</sub>. The experimental results can be reproduced using a combination of band structure and model calculations. This approach could be extended to study the electronic structure of many other interesting 4d alloys and compounds.

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