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¹ Effective Anisotropic Interactions in Spin Pairs Containing High-Spin ² Ions with Large Zero-Field Splitting

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6 Supporting Information

ABSTRACT: Analytic and numeric derivations are made of the 7 8 effective exchange and dipolar magnetic interactions between 9 spin pairs containing S = 3/2 ions, such as high-spin Co(II), S =5/2 ions, such as high-spin Fe(III) ions, experiencing zero-field 10 splittings much larger than the interion interactions, or J = 15/211 ions such as Dy(III) with crystal-field splittings much larger than 12 the interion interaction. These formulas allow for a simpler 13 analysis of the magnetic properties of dimers containing high-14 spin ions. 15



16 INTRODUCTION

17 The study of magnetic interactions between paramagnetic ions 18 or molecules, molecular magnetism, has been of great interest 19 in the fields of inorganic 1,2 and bioinorganic chemistry. $^{3-7}$ EPR 20 spectroscopy has played an important role in the first and a 21 major role in the second discipline. In the case of interactions 22 between S = 1/2 spins, the theoretical framework allowing 23 interpretation of EPR results is well understood, remaining 24 mostly improvements in quantum-mechanical calculations of 25 the magnitudes of different magnetic interactions and further 26 development of advanced methods (i.e., high-frequency and 27 pulsed EPR) for elucidation of subtle or difficult to resolve 28 interactions. The story is very different in the case of high-spin 29 ions such as Co(II) (S = 3/2) and Fe(III) (S = 5/2) which, in 30 addition to having a large spin, usually present partially 31 unquenched orbital momentum, which under the effect of 32 spin-orbit coupling, produce a zero-field splitting (ZFS) of the 33 spin sublevels. This ZFS does not only give rise to very 34 anisotropic effective g' matrices, but also conveys high 35 anisotropy to magnetic interactions between these types of 36 ions. A similar situation is observed for some lanthanides such $_{37}$ as Dy, which present highly anisotropic effective g' matrices 38 $(g'_{x,y} \approx 0, g'_z \approx 10-20)$ and have lately been the focus of ³⁹ intense research efforts.⁸⁻¹⁰ One of the problems in studying 40 anisotropic magnetic interactions is that magnetic suscepti-41 bility or magnetization studies, which are usually performed on 42 powder samples, do not always allow us to reliably obtain the 43 anisotropic parts of the interactions, with some exceptions.¹¹ 44 While EPR performed on powder samples does give 45 anisotropic information, when systems with more than one 46 ion with a high spin are studied, usually the number of 47 parameters associated with the magnetic interactions is so large 48 that approximations have to be made and the extracted 49 information is limited. Finally, although single-crystal EPR can

be used to extract full anisotropic interactions, the difficulties $_{50}$ associated with measurements (large (~0.5³ mm³) stable $_{51}$ crystals have to be grown, morphology has to be determined to $_{52}$ align the crystals, and long times of data analysis) have $_{53}$ diminished the number of this kind of studies. In the case of $_{54}$ metalloproteins, single-crystal EPR studies are very scarce. Our $_{55}$ aim in this work is to derive general equations for anisotropic $_{56}$ effective interactions in pairs involving high-spin ions, which $_{57}$ can confidently be used to analyze EPR, thus considerably $_{58}$ diminishing the number of unknown parameters and $_{59}$ simplifying the problem. Furthermore, these relations could $_{60}$ be used to more reliably relate EPR results, which are usually $_{61}$ (but not always) analyzed as effective spin 1/2 systems, to $_{62}$ magnetization and magnetic susceptibility results, which are $_{63}$ usually treated using real-spin Hamiltonians.

Half-integer spins, in the presence of an axial ZFS, have $(2S \ 6S \ +1)/2$ Kramers doublets separated by an energy difference 66 $D(2|M_S|+1)$ between the $(|M_S|+1)$ and $|M_S|$ doublets. When 67 the ZFS is large enough, each of these doublets can be 68 considered an effective S' = 1/2 with an effective \mathbf{g}' matrix. 69 When the ZFS is not axial the doublets are not pure M_S states 70 and the energy separation between them is modified, but still 71 an effective spin and \mathbf{g}' matrix can be assigned to each doublet. 72 Pilbrow's equations¹² relate the real \mathbf{g} matrix associated with 73 the real spin (S = 3/2 or in some cases S = 5/2) and the 74 rhombicity parameter E/D of the zero-field splitting \mathbf{D}^{ZFS} 75 tensor with the effective \mathbf{g}' matrix associated with each of 76 the Kramer's doublets arising from the total spin and separated 77 in energy by the ZFS. The general form of these equations is 78

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$$g'_{u} = g_{u} \alpha_{u} (E/D) \tag{1}$$

so where u = x, y, z and the expressions for $\alpha_u(E/D)$ for S = 3/2⁸¹ are given in the SI. For S = 5/2 and higher the α_{μ} factors have 82 to be calculated numerically and plotted in the form of ¹ s3 rhombograms^{13,14} The same can be stated in matrix form as g'84 = gP, where P is the diagonal matrix with α_{μ} coefficients. In 85 this work we extend our previous relations¹⁵ to obtain general 86 formulas for the effective anisotropic exchange and magnetic ⁸⁷ dipolar interactions within pairs containing S = 3/2 and S = 5/288 2 ions and, more preliminarily, to J = 15/2 ions such as 89 Dy(III). The form of these interactions makes it easy to see 90 that the expressions are valid for any kind of spin pair and 91 allows us to more accurately study interactions in any magnetic 92 system with more than one spin. In particular, these equations 93 will allow researchers to more accurately measure distances in 94 spin-labeled metalloproteins using the dipolar interaction as a 95 ruler.

96 THEORETICAL METHODS

97 There is a simple theoretical deduction, which will be shown 98 below, of the equations relating the real interspin interactions 99 $J_{\rm iso}^{\rm HS}$, $J_{\rm ani}^{\rm HS}$, and $\mathbf{D}_{\rm dip}^{\rm HS}$ (isotropic and anisotropic exchange and 100 magnetic dipolar interaction, respectively), with the effective 101 interion interactions $\mathbf{J}_{\rm ani}^{1/2}$ and $\mathbf{D}_{\rm dip}^{1/2}$ (the effective isotropic 102 interaction $J_{\rm iso}^{1/2}$ can be obtained as 1/3 of the trace of the 103 effective anisotropic interactions). Assuming for simplicity that 104 all matrices involved are coaxial and given the real Hamiltonian 105 for S = 3/2 or 5/2 ions

$$\hat{H}^{3/2} = \mu_{\rm B} \mathbf{B}^{\rm T} \mathbf{g} \hat{\mathbf{S}}^{\rm HS} + D \left((S_z^{\rm HS})^2 - \frac{S(S+1)}{3} \right) + E ((S_x^{\rm HS})^2 - (S_y^{\rm HS})^2)$$
(2)

¹⁰⁷ with HS meaning high spin; if $D \gg h\nu$, each Kramers' doublet ¹⁰⁸ can be described by an effective S' = 1/2 spin and the ¹⁰⁹ following Zeeman Hamiltonian

$$\hat{H}_{Z}^{1/2} = \mu_{\rm B} \mathbf{B}^{\rm T} \mathbf{g}' \hat{\mathbf{S}}^{1/2}$$
$$= \boldsymbol{\mu}_{\rm B} \mathbf{B}^{\rm T} \mathbf{g} \mathbf{P} \cdot \hat{\mathbf{S}}^{1/2}$$
(3)

¹¹¹ The **P** matrix contains the coefficients α_{μ} relating the real and ¹¹² effective **g** matrices, and it is written explicitly to show that the ¹¹³ real $\hat{\mathbf{S}}^{\text{HS}}$ operator can be replaced by the effective **P** $\hat{\mathbf{S}}^{1/2}$ on its ¹¹⁴ action within each Kramer's doublet. Therefore, in the effective ¹¹⁵ spin-1/2 space of two interacting ions the real-spin operators ¹¹⁶ in the pair spin Hamiltonian

$$\hat{H}_{12} = (\hat{H}_{Z,1} + ZFS_1) + (\hat{H}_{Z,2} + ZFS_2) + (\hat{\mathbf{S}}_1^{\text{HS}})^T \mathbf{D}_{12}^{\text{HS}} (\hat{\mathbf{S}}_2^{\text{HS}})$$

¹¹⁷ where $ZFS_{1,2}$ represents the zero-field splitting terms for each ¹¹⁸ center, can be replaced by the effective spin operator as follows ¹¹⁹ (note that the $ZFS_{1,2}$ terms will now contribute a constant ¹²⁰ value because of the effective spin-1/2 operators and so can be ¹²¹ disregarded)

$$\hat{H}_{12} = \hat{H}_{Z,1} + \hat{H}_{Z,2} + (\mathbf{P}_{1}\hat{\mathbf{S}}_{1}^{1/2})^{T}\mathbf{D}_{12}^{\mathrm{HS}}(\mathbf{P}_{2}\hat{\mathbf{S}}_{2}^{1/2})$$

$$= \hat{H}_{Z,1} + \hat{H}_{Z,2} + (\hat{\mathbf{S}}_{1}^{1/2})^{T}\mathbf{P}_{1}^{T}\mathbf{D}_{12}^{\mathrm{HS}}\mathbf{P}_{2}(\hat{\mathbf{S}}_{2}^{1/2})$$

$$= \hat{H}_{Z,1} + \hat{H}_{Z,2} + (\hat{\mathbf{S}}_{1}^{1/2})^{T}\mathbf{D}_{12}^{1/2}(\hat{\mathbf{S}}_{2}^{1/2})$$
(4)

where $\hat{H}_{z,i}$ represents the effective Zeeman interaction for each 123 spin and $\mathbf{D}_{12}^{\text{HS}}$ and $\mathbf{D}_{12}^{1/2}$ represent the real and effective 124 interaction between the spins. The effect of multiplication by 125 the \mathbf{P}_i matrices on the interaction matrix is shown in eq 5 for a 126 general exchange (isotropic plus anisotropic) and dipolar 127 interaction. If one of the spins is a 1/2 spin, then its effective 128 and real **g** matrices will be the same (or its \mathbf{P}_i matrix will be the 129 unit matrix). 130

RESULTS AND DISCUSSION

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The equations derived in the SI and in the preceding section 132 allowed us to obtain the following general formulas for the 133 effective anisotropic exchange interaction and the effective 134 magnetic dipolar interaction. 135

$$\mathbf{J}_{\text{ani,full}}^{1/2} = \mathbf{R}_{1} \cdot \begin{pmatrix} \frac{g'_{xA}}{g_{xA}} \\ & \frac{g'_{yA}}{g_{yA}} \\ & \frac{g'_{yA}}{g_{yA}} \end{pmatrix} \cdot \mathbf{R}_{1}^{T} \cdot \mathbf{J}_{\text{ani,full}}^{3/2} \cdot \mathbf{R}_{2} \cdot \begin{pmatrix} \frac{g'_{xB}}{g_{xB}} \\ & \frac{g'_{yB}}{g_{yB}} \\ & \frac{g'_{yB}}{g_{yA}} \end{pmatrix} \cdot \mathbf{R}_{2}^{T}$$
$$D_{\text{dip,eff}}^{1/2} = \mu_{B}^{2} \hbar^{2} \frac{\mu_{0}}{4\pi r^{3}} (\mathbf{g}_{1}^{\prime T} \cdot \mathbf{g}_{2}^{\prime} - 3 \cdot (\mathbf{g}_{1}^{\prime T} \cdot \hat{\mathbf{r}}) (\hat{\mathbf{r}}^{T} \cdot \mathbf{g}_{2}^{\prime}))$$
(5) 136

where $\mathbf{R}_{A,B}$ are the column eigenvector matrices of the effective 137 \mathbf{g}' matrices from sites A and B, $\hat{\mathbf{r}}$ is the unit vector connecting 138 spins A and B, and all other symbols have the usual meanings. 139 These equations have been derived for cases of high symmetry, 140 in which the xyz system defined by the ZFS matrix is 141 coincident with the axis system of either the anisotropic or 142 dipolar interaction, but the shape of the equations obtained 143 made us propose, in a heuristic approach, that they are indeed 144 general to any arbitrary mutual orientation of each site's ZFS 145 matrices. The easiest way to prove if this was indeed so was 146 simulation of the energy levels in both the full $S \otimes S$ space 147 (with S = 5/2 or S = 3/2) and the effective $(S' = 1/2) \otimes (S' = 148)$ 1/2) subspace, choosing arbitrary orientations for the different 149 matrices. Figure 1 shows a simulation of the energy levels of a 150 fl pair of high spin Co(II) ions with positive ZFS, treated as real 151 S = 3/2 spins and as effective S' = 1/2 spins. The simulation 152 parameters are listed in Table 1. 153 tl

Equation 5, which is the primary result of this study, is the 154 equivalent of Pilbrow's equations for the interaction between 155 spins, at least one of which is higher than S = 1/2. The idea of 156 an effective exchange interaction is not at all new, 15,17-20 but 157 previous formulas were limited to special symmetries or either 158 only discussed an effective isotropic exchange or used effective 159 g' matrices in dipolar interactions without a theoretical 160 justification. In particular, Lines' important 1971 paper¹⁷ 161 studied the effective exchange interaction between S = 3/2 162 Co(II) ions in cubic symmetry, arriving to an effective 163 isotropic interaction $J_{iso}^{1/2} = (25/9)J_{iso}^{3/2}$. Lines' approach has 164 been applied to many studies, but this work is not useful in 165 most systems containing HS Co(II), which are distorted from 166 cubic symmetry, particularly in biological systems. Specifically, 167 the isotropic part of the effective interaction is not 25/9 times 168 larger than the real isotropic interaction. In general, the factor 169 $\alpha_{\rm iso}$ in the formula $J_{\rm iso}^{1/2} = \alpha_{\rm iso} J_{\rm iso}^{\rm HS}$ depends on the rhombicity 170 factor (E/D), as shown in Table 2, but for a pair of high-spin $171 t_2$ Co(II) ions this factor equals 3 independently from E/D. The 172 ratios between the isotropic part of the effective exchange and 173

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Figure 1. Energy levels of a dimer of S = 3/2 ions and the corresponding dimer of effective S' = 1/2 ions, with the magnetic field in the *x*, *y*, and *z* directions. Simulation parameters are given in Table 1.

Table 1. Simulation Parameters for the Pair of High-Spin Co(II) Ions Corresponding to Figure 1

	ZFS	Euler rotatio angles ^a	on	real g matrix	effective g' matrix	
Co1	$D = 70 \text{ cm}^{-1}$	$\alpha = 30^{\circ}$	ş	$g_x = 2.5$	4.8791	
	E/D = 0.016	$\beta = -40^{\circ}$	ş	$g_y = 2.5$	5.1190	
		$\gamma = 0^{\circ}$	ş	$g_z = 2.61$	2.6080	
Co2	$D = 70 \text{ cm}^{-1}$	$\alpha = 50^{\circ}$	ş	$g_x = 2.4$	4.3875	
	E/D = 0.056	$\beta = 50^{\circ}$	Į	$g_y = 2.4$	5.1901	
		$\gamma = 0^{\circ}$	Į	$g_z = 2.5$	2.4766	
real interaction $(MHz)^b$ effective interaction $(MHz)^b$						
3011.3	3 0	-187.4	7919	1000	6529	
0	2884.5	0	826	11506	1030	
185.4	ŧ 0	-6021.1	-5945	-1331	-13960	

^{*a*}**D** tensors and real **g** matrices for each Co(II) ion are assumed coincident, and the Euler rotation angles are randomly chosen to illustrate the generality of the model. These angles are used in the same way as described in the documentation of the EasySpin program. ¹⁶ ^{*b*}Simulated interaction is magnetic dipolar (eq 5), with the Co(II)–Co(II) vector chosen in the common frame *z* direction and an interion distance of 3 Å. Interaction matrices are both anisotropic and antisymmetric due to the noncoincidence and anisotropy of the **g** and **g**' matrices.

174 the real exchange for several possible pairs of Co(II) (S = 3/2), 175 Fe(III) (S = 5/2) and Cu(II) (S = 1/2) are shown in Table 2, 176 involving different doublets. Where relevant, the *xyz* axes of 177 each *ZFS* matrix are considered coincident. There are many 178 other possibilities, but we consider that inclusion of more of 179 these would not facilitate understanding.

Furthermore, Lines'¹⁷ and others'²⁰ formulations followed from considerations that included crystal field and spin-orbit terms and therefore required a number of implicit and explicit approximations and assumptions. The formulas presented in this work are much more general and arise from the spin Hamiltonian formalism, with the only assumption that $|D_{ij}| \gg |$ 186 $J_{ij}|$, which is not necessarily true for molecular magnets but is almost always true for metalloproteins or any system with highspin ions further away than 4 to 5 Å.

In the case of the effective dipolar interaction, the result that 189 effective g' matrices have to be used will make it much easier 190 to interpret and extract distance information from EPR 191 measurements in systems with high-spin ions at large distances. 192 If HS Co(II) or Fe(III) spin labels were developed and used in 193 combination with S = 1/2 spin labels such as nitroxides, then 194 distances up to about three times larger could be determined, 195 depending on the relative orientation of the spin-spin vector 196 and the metal effective g' matrix. This is due to the relation 197 between effective g' and real g values for high-spin Co(II) or 198 Fe(III). Figure S5 shows simulated EPR spectra of M-nitroxide 199 spin pairs (M = HS Co(II), HS Fe(III), and Cu(II)) with 200 metal-nitroxide distances of 15 Å, considering only the 201 magnetic dipolar interaction. It can be observed that high-spin 202 ions with effective g' matrices cause a larger splitting of the 203 nitroxide signal. The splitting of the high-field nitroxyl signal in 204 the Fe(III)-nitroxyl pair is ~ 2 times larger than in the case of 205 Cu(II)-nitroxyl. This ratio, of course, depends on the relative 206 orientation between the largest g' values in the metal and the 207 spin-spin vector, and, in principle, up to three to four times 208 larger splittings could be observed in Fe(III)-nitroxyl or 209 Co(II)-nitroxyl pairs in comparison with Cu(II)- or 210 nitroxyl-nitroxyl pairs. The observed splittings for the 211 Fe(III)-nitroxyl pair are in line with results by Eaton and 212 Eaton on iron(III) porphyrins with attached spin labels.^{21,22} 213 Fielding et al. analyzed the magnetic interactions present in 214 this Fe(III)-nitroxyl dimer using a dipolar interaction with the 215 Fe(III) anisotropic g matrix. However, to the best of our 216 knowledge, neither in this paper nor in other work was a 217 theoretical justification given for the use of this effective 218 dipolar tensor. This work provides that justification. 219

Given the success of the effective interaction approach for 220 HS Co(II) and Fe(III) ions, as evidenced by the simulations 221 shown in Figure 1 and the SI, it is natural to ask if the model 222 would be useful for highly anisotropic lanthanide ions such as 223 Dy(III) (I = 15/2). The discussion and simulations presented 224 in Section 7 of the SI show that if the crystal-field splittings 225 acting on Dy(III) ions in a dimer are large compared with the 226 interion magnetic interaction then the effective interaction 227 provides a very good description of the lower energy levels of 228 the dimer. Because of the extremely large anisotropy of Dy(III) 229 in some situations ($g'_{x,y} \approx 0$, $g'_z \approx 10-20$), some very small $_{230}$ deviations occur between the real-spin and effective-spin 231 simulations; but in the direction of g'_{z} , where the splitting is 232 largest, these differences are negligible. Because Dy(III) is the 233 most anisotropic lanthanide with the highest J, it is almost 234 certain that the usefulness of the model presented in this work 235 can be extended to other lanthanide ions. 236

Finally, we think necessary to discuss covalency effects, $_{237}$ which at first sight would seem important for ions with highly $_{238}$ anisotropic g' matrices, because spin delocalization into the $_{239}$ ligands will reduce the orbital angular momentum contribu- $_{240}$ tions to the total electronic angular momentum. The $_{241}$ distributed dipole approximation, $^{23-27}$ which is an intermedi- $_{242}$ ate between the point-dipole and full quantum-mechanical $_{243}$ treatment of the dipolar interaction, assumes nonzero spin $_{244}$ populations on the metal ligands and calculates the dipolar $_{245}$ interaction as a sum of point-dipole terms in the following way 246

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Table 2. Effective α_{iso} Factor Relating the Isotropic Part of the Effective and Real Exchange Interactions as a Function of Rhombicity for Several Types of Spin Pairs^{*a*}

	3/2,3/2 ^b	3/2,1/2	5/2,5/2 ^c	$5/2, 1/2^{c}$	5/2,3/2 ^d
	$(\pm 1/2, \pm 1/2)$				
E/D	$(\pm 3/2, \pm 3/2)$	$(\pm 3/2, \pm 1/2)$	$(\pm 3/2, \pm 3/2)$	$(\pm 3/2, \pm 1/2)$	$(\pm 3/2, \pm 1/2)$
0	3	1.666	6.333	2.516	4.333
	3	1.000	3.000	1.732	1.000
0.05	3	1.661	6.332	2.516	4.295
	3	1.097	3.068	1.752	1.743
0.10	3	1.647	6.631	2.575	4.325
	3	1.187	3.422	1.849	2.326
0.15	3	1.624	7.053	2.656	4.395
	3	1.269	3.876	1.969	2.752
0.20	3	1.593	7.453	2.730	4.476
	3	1.341	4.282	2.069	3.028
0.25	3	1.557	7.719	2.778	4.542
	3	1.404	4.531	2.129	3.165
0.30	3	1.516	7.777	2.789	4.569
	3	1.457	4.552	2.133	3.174
1/3	3	1.488	7.678	2.771	4.556
	3	1.488	4.419	2.102	3.115

^{*a*}Parentheses indicate the considered doublets. ^{*b*}High-spin Co(II) (S = 3/2) only shows signals arising from the lower energy doublet. If D > 0, this is the $M_S = \pm 1/2$, and if D < 0, it is the $M_S = \pm 3/2$, which is rarely observed. Nevertheless, the $M_S = \pm 1/2$ option is given above the $M_S = \pm 3/2$ case. ^{*c*}High-spin Fe(III) (S = 5/2) usually shows EPR signals arising from the lower $M_S = \pm 1/2$ doublet and the middle $M_S = \pm 3/2$ doublet, which have different g' values. Both options are given, the $M_S = \pm 3/2$ case below the $M_S = \pm 1/2$ case. ^{*d*}The top possibility is the Co(II) $M_S = \pm 1/2$, Fe(III) $M_S = \pm 1/2$ case and the bottom possibility is the Co(II) $M_S = \pm 1/2$, Fe(III) $M_S = \pm 3/2$ case.

$$\mathbf{D}_{\text{total}} = \sum_{i=1,j=1}^{N_i, N_j} \rho_i \rho_j \mathbf{D}_{ij}(\mathbf{g}_{i'} \ \mathbf{g}_{j'} \ \mathbf{r}_{ij})$$
(6)

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248 where i(j) runs over the atoms in which there is a spin-249 population ρ_i (ρ_i) due to electron 1 (2). \mathbf{r}_{ij} is the distance 250 between the corresponding nuclei. One problem with this 251 equation occurs when there are bridging atoms that have spin-252 populations arising from both paramagnetic centers. In this 253 case, $r_{ii} = 0$ and the dipolar interaction diverges. This is an 254 artifact caused by the semiclassical treatment and could be 255 resolved by an artificial splitting of the spin population of the 256 bridging atom into near but not coincident points. We have 257 previously obtained²⁷ metal and ligand spin populations ρ_i in a 258 copper(II) dimer with Cu-Cu separation of 3.34 Å using the 259 distributed dipole approximation and density functional 260 calculations and found for the metals $\rho_{\rm Cu} \approx 0.65$ and for the ²⁶¹ ligands $\rho_{\rm N,O} \approx 0.6$ to 0.10. This is a large amount of covalency, 262 and indeed we found that the point dipole approximation was 263 not nearly satisfactory, while the distributed dipole approx-264 imation was much better. Density functional theory calcu-265 lations on a nearly isostructural high-spin Co(II) dimer 266 (unpublished results) show a relatively higher spin-population ²⁶⁷ on the Co(II) atoms ($\rho_{Co} \approx 2.7$). When the spin S = 3/2 is 268 projected into an S = 1/2 pseudospin, the effective spin ²⁶⁹ population of the Co(II) ions becomes ${\rho'}_{\rm Co} pprox$ 0.9. For the 270 relatively long Co(II)-Co(II) distances for which the 271 interaction is dominated by a dipolar contribution, this 272 relatively large effective spin population should make the 273 point-dipole approximation more valid. However, given the 274 vast diversity of possible situations and varying structural 275 characteristics, simple rules-of-thumb should wait until enough 276 weakly coupled dimeric systems containing high -spin ions are 277 characterized by techniques such as single-crystal or high-field/ 278 frequency EPR.

CONCLUSIONS

We have developed a general set of formulas for analyzing 280 magnetic interactions of exchange and dipolar origin in systems 281 including high-spin ions that can be treated as effective S' = 1/2 282 spins, including HS Co(II), Fe(III), and Dy(III). The 283 presented formulas do not rely on explicit crystal-field 284 arguments and are easy to apply, given that the effective \mathbf{g}' 285 matrices can be determined or estimated for the individual 286 sites. These formulas provide a bridge between standard 287 frequency (X-band, Q-band) EPR experimental results, usually 288 treated with effective spin-1/2 Hamiltonians, and bulk 289 magnetic measurements, which are usually treated as real- 290 spin systems. We hope that these formulas prove useful to 291 researchers studying magnetic interactions in (bio)inorganic 292 systems containing high-spin ions.

ASSOCIATED CONTENT 294

Supporting Information

The Supporting Information is available free of charge on the 296 ACS Publications website at DOI: 10.1021/acs.jpca.7b11908. 297

Derivation of empirical formulas for the effective *g* values $_{298}$ of high spin Fe(III) ions; derivations of formulas for the $_{299}$ effective magnetic interactions between high-spin ions $_{300}$ treated as 1/2 pseudospins, simulations of some spin $_{301}$ pairs, and analysis of effective interactions in Dy(III) $_{302}$ pairs. (PDF) 303

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313 **Notes**

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319 **REFERENCES**

320 (1) Kahn, O. *Molecular Magnetism*; VCH Publishers: New York, 321 1993.

322 (2) Bencini, A.; Gatteschi, D. Electron Paramagnetic Resonance of 323 Exchange Coupled Systems; Springer: Berlin, 1990; p 287.

324 (3) Eaton, G. R.; Eaton, S. S. In Spin Labeling; Springer: 1989; pp 325 339-397.

326 (4) More, C.; Camensuli, P.; Dole, F.; Guigliarelli, B.; Asso, M.; 327 Fournel, A.; Bertrand, P. A New Approach for the Structural Study of 328 Metalloproteins: the Quantitative Analysis of Intercenter Magnetic 329 Interactions. *JBIC, J. Biol. Inorg. Chem.* **1996**, *1*, 152–161.

(5) Cheesman, M. R.; Oganesyan, V. S.; Watmough, N. J.; Butler, C.
S.; Thomson, A. J. The Nature of the Exchange Coupling between
High-Spin Fe(III) Heme o3 and CuB(II) in Escherichia coli Quinol
Oxidase, Cytochrome bo3: MCD and EPR Studies. J. Am. Chem. Soc.
2004, 126, 4157–4166.

(6) Berliner, L.; Eaton, S. S.; Eaton, G. R. Distance Measurements in
Biological Systems by EPR; Springer Science & Business Media: 2006;
Vol. 19.

(7) Rizzi, A. C.; Neuman, N. I.; González, P. J.; Brondino, C. D. EPR
as a Tool for Study of Isolated and Coupled Paramagnetic Centers in
Coordination Compounds and Macromolecules of Biological Interest. *Eur. J. Inorg. Chem.* 2016, 2016, 192–207.

342 (8) Cucinotta, G.; Perfetti, M.; Luzon, J.; Etienne, M.; Car, P. E.; 343 Caneschi, A.; Calvez, G.; Bernot, K.; Sessoli, R. Magnetic Anisotropy 344 in a Dysprosium/DOTA Single-Molecule Magnet: Beyond Simple 345 Magneto-Structural Correlations. *Angew. Chem.* **2012**, *124*, 1638– 346 1642.

(9) Moreno Pineda, E. M.; Chilton, N. F.; Marx, R.; Dörfel, M.;
Sells, D. O.; Neugebauer, P.; Jiang, S.-D.; Collison, D.; van Slageren,
J.; McInnes, E. J.; et al. Direct Measurement of Dysprosium (III)...
Dysprosium (III) Interactions in a Single-Molecule Magnet. *Nat.*Commun. 2014, 5, 5243.

(10) Long, J.; Habib, F.; Lin, P.-H.; Korobkov, I.; Enright, G.;
Ungur, L.; Wernsdorfer, W.; Chibotaru, L. F.; Murugesu, M. SingleMolecule Magnet Behavior for an Antiferromagnetically Superexchange-Coupled Dinuclear Dysprosium (III) Complex. J. Am. *Chem. Soc.* 2011, 133, 5319–5328.

357 (11) Hudák, J.; Boča, R.; Moncol, J.; Titiš, J. Magnetism of 358 Dinuclear Benzoato Cobalt(II) Complexes Modeled by a General 359 Bilinear Exchange. *Inorg. Chim. Acta* **2013**, *394*, 401–409.

360 (12) Pilbrow, J. R. Effective g values for S = 3/2 and S = 5/2. J.
361 Magn. Reson. (1969-1992) 1978, 31, 479-490.

362 (13) Hagen, W. R. *Biomolecular EPR Spectroscopy*; CRC Press: 2008. 363 (14) We have spent some time trying to derive accurate analytical 364 formulae for the effective g' values for S = 5/2 through a combination 365 of matrix diagonalization and perturbation theory and have reached 366 the conclusion that analytical expressions can be obtained but are too 367 long and complicated to be practical. Instead, we have fit the 368 rhombograms with second-order polynomials to provide compact 369 formulae to use instead of the graphics.

(15) Neuman, N. I.; Winkler, E.; Peña, O.; Passeggi, M. C.; Rizzi, A.
Brondino, C. D. Magnetic Properties of Weakly ExchangeCoupled High Spin Co (II) Ions in Pseudooctahedral Coordination

Evaluated by Single Crystal X-Band EPR Spectroscopy and Magnetic 373 Measurements. Inorg. Chem. 2014, 53, 2535–2544. 374

(16) Stoll, S.; Schweiger, A. EasySpin, a Comprehensive Software 375 Package for Spectral Simulation and Analysis in EPR. *J. Magn. Reson.* 376 **2006**, 178, 42–55. 377

(17) Lines, M. Orbital Angular Momentum in the Theory of 378 Paramagnetic Clusters. J. Chem. Phys. **1971**, 55, 2977–2984. 379

(18) Abragam, A.; Bleaney, B. Electron Paramagnetic Resonance of 380 Transition Ions; Clarendon Press: Oxford, U.K., 1970. 381

(19) Bates, C. A.; Passeggi, M. C. G.; Stevens, K. W. H.; Wood, P. 382 H. Exchange Interaction Between Two Cobalt Ions and a Calculation 383 of the g-shift. *J. Phys. C: Solid State Phys.* **1976**, *9*, 1511. 384

(20) Palii, A. V.; Tsukerblat, B. S.; Coronado, E.; Clemente-Juan, J. 385 M.; Borrás-Almenar, J. J. Microscopic Approach to the Pseudo-Spin- 386 1/2 Hamiltonian for Kramers Doublets in Exchange Coupled Co(II) 387 Pairs. *Inorg. Chem.* **2003**, *42*, 2455–2458. 388

(21) Fielding, L.; More, K. M.; Eaton, G. R.; Eaton, S. S. Metal- 389 nitroxyl interactions. 51. Collapse of Iron-Nitroxyl Electron-Electron 390 Spin-Spin Splitting due to the Increase in the Electron Spin 391 Relaxation Rate for High-Spin Iron (III) when Temperature is 392 Increased. J. Am. Chem. Soc. **1986**, 108, 8194–8196. 393

(22) Rakowsky, M. H.; Zecevic, A.; Eaton, G. R.; Eaton, S. S. 394 Determination of High-Spin Iron(III)–Nitroxyl Distances in Spin- 395 Labeled Porphyrins by Time-Domain EPR. J. Magn. Reson. **1998**, 131, 396 97–110. 397

(23) Bertrand, P.; Camensuli, P.; More, C.; Guigliarelli, B. A Local 398 Spin Model To Describe the Magnetic Interactions in Biological 399 Molecules Containing [4Fe-4S]+ Clusters. Application to Ni- Fe 400 Hydrogenases. J. Am. Chem. Soc. **1996**, 118, 1426–1434. 401

(24) Riplinger, C.; Kao, J. P.; Rosen, G. M.; Kathirvelu, V.; Eaton, G. 402 R.; Eaton, S. S.; Kutateladze, A.; Neese, F. Interaction of Radical Pairs 403 Through-Bond and Through-Space: Scope and Limitations of the 404 Point-Dipole Approximation in Electron Paramagnetic Resonance 405 Spectroscopy. J. Am. Chem. Soc. **2009**, 131, 10092. 406

(25) Smith, S.; Owen, J. EPR Measurements on Exchange-Coupled 407 Pairs of V2+ Ions in KMgF3. J. Phys. C: Solid State Phys. **1971**, 4, 408 1399. 409

(26) Harris, E. EPR of Mn2+ Pairs in MgO and CaO. J. Phys. C: 410 Solid State Phys. **1972**, 5, 338. 411

(27) Neuman, N. I.; Burna, E.; Baggio, R.; Passeggi, M. C.; Rizzi, A. 412
C.; Brondino, C. D. Transition from Isolated to Interacting Copper 413
(II) Pairs in Extended Lattices Evaluated by Single Crystal EPR 414
Spectroscopy. *Inorg. Chem. Front.* 2015, *2*, 837–845.