



Bioscience, **Biotechnology**, and **Biochemistry**

ISSN: 0916-8451 (Print) 1347-6947 (Online) Journal homepage: http://www.tandfonline.com/loi/tbbb20

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To cite this article: Yoshitaka Ano, Roque A. Hours, Yoshihiko Akakabe, Naoya Kataoka, Toshiharu Yakushi, Kazunobu Matsushita & Osao Adachi (2016): Membrane-bound glycerol dehydrogenase catalyzes oxidation of D-pentonates to 4-keto-D-pentonates, D-fructose to 5-keto-D-fructose, and D-psicose to 5-keto-D-psicose, Bioscience, Biotechnology, and Biochemistry, DOI: 10.1080/09168451.2016.1254535

To link to this article: http://dx.doi.org/10.1080/09168451.2016.1254535



Published online: 16 Nov 2016.



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Membrane-bound glycerol dehydrogenase catalyzes oxidation of D-pentonates to 4-keto-D-pentonates, D-fructose to 5-keto-D-fructose, and D-psicose to 5-keto-D-psicose

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Received September 7, 2016; accepted October 24, 2016 http://dx.doi.org/10.1080/09168451.2016.1254535

A novel oxidation of D-pentonates to 4-keto-Dpentonates was analyzed with Gluconobacter thailandicus NBRC 3258. D-Pentonate 4-dehydrogenase activity in the membrane fraction was readily inactivated by EDTA and it was reactivated by the addition of POO and Ca²⁺. D-Pentonate 4-dehydrogenase was purified to two different subunits, 80 and 14 kDa. The absorption spectrum of the purified enzyme showed no typical absorbance over the visible regions. The enzyme oxidized D-pentonates to 4-keto-D-pentonates at the optimum pH of 4.0. In addition, the enzyme oxidized D-fructose to 5-keto-D-fructose, D-psicose to 5-keto-D-psicose, including the other polyols such as, glycerol, D-ribitol, D-arabitol, and D-sorbitol. Thus, D-pentonate 4-dehydrogenase was found to be identical with glycerol dehydrogenase polyol dehydrogenase in (GLDH), a major Gluconobacter species. The reaction versatility of quinoprotein GLDH was notified in this study.

Key words: acetic acid bacteria; glycerol dehydrogenase; 4-keto-D-arabonate production; 4keto-D-pentonate; oxidative fermentation

The newly isolated *Gluconacetobacter liquefaciens* (*Ga. liquefaciens*) RCTMR strains catalyze further oxidation in D-glucose oxidation pathway beyond 2,5-diketo-D-gluconate yielding novel carbonyl sugars.¹⁾ Formation of 4-keto-D-pentonate by the oxidative fermentation of acetic acid bacteria was indicated in the previous papers.^{1–3)} There are three different metabolic pathways forming 4-keto-D-arabonate. Enzymatic oxidation of 4-keto-D-gluconate.¹⁾ Alternatively, 4-keto-D-arabinose is formed after oxidation of D-arabinose by a putative D-aldopentose 4-dehydrogenase.²⁾ The third pathway is the direct oxidation of D-arabonate at the C4-postion.³⁾ There are two different

metabolic pathways forming 4-keto-D-ribonate, one from D-ribose via 4-keto-D-ribose²⁾ and the direct oxidation at the C4-postion of D-ribonate.³⁾ There was no report about D-pentonate 4-oxidizing enzyme until we reported the existence of the novel enzyme in the membrane of acetic acid bacteria.¹⁾ D-Pentonate 4-dehydrogenase is significant because it catalyzes the formation of oxidation products, 4-keto-D-pentonates. It implies that the sequential oxidation pathway of D-glucose by acetic acid bacteria is extending beyond 2,5-diketo-Dgluconate, which had been thought to be the final product in D-glucose oxidation for more than past 50 years. A strong oxidation of D-pentonate was observed in Ga. liquefaciens RCTMR 10¹), whereas other strains belonging to Gluconobacter and Gluconacetobacter genera show variable D-pentonate 4-dehydrogenase activity.2)

Interestingly, the same membrane fraction oxidized D-fructose to 5-keto-D-fructose, as preliminarily shown in the previous paper.²⁾ With respect to D-fructose oxidation to 5-keto-D-fructose, a flavoprotein D-fructose dehydrogenase (FDH, EC 1.1.99.11) from Gluconobacter industrius IFO 3260 (presently Gluconobacte japonicus NBRC 3260) is the sole example of FDH so far reported.⁴⁾ No FDH has been reported in other species over Gluconobacter and Gluconacetobacter. It is also important to solve the question whether a similar enzyme to FDH exists in acetic acid bacteria or an alternative FDH, with a different coenzyme from flavoprotein FDH, catalyzing D-fructose oxidation to 5-keto-D-fructose. Similarly to the case of D-sorbitol oxidation forming L-sorbose, a bound-FAD containing D-sorbitol dehydrogenase (SLDH) from Gluconobacte suboxydans var. a IFO 3254 (presently Gluconobacte thailandicus [G. thailandicus] NBRC 3254) has been characterized.⁵⁾ The bound flavin-dependent SLDH is the sole example of SLDH, which shows a narrow substrate specificity and oxidizes D-sorbitol exclusively. Except for a few examples, membrane-bound dehydrogenases

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A part of this work was presented at the 2016 annual meeting of JSBBA, Sapporo, as 3D014, on March 29 2016.

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in acetic acid bacteria have been mainly classified into two categories, PQQ- or FAD-dependent dehydrogenases.⁶⁾ As an alternative SLDH, PQQ-dependent D-arabitol dehydrogense and SLDH have been purified.^{7,8)} The two PQQ-dependent enzymes show similar and broad substrate specificity to several different polyols. These quinoproteins have been shown to be identical to each other based on their immuno-crossreactivity and also on gene disruption and finally it was suggested to be the same as the previously isolated glycerol dehydrogenase (GLDH) (EC 1.1.99.22).⁹⁾

In this paper, we wish to note that the oxidation of D-pentonate, D-fructose, and D-psicose yielding novel carbonyl sugars is catalyzed by GLDH, a major polyol dehydrogenase, showing novel reaction versatility of the enzyme.

Materials and methods

Chemicals. Potassium D-arabonate was prepared by bromine oxidation of D-arabinose.¹⁰⁾ D-Ribonate and D-erythronate were prepared by delactonization of D-ribono-1,4-lactone (Wako, Osaka, Japan) and D-erythronolactone (Tokyo Kasei, Tokyo), respectively. CM-Toyopearl C-650 was purchased from Tosoh Mfg. (Yamaguchi, Japan). *n*-Dodecyl-β-D-maltoside, *n*-octylβ-D-thioglucoside, and *n*-octyl-β-D-glucoside were purchased from Dojindo (Kumamoto, Japan). Tween 20, Brij 35, Na-cholate, Triton X-100 and sucrose monolarurate were commercially obtained. Mydol 10 and Mydol 12 were kind gift from Kao (Tokyo, Japan). PQQ was prepared and purified as reported previously.¹¹

Microbial strain and culture medium. G. thailandicus NBRC 3258 was mainly used in this study. A basal medium was composed of 2% Na-gluconate, 0.5% glucose, 0.3% glycerol, 0.3% yeast extract (Oriental Yeast, Co, Tokyo), and 0.2% polypeptone. The bacterial culture (150 mL) in 500 mL of an Erlenmeyer flask was done at 30°C under shaking at 200 rpm.

Preparation of membrane and cytoplasmic frac-G. thailandicus NBRC 3258 was cultured in tions 20 L of the basal medium. Cultivation was terminated when the turbidity of the culture medium reached at 350 Klett units, which corresponded to the early stationary phase. A part of cell paste (50 g wet wt.) was suspended in 300 mL of 5 mM acetate buffer, pH 5.0, and disrupted mechanically with a French pressure cell press (Aminco, Silver Spring, MD) at 1000 kg/cm². To the cell homogenate, phenylmethylsulfonyl fluoride was added to 0.5 mM. After unbroken cells were removed by low-speed centrifugation, 3000 rpm for 5 min, the resulting crude cell-free extract was further centrifuged at $10^5 \times g$ for 90 min to separate the membrane fraction from the cytoplasmic fraction.

Enzyme assays. The enzyme activity of D-pentonate 4-dehydrogenase was measured using K-ferricyanide as an electron acceptor, according to the method described by Ameyama.12) The reaction mixture contained 50 µmol of substrate, 100 µmol of acetate buffer, pH 5.0, 10 µmol of K-ferricyanide, and enzyme solution in 1.0 mL. The reaction was started by the addition of K-ferricyanide. The reaction was maintained at 25°C and terminated by adding 0.5 mL of ferric-Dupanol solution. To the reaction mixture, 3.5 ml of water was added and the reaction mixture stood at room temperature for 20 min, before reading the absorbance of the solution at 660 nm. One unit of enzyme activity was defined as the amount of enzyme that catalyzed the oxidation of 1 µmol of D-arabonate per min under the above conditions, which was equivalent to 4.0 of absorbance unit at 660 nm. Alternatively, the enzyme activity was measured with phenazine methosulfate (PMS) and 2,6-dichlorophenol indophenol (DCIP) as the electron acceptors and decrease in absorbance at 600 nm observed at 25°C was recorded by a spectrophotometer. The reaction mixture (1 mL) contained 100 µmol of acetate buffer, pH 5.0, 50 µmol of substrate, 0.3 µmol of PMS, and 0.1 µmol of DCIP. The reaction was started by the addition of PMS and DCIP. In the acetate buffer used, CaCl₂ and PQQ were added to 5 mM and 0.5 µM, respectively, to certify the enzyme as holo-enzyme during the assay. One unit of enzyme activity was defined as the enzyme catalyzing oxidation of 1 µmol of substrate per min under the above conditions. The extinction coefficient of DCIP at pH 5.0, of 3.5 mM⁻¹cm⁻¹ was used. Protein concentration measurement was done by reading optical density at 280 nm and 1% protein solution (10 mg/mL) was arbitrarily defined to be 10.0. The specific activity was defined as enzyme activity units/mg protein. Enzyme activity of D-aldopentose 4-dehydrogenase, aldehyde dehydrogenase (ALDH), and alcohol dehydrogenase (ADH) was assayed with K-ferricyanide under the similar conditions described above using D-ribose, acetaldehyde, and ethanol used as substrate, respectively.

EDTA treatment of membrane. The membrane suspension (20 mg of protein/mL in 5 mM acetate buffer, pH 5.0) was dialyzed overnight in a cold room against the same buffer containing 20 mM EDTA. The membrane suspension was centrifuged three times to remove excess EDTA at $10^5 \times g$ for 60 min. The resulting precipitate was resuspended in 5 mM acetate buffer, pH 5.0, and designated as apo-enzyme. The apo-enzyme was reactivated to the holo-enzyme by the exogenous addition of PQQ and CaCl₂ at 5 μ M and 5 mM, respectively. To complete the holo-enzyme formation, the membrane suspension was maintained at 20°C for 30 min.

TLC chromatography. Analytical TLC cellulose plate (aluminum sheet 20×20 cm, Merck, Darmstadt, Germany) was used. It was done at 25° C with a solvent system of *t*-butanol:formic acid:water = 4:1:1.5. Detection of reaction products, carbonyl sugars, and carbonyl sugar acids possessing intramolecular ketones was done at room temperature by spraying an alkaline ethanol solution of 2,3,5-triphenyltetrazolium chloride over TLC plate.¹³⁾

Purification of D-pentonate 4-dehydrogenase. A11 operations were done at 4°C. The membrane fraction of G. thailandicus NBRC 3258 was homogenized by a glass homogenizer in 5 mM acetate buffer, pH 4.0 and the protein concentration was adjusted to 20 mg/mL. Before enzyme solubilization, it was brought to holoenzyme by adding PQQ and CaCl₂ to the membrane suspension to 5 µM and 5 mM, respectively. Enzyme solubilization was done by adding 1.0% (v/v) Mydol 10 to the membrane suspension. The membrane suspension was stirred gently overnight and centrifuged at $10^5 \times g$ for 60 min. The supernatant was dialyzed for 6 h against 5 mM acetate buffer, pH 4.0 containing 0.2% Mydol 10. Insoluble precipitate was removed by centrifugation before applying the enzyme solution to a CM-Toyopearl C-650 column (1.5×15 cm), which was equilibrated with the same acetate buffer. After washing the column with the buffer, the enzyme was eluted by a linear gradient made by 10 to 35 mM CaCl₂ in the same acetate buffer. Then, the pooled enzyme fraction was concentrated by a centrifugal filter unit (Millipore Centricon, MW 50,000, Millipore Japan) and applied to a Sephadex G-200 column (1.5 \times 120 cm), which had been equilibrated with 5 mM acetate buffer, pH 4.0, containing 5 mM CaCl₂ and 0.2% Mydol 10, and fractionated by every 3 mL fraction.

SDS-polyacrylamide gel electrophoresis. SDS-polyacrylamide gel electrophoresis (SDS-PAGE) was done by a slab gel of polyacrylamide (12.5%, w/v) by the method described by Laemmli.¹⁴⁾ The dual color standard proteins of molecular masses from 250 kDa to 15 kDa (Bio-Rad, Hercules, CA, U.S.A.) were used as references for measurement of the molecular mass of the enzyme.

Amino acid sequencing of D-pentonate 4dehydrogenase. D-pentonate 4-dehydrogenase was electrophoresed in SDS-PAGE, and then the protein bands were electroblotted on to a PVDF membrane. After visualization of protein bands with Ponceau S, the 80-kDa subunit was cut out, and then subjected to amino acid sequencing. Amino acid sequencing was done with a Shimadzu model protein sequencer, PPSQ-21A, according to the instructions of the manufacturer.

Preparation of GLDH mutants. GLDH mutants $(\Delta GLDH \text{ and } \Delta GLDH \text{ paralog})$ were prepared from G. thailandicus NBRC 3255. The gene encoding GLDH (NBRC3255 0235) was disrupted by the same method described previously,15) while knock out mutant of GLDH paralog was prepared as follows: for disruption of GLDH paralog gene (NBRC3255 0026), the disruption vector, pSK-\DeltaParaSldA, was constructed. The 3.1 kb fragment containing NBRC3255 0026 was PCR-amplified with the primer set, 41 ParaSldBA-5' (5'-AAAG-CGGCCGCCGGTGCCACGGGACTTTAT-3') and 42 ParaSldBA-3'(5'-CCGCTCGAGGACGTGACGCTGG-GCTATC-3') from genomic DNA of G. thailandicus NBRC 3255 with TaKaRa Ex Tag DNA polymerase (Takara Bio Inc., Shiga, Japan). The purified PCR

product was digested with NotI/XhoI and ligate to pBluescript SK⁺ vector (Stratagene, CA, USA). The resultant plasmid, pSK-ParaSldBA, was linearlized with EcoRI digenstion and treated with Calf intestine alkaline phosphatase, then ligated with the EcoRI-digested tetracycline-resistant cassette provided from pKRP12 to yield pSK- Δ ParaSldA. To perform homologous recombination in the NBRC3255_0026 locus of genomic DNA, circular pSK-ΔParaSldA was electroporated into G. thailandicus NBRC 3255 at the setting 2.5 kV, 600 Ω , 10 μ F, using a 2 mm gap electrocuvette using MicroPluser (Bio-Rad Laboratories, Hercules, California, USA). The tetracycline-resistant colonies confirmed gene disruption by PCR-amplification with same primer set using genome DNA as a template, comparing with wild-type cells (3.1 kb) and $\triangle GLDH$ paralog cells (5.1 kb).

Results and discussion

Resolution of D-pentonate 4-dehydrogenase with When the membrane fraction of G. thailandicus EDTA. was incubated with D-arabonate and D-ribonate, 4-keto-D-arabonate and 4-keto-D-ribonate predominated as the sole reaction product, as indicated in the previous paper.²⁾ To characterize the enzyme in the membrane fraction whether it was a quinoprotein or a flavoprotein, the effect of EDTA to the enzyme activity was examined with D-arabonate and D-ribonate. EDTA-treatment of the membrane fraction caused a clear decrease in D-pentonate 4-dehydrogenase activity and the decreased enzyme activity was reactivated to almost the original level by the addition of PQQ and Ca²⁺ (Fig. 1). Other species of divalent metallic ions, such as Mg^{2+} , Sr^{2+} , Ba^{2+} , and several kinds of lantanides, were ineffective and no appreciable reactivation of the apo-enzyme was observed. Thus, Dpentonate 4-dehydrogenase was confirmed tentatively to be a quinoprotein, to which PQQ is attached by the aid of Ca^{2+} . On the contrary, in the case of flavoproteins such as D-gluconate dehydrogenase (GADH),^{16, 17)} D-fructose dehydrogenase,⁴⁾ SLDH,⁵⁾, and 2-keto-D-gluconate dehydrogenase,¹⁸⁾ enzyme activity was not affected with EDTA treatment at all.

Resolution of D-fructose 5-dehydrogenase with Using the same membrane fraction that had EDTA. been resolved with EDTA, the profile of D-fructose oxidation was examined under the similar criteria employed above. The native membrane showed a strong D-fructose oxidation irrespective to the presence of PQQ or Ca²⁺ and 5-keto-D-fructose was obtained as the direct oxidation product (Fig. 2). On the other hand, D-fructose oxidation was lost after EDTA treatment. Since the enzyme activity was not recovered only by the addition of PQQ, Ca^{2+} seemed to be essential. These data demonstrate that Ca²⁺ plays an important role to link PQQ with PQQ-binding site in the apo-enzyme. D-Fructose 5-dehydrogenase predominated over Gluconobacter and Gluconacetobacter, suggesting its profound physiological roles in D-fructose metabolism. Enzyme activity of the flavin-dependent FDH⁴⁾ was not resolved with EDTA, while D-fructose 5-dehydrogenase in this study was readily converted to the apo-enzyme. Different from the case of membrane



Fig. 1. Effects of EDTA on D-pentonate 4-dehydrogenase. Notes: D-arabonate and D-ribonate were incubated with native (N), EDTA-treated (E), and the reactivated membrane with PQQ and Ca²⁺ (R). The reaction mixture contained 50 μ L of membrane solution of 20 mg of protein/mL in 10 mM acetate buffer pH 5.0, and 50 μ L of 0.5 M substrate. The enzyme reaction was carried out for 18 h at 30°C with gentle shaking. To the reaction mixture, trichloroacetic acid was added to 10% (v/v). The reaction mixtures were spun down by a tabletop centrifuge. An aliquot of the supernatant was spotted on a TLC plate and developed with a solvent system of *n*-butanol : formic acid : water = 4 : 1 : 1.5. Detection of reaction products possessing ketones was done by spraying an alkaline ethanol solution of 2,3,5triphenyltetrazolium chloride over the TLC plate.



Fig. 2. Effects of EDTA on D-fructose 5-dehydrogenase.

Notes: The native membrane and EDTA-treated membrane were incubated with D-fructose by adding either PQQ or Ca^{2+} , and both of PQQ and Ca^{2+} . D-fructose (F) and 5-keto-D-fructose (KF) were used as the authentic standard. Analytical conditions are the same as in Fig. 1.

enzyme, once the solubilized D-fructose 5-dehydrogenase was resolved with EDTA, it was difficult to resume the lost enzyme activity to the original level by adding PQQ and Ca^{2+} . Though the data are not shown here, D-psicose was also oxidized to 5-keto-D-psicose.



Fig. 3. CM-Toyopearl chromatography of D-pentonate 4-dehydrogenase.

Notes: After the solubilized enzyme solution was dialyzed against 5 mM acetate buffer, pH 4.0, containing 5 mM CaCl₂ and 0.2% Mydol 10, the enzyme solution was applied to a CM-Toyopearl column (1.5 x 15 cm), which was equilibrated with the same acetate buffer. After washing the column with the buffer, the enzyme was eluted by a linear gradient made by 750 ml each of 10 mM to 35 mM CaCl₂. D-Pentonate 4-dehydrogenase was assayed with PMS-DCIP as the electron acceptors. Enzyme activities of 4-aldopentose 4-dehydrogenase (ADH) were assayed with K-ferricyanide as the electron acceptor.¹²⁾

Since a standard 5-keto-D-psicose is not available from commercial sources, the oxidation product was judged putatively to be 5-keto-D-psicose.

Solubilizatioin of D-pentonate 4-dehydrogenase from the membrane fraction. Before solubilization of Dpentonate 4-dehydrogenase, PQQ and CaCl₂ were added to the membrane fraction (20 mg of protein/mL in 5 mM of acetate buffer, pH 4.0) to 5 μ M and 5 mM, respectively, preventing partial resolution of the target quinoproteins. Solubilization of D-pentonate 4-dehydrogenase was done with 1% of various detergents as shown in Table 1. *n*-Octyl-β-D-thioglucoside, *n*-octyl-β-D-glucoside, Mydol 10, and Mydol 12 were effective detergents to solubilize the enzyme. From this result, we selected Mydol 10 for solubilization of D-pentonate 4-dehydrogenase since we had been successful in purification of various quinoproteins using this detergent.^{7,15,19} No appreciable enzyme activity remained in the membrane residue precipitated after solubilization. Solubilized enzyme activity of D-pentonate 4-dehydrogenase was lost almost completely when it was dialyzed against acetate buffer, pH 4.0, in the absence of detergent. These observations strongly indicate that D-pentonate 4-dehydrogenase was a typical membranebound enzyme.

Purification of D-pentonate 4-dehydrogenase. D-Pentonate 4-dehydrogenase was eluted from CM-Toyopearl C-650 at the initial stage of linear gradient of $CaCl_2$ and the enzyme activity appeared as a sharp peak as shown in Fig. 3. As judged from SDS-PAGE



Fig. 4. Gel filtration, absorption spectrum (A) and SDS-PAGE (B) of purified D-pentonate 4-dehydrogenase.

Notes: (A) After the pooled enzyme fraction was concentrated with sucrose, a part of the enzyme solution was applied to a Sephadex G-200 column (1.5×120 cm), which had been equilibrated with 5 mM acetate buffer, pH 4.0, containing 5 mM CaCl₂ and 0.2% Mydol 10, and fractionated by every 3 mL fraction. Absorption spectrum of purified D-pentonate 4-dehydrogenase is shown in the panel. Purified enzyme solution (1.7 mg/mL) in acetate buffer, pH 4.0, was used. Absorption spectrum was taken by a Hitachi Model 200-10 spectrophotometer. (**B**) SDS-PAGE of purified D-pentonate 4-dehydrogenase (5 µg) was developed. The dual color standard proteins of molecular masses from 250 kDa to 15 kDa (Bio-Rad, Hercules, CA, U.S.A.) were used as references for measurement of the molecular masse of the enzyme.



Fig. 5. Reaction versatility of GLDH to D-pentonate, D-fructose, D-psicose, and D-erythronate on the basis of the Bertrand–Hudson's rule. Notes: GLDH has been known to oxidize various compounds, such as glycerol, *meso*-erythritol, D-arabitol, D-sorbitol, D-mannitol, and D-gluconate to dihydroxyacetone, L-erythrulose, D-xylulose, L-sorbose, D-frucose, and 5-keto-D-gluconate according to Bertrand–Hudson's rule.¹⁵⁾ The present study revealed that D-pentonate, D-fructose, D-psicose, and D-erythronate are also oxidized to 4-keto-D-pentonate, 5-keto-D-fructose,²⁵⁾ 5-keto-D-psicose, and 3-keto-D-erythronate by GLDH reaction. Asterisks indicate the putative structure because the authentic standards are not available.

preliminarily done at this stage, D-pentonate 4-dehydrogenase appeared to be highly purified from the purification procedures employed in this study. At this stage of enzyme purification, enzyme activity assayed with K-ferricyanide was lost rapidly. It was caused by the release of co-existed heme c components in the solubilized enzyme solution during the CM-Toyopearl chromatography. Therefore, the enzyme activity of D-pentonate 4-dehydrogenase was no longer assayed with K-ferricyanide and, instead of that, it was assayed with PMS-DCIP as the electron acceptors hereafter, as similarly seen with other examples of membrane-bound quinoproteins.^{9, 19, 20)} On the contrary, D-aldopentose 4-dehydrogenase, ALDH, and ADH were assayed with



Fig. 6. Immunoblotting analysis of D-pentonate 4-dehydrogenase and glycerol dehydrogenase (GLDH).

Notes: The purified enzyme (5 μ g) was developed on SDS-PAGE and then transferred to a PVDF membrane. The enzyme band was visualized using anti-GLDH antibody prepared with purified GLDH from *G. thailandicus* NBRC 3257.



Fig. 7. Effect of gene disruption on D-pentonate 4-dehydrogenase activity.

Notes: The membrane fraction of *G. thailandicus* NBRC 3255 and it variants were prepared by suspending with 5 mM Tris-HCl, pH 7.0. The enzyme activity for D-arabonate (final conc. 50 mM) was measured by K-ferricyanide reductase assay in 50 mM acetate buffer, pH 4.0, after reactivation with 5 mM CaCl₂ and 5 μ M PQQ.

Table 1. Solubilization of D-pentonate 4-dehydrogenase from the membrane fraction of *G. thailandicus* NBRC 3258.

	D-Pentonate 4-dehydrogenase activity (units/mL) assayed with		
Detergent	D-Ribonate	D-Arabonate	
Tween 20	nd	nd	
Brij 35	nd	nd	
Na-cholate	0.095 (28%)	0.093 (14%)	
n-dodecyl-β-D-maltoside	0.065 (19%)	0.012 (2%)	
<i>n</i> -octyl-β-D-thioglucoside	0.347 (102%)	0.605 (9%)	
<i>n</i> -octyl-β-D-glucoside	0.412 (121%)	0.599 (88%)	
Mydol 10	0.370 (108%)	0.576 (85%)	
Mydol 12	0.428 (125%)	0.805 (119%)	
Triton X-100	nd	nd	
Sucrose-monolaurate	nd	nd	
Original membrane	0.341 (100%)	0.676 (100%)	

Notes: Membrane suspension (20 mg of protein/mL in 5 mM acetate buffer, pH 4.0) was mixed with detergent at 1% (v/v). The membrane suspension was gently stirred overnight on a rotary shaker in a cold room. The enzyme activity of D-pentonate 4-dehydrogenase was measured with an aliquot of supernatant (20 μ L) after spinning the membrane suspension down at 10⁵ ×*g* for 60 min. The numbers in parentheses mean the recovery of enzyme activity after solubilization.

K-ferricyanide. To certify the purity of enzyme preparation, active enzyme fractions from CM-Toyopearl chromatography were pooled and concentrated by a Centricon and applied to Sephadex G-200 gel filtration. As shown in Fig. 4, elution of protein by the absorbance at 280 nm coincided with enzyme activity. The enzyme activities oxidizing glycerol, D-sorbitol, D-arabitol, and ribitol were purified for about 50 times from the membrane fraction with 50-60% recovery as summarized in Table 2. The purified enzyme was recovered with a high yield after the simple enzyme purification method employed. The purification and recovery of the minor enzyme activities oxidizing Dpentonates and D-fructose were deduced based on the purification yields against polyol oxidation. Since enzyme activity ratio of oxidizing glycerol to Darabonate or glycerol to D-fructose was constant to be about 10 through the purification steps from the membrane fraction to the purified enzyme, the data strongly suggest that a same single enzyme catalyzed oxidation of all these substrates. It should be noted that D-pentonate 4-dehydrogenase was distinct from GADH of oxidative bacteria, such as *Serratia marcescens*,¹⁶, *Pseudomonas aeruginosa*,¹⁷⁾ and acetic acid bacteria.²¹⁾ D-Pentonate 4-dehydrogenase contained PQQ as the primary coenzyme, while GADH was a flavoprotein containing a covalently bound flavin.²²⁾

Physicochemical and catalytic properties of D-pentonate 4-dehydrogenase. Absorption spectrum of the purified D-pentonate 4-dehydrogenase did not show any appreciable absorption peak over the visible regions (Fig. 4 (A)) and it was alike to that of quinoprotein D-glucose dehydrogenase.^{23,24)} When examined the molecular size of the enzyme with SDS-PAGE, the

Table 2. Summary of purification of D-pentonate 4-dehydrogenase from *G. thailandicus* NBRC 3258.

Step	Substrates	Total units	Total mg	units/mg
Membra	ane fraction			
	Glycerol	159	2580	0.062
	D-sorbitol	191		0.074
	D-arabitol	232		0.09
	D-ribitol	100		0.036
	D-arabonate	15		0.006
	D-fructose	10		0.004
Solubili	ized fraction			
	Glycerol	126	1440	0.087
	D-sorbitol	180		0.125
	D-arabitol	201		0.139
	D-ribitol	88		0.061
	D-arabonate	15		0.011
	D-fructose	13		0.009
CM-To	yopearl			
	Glycerol	99	75	1.328
	D-sorbitol	76		1.013
	D-arabitol	164		2.186
	D-ribitol	59		0.786
	D-arabonate	10		0.134
	D-fructose	8		0.107
Gel filt	ration			
	Glycerol	96	41	2.341
	D-sorbitol	89		2.17
	D-arabitol	146		3.56
	D-ribitol	53		1.292
	D-arabonate	13		0.317
	D-ribonate	6		0.146
	D-fructose	11		0.268
	D-psicose	19		0.463

enzyme was composed of two different subunits of 80 and 14 kDa (Fig. 4 (B)). In addition to D-arabonate, Dribonate, and D-erythronate, D-fructose and D-psicose were oxidized with D-pentonate 4-dehydrogenase. The optimum pH of D-pentonate oxidation was found at 4.0 and apparent km was determined to be 10.5 mM for D-arabonate and 9.7 mM for D-ribonate. D-Erythronate was also oxidized with the enzyme giving 3keto-D-erythronate as a putative reaction product, since the purity of the substrate from a commercial source was not guaranteed. Surprisingly, when reactivity of Dpentonate 4-dehydrogenase was examined with different polyols, a higher enzyme activity was observed with glycerol, D-sorbitol, D-arabitol, and ribitol, as suggested above (Table 2). D-Gluconate was also oxidized at the similar velocity with that of D-arabonate (Data not shown). On the basis of these results from substrate specificity, it was suggested that D-pentonate 4-dehydrogenase was identical to GLDH, a major polyol dehydrogenase, in acetic acid bacteria.¹⁵⁾ D-Pentonate 4-dehydrogenase catalyzed oxidation of a wide variety of substrates which have an erythro-configuration to the bottom primary alcohol residue basing on the Bertrand-Hudson's rule.⁶⁾ D-Pentonate 4-dehydrogenase is different from the membrane-bound SLDH, which is composed of three different subunit structure involving heme c subunit.⁸⁾

Genomic characterization of D-pentonate dehydroge-We prepared the disruptant of pqq1 gene nase. (GOX1857) from G. oxydans 621H and GLDH-like gene (NBRC3255 0026) from G. thailandicus NBRC 3255 and their activities for D-aldopentoses were compared with those of the wild-type strain. Both of the wild-type strains did not show the enzyme activity for D-fructose, indicating that these strains do not have Dfructose 5-dehydrogenase. On the contrary, enzyme activities for D-ribose and D-arabinose were detected in these strains and they are enhanced by the addition of PQQ and CaCl₂. However, the loss of enzyme activity in the mutant was not observed, suggesting that GOX1857 and NBRC3255 0026 were not responsible for D-aldopentose 4-dehydrogenase. From our recent investigations about the substrate specificity, D-aldopentose 4-dehydrogenase was different from GDH. GDH oxidized the C1-position of D-glucose giving D-gluconate but did not oxidize the C4-position of D-aldopentose, while D-aldopentose 4-dehydrogenase oxidized the C4-position of D-aldopentose but not C1-position. As shown in Fig. 3, D-pentonate 4-dehydrogenase was separated clearly from D-aldopentose 4-dehydrogenase in CM-Toyopearl chromatography. Purification and characterization of D-aldopentose 4-dehydrogenase would be the subject in our forthcoming study.

D-Pentonate 4-dehydrogenase oxidized a variety of polyols in accordance with Bertrand–Hudson's rule and D-pentonate was also included among the oxidizable substrates. In addition, SDS-PAGE showed that D-Pentonate 4-dehydrogenase composed of 80- and 14-kDa subunit. Membrane-bound quinoprotein consisting of two subunits is limited in *Gluconobacter* species.

Thus, D-pentonate 4-dehydrogenase could be similar to the major polyol dehydrogenase in acetic acid bacteria such as GLDH (Fig. 5). Alternative confirmation showing D-pentonate 4-dehydrogenase is identical to GLDH was conducted with the following criteria: immuno-blotting analysis of the purified enzyme with GLDH and gene disruption of D-pentonate 4-dehydrogenase activity. To determine the identity of D-pentonate 4-dehydrogenase with GLDH, immuno-blotting analysis was carried out by the method described previously (Fig. 6).¹⁵⁾ The purified D-pentonate 4-dehydrogenase showed immune-cross-reactivity to anti-GLDH antibody with the same level. Moreover, to investigate whether GLDH is involved in D-pentonate 4-dehydrogenation, the enzyme activity was compared with GLDH mutants of G. thailandicus NBRC 3255 (Fig. 7). Two sets of GLDH genes (GLDH, NBRC3255 0235; sldA and GLDH paralog, NBRC3255 0026) are known in G. thailandicus NBRC 3255. The enzyme activity was lost completely in GLDH disruptant ($\Delta GLDH$), whereas the mutant of GLDH paralog ($\Delta GLDH$ paralog) did not affect the enzyme activity. These data indicate GLDH catalyzes D-pentonate 4-oxidation. The N-terminal amino acid sequencing of 80-kDa subunit of D-pentonate 4-dehydrogenase, GAGGE, accorded with the sequence encoded by GLDH gene (NBRC3255 0235; sldA). Provided D-pentonate 4-dehydrogenase is identical to the quinoprotein GLDH, this study has given a complete purification of GLDH for the first time. Because it was not realized in the past 40 years after GLDH purification that became a significant research subject in the research of membrane-bound dehydrogenases.

Author contributions

Osao Adachi conceived and designed the experiments. Yoshitaka Ano and Osao Adachi performed the experiments and prepared the manuscript. Roque A. Hours, Yoshihiko Akakabe, and Naoya Kataoka helped in interpreting data. Toshiharu Yakushi and Kazunobu Matsushita participated in critical revision of the manuscript.

Acknowledgments

The authors wish to thank Dr. N. Nontapong, Ms. A. Kazui, Ms. A. Nishibara and Ms. M. Fujii, for their technical assistances.

Disclosure statement

No potential conflict of interest was reported by the authors.

Funding

This work was supported by Japan Society for the Promotion of Science (JSPS KAKENHI) [grant number 24780074] (to YA), [26450094] (to OA).

References

- Adachi O, Hours RA, Akakabe Y, et al. Production of 4-keto-Darabonate by oxidative fermentation with newly isolated gluconacetobacter liquefaciens. Biosci Biotechnol Biochem. 2010; 74:2555–2558.
- [2] Adachi O, Hours RA, Shinagawa E, et al. Formation of 4-keto-D-aldopentoses and 4-Pentulosonates (4-Keto-D-pentonates) with unidentified membrane-bound enzymes from acetic acid bacteria. Biosci Biotechnol Biochem. 2011;75:1801–1806.
- [3] Adachi O, Hours RA, Shinagawa E, et al. Enzymatic synthesis of 4-pentulosonate (4-keto-D-pentonate) from D-aldopentose and D-pentonate by two different pathways using membrane enzymes of acetic acid bacteria. Biosci Biotechnol Biochem. 2011;75: 2418–2420.
- [4] Ameyama M, Shinagawa E, Matsushita K, et al. D-Fructose dehydrogenase of *Gluconobacter industrius*: Purification, characterization and application to enzymatic micro-determination of D-fructose. J Bacteriol. 1981;145:814–823.
- [5] Shinagawa E, Matsushita K, Adachi O, et al. Purification and characterization of D-sorbitol dehydrogenase from membrane of *Gluconobacter suboxydans* var. a. Agric Biol Chem. 1982;46:135–141.
- [6] Adachi O, Ano Y, Toyama H, et al. Biooxidation with PQQand FAD-dependent dehydrogenase. In Schmid RD, Urlacher VB, editors. Modern biooxidation. Enzymes, reactions and applications. Weinheim: Wiley-VCH; 2007. p. 1–41.
- [7] Adachi O, Fujii Y, Ghaly MF, et al. Membrane-bound quinoprotein D-arabitol dehydrogenase of gluconobacter suboxydans IFO 3257: a versatile enzyme for the oxidative fermentation of various ketoses. Biosci Biotechnol Biochem. 2001;65:2755–2762.
- [8] Choi ES, Lee EH, Rhee SK. Purification of a membrane-bound sorbitol dehydrogenase from *Gluconobacter suboxydans*. FEMS Microbiol Lett. 1995;125:45–49.
- [9] Ameyama M, Shinagawa E, Matsushita K, et al. Solubilization, purification and properties of membrane-bound glycerol dehydrogenase from *Gluconobacter industrius*. Agric Biol Chem. 1985; 49:1001–1010.
- [10] Isbell HS. Oxidation of aldoses with bromine. J Res Nat Bur Stand. 1962;66A:233–239.
- [11] Ameyama M, Hayashi M, Matsushita K, et al. Microbial production of pyrroloquinoline quinone. Agric Biol Chem. 1984;48:561–565.
- [12] Ameyama M. Enzymatic microdetermination of D-glucose, D-fructose, D-gluconate, 2-keto-D-gluconate, aldehyde, and alcohol with membrane-bound dehydrogenases. Meth Enzymol. 1982;89:20–29.
- [13] Lisboa BP. Thin-layer chromatography of steroids, sterols, and related compounds. Meth Enzymol. 1969;15:38–43.
- [14] Laemmli UK. Cleavage of structural proteins during the assembly of the head of bacteriophage T4. Nature (London). 1970; 227:680–685.

- [15] Matsushita K, Fujii Y, Ano Y, et al. 5-keto-D-gluconate production is catalyzed by a quinoprotein glycerol dehydrogenase, major polyol dehydrogenase, in gluconobacter species. Appl Environ Microbiol. 2003;69:1959–1966.
- [16] Shinagawa E, Matsushita K, Adachi O, et al. Membrane-bound D-gluconate dehydrogenase of *Serratia marcescens*. Purification and properties. Agric Biol Chem. 1978;42:2355–236.
- [17] Matsushita K, Shinagawa E, Adachi O, et al. Membrane-bound D-gluconate dehydrogenase from *Pseudomonas aeruginosa*. Purification and structure of cytochrome-binding form. J Biochem. 9179;85:1173–1181.
- [18] Shinagawa E, Matsushita K, Adachi O, et al. D-Gluconate dehydrogenase, 2-keto-D-gluconate yielding, from *Gluconobacter dioxyacetonicus*. Purification and characterization. Agric Biol Chem. 1984;48:1517–1522.
- [19] Moonmangmee D, Fujii Y, Toyama H, et al. Purification and characterization of membrane-bound cyclic alcohol dehydrogenase from Gluconobacter frateurii CHM 9. Biosci Biotechnol Biochem. 2001;65:2763–2772.
- [20] Moonmangmee D, Adachi O, Shinagawa E, et al. L-erythrulose production by oxidative fermentation is catalyzed by PQQ-containing membrane-bound dehydrogenase. Biosci Biotechnol Biochem. 2002;66:307–318.
- [21] Shinagawa E, Matsushita K, Adachi O, et al. D-Gluconate dehydrogenase, 2-keto-D-gluconate yielding, from *Gluconobacter dioxyacetonicus*. Purification and characterization. Agric Biol Chem. 1984;48:1517–1522.
- [22] McIntire W, Singer TP, Ameyama M, et al. Identification of the covalently bound flavins of D-gluconate dehydrogenase from Pseudomonas aeruginosa and Pseudomonas fluorescens and of 2-keto-D-gluconate dehydrogenase from Gluconobacter melanogenus. Biochem J. 1985;231:651–654.
- [23] Matsushita K, Ohno Y, Shinagawa E, et al. Membrane-bound D-glucose dehydrogenase from *Pseudomonas* sp.: solubilization, purification and characterization. Agric Biol Chem. 1980;44: 1505–1512.
- [24] Ameyama M, Shinagawa E, Matsushita K, et al. D-glucose dehydrogenase of *Gluconobacter suboxydans*. Solubilization, purification and characterization. Agric Biol Chem. 1981;45: 851–861.
- [25] Terada O, Tomizawa K, Suzuki S, et al. Formation of 5-dehydrofructose by members of *Acetobacter* species. Part II. Studies on the fermentation, purification and properties of crystal. J Agr Chem Soc Jpn. 1961;35:131–134.
- [26] Adachi O, Matsushita K, Shinagawa E, et al. Purification and properties of methanol dehydrogenase and aldehyde dehydrogenase from *Methylobacillus glycogens*. Agric Biol Chem. 1990; 54:3123–3129.