Fused azulenes as possible organic multiferroics

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We present compelling theoretical results showing that fused azulene molecules are strong candidates for exhibiting room temperature multiferroic behavior, i.e., having both ferroelectric and ferromagnetic properties. If this is experimentally proved, these systems will be organic multiferroic materials with important potential applications.

DOI: 10.1103/PhysRevB.86.180403

PACS number(s): 77.84.Jd, 75.50.Xx, 77.55.Nv

I. INTRODUCTION

Organic electronic materials have come to the forefront since the provocative suggestion of Little,¹ that a high temperature excitonic superconductor can be realized in conjugated molecular systems. This has led to the discovery of organic systems which exhibit a wide range of properties ranging from superconductivity to ferromagnetism. In recent years, in the field of materials, multiferroics have become increasingly important because of the possibility of a wide range of applications. Most common multiferroics exhibit both ferroelectric and ferromagnetic (FM) properties. A few inorganic materials are known to be multiferroics and the most widely known are ferrites and manganites with perovskite structures.² A few purely organic ferroelectrics or exciton multiferroics³ and ferromagnets⁴ are known. However, to the best of our knowledge, there are no organic multiferroic materials. In the quest for organic systems that are potential multiferroic materials, we have discovered theoretically that fused azulene, a π -conjugated system, can exhibit multiferroic behavior.

A simple frustrated conjugated molecule that has been well studied is azulene ($C_{10}H_8$), a molecule with fused five- and seven-membered rings. It also has a dipole moment ~1.0 D attributed to the Hückel "4n + 2" rule.⁵ Azulene oligomers (*n*-azulene) can be made up in multiple geometries, particularly in fused (*n*-C₈H₄) azulene form [see Fig. 1(a)]. Because of the five- and seven-membered rings of the base monomer, all these oligomers are expected to have long-range frustration and show a nontrivial magnetic state with possible low energy spin excitations. Besides, depending on the oligomer, the dipole moment of the azulene monomer can also align, resulting in a molecular ferroelectric phase.

To explore these molecules, we study the low-lying charge and spin energy gaps of azulene oligomers using the long-range interacting Pariser-Parr-Pople (PPP) model and the density matrix renormalization group method (DMRG) technique.⁶ We show that the existence of a magnetic state in this case does not require the presence of flat energy bands, since it also arises within a spin-1/2 antiferromagnetic Heisenberg model (AHM), thus suggesting magnetic frustration as the origin.

We find that the fused azulene ground state is a singlet for oligomers with up to five azulene units, while with more azulene units the ground state spin increases. Our studies show that with more than five units and up to 11 azulene units the oligomers have a triplet ground state. From these results we predict that the ground state spin of the fused azulene increases with the number of azulene units. In the thermodynamic or polymer limit $(n \to \infty)$ we expect fused azulene to be a (nonsaturated) ferromagnet. We suggest that the ferromagnetism in the model comes from the magnetically frustrated geometry of the chain.

In addition, our charge-density calculations show that the ground state of the system has a ferroelectric alignment of the electric dipoles of the monomeric units and that the charge gap is finite but small in the polymer limit of fused azulene. Fused azulenes are large but finite molecules and do not have a center of inversion. This leads to the appearance of an electric dipole moment in one direction (sometimes referred to as a pyroelectric state).

These results show that fused azulenes are excellent candidates to exhibit multiferroic behavior. In the next section, we briefly discuss the model Hamiltonian used in our study. In the last section we present our results and discussions.

II. MODEL HAMILTONIAN FOR FUSED AZULENES

The π -conjugated systems in fused azulene [Fig. 1(a)], azulene polymer [Fig. 1(c)], and oligoacene [Fig. 1(d)] are modeled by considering one p_z orbital on each carbon atom. The PPP model⁷ Hamiltonian is given by

$$H_{PPP} = \sum_{\langle i,j \rangle} t(\hat{a}_{i,\sigma}^{\dagger} \hat{a}_{j,\sigma} + \text{H.c.}) + \frac{U}{2} \sum_{i} \hat{n}_{i}(\hat{n}_{i} - 1) + \sum_{i>j} V_{ij}(\hat{n}_{i} - 1)(\hat{n}_{j} - 1), \qquad (1)$$

where $\hat{a}_{i,\sigma}^{\dagger}$ ($\hat{a}_{i,\sigma}$) creates (destroys) an electron on site *i* with spin σ and \hat{n}_i is the number operator and $\langle i, j \rangle$ implies summation over nearest neighbors.

For the fused azulene case, the pentagons and the heptagons are treated as regular polygons of side 1.397 Å. The transfer integral between bonded sites is taken to be -2.4 eV. The Hubbard parameter U for carbon is fixed at 11.26 eV. The intersite interactions are parametrized using the Ohno⁸ formula. The PPP model with these standard parameters has been

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FIG. 1. Structure of (a) fused azulene, (b) DMRG ladderlike scheme for building fused azulene, (c) polyazulene, and (d) oligoacene.

extensively used for successfully modeling the excitations in a host of conjugated molecules. This model is used throughout this Rapid Communication unless otherwise stated.

The relatively large value of U/t allows us to make the approximation of the active phase space by a single electron per site. This allows checking the magnetic excitations by using a spin-1/2 AHM,

$$H_H = J \sum_{\langle i,j \rangle} \vec{S}_i \cdot \vec{S}_j \quad (J > 0)$$
⁽²⁾

We study this model to compare magnetic excitations in oligoacenes and fused azulenes.

The DMRG⁶ method has proved to be ideally suited to study conjugated polymeric systems, within model Hamiltonian approaches. It has been shown that the DMRG method is accurate even in the presence of long-range interactions if these interactions are diagonal in real space, as with the PPP model.⁹ Many of the interesting conjugated polymers consist of ring systems, such as phenyl, pyridine, thiophene, furan, or pyrrole rings. Building the desired oligomers of such ring systems by adding two sites at each infinite DMRG step is nontrivial and the general method was first illustrated for the case of poly-para-phenylenes.¹⁰ A similar approach has been used for building the molecular systems in this study. We have checked our finite DMRG calculations against the noninteracting model results for all oligomers and for different cutoffs in the number of dominant density matrix eigenvectors, m. Based on these studies, we find m = 500 is optimal.

III. RESULTS AND DISCUSSION

We have carried out the DMRG calculations on the structures shown in Fig. 1(a) (fused azulene), using a ladderlike building block [Fig. 1(b)], Fig. 1(c) (polyazulene), and Fig. 1(d) (oligoacene). For fused azulene the energy per azulene unit linearly extrapolates to -19.117 eV in the thermodynamic limit.



FIG. 2. Spin gaps (in eV) vs $\frac{1}{n}$ defined as $\Delta_{M_S} = E(M_S) - E(0)$, $M_S = 1$, 2, or 3 for the PPP model (a) and for the Heisenberg approximation (b). The shaded regions represent the ground state of a particular spin for the chain lengths above which they occur. The spins of states in the shaded region progressively increase in units of 1 from right to left, for fused azulenes. In the case of oligoacenes, Δ_1 is always positive, implying a singlet ground state.

The oligomers we consider have an even number of sites and a half-filled band, implying states with an integer total spin S. These systems have SU(2) symmetry so we take advantage of the degeneracy of the different spin projections M_S for a total spin S to determine the total spin of the system in its ground state. Using the DMRG method it is straightforward to calculate the energy of states with fixed M_S . The ground state spin of the system is S if the lowest energy states in the $M_S = 0, 1, \ldots, S$ subspaces are degenerate and a gap exists from these to the $M_S = S + 1$ state.

Within the PPP model we computed the lowest energy $E(M_S)$ for spin projections $M_S = 0, 1, 2, \text{ and } 3$ for different n in fused azulenes. In Fig. 2(a) we show the spin gap defined as the energy difference $\Delta_{M_S} = E(M_S) - E(0)$. Fused azulene oligomers with less than five monomers ($n \leq 5$) have a total spin S = 0. For larger oligomers, i.e., $5 < n \leq 11$, the lowest $M_S = 1$ state becomes degenerate (within the accuracy of the calculation), with the lowest $M_S = 0$ state indicating a total spin S = 1 for the ground state. For even larger oligomers, we have strong indications that lowest energy states in higher M_S sectors will become degenerate, showing transitions to larger spin values for the ground state. For example, from the behavior of the gap between the lowest $M_S = 2$ and $M_S = 0$ level, we see that this gap will also vanish when the system

size increases to about ten or 11 azulene units, leading to the spin S = 2 ground state. The gap Δ_3 between the lowest $M_S = 3$ and $M_S = 0$ levels also appears to vanish for even larger systems, making S = 3 the total spin of the ground state. Therefore, it is likely that in the polymer limit, a (nonsaturated) ferromagnetic ground state will result.

Polymers with a triplet ground state are, to the best of our knowledge, not known. Monkman and co-workers¹¹ predicted that a polymer will have a triplet ground state if the lowest singlet exciton energy is below 1.3 eV. This conclusion was arrived at since triplet exciton in their systems were ~ 1.3 eV below the singlet exciton.

To see if the system is a polymeric example of flat-band ferromagnetism, we examined the noninteracting one-particle states of the systems of up to 20 azulene units. We find that the highest occupied molecular orbital (HOMO – k) and lowest unoccupied molecular orbital (LUMO + k) levels come within an energy gap <0.1t with k = 0 for $5 \le n < 10$, k = 1 for $10 \le n < 15$, and k = 2 for $15 \le n < 20$. However, the degeneracy is not sufficiently close to expect exchange correlation to yield a high spin ground state.

An alternative mechanism for the existence of ferromagnetism for larger systems is magnetic frustration stemming from the lattice geometry.¹² To illustrate this we considered a simple spin-1/2 AHM for fused azulenes with a spin-1/2 entity at each site. As seen in Fig. 2(b) the existence of high spin ground states are further confirmed with larger chains using this simplified model. While the number of monomers needed to increase the ground state spin is different compared to the PPP model, both models share the same type of transition. The fact that the purely AHM reproduces these results is a clear indication that this FM state does not have its origin in Fermi level degeneracies.

To further confirm the geometric nature of this FM state we have computed the spin gap Δ_1 using the spin-1/2 AHM on the oligoacene lattice [Fig. 1(d)]. This differs from fused azulene in the fact that all the fused rings are hexagons. The geometry allows for a Ising-like nonfrustrated antiferromagnetic ground state (which is impossible in systems with odd membered rings). In this case, we found that the singlet was always the ground state, with a finite gap to the lowest triplet state

which remained finite in the thermodynamic limit. Similar calculations were earlier performed on oligoacenes using the PPP model.¹³ It was found that the spin gap in the PPP model remained finite in the polymer limit. To explore if extended frustration is essential for a ferromagnetic phase, we carried out calculations on polyazulene [Fig. 1(c)]. In this case we found a singlet ground state for the largest system with eight azulene units in our study and a finite spin gap for the thermodynamic system. These results show the complex nature of the origin of the high spin states in these geometrically frustrated systems. Further studies are necessary to elucidate the nature of the ferromagnetism in these cases.

It is well known that low-dimensional systems can undergo a distortion of the Peierls' type which could lead to stabilization of the singlet state relative to the triplet state.¹⁴ So for a real molecule it is important to see if distortions could destroy the magnetic state. Bond order of a bond (i, j) in the ground state is defined as¹⁵ $\frac{1}{2} \langle \sum_{\sigma} \{ a_{i,\sigma}^{\dagger} a_{j,\sigma} + a_{j,\sigma}^{\dagger} a_{i,\sigma} \} \rangle$. Bonds with large (small) bond orders will have a tendency to become shorter (longer) in the equilibrium geometry. We note that in both the singlet and the triplet states, the bond orders for the peripheral bonds are nearly uniform. This shows that the system is unlikely to undergo a structural distortion. We also note that the bond orders for the bonds that connect the upper and lower polyenelike chains are the smallest, implying that any distortion of the system will only uniformly increase the distance between the upper and lower chains, which is unlikely to alter the energy level ordering of the low-lying states.

To determine the polarized nature of the ground state of fused azulene, we investigate the charge distribution and charge gap in these oligomers within the PPP model. We define the charge of a ring as (p-c) (p = number of sites in the ring; c = total charge in the ring, $\sum_{i \in \text{ring}} \langle n_i \rangle$) for oligomers with up to ten azulene units. We show the net charge in the seven- and five-membered rings in the oligomers in Fig. 3(a). In all these oligomers, seven-membered rings have positive charge while five-membered rings have negative charge. The amplitude of the charge density wave in the system remains approximately the same, independent of the oligomer size, hence we expect a ferroelectric state in the polymer limit. In Fig. 3(b) we give the dipole moment $[\vec{\mu} = \sum_i (\langle n_i \rangle - 1)\vec{r_i}]$ in



FIG. 3. (a) Net charge (p-c) in each ring of fused azulenes with n = 10, 9, and 8 azulene units (top to bottom)—see text. The even (odd) ring index corresponds to a five- (seven-) membered ring. (b) shows dipole moment in the ground state for a different number of azulene units.

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the ground state. We note that when the spin of the ground state changes, there is a drop in the dipole moment. This is due to an increase in covalency of the ground state as the spin of the state increases. Our results show that the fused azulene system in the polymer limit also exhibits a spontaneous polarization. Notice that fused azulene does not have a center of inversion, a necessary condition for the emergence of electric dipolar moment. It consists of a sequence of azulene monomers containing seven- and five-membered rings in a definite order. Thus the polymer necessarily has a seven-membered ring on one end and a five-membered one on the other.

The charge gap measures the energy required to create an independent electron-hole pair in a system. The charge gap for a polymeric system can be obtained by extrapolating $\Delta_c(n) = E^+(n) + E^-(n) - 2E_{gs}^0(n)$ (where *n* is the number of monomers and superscripts "+" and "–" refer to cation and anion while "0" refers to the neutral species) to the thermodynamic limit. The quantity $\Delta_c(n)$ gives the energy required to create a pair of free moving electrons and holes in the ground state.¹⁶ The plot of the charge gap versus $\frac{1}{n}$ is shown in Fig. 4. Extrapolating the charge gap to the thermodynamic limit, we obtain the value $\Delta_c(\infty) = 0.403$ eV. This is smaller than typical charge gaps found for conjugated polymers by almost an order of magnitude.¹³ The existence of a charge gap reinforces the conclusion obtained from the ring charge disproportionation on the existence of a ferroelectric state in the thermodynamic limit.

To conclude, the results presented in this Rapid Communication for the ground state of fused azulenes using the PPP

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FIG. 4. Charge gap (in eV) for fused azulenes vs $\frac{1}{n}$.

and Heisenberg models, namely, the increasing ground state magnetic moment with system size, the presence of a finite but small charge gap for all lengths, and a dipolar moment arising from positively charged seven-membered and negatively charged five-membered rings, show that this system is both ferromagnetic and ferroelectric. Due to the fact that these molecules can be large but finite, we expect that real polymers will behave more as single molecular magnets and electrets. These are compelling results, indicating that the fused azulene system could have multiferroic behavior. This would not only signal these materials as organic multiferroics, but would be also of great importance for future organic device applications.

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