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CHARACTERIZATION OF A PHENOL-DEGRADING BACTERIUM ISOLATED FROM AN INDUSTRIAL EFFLUENT AND ITS POTENTIAL APPLICATION FOR BIOREMEDIATION

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The use of native microorganisms is a useful strategy for phenol bioremediation. In the 25 present work, a bacterial strain, named RTE1.4, was isolated from effluents of a chemical 26 industry. The strain was able to grow at high concentrations of phenol and its derivatives, 27 such as guaiacol, 2.4-dichlorophenol and pentachlorophenol, as well as in a medium 28 containing industrial effluents. This bacterium was identified as Acinetobacter sp. using 29 morphological, physiological, biochemical and 16S rRNA gene analysis. Acinetobacter sp. 30 RTE1.4 degraded phenol (200 to 600 mg/L) at wide pH range and temperature (5-9 and 25-31 37 °C, respectively) demonstrating high adaptation ability to different conditions. The strain 32 would metabolize phenol by the ortho-pathway since catechol 1,2-dioxigenase activity was 33 detected. When bacteria were grown in medium containing phenol, an altered whole-cell 34 protein pattern was observed by SDS-PAGE, with the lack of some low-molecular mass 35 polypeptides and an increase in the relative abundance of high-molecular mass proteins after 36 treatment. Considering that the use of native strains in bioremediation studies shows several 37 ecological advantages and that the studied bacterium showed high tolerance and 38 biodegradation capabilities, Acinetobacter sp. RTE1.4 could be an 39 appropriate microorganism to improve bioremediation and biotreatment of areas polluted with phenol 40 and/or some of its derivatives. Moreover, the establishment of the optimal growth conditions 41 (pH, temperature, concentration of the pollutant) would provide base line data for bulk 42 production of the strain and its use in bioremediation processes. 43

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Keywords: phenol removal, optimal growth conditions, protein patterns, catechol 1,2 dioxigenase, *ortho* pathway

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51 **1. INTRODUCTION**

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53 Phenol is a toxic substance frequently found in aquatic environments due to 54 contamination derived from many industrial processes, such as oil refineries, chemical 55 industries, explosives and dye manufacturing, wood preservation, pulp and paper production, 56 and coke ovens.

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Phenol is toxic to several aquatic organisms including anuran amphibians, fishes and 57 crustaceans [1,2]. It is also toxic to humans, inducing carcinogenicity and causing 58 reproductive and developmental toxicity, neurotoxicity, and acute toxicity. Accidental 59 exposure to high phenol concentrations may cause severe local effects, systemic 60 intoxications and even death [3]. Consequently, it is reasonable to assume that living 61 organisms may be at risk when they are exposed to water contaminated with phenol. In this 62 context, due to its high toxicity, phenol has been listed as a priority pollutant by different 63 regulatory agencies such as the US Environmental Protection Agency [4] and the Agency for 64 Toxic Substances and Disease Registry [5]. Therefore, phenol remediation is important for a 65 safe environment. 66

Current chemical procedures for phenol removal from fluid streams involve expensive 67 treatments such as distillation, liquid-liquid extraction with different solvents, adsorption, 68 membrane pervaporation and membrane-solvent extraction [6]. However, in the past few 69 decades, with advances in biotechnology, bioremediation has become one of the most 70 rapidly developing fields of environmental restoration, utilizing microorganisms to reduce the 71 concentration and toxicity of various chemical pollutants. This approach represents a good 72 alternative compared with physico-chemical strategies, which have high costs and produce 73 other toxic end products [7]. Thus, the role of biotechnology in bioremediation is to efficiently 74

apply existing resources available in nature to clean up contaminated environments. In this 75 sense, scientists are making efforts to find adequate microorganisms, with high metabolic 76 activities that would be able to efficiently remove pollutants. Bioremediation must address 77 multiphasic and heterogenous environments, including clean-up of ground water, soils, 78 lagoons, sludges, and process-waste streams, where there has been either accidental or 79 intentional release of pollutants that pose a risk to human, animal or ecosystem health. 80 Because of this complexity, successful bioremediation is dependent on having proper 81 microorganisms in the right place with an adequate control of environmental factors for 82 degradation to occur [8]. Among the several factors that directly impact on bioremediation are 83 energy sources (electron donors), electron acceptors, nutrients, pH, temperature, and 84 inhibitory substrates or metabolites. Even with some obstacles, bioremediation has been 85 successfully applied for efficient clean-up of environments contaminated with inorganic or 86 organic xenobiotics. Regarding organic compounds, phenol and its derivatives has a high 87 toxicity, as it was pointed out before, hence, there is an increasing interest in isolating and 88 identifying microorganisms with phenol metabolizing capacity. Bioremediation of phenolic 89 compounds can be carried out by many microorganisms including bacteria and fungi, which 90 have been notified as phenol degrading agents and they are often able to completely 91 degrade this pollutant. Among them, bacteria from different genera, such as Pseudomonas, 92 Vibrio, Serratia, Bacillus and Achromobacter are able to metabolize phenol [9]. Some of 93 these microorganisms have the ability to completely mineralize phenol through the action of 94 several enzymes. In fact, phenol can be metabolized by two major pathways known as ortho 95 and meta cleavage [10]. Phenol hydroxilase represents the first enzyme in the metabolic 96 pathway of phenol degradation [11]. This enzyme catalyzes the oxidation of phenol to form 97 catechol. In the next step, two enzymes can be induced, catechol 1.2- or 2.3-dioxigenase, 98 which belong to the ortho and meta ring fission pathways, respectively [10,11]. This ability to 99 use phenol as carbon and energy source is exploited to bioremediate soil, water or effluents 100

101 contaminated with this compound. However, phenolic compounds can frequently inhibit 102 microbial growth. Thus, the optimization of degradation conditions (pH, temperature, 103 concentration of pollutants) is necessary for bio-oxidative processes in industrial and 104 environmental applications.

Phenol degrading microorganisms are frequently isolated from phenol-polluted 105 environments. However, to our knowledge, the isolation of native strains from the proximity of 106 Ctalamochita River (Río Tercero, Argentina) and its use for phenol bioremediation has not 107 been documented. This river receives effluents from domestic activities and mainly from 108 several chemical and petrochemical industries and represents a dangerous polluted area. 109 So, this site was selected to isolate microbes which, after a deep study, could be returned to 110 the same place for bioremediation/bioaugmentation purposes and even more for the 111 treatment of other polluted areas. 112

Thus, the aims of the present study were to (a) isolate and characterize, indigenous bacteria capable to tolerate and degrade phenol, from industrial effluents near Ctalamochita River; (b) investigate phenol degradation as well as establish optimal degradation conditions; (c) study the kinetics of biodegradation, the possible degrading pathways and changes in total protein profiles, in the presence of the pollutant.

- 119 2. MATERIALS AND METHODS
- 121 2.1 Sample collection
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Bacterial strains were obtained from effluents of chemical industries located near Río Tercero city (Córdoba Province), Argentina. Effluent samples were collected in sterile bottles from the end of a channel discharging wastewater into Ctalamochita River.

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For selection of phenol-degrading strains, appropriately diluted water samples (10⁻³) 129 were spread on agar plates with TY medium $[(g/L): 5 \text{ tryptone}; 3 \text{ yeast extract}; 0.65 \text{ CaCl}_2; 13$ 130 agar] supplemented with 100 mg/L phenol (Merck). Cultivation was carried out for 48 h at 131 28±2 °C. Morphologically different colonies obtained from the plates were transferred to agar 132 mineral medium (MM) plates with phenol (100 mg/L). Subsequently, isolated colonies 133 exhibiting visible growth and different morphology were selected and stored for further 134 studies. MM used in these assays contained (g/L): 2.8 Na₂PO₄H; 9 KPO₄H₂; 2.5 NaCl; 1 135 NH₄Cl; 13 agar. Phenol was used as sole source of carbon and energy. 136

An isolated colony, named RTE1.4 until its identification, with capability to fast growth in MM agar containing phenol was selected and identified for further characterization and phenol degradation studies.

2.3 Screening of tolerance for phenol and its derivatives

Phenol tolerance as well as tolerance to different phenolic compounds such as 2,4dichlorophenol (2,4-DCP), guaiacol and pentachlorophenol (PCP) was evaluated. For that, the isolated bacterium was spread in MM agar supplemented with phenol (100 to 1000 mg/L), 2,4-DCP (100-1000 mg/L), guaiacol (50-1000 mg/L) and PCP (20-400 mg/L). The maximum tolerated concentration (MTC) was established as the highest concentration of the contaminant at which bacteria could grow, after 7 d at 28±2 °C.

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150 2.4 Bacterial tolerance in a cork processing wastewater

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RTE1.4 strain was tested for its ability to grow in the presence of an effluent derived from the cork industrial process. This effluent usually contains some cork wood extracts, such as phenolic acids, tannic fraction, 2,4,6-trichloroanisol and PCP [12]. Exhausted cork boiling wastewater (after being used for 20-30 cycles) was collected from cork transformation industry. Solid MM was supplemented with 10, 25, 50 and 100% of this liquid effluent. Plates were spread with the strain and incubated at 28±2 °C for 72 h. The bacterial growth in these plates was evaluated.

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160 2.5 Characterization and identification of bacterial strain

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RTE1.4 strain was subjected to several morphological, physiological and biochemical tests for its characterization and identification.

The isolate was characterized by Gram staining kit (Britania) using an Axiolab (Zeiss) microscope. To determine oxidase activity, isolated single colonies were picked from the plate and were gently scratched on oxidase disks (Britania). The change to purple colour of the disc was indicative of a positive test. On one single colony, suspended in 0.9% NaCl, 3% H_2O_2 was dropped and the appearance of effervescence caused by free oxygen release demonstrated the presence of catalase enzyme.

A commercially available kit for bacterial biochemical analysis was used in this study (API 20 NE system, BioMerieux[®] SA). The protocols followed the manufacturer's instructions. The kit monitors 8 conventional and 12 assimilation tests. Results of the biochemical test were analyzed with API web program, which allows identification of the studied microorganism.

A genotypic study was used to confirm the identification of RTE1.4 strain. For that, a PCR analysis for 16S rRNA gene was performed with a pair of forward (5' -CCAGCAGCCGCGGTAATACG -3') and reverse (5' -TACCAGGGTATCTAATCC- 3') primers

- and the PCR product was sequenced. All the procedures were carried out by "MacroGen"
 company (Korea). The sequence was compared using BLAST program [13,
 <u>http://www.ncbi.nlm.lih.gov</u>] and RDP [14, <u>http://www.rdp.cme.msu.edu</u>] and the results were
 deposited in GenBank, under the accession number JF304543.
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183 2.6 Phylogenetic analysis

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A phylogenetic tree was constructed based on the 16S rRNA gene sequences. First, multiple alignments of the different 16S rRNA gene sequences downloaded from GenBank were performed using CLUSTAL-X software. The evolutionary distances were calculated by the Tamura-Nei model. Phylogenetic tree was constructed by Neighbor-Joining method by MEGA4 software [15] and a bootstrap analysis of up to 1000 iterations were carried out.

2.7 Phenol biodegradation experiments

Biodegradation studies of RTE1.4 strain were carried out at different pH, temperature 193 and initial phenol concentrations. For all the experimental designs, the bacterial strain was 194 grown in TY liquid medium until late exponential phase. Then, this culture was used to 195 inoculate Erlenmeyer flasks (to give an initial optical density (OD) of 0.05) containing 30 mL 196 of liquid MM with phenol as sole carbon source. The flasks were placed in a shaker at 100 197 rpm at controlled temperature. Samples were taken at different intervals depending on each 198 experiment and analyzed for bacterial biomass and phenol consumption, until phenol was 199 completely removed. All experiments were carried out by triplicate set for checking the 200 consistency of data. 201

To establish optimum growth and phenol biodegradation conditions, bacterium was incubated in liquid MM plus 200 mg/L phenol under different temperatures (from 25 to 37 °C) concentration was determined as described in section 2.8.
 Finally, the potential of RTE1.4 for phenol degradation and the effect of the initial
 concentration on its degradation ability were assayed under optimal pH and temperature
 previously determined, with phenol concentrations from 200 to 700 mg/L.

and pH (from 5 to 11). Assays were performed for two days and then, residual phenol

- 209 All experiments were carried out by triplicate and repeated three times.
- 210 2.8 Phenol determination
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In all experiments, phenol concentration was estimated by a colorimetric assay [16] based on its rapid condensation with 4-aminoantypirine, followed by oxidation with potassium ferricyanide under alkaline conditions to give a red colour product, which was measured at 510 nm wavelength using Beckman DU640 UV-Vis spectrophotometer. Phenol concentrations were calculated using a calibration curve, which was carried out with known concentrations of pure phenol.

2.9 Elucidation of phenol degradation pathway

RTE1.4 strain was grown in liquid MM containing 600 mg/L of phenol until it reached 221 50% of phenol removal. This culture was centrifuged at 12,000g at 4 °C for 10 min. To 222 prepare crude extracts, cell pellets were washed twice with 0.9% NaCl and resuspended in 223 Tris-HCl buffer (pH 9) obtaining 0.1 g/mL of wet cell biomass. These cells were disrupted at 4 224 $^{\circ}$ C for 6 cycles of 0.5 min each, with a gap of 0.5 min between the cycles, using a 225 Sonic&Materials VibraTMCells sonicator. Unbroken cells and cell debris were removed by 226 centrifugation at 24,000g at 4 °C for 30 min. The supernatant was used as enzyme crude 227 source for all the assays. The enzyme activity was spectrophotometrically determined in 228 potassium phosphate buffer (0.1 M, pH 7.5) adding 15 µl of the crude extract and 7.5 µl of 229

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catechol (10 mM) as substrate. Catechol 1,2-dioxigenase and catechol 2,3-dioxigenase
activities were estimated by measuring the reaction products cis,cis-muconic acid at 260 nm
and 2-hydroxy muconic semialdehyde at 375 nm, respectively, as described by Pradhan and
Ingle [17]. One unit (U) of enzyme activity was defined as the amount which catalyzed the
formation of 1 µmol of product/min at 25 °C. Specific activity was calculated as Units per mg
of protein. Proteins were determined by Bradford method with bovine serum albumin as a
standard [18].

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238 2.10 Analysis of changes in protein profiles induced by phenol

240 Cultures grown in liquid MM with different initial phenol concentrations (200, 400 and 600 mg/L) and a control culture, grown with 500 mg/L of L-lactate as a carbon source, were 241 collected, in exponential growth phase, by centrifugation at 10,000g for 10 min at 4 °C. Then, 242 the pellet was washed twice with 0.9% NaCl and centrifuged again in the same conditions. 243 The pellets were resuspended in Tris-HCI buffer (pH 7.5) and disrupted by sonication on ice 244 using 6 cycles of 0.5 min each, with a gap of 0.5 min between the cycles. After sonication, 245 suspensions were centrifuged at 20,000 for 20 min at 4 °C, and the supernatants were 246 collected. Protein concentrations were measured as was described before. 247

The protein patterns of cell extracts were analyzed by polyacrilamide gel electrophoresis (PAGE) under denaturizing conditions according to Laemmli [19], using 25 µg of protein per lane. The acrylamide/bis-acrylamide concentration was 4% and 12% (W/V) in the stacking and resolving gels, respectively. The gel was stained with Coomasie Brilliant Blue R-250. To determine molecular mass values, a Kaleidoscope Prestained Standard (Bio-Rad) was used.

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255 2.11 Statistical analysis

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257 Statistical analysis was performed using STATISTICA 7.1 software package. All the 258 data were analyzed using ANOVA. In all cases $p \le 0.05$ was statistically significant. Dunnett 259 test was used for comparing several treatment groups with a control.

3. RESULTS

3.1 Isolation of bacterial strains and phenol tolerance

Four morphologically different microbial isolates were able to grow on TY agar medium 268 plates supplemented with 100 mg/L of phenol. These strains were tested for their capability 269 to grow in the presence of phenol (100 mg/L) as sole carbon source, in MM agar plates. One 270 of them, RTE1.4, was selected for further studies because it showed a fast and visible growth 271 in this medium, whereas the other three isolates were not able to grow in this condition. 272 When this isolate was spread in plates containing MM agar with different phenol 273 concentrations (100 to 1000 mg/L), RTE1.4 was able to tolerate concentrations as high as 274 1000 mg/L after 48 h at 28±2 °C. 275

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3.2 Tolerance assay using phenol derivatives and an industrial wastewater

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Tolerance of the isolated strain to various phenolic derivatives was also analyzed. After 7 days, the MTC for RTE1.4 were 1000 mg/L for guaiacol, 250 mg/L for 2,4-DCP and 200 mg/L for PCP.

In addition, growth of RTE1.4 in the presence of cork processing wastewater, containing 283 20 mg/L of total phenolic compounds was evaluated. After 72 h of incubation, this strain 284 proved to be tolerant to this wastewater and showed ability to grow even in plates containing 285 pure effluent.

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3.3 Characterization and identification of bacterial strain

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289 RTE1.4 strain was characterized based on morphological, physiological and 290 biochemical assays. On MM agar colonies were 1-2 mm in diameter and white-yellow. After 291 single colony isolation the strain was characterized as Gram negative, oxidase-negative and 292 catalase-positive, bacilli. Table 1 shows full details of biochemical and physiological 293 characteristics of RTE1.4 strain. Based on the results of the API 20 NE biochemical test, this 294 microorganism was identified as *Acinetobacter Iwoffi* with a 99% of identity.

By PCR amplification, 1388 bp 16S rRNA gene fragment of RTE1.4 strain was obtained and submitted for sequencing. The BLAST and RDP search indicated that RTE1.4 was closely related to species of *Acinetobacter* genus.

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3.4 Phylogenetic analysis

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Figure 1 shows the phylogenetic relationship of the RTE1.4 strain with other *Acinetobacter* strains, based on 16S rRNA gene sequences. As can be seen, the 16S rRNA gene sequence of RTE1.4 strain showed closest relation to three strains of *Acinetobacter sp* and two strains of *A. tandoii*, forming a cluster in which the homology between these species was high (97-99%). RTE1.4 also showed high homology with several *Acinetobacter* species (such as *A. schlinderi*, *A. lwoffii*, *A. baumanii*, among other) but these bacterial species were

- included in separate clusters. Taking into account these results, we consider that RTE1.4 307 strain is a typical member of Acinetobacter genus. 308
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3.5 Effect of pH, temperature and initial phenol concentration on biodegradation 310

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Factors affecting phenol degradation capability of Acinetobacter sp. RTE1.4 strain such 312 as pH, temperature and initial phenol concentration were investigated. 313

The strain was able to degrade phenol (200 mg/L) in all the tested pH (5 to 11) at 28 ± 2 314 °C and the highest phenol degradation (100%) was achieved at the neutral condition (Figure 315 2). At pH 5, 6, 8 and 9, degradation was still high (above 50%). However, at pH 10 and 11, a 316 very low phenol degradation was observed and only 10 mg/L of the pollutant were removed. 317

Similarly, optimal temperature for phenol removal was determined incubating bacterial 318 strain at different temperatures (25-40 °C), at pH 7, with 200 mg/L of the pollutant and 319 monitoring residual phenol concentration (Figure 3). The optimal temperature registered was 320 28 °C, reaching 100% of phenol removal after 48 h and showing statistically significant differences respect to the other assayed temperatures. In addition, at 25 °C phenol 322 degradation was around 80%. Nevertheless, when incubation was carried out at 33 or 37 °C, 323 low phenol degradation was observed (40%), while at 40 °C the removal was as low as 5%. 324 According to these results, the optimal conditions for phenol removal by Acinetobacter sp. 325 RTE1.4 strain were pH 7 and 28 °C, and they were selected for further experiments. 326

As it is shown in Figure 4a, initial phenol concentrations were varied from 200 to 700 327 mg/L to analyze and compare the degree of biodegradation and the time required for 328 complete degradation whereas Figure 4b shows growth curves at different phenol 329 concentrations. The time required for phenol degradation significantly increased as a function 330 of initial phenol concentration in the culture medium. Total degradation of 200 mg/L phenol 331 occurred after 48 h and a short lag phase was observed (6 h, data not shown). The 332

degradation time was prolonged to 7 and 11 d, for 400 and 600 mg/L of phenol, respectively,
whereas only 28% of phenol degradation was achieved for 700 mg/L after 13 d. In this case,
the strain showed a 6 d *lag* phase to grow and consume phenol.

Thus, the strain was able to tolerate and degrade high phenol concentrations, removing completely 200 to 600 mg/L of this contaminant at 28 °C and pH 7, which were determined as optimal conditions.

- 339
- 340 3.6 Elucidation of phenol degradation pathway
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The catechol 1,2-dioxigenase and catechol 2,3-dioxigenase activities were examined in cell extracts obtained from *Acinetobacter* sp. RTE1.4. The increase in absorbance at 260 nm was observed in the supernatant of cell suspensions containing catechol as substrate, indicating formation of cis,cis-muconic acid. Specific activity of catechol 1,2-dioxigenase was 0.058 U/mg protein whereas 2,3-dioxigenase activity was not detected. These results indicated that *Acinetobacter* sp. RTE1.4 metabolized phenol via *ortho*-pathway while the *meta*-cleavage pathway was discarded.

3.7 Analysis of changes in protein profiles induced by phenol

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The SDS-PAGE showed several polypeptides which ranged from 17 to 200 kDa. Growth in the presence of phenol altered the whole-cell protein patterns compared with cell growing in a preferential carbon source such as L-lactate. Some protein bands disappeared, other appeared and some of them varied their intensity (Figure 5). For instance, the control pattern showed a small protein of 18.6 kDa, which was absent in phenol treated cells. Moreover, this pattern presented a lower number of bands. Despite the concentration used, phenol treated cells showed similar protein patterns. The most relevant differences were observed in high molecular mass proteins (120 and 145 kDa), which did not appear in control cells. Besides, 600 mg/L of phenol produced a protein pattern with many differences compared with 200 and 400 mg/L, such as less intensity in bands of 28.2 and 145 kDa and more intensity in a band of 20 kDa.

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364 4. DISCUSSION

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During the past few decades the incidence and threat of phenol pollution has resulted in extensive research. In this context, the use of native microbial communities constitutes a useful strategy for phenol bioremediation. In the present study, four microorganisms with ability to grow *in vitro* in liquid MM supplemented with phenol, have been isolated from an industrial effluent, which is discharged into Ctalamochita River. One of them (RTE1.4) was selected by its high phenol tolerance.

In industrial processes, a number of other toxic compounds besides phenol are formed, 372 giving multicomponent composition to wastewaters. Therefore, the strains used for 373 decontamination should not only be tolerant and efficient for phenol removal but they should 374 also be tolerant to the remainder of the pollutants. In this sense, bacteria have adaptively 375 evolved by developing catabolic pathways to utilize a variety of compounds that are available 376 in the natural environment [20]. Many bacteria have the metabolic potential to degrade 377 several aromatic compounds. Although the metabolic pathways are very diverse, they are 378 channeled into a limited number of key intermediates, such as catechol and substituted 379 catechols [21]. Exploring the tolerance of Acinetobacter sp. RTE1.4 to different phenol 380 compounds such as guaiacol, 2,4-DCP and PCP as well as to effluents from the cork 381 industry, we found that this strain tolerated them at high concentrations and also showed 382 tolerance to pure effluent. Taking into account that the above mentioned wastewaters 383 contained phenolic compounds (20 mg/L), the results obtained are consistent with bacterial 384

ability to degrade several phenolic compounds. Considering that polyphenols in cork processing wastewater may form chemical complexes which may have an inhibitory effect on the microbial growth [22], the obtained results are promissory. Therefore, the tolerance showed by the isolated strain could indicate that it could be used to remediate complex effluents, prior to discharge into the environment.

Using different approaches such as 16S rRNA gene sequencing and phylogenetic analysis, this strain was identified as *Acinetobacter* sp. In addition, through the use of biochemical tests, this microorganism was identified as *Acinetobacter Iwoffi* with a 99% of identity. However, for the identification of species from a specific genus, sequencing of different genes should be performed. These experiments are currently undergone in our laboratory. Thus, until now, we tentatively classified this strain as *Acinetobacter* sp RTE1.4.

In the present work, several factors which could affect phenol degradation capability of 396 the bacterial strain, such as pH, temperature and initial phenol concentration were 397 investigated. When Acinetobacter sp. RTE1.4 was exposed to the lowest phenol 398 concentration assayed (200 mg/L), a short lag phase was observed (6 h) and complete 399 degradation of phenol was reached after 48 h, whereas for 600 mg/L of the contaminant, the 400 strain showed a 4-d-lag phase to grow and 11 d to complete phenol consumption. This 401 prolonged lag phase could be attributed to the toxic effect of phenol on cells. Pradhan and 402 Ingle [17] showed that Serratia plymuthica strain GC had 4-d-lag phase when it was grown in 403 the presence of high phenol concentrations (1050 mg/L). Similar events were shown by 404 Wang et al. [23], which also remarked that phenol toxicity, at high concentrations, could 405 inhibit the growth of free cells resulting in a lower removal efficiency of the pollutant. 406 Additional support was given by Geng and Lim [24], which indicated that high phenol 407 concentration treatments include not only inhibition of bacterial growth but also the up-408 regulation of oxidative stress proteins, heat shock proteins, and a membrane protein, the 409 ABC type sugar transporter. Based on these findings, it is clear that phenol consumption is 410

strongly dependent on the initial phenol concentration due to toxic effects induced by the 411 pollutant. Despite of phenol toxicity. Acinetobacter sp. RTE1.4 was able to tolerate and 412 degrade high phenol concentrations, removing completely 200 to 600 mg/L of this 413 contaminant at 28 °C and pH 7, which were determined as optimal conditions. However, 414 Acinetobacter sp. RTE1.4 could also degrade phenol at wide pH range and temperature 415 conditions (5-9 and 25-37 °C, respectively) demonstrating high adaptation ability to different 416 conditions. This is a suitable feature for microorganisms selection to be used for the 417 biotreatment of effluents, because they could have periodic fluctuations in the inlet substrate 418 concentration, influent flow rate, aeration rate, pH and temperature. Some problems in the 419 bioremediation process are frequently associated with such fluctuations since these factors 420 421 play an important role in the establishment and maintenance of the microbial communities which can remove the organic constituents [25]. 422

As it is well known, the aerobic metabolism of phenol is initiated by two different 423 pathways, either the ortho- or meta-pathway. The final products of both pathways are 424 molecules that can enter the tricarboxylic acid cycle. In this study, the enzymatic 425 determinations showed that Acinetobacter sp. RTE1.4 only exhibited catechol 1,2-426 dioxigenase activity, which indicated that this strain would metabolize phenol by the ortho-427 pathway. This result is in agreement with those described by Zaki [26], which showed that 428 high phenol hydroxylase and catechol 1,2-dioxigenase activities were detected in 429 Acinetobacter isolates. 430

One approach to understand the ability of bacteria to tolerate xenobiotic stress has been the identification of stress-induced changes of individual proteins under the assumption that stress adaptation results from alterations in gene expression. When we analyzed SDS-PAGE profiles, many changes were detected after phenol treatment. A large number of proteins were observed in the presence of this contaminant in all the assayed concentrations, compared to control cells growing in L-lactate medium. Therefore, both qualitative and

quantitative differences were detected, mainly the lack of some low-molecular mass 437 polypeptides and an increase in the relative abundance of high-molecular mass proteins after 438 439 phenol treatment. In addition, protein profiles derived from cells treated with 200 and 400 mg/L of phenol, showed slight differences, mainly in the intensity of bands, compared with 440 those grown with 600 mg/L of phenol. Similarly, Soussi et al. [27] and Unni and Rao [28] 441 described the increase of some high-molecular mass proteins in Mesorhizobium ciceri and 442 Rhizobium sp strains, respectively under other stress conditions. They concluded that these 443 proteins along with other factors may be responsible for growth and hence survival under 444 stress conditions. It must be noted that, the results presented here were a preliminary study 445 of proteome analysis. Thus, an exhaustive study would be necessary to provide a more 446 447 comprehensive approach related to particular proteins associated with tolerance and degradation of this harmful chemical. However, the presence of several proteins induced in 448 response to phenol stress may suggest that they could have an important role in 449 homeostasis and maintenance of vital cellular functions under this adverse condition. 450

One of the best ways to minimize the ecological risks and regulatory problems 451 associated with the introduction of non-native biota, including genetically modified species, 452 into ecosystems is the use of native species for bioremediation. The addition of specifically 453 selected bacteria from polluted areas have ecological advantages, such as already 454 adaptation of these strains to the environmental conditions of a given site and the reduction 455 of competition pressure. Taking into account these aspects, Acinetobacter sp. RTE1.4 could 456 be an appropriate strain for bioremediation of industrial zones because it belongs to a 457 polluted environment and has not been genetically engineered. 458

- 460 **5. CONCLUSION**
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Bioremediation uses the resources available in nature to clean up contamination and 462 usually involves lower costs and additional advantages compared to chemical treatment 463 processes. In this work, a bacterial strain was isolated from effluents of a chemical industry 464 and identified as Acinetobacter sp. RTE 1.4 using morphological, physiological, biochemical 465 and 16S rRNA gene analysis. The strain was able to tolerate phenol and phenol derivatives, 466 such as guaiacol, 2,4-dichlorophenol and pentachlorophenol and to grow in a medium 467 containing industrial effluents. The strain was also able to efficiently metabolize high phenol 468 concentrations by the ortho-pathway, at wide pH range and temperature (pH 5-9 and 25-37 469 °C, respectively). Protein patterns of Acinetobacter sp. RTE 1.4 changed in response to 470 phenol stress, showing mainly the induction of several proteins. 471 The biochemical and enzymatic findings, as well as phenol compounds tolerance and 472 removal capability of this native bacterium, would be helpful and provide base line data for 473 further studies associated with its application for effluent bioremediation or bioaugmentation 474

of areas polluted with phenolic compounds.

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Table 1. Physiological and biochemical characteristics of RTE1.4 strain (+, positive reaction; -, negative reaction).

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Characteristics

RTE1.4 strain

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| | Cell morphology Gram reaction | Bacilli - |
|---------------------|----------------------------------|--------------|
| | Cell size (µm) | 1-3 |
| | Indele production | + |
| | D-alucose fermentation | |
| | D glucose lemientation | |
| | Enzyme production | |
| | Oxidase | - |
| | Catalase | + |
| | Arginine dihydrolase | - |
| | Urease | - |
| | β-glucosidase | - |
| | Protease (Hydrolisis of gelatin) | C |
| | β-galactosidase | |
| | Sugar assimilation | |
| | Glucose | |
| | Arabinose | - |
| | Mannose | - |
| | Mannitol | - |
| | N-acetyl-glucosamine | - |
| | Maltose | - |
| | Potassium gluconate | - |
| | Acid assimilation | |
| | Capric acid | + |
| | Adipic acid | - |
| | Malic acid | + |
| | Trisodium citrate | - |
| | Phenylacetic acid | + |
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| 568 FIGURE CAPTIONS | S | |

Figure 1. Phylogenetic tree based on 16S rRNA gene sequence analysis (1000 bootstrap for the confidence level) showing the relationship between RTE1.4 strain and representative species of *Acinetobacter* genus. The bacterial species are indicated in the

- figure by their accession number between parentheses followed by the taxonomical
 identification. The scale bars represents 5 substitutions per 100 nucleotides.
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Figure 2. Phenol (200 mg/L) degradation by *Acinetobacter* sp. RTE1.4 strain at different pH and 28 ± 2 °C, after 48 h of incubation. Values are mean of three independent experiments and asterisk (*) represents significant differences respect to pH 7, according to Dunnett's test (p < 0.05).

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Figure 3. Effect of temperature on degradation of 200 mg/L of phenol in liquid MM at pH 7 and after 48 h of incubation. Values are mean of three independent experiments and asterisk (*) represents significant differences respect to 38 °C, according to Dunnett's test (p <0.05).

Figure 4. Effect of initial phenol concentrations on: biodegradation (a) and growth (b) of *Acinetobacter* sp. RTE1.4.

Figure 5. SDS-PAGE of total proteins from *Acinetobacter* sp. RTE1.4 grown in: control medium (L-lactate) (Lane 2), and in culture medium containing phenol at different concentrations: 200 mg/L (Lane 3); 400 mg/L (Lane 4); 600 mg/L (Lane 5). Lane 1: Molecular mass marker. Each lane was loaded with the same protein amount (25 µg).

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