G331.512-0.103: AN INTERSTELLAR LABORATORY FOR MOLECULAR SYNTHESIS I THE ORTHO-TO-PARA RATIOS FOR CH_3OH AND CH_3CN

Edgar Mendoza,¹ Leonardo Bronfman,² Nicolas U. Duronea,³ Jacques R. D. Lépine,¹ Ricardo Finger,² Manuel Merello,⁴ Carlos Hervías-Caimapo,^{2,5} Diana R. G. Gama,¹ Nicolas Reyes,⁶ and Lars Åke-Nyman⁷

¹Universidade de São Paulo, IAG Rua do Matão, 1226, Cidade Universitária, 05508-090, São Paulo, Brazil

²Departamento de Astronomía, Universidad de Chile, Casilla 36-D, Santiago de Chile, Chile

³Instituto Argentino de Radioastronomía, IAR CONICET, CCT-La Plata, C.C.5., 1894, Villa Elisa, Argentina

⁴ Istituto di Astrofisica e Planetologia Spaziali-INAF Via Fosso del Cavaliere 100, I-00133 Roma, Italy

⁵ Jodrell Bank Centre for Astrophysics, School of Physics and Astronomy, University of Manchester, Oxford Road, Manchester M139PL, UK

⁶Electrical Engineering Department, Universidad de Chile, Av. Tupper 2007, Santiago, Chile

⁷ Joint ALMA Observatory, JAO Alonso de Córdova 31070, Vitacura, Santiago de Chile, Chile

(Received; Revised; Accepted)

Submitted to The Astrophysical Journal

ABSTRACT

Spectral line surveys reveal rich molecular reservoirs in G331.512-0.103, a compact radio source in the center of an energetic molecular outflow. In this first work, we analyse the physical conditions of the source by means of CH_3OH and CH_3CN .

The observations were performed with the APEX telescope. Six different system configurations were defined to cover most of the band within (292–356) GHz; as a consequence we detected a forest of lines towards the central core. A total of 70 lines of A/E-CH₃OH and A/E-CH₃CN were analysed, including torsionally excited transitions of CH₃OH (ν_t =1). In a search for all the isotopologues, we identified transitions of ¹³CH₃OH. The physical conditions were derived considering collisional and radiative processes. We found common temperatures for each A and E symmetry of CH₃OH and CH₃CN; the derived column densities indicate an A/E equilibrated ratio for both tracers. The results reveal that CH₃CN and CH₃OH trace a hot and cold component with $T_k \sim 141$ K and $T_k \sim 74$ K, respectively. In agreement with previous ALMA observations, the models show that the emission region is compact ($\leq 5.5''$) with gas density $n(\text{H}_2)=(0.7-1) \times 10^7$ cm⁻³. The CH₃OH/CH₃CN abundance ratio and the evidences for pre-biotic and complex organic molecules suggest a rich and active chemistry towards G331.512-0.103.

Keywords: ISM: molecules – Molecular processes – Radio lines: ISM

1. INTRODUCTION

G331.512-0.103 is one of the most luminous and energetic molecular outflows known in the Galaxy. All the system is embedded in a star forming region known as G331.5-0.1, located in the Norma spiral arm at a distance of ~ 7 pc. This source exhibits typical H II region properties and intense maser emission of OH and CH_3OH (Caswell 1998; Nyman et al. 2001; Bronfman et al. 2008). At the center of the system, a young and massive stellar object drives a bipolar flow of around $55M_{\odot}$ with a momentum of ~ 2.4 × 10³ M_{\odot} km s⁻¹. Radio continuum observations have revealed a compact and central structure associated to a dust core. The central region has been also mapped with ALMA observations with transitions of SiO (8-7), H¹³CO⁺ (4-3), HCO⁺ (4-3) and CO (3-2). The observations revealed a ring-like structure, consistent with a cavity, and the existence of a high velocity outflow emission confined in a region lower than 5''in size (Merello et al. 2013a,b; Hervías et al. 2015). In this work, we present new results on the emission of CH₃OH and CH₃CN detected towards the central core, denominated hereafter as G331.

As an interstellar laboratory, G331 exhibits active chemistry in pre-biotic and complex organic molecules likely stimulated by physical processes in the molecular core and outflow. With the Atacama Pathfinder EXperiment APEX telescope, we have detected a vast number of molecular lines potentially linked to reservoirs traced by CH₃OH and CH₃CN. Those tracers are excellent candidates to unveil the different physical components that O- and N-bearing molecules can reveal towards hot molecular cores (Sandell et al. 1994; Wyrowski et al. 1999; Jørgensen et al. 2004; Beuther et al. 2005; Fontani et al. 2007; Fuente et al. 2014; Giannetti et al. 2017). In addition, from a chemical point of view, CH₃OH and CH₃CN are considered as parent and daughter molecules, respectively, in the route of interstellar molecular formation (e.g. Nomura & Millar 2004). With that motivation, we started a systematic study of G331 beginning with spectral analyses of A/E-CH₃OH and A/E-CH₃CN.

1.1. The tracers CH₃CN and CH₃OH

CH₃CN (methyl cyanide) is a symmetric top molecule. The nuclear spin state of the hydrogen atoms defines whether the molecule has an ortho (E) or para (A) symmetry. Rotational levels of CH₃CN are characterized by two quantum numbers: the total angular momentum (J) and its projection on the axis of symmetry (K). Spectral signatures are associated to K-ladder structures whose notation for the A and E configurations are K = 3n and $K = 3n\pm 1$, respectively (Solomon et al. 1971; Boucher et al. 1980; Cummins et al. 1983).

 CH_3OH (methanol) is one of the simplest asymmetrictop molecules, however its spectrum is quite complicated since there is a strong coupling between torsional and vibrational modes. Methanol also exists in two species denoted as A and E depending on the nuclear spin alignment of the three H-atoms of the methyl group (CH₃). The energy levels of methanol can be assigned with the total angular momentum J and its component K along the symmetry axis (Lees & Baker 1968; Lovas et al. 1982).

The excellent spectral resolution of the used instrument allowed us to study separately the A and E isomers of CH₃CN and CH₃OH. Nuclear spin conversion of A and E symmetries are considered as rare events, being affected by chemical reactions, non-reactive collisions and grain-surface mechanisms (Willacy et al. 1993; Hugo et al. 2009). However, the spin conversion has a particular relevance in molecular astrophysics. Important questions remain unsolved, for instance, how collisions stimulate conversions and, as a consequence, if that property can be used as an astronomical clock (Lee et al. 2006; Sun et al. 2015). One of our goals is to verify if the A and E pairs are equally populated at the local temperature of the source (e.g Andersson et al. 1984; Minh et al. 1993; Wirström et al. 2011).

We organized this paper as follows: in section 2 the observational procedure is described. In section 3 results about the line identification are presented. In section 4 we discuss the radiative analysis. The discussion, conclusions and perspectives are presented in sections 5 and 6.

2. OBSERVATIONS AND METHODOLOGY

The observations were carried out in March 2016 with the Atacama Pathfinder Experiment Telescope (APEX), located at *Llano de Chajnantor* (Chilean Andes). The spectra were obtained using the single point mode towards the coordinates RA, $\text{Dec.}(J2000) = 16^{h} 12^{m} 10.1^{s}$, $-51^{\circ}28'38.1''$. We used the APEX-2 receiver of the Swedish Heterodyne Facility Instrument (SHeFI) as frontend (Güsten et al. 2006; Risacher et al. 2006). The employed backend was the eXtended bandwidth Fast Fourier Transform Spectrometer2 (XFFTS2), which consists of two units with a bandwidth of 2.5 GHz divided into 32768 channels. We selected six set-ups to cover a band ranging from 292 GHz to 356 GHz; they are summarized in Table 1. The integration time was estimated to obtain a conservative rms noise of \sim 25 mK; then each set-up expended integration times between 0.7 and 2.6 hs, with system temperatures in the range of $T_{sys} \sim (200-380)$ K.

The calibration was done applying the chopper-wheel technique. The output intensity provided by the system was obtained in a scale of $T_{\rm A}^*$, which represents the antenna temperature corrected by atmospheric attenuation. The observed intensities were converted to the main-beam temperature scale using $\eta_{\rm mb} = 0.73$, the main beam efficiency for APEX-2 (Güsten et al. 2006).

All the K-ladders structures appeared closely spaced in a wide range of frequencies, so that they were ob-

Set-up	Frequency	Beam	Resolution
centred at	(GHz)	('')	$(10^{-2} \text{ km s}^{-1})$
CH_3CN (16–15)	292 - 296	21.4	7.7
CH_3CN (19–18)	329 - 333	19.0	6.9
CH ₃ OH (7–6)	336 - 340	18.6	6.8
SO(8-7)	343 - 347	18.2	6.6
CH ₃ OCH ₃ (11–10)	347 - 351	17.9	6.5
$HCOOCH_3$ (33–32)	352 - 356	17.7	6.4

Table 1. List of set-ups observed with the APEX-2 SHeFIinstrument towards G331.

served with the same receiver. This fact helped us with the inspection of irregularities due to calibration uncertainties, for which we adopted an error of 20 %.

The data reduction was carried out with the CLASS package of the GILDAS software.¹ The line identification was performed using the NIST² (Lovas et al. 2009), CDMS³ (Müller et al. 2005) and JPL⁴ (Pickett et al. 1998) spectroscopy databases. Specifically, the CDMS and JPL catalogues were interactively loaded on the survey using the Weeds extension of CLASS. The radiative analyses were performed using the CASSIS software.⁵

3. RESULTS

3.1. A/E- CH_3CN

We observed the transitions CH₃CN J=16-15, J=18-17 and J=19-18; a set of them is displayed in Figure 1. Within those levels, the spectral resolution was high enough to separate the components K=0, 3, 6 and 9 from K=1, 2, 4, 5, 7 and 8, which correspond to A-CH₃CN and E-CH₃CN, respectively (see Table 2).

The line identification was performed overlapping the laboratory values on the observed spectral bands at the rest velocity of the source. A high coincidence was obtained, since the K-ladder structures appeared separately and well centred, at the source's rest frequency (with V_{lsr} =-90 km s⁻¹), as predicted by the spectroscopic databases (Figure 1). This inspection revealed that most of the lines have a Gaussian profile, but only few of them presented broad blue and red wings due to the outflow activity. We applied Gaussian functions to fit and model the emission. For that, the spectral line processing was carried out using the CLASS package, by means of which we fitted baselines and Gaussian functions for each processed line. As a result, we have listed in Table 2 the Gaussian fit parameters and uncertainties of each analysed line; those parameters are the integrated line flux ($\int T_{mb}dV$ in K km s⁻¹), local standard of rest (LSR) velocity (V_{lsr} in km s⁻¹) and the line width given by the full width at half maximum (FWHM in km s⁻¹).

From a simple comparison of the A-CH₃CN fluxes, we observed a stiff increase of them as the lines go from the K=6 to K=0 transitions. This behaviour is in agreement with the probabilities predicted by the Einstein coefficients (A_{ij}) . For the transitions with $J_{up}=16$, 18 and 19, the obtained patterns are $16_6:16_3:16_0 \approx 1:3:4$, $18_6:18_3:18_0 \approx 1:4:6$ and $19_6:19_3:19_0 \approx 1:1:2$, respectively. Only few transitions diverged from this pattern, namely, A-CH₃CN $J=18_6-17_6$ and $J=19_3-18_3$, whose emission appeared blended with HNCO $(15_{1,14}-14_{1,13})$ and C₂H $(4_{4,4}-3_{3,2})$ at 330848 MHz and 349400 MHz, respectively. Those contaminant transitions were previously detected by Jewell et al. (1989) and Sutton et al. (1991) in Orion A and OMC-1.

For $E-CH_3CN$ lines, the derived fluxes also increase when the K-ladders vary from 7 to 1. That is in agreement with the spontaneous decay rates predicted by Aij. For the transitions with $J_{up}=16$, 18 and 19, the complete patterns are $16_7:16_5:16_4:16_2:16_1 \approx 1:2:3:5:10$, $18_7:18_5:18_4:18_2:18_1 \approx 1:3:4:8:12$ and $19_7:19_5:19_4:19_2:19_1 \approx 1:7:7:12:26$, respectively. Here the transition E-CH₃CN $J=19_4-18_4$ at 349346 MHz appeared blended with C₂H ($4_{5,4} - 3_{4,3}$) at 349339 MHz. We also found that the transition E-CH₃CN $J=19_7 - 18_7$ at 349125 MHz might be blended with CH₃OCHO at 349124 MHz.

The search for the high levels K=8 and K=9 yielded negative results, we did not detect lines of these Kladder numbers above the detection limit of the APEX data. For the cases A-CH₃CN (16₉-15₉) and (19₉-18₉), we did not observe emission at all. In the case of the 18₉-17₉ line, we obtained a marginal flux from the Gaussian fits. This line also might be blended with c-C₃HD and/or weeds,⁶ see Table 2. For the cases E-CH₃CN (16₈-15₈) and (18₈-17₈), the emission appeared fully and partially blended with CS (6–5) and ³⁴SO₂ (21₂-21₁), respectively. For the 19₈-18₈ transition, we did not detect emission. The results are listed in Table 2. In addition, a search for emission of CH₃CN ($v_8 = 1$) was performed with the current dataset; however we did not observe clear evidences to conclude about its presence in G331.

3.2. $A/E-CH_3OH$

As a forest of spectral lines (Figure 1), the main part of the CH_3OH emission has appeared only in an interval of 3 GHz along the whole survey. The lines were identified

¹ https://www.iram.fr/IRAMFR/GILDAS/

² http://physics.nist.gov/cgi-bin/micro/table5/start.pl

³ https://www.astro.uni-koeln.de/cdms

⁴ https://spec.jpl.nasa.gov/

⁵ http://cassis.irap.omp.eu/

⁶ Conventional term adopted to label species with numerous spectral lines in the mm/sub-mm regime.



Figure 1. Lines of A/E-CH₃CN and A/E-CH₃OH as observed with the APEX-2 SHeFI instrument towards G331. Lines were adjusted to their rest frequencies using the CDMS and JPL catalogues. Due to the high intensity of the methanol lines, the range of the *y*-axis is different in both panels.

by a simple comparison of their frequencies, corrected by the source V_{lsr} =-90 km s⁻¹, with the rest frequencies reported in molecular databases. The methanol lines exhibit a symmetrical distribution which were fitted by simple Gaussian functions. The spectroscopic and fit parameters estimated for these lines are listed in Table 3. We find a FWHM of around 5.5 km s⁻¹ and a V_{lsr} =-90 km s⁻¹ with a dispersion of ± 2km s⁻¹, mainly due to lines which exhibited a low signal-to-noise ratio. In Figure 1 we show a set of A and E-CH₃OH lines with J=7–6.

Although most of the methanol emission corresponds to the torsional ground state, we also identified lines corresponding to the first excited state (ν_t =1). Such is the case for the lines displayed in Figure 2 (upper panel), where it is observed a broad line of ³⁴SO at ~ 337580 MHz blended with the transition *E*-CH₃OH (ν_t =1) 7₄ - 6₄ at a frequency, corrected by the source V_{lsr} , of 337581.663 MHz. Around the ³⁴SO peak, minor satellite emission also was identified as corresponding to (A/E)-CH₃OH ν_t = 1; the spectroscopic parameters are summarized in Table 4. In the bottom panel of Figure 2 we display the only excited line without contaminant emission, which corresponds to *A*-CH₃OH (ν_t =1) 7₁ - 6₁ at a frequency, corrected by the source V_{lsr} , of 337969.414 MHz.

Emission of excited methanol is typically observed towards hot molecular cores. For instance, Menten et al. (1986), Sutton et al. (1995) and Schilke et al. (2001) have reported its emission in OMC-1, W3(OH) and

W51.

The observed lines of excited methanol constitute an evidence for its presence in G331. However, in order to solve questions as whether the excited emission is populated or not at the same conditions of the ground state, it is necessary to carry out complementary observations; for instance, at frequencies between 600–720 GHz. These observations could reveal a larger number of CH₃OH $\nu_t=1,2$ lines favouring a better estimation on the physical conditions. In the bottom panel of Figure 2, we displayed a broad line close to 337900 MHz which is associated to SO₂; however, that line could be also blended with CH₃OH ($\nu_t=2$) J=7-6 at 337877 MHz.

3.3. Isotopologues

We searched for lines of the D, 13 C, 15 N and 18 O isotopologues of CH₃CN and CH₃OH. As predicted by the spectroscopic databases, several frequencies of those isotopologues fall across the bands; however we only confirm the presence of 13 CH₃OH. Emission related to 13 CH₃CN J=19–18 was also identified; however, the lines are blended with SO J=3–2 at 339341 MHz. Those lines were also reported by Jewell et al. (1989) in Orion A.

We detected a couple of transitions of A^{-13} CH₃OH. The associated spectroscopic parameters are presented in Table 5. The lines were identified at the rest frequencies 330252.798 MHz and 350103.118 MHz with signal-to-noise ratios of $S/N \geq 5$. Such detections

AASTEX

Table 2. Spectral line analysis of A-CH₃CN and E-CH₃CN in G331. Columns 1–4 list the spectroscopic parameters collected from databases such as CDMS and JPL. Columns 5–7 contain the resulting parameters and uncertainties derived from Gaussian fits, they are line flux ($\int T_{mb}dV$), local standard of rest velocity (V_{lsr}) and the line width (FWHM).

Transition	Frequency	E_u	A_{ij}	$\int T_{mb} dV$	V_{lsr}	FWHM
	(MHz)	(K)	(10^{-3} s^{-1})	$({\rm K~km~s^{-1}})$	$(\mathrm{km}~\mathrm{s}^{-1})$	$({\rm km~s}^{-1})$
A-CH ₃ CN						
$^{\dagger}16_{9}{-}15_{9}$	293845.164	697.83	1.51	_	_	_
$16_6 - 15_6$	294098.867	377.0721	1.9	2.065 ± 0.006	-90.7 ± 0.2	10.3 ± 0.5
$16_3 - 15_3$	294251.461	184.3532	2.13	5.9 ± 0.3	-90.4 \pm 0.2	8.2 ± 0.6
$16_0 - 15_0$	294302.388	120.0696	2.21	7.9 ± 0.2	-89.69 ± 0.07	8.8 ± 0.2
$^{a}18_{9}-17_{9}$	330557.569	728.67	2.36	0.15 ± 0.04	$\textbf{-91.2}\pm0.1$	0.9 ± 0.2
$^{\rm b}18_6 - 17_6$	330842.762	407.9468	2.8	1.53 ± 0.07	$\textbf{-88.2}\pm0.4$	10 ± 0.1
$18_3 - 17_3$	331014.296	215.2437	3.07	6.6 ± 0.5	-90.3 ± 0.3	8.9 ± 0.8
$18_0 - 17_0$	331071.544	150.9654	3.16	9.1 ± 0.3	-89.5 ± 0.1	9.4 ± 0.3
$^{\dagger}19_{9}{-}18_{9}$	348911.401	745.42	2.87	_	_	_
$19_{6} - 18_{6}$	349212.311	424.7065	3.34	3.01 ± 0.01	-90.4 ± 0.2	11.5 ± 0.5
$^{\rm c}19_3-18_3$	349393.297	232.012	3.63	3.66 ± 0.07	-91.7 ± 0.3	3.8 ± 0.6
$19_0 - 18_0$	349453.7	167.7368	3.72	5.5 ± 0.3	-90.5 ± 0.1	6.8 ± 0.4
E-CH ₃ CN						
$^{\rm d}16_8 - 15_8$	293940.916	568.69	1.65	_	_	_
$16_7 - 15_7$	294025.496	461.7635	1.78	0.8 ± 0.1	-88.7 ± 0.6	10 ± 2
$16_5 - 15_5$	294161.001	290.5519	1.99	2.14 ± 0.09	-89.2 ± 0.2	9.8 ± 0.5
$16_4 - 15_4$	294211.873	226.3073	2.07	2.93 ± 0.08	-90.2 ± 0.1	9.3 ± 0.3
$16_2 - 15_2$	294279.75	140.6138	2.17	4.19 ± 0.09	-90.79 ± 0.08	7.9 ± 0.2
$16_1 - 15_1$	294296.728	119.1843	2.2	7.74 ± 0.07	-91.87 ± 0.07	9.4 ± 0.1
$^{\rm e}18_8 - 17_8$	330665.206	599.54	2.52	1.21 ± 0.07	-91.4 ± 0.2	8.1 ± 0.5
$18_7 - 17_7$	330760.284	492.6304	2.67	0.8 ± 0.1	-90.8 ± 0.7	11 ± 2
$18_5 - 17_5$	330912.608	321.4329	2.91	2.9 ± 0.1	-89.4 ± 0.2	11.6 ± 0.6
$18_4 - 17_4$	330969.794	257.1937	3	3.3 ± 0.1	-90.5 ± 0.1	9.2 ± 0.4
$18_2 - 17_2$	331046.096	171.5073	3.12	6.66 ± 0.09	-90.87 ± 0.07	10.4 ± 0.2
$18_1 - 17_1$	331065.181	150.0797	3.15	10.252 ± 0.007	-91.62 ± 0.06	10.6 ± 0.1
$^{\dagger}19_{8}18_{8}$	349024.971	616.29	3.05	_	_	_
$^{\rm f}19_{7}-18_{7}$	349125.287	509.38	3.2	0.304 ± 0.003	-90.1 ± 0.6	5.9 ± 0.7
$19_{5} - 18_{5}$	349286.006	338.1962	3.45	2.106 ± 0.008	-89.6 ± 0.2	9.5 ± 0.5
$^{\rm g}19_4 - 18_4$	349346.343	273.9599	3.55	2.091 ± 0.002	-89.76 ± 0.09	6.1 ± 0.1
$19_2 - 18_2$	349426.85	188.2773	3.67	5.3 ± 0.1	-90.61 ± 0.9	9.8 ± 0.2
$19_1 - 18_1$	349446.987	166.8506	3.71	8.018 ± 0.001	-91.76 ± 0.05	9.39 ± 0.08

NOTE—Likely blended lines with (a) HCOOCH₂CH₃ (330557 MHz) and/or c-C₃HD (330558 MHz), (b) HNCO (330848 MHz), (c) C₂H (349400 MHz), (d) CS (293912 MHz), (e) ³⁴SO₂ (330667 MHz), (f) CH₃OCHO (349124 MHz) and (g) C₂H (349339 MHz). †Marginal emission (< 5σ).

have been fundamental to analyse the opacity of the detected emission. For instance, we estimated flux ra-

tios using lines of A-¹³CH₃OH and A-CH₃OH with comparable spectroscopy, namely, using those transitions

Table 3. The table 2 caption applies here for A-CH₃OH and E-CH₃OH.

Transition	Frequency	E_u	A_{ij}	$\int T_{mb} dV$	V_{lsr}	FWHM
J_k	(MHz)	(K)	(10^{-5} s^{-1})	$({\rm K~km~s^{-1}})$	$(\mathrm{km}~\mathrm{s}^{-1})$	$({\rm km~s^{-1}})$
A-CH ₃ OH						
$6_1 - 5_1$	292672.889	63.707	10.6	6.95 ± 0.06	-90.67 ± 0.02	5.24 ± 0.05
$3_2 - 4_1$	293464.055	51.6381	2.89	3.39 ± 0.09	-90.35 ± 0.09	6.8 ± 0.2
$12_2 - 11_3$	329632.881	218.8033	6	0.59 ± 0.07	-89.8 ± 0.3	4.1 ± 0.6
$11_1 - 11_0$	331502.319	169.0101	19.6	3.394 ± 0.001	-90.17 ± 0.02	2.82 ± 0.03
$^{\dagger}7_{3}$ – 8_{1}	332920.822	114.794	1.15×10^{-5}	_	_	_
$12_1 - 12_0$	336865.149	197.0765	20.3	6.9 ± 0.5	-91.9 \pm 0.2	6.9 ± 0.3
7_0-6_0	338408.698	64.9817	17	10.83 ± 0.07	-90.51 ± 0.02	5.31 ± 0.04
7_6-6_6	338442.367	258.6994	4.53	1.09 ± 0.06	-91.3 \pm 0.1	5.7 ± 0.4
$7_{5}-6_{5}$	338486.322	202.8864	8.38	2.11 ± 0.08	-90.98 ± 0.09	4.9 ± 0.2
$7_2 - 6_2$	338512.853	102.7032	15.7	7.3 ± 0.4	-90.5 \pm 0.1	5.5 ± 0.4
7_3-6_3	338543.152	114.7948	13.9	9.21 ± 0.08	-89.77 ± 0.02	5.62 ± 0.05
7_2-6_2	338639.802	102.7179	15.7	7.4 ± 0.7	-90.6 \pm 0.2	5.6 ± 0.6
$2_2 - 3_1$	340141.143	44.6729	2.79	2.14 ± 0.06	-90.71 ± 0.08	5.5 ± 0.2
$^{\dagger}16_{1}$ –-15 $_{2}$	345903.916	332.6545	9.02	_	_	_
5_4-6_3	346204.271	115.1625	2.17	2.31 ± 0.07	-90.74 ± 0.09	5.9 ± 0.2
$14_1 - 14_0$	349106.997	260.2068	22	4.25 ± 0.06	-90.77 ± 0.04	5.41 ± 0.09
$1_1 - 0_0$	350905.1	16.841	33.1	12.19 ± 0.05	-90.58 ± 0.01	5.13 ± 0.03
$^{\dagger}8_{2}$ –90	351685.48	121.2915	16.2	-	-	_
$^{\dagger}10_{4}11_{2}$	355279.539	207.9929	1.22×10^{-3}	-	-	_
$13_0 - 12_1$	355602.945	211.0285	12.6	4.02 ± 0.04	-91.01 ± 0.02	4.76 ± 0.06
$15_1 - 15_0$	356007.235	295.2675	23	2.68 ± 0.05	-90.46 ± 0.05	5.8 ± 0.1
E-CH ₃ OH						
8_{-3} - 9_{-2}	330793.887	138.3774	5.38	1.96 ± 0.06	-91.03 ± 0.07	4.6 ± 0.1
$16_{-1} - 15_{-2}$	331220.371	312.7333	5.23	0.55 ± 0.06	-91.4 \pm 0.3	5.3 ± 0.6
$3_3 - 4_2$	337135.853	53.7412	1.58	1.09 ± 0.06	-91.1 \pm 0.1	4.7 ± 0.3
$^{\dagger}7_{2}$ 6 $_{2}$	337671.238	456.8285	15.5	-	-	_
$^{\dagger}7_{5}$ – 6_{5}	337685.248	486.0568	8.28	-	-	_
$^{\dagger}7_{-1}-6_{-1}$	337707.568	470.3161	16.5	-	-	_
7_0-6_0	338124.488	70.1794	16.9	9.09 ± 0.06	-90.74 ± 0.01	5.01 ± 0.04
$7_{-1} - 6_{-1}$	338344.588	62.6521	16.6	9.61 ± 0.08	-90.63 ± 0.02	4.87 ± 0.05
$^{\dagger}7_{6}$ – 6_{6}	338404.61	235.8941	4.51	-	-	_
$^{\dagger}7_{-6}$ –6 $_{-6}$	338430.975	246.0504	4.54	-	_	_
$7_{-5}-6_{-5}$	338456.536	181.1017	8.34	1.3 ± 0.4	-90.9 ± 0.7	5 ± 2
$7_{5}-6_{5}$	338475.226	193.1625	8.34	1.5 ± 0.6	-91 ± 0.1	5 ± 3
7_{-4} - 6_{-4}	338504.065	144.9961	11.5	4 ± 2	-91.3 ± 0.04	6 ± 1
7_4-6_4	338530.257	153.0934	11.5	3.1 ± 0.6	-91.1 ± 0.4	5 ± 1
7_{-3} - 6_{-3}	338559.963	119.8084	14	4.5 ± 0.1	-90.93 ± 0.6	5.2 ± 0.1
7_3-6_3	338583.216	104.8115	13.9	6.9 ± 0.2	-90.99 ± 0.08	6.6 ± 0.2
$7_1 - 6_1$	338614.936	78.1537	17.1	21.4 ± 0.8	-89.5 ± 0.2	9.9 ± 0.5
$7_{-2}-6_{-2}$	338722.898	83.0148	15.7	12.53 ± 0.07	-90.04 ± 0.01	5.61 ± 0.03
$^{\dagger}18_{2}{-}17_{3}$	344109.039	411.5062	6.79	_	_	_

Note—†Marginal emission (< 5 σ).

AASTEX

Transition	Frequency	E_u	A_{ij}	$\int T_{mb} dV$	V_{lsr}	FWHM
J_k	(MHz)	(K)	$(10^{-5} \ {\rm s}^{-1})$	$(\rm K\ km\ s^{-1})$	$({\rm km~s^{-1}})$	$({\rm km~s^{-1}})$
A-CH ₃ OH ($\nu_t = 1$)						
$a_{75}-6_{5}$	337546.048	485.4	8.13	1.047 ± 0.004	-92.8 ± 0.3	9.1 ± 0.5
^a 7 ₂ -6 ₂	337625.679	363.5	15.5	10.2 ± 0.9	-90.5 ± 0.3	8 ± 1
^a 7 ₂ -6 ₂	337635.655	363.5	15.5	1.2 ± 0.2	-92.6 ± 0.6	9 ± 2
$7_1 - 6_1$	337969.414	390.1	16.6	0.39 ± 0.05	-90.8 ± 0.3	5.2 ± 0.7
E -CH ₃ OH ($\nu_t = 1$)						
$a_{7_4-6_4}$	337581.663	428.2	11.3	_	_	-
$a_{7-2}-6_{-2}$	337605.255	429.4	15.6	1.513 ± 0.003	-89.8 ± 0.3	11.7 ± 0.5
^a 7 ₀ –6 ₀	337643.864	365.4	16.9	1.1 ± 0.8	-90.2 ± 0.2	5.9 ± 0.5
$^{b}10_{-2}$ –113	344312.267	491.91	17.7	_	_	_

Table 4. The table 2 caption applies here for A-CH₃OH ($\nu_t = 1$) and E-CH₃OH ($\nu_t = 1$).

NOTE—Transitions and quantum numbers from Xu & Lovas (1997) and references therein. Labels indicate the lines partially and totally blended with (a) ³⁴SO 8₈-7₇ (337580 MHz) and (b) SO 8₈-7₇ (344310 MHz).

Table 5. Spectroscopic and observational parameters of the of the A^{-13} CH₃OH lines analysed in G331. Similarly, the table 2 caption applies here.

Transition	Frequency	E_u	A_{ij}	$\int T_{mb} dV$	V_{lsr}	FWHM	$^{\dagger}\mathrm{C}/^{13}\mathrm{C}$
J_k	(MHz)	(K)	$(10^{-5} \ {\rm s}^{-1})$	$({\rm K~km~s^{-1}})$	$({\rm km~s^{-1}})$	$({\rm km~s^{-1}})$	
7_0-6_0	330252.798	63.4	15.8	0.60 ± 0.07	-90.7 ± 0.3	5.0 ± 0.6	18.1 ± 0.7
$1_1 - 0_0$	350103.118	16.8	32.9	0.69 ± 0.05	-91.2 ± 0.2	4.3 ± 0.3	17.7 ± 0.1

Labels— \dagger Flux ratios derived by considering the integrated fluxes of A-CH₃OH (7₀-6₀) and (1₀-0₀) listed in Table 3.

which have exhibited similar frequencies, A_{ij} and E_u values. Then, by selecting the 7_0-6_0 and 1_1-0_0 transitions both of A-CH₃OH and A-¹³CH₃OH (Tables 3 and 5), whose A_{ij} values are in agreement almost by a factor 1, we obtained the flux ratios $C/^{13}C= 18.1 \pm 0.7$ and $C/^{13}C= 17.7 \pm 0.1$, respectively. Such flux ratios are similar to the values reported in sources towards the Galactic center, where $C/^{13}C= 20$ (Wilson & Rood 1994; Requena-Torres et al. 2006). These ratios imply finite values of opacities, therefore in the subsequent sections we will take into account optical depth corrections to estimate the physical conditions derived from the CH₃OH and CH₃CN emission.

Two lines of E^{-13} CH₃OH were also identified, although they exhibited contamination. Close to their rest frequencies, lines of 34 SO₂ J=8-7 and CH₃OCH₃ J=16-15 are candidates for blended emission. These contaminants were also detected towards the Hot Molecular Core G34.3+0.15 (MacDonald et al. 1996).

4. PHYSICAL CONDITIONS

4.1. Statistical equilibrium calculations

In this section we present results based on statistical equilibrium models when considering collisions between the analysed molecules and molecular hydrogen. To estimate the kinetic temperatures and column densities, we ran the RADEX code into CASSIS, whose formalism is analogous to the Large Velocity Gradient approximation (van der Tak et al. 2007). RADEX considers the probability that a photon escape in an isothermal and homogeneous medium without large-scale velocity fields. Collision rate data were collected from the LAMDA database.⁷

Conjointly with RADEX/CASSIS, we used the Markov Chain Monte Carlo (MCMC) method to simulate the observed lines from the resulting best solutions, which are given by the minimal χ^2 value. The MCMC algorithm operates on values chosen randomly within a sample defined by a minimum and maximum limit for an N-dimensional parameter space. This characteristic is the most relevant when calculations require as inputs

 $^{^7}$ Leiden Atomic and Molecular Database LAMDA, http://home.strw.leidenuniv.nl/moldata/

Table 6. Parameters and physical conditions derived from the statistical equilibrium calculations which have yielded hydrogen densities between 0.7 and 1×10^7 cm⁻³. The number of computed lines and the reduced χ^2 values have also been included.

Parameters	A-CH ₃ CN	E-CH ₃ CN	A-CH ₃ OH	E-CH ₃ OH
Modelled lines	7	13	10	13
$\chi^2 \min$	5.69	7.77	6.23	4.42
$\rm FWHM~(km~s^{-1})$	5.49 ± 0.01	5.49 ± 0.01	4.9 ± 0.1	4.88 ± 0.01
$N \ (10^{14} \ {\rm cm}^{-2})$	4.77 ± 0.05	3.69 ± 0.04	85 ± 20	98 ± 0.1
T_k (K)	142.1 ± 0.9	140.5 ± 0.3	64 ± 2	84.8 ± 0.1
θ (")	4.51 ± 0.01	4.53 ± 0.02	5.7 ± 0.3	5.01 ± 0.04



Figure 2. Panels showing the torsional excited lines of A-CH₃OH (red dash-dotted line) and E-CH₃OH (blue dashed line) against the spectra of G331 (black histogram). In increasing order, the Gaussian fits are centred at the rest frequencies ~ 337546 MHz, 337581 MHz (blended with ³⁴SO), 337605 MHz, 337625 MHz, 337635 MHz, 337643 MHz and 337969 MHz.

several free parameters. In detriment, the quality of the fit is a monotonically increasing function of the number of iterations introduced in the chain, which in turn affects the "computational time" to achieve convergence (e.g. Guan & Krone 2007; Foreman-Mackey et al. 2013; Tahani et al. 2016).

We prepared MCMC routines for each A and Especies. The numerical sample explored by the algorithm was delimited by five free parameters: kinetic temperature (T_k) , column density (N), source size (θ) , FWHM and hydrogen density $(n_{\rm H_2})$. To guarantee that the code has visited the sample, we tested routines with number of iterations of the order of 10⁵. Once the algorithm achieves convergence, the minimal χ^2 is computed printing the statistical results for each parameter. This methodology implied large run-times. As a total, we modelled 7 and 13 lines of A-CH₃CN and E-CH₃CN, respectively, and 10 and 13 lines of A-CH₃OH and E-CH₃OH, respectively. The statistical calculations are summarized in Table 6. Likewise, the best fit model lines are exhibited in Figure 3.

There are two aspects which were common for the models of A/E-CH₃OH and A/E-CH₃CN. First, we obtained H₂ densities in the interval $n_{\rm H_2}$ = (0.7–1) × 10^7 cm^{-3} , which is in agreement within a factor $\gtrsim 1.5$ with the density derived from the dust continuum (Hervías-Caimapo et al. submitted). Also, we intended to differentiate the densities of the ortho-H₂ and para- H_2 collisional partners (o- H_2 and p- H_2 respectively). Although studies explain the differences when A and Especies collide with $o-H_2$ and $p-H_2$ (e.g. Rabli & Flower 2010), for our purposes we just conclude that the o- $H_2/p-H_2$ ratio varied between ~ 0.6 and 1.0 for the different MCMC computations. That interval is consistent with the ortho-to-para limit of 0.2, when H_2 is thermalized at the CO temperature, and ortho-to-para limit of 3 when H₂ first forms (Flower & Watt 1984; Lacy et al. 1994).

The second aspect is that the modelled spectral lines yielded source sizes in agreement with a compact emitter region. For CH₃CN and CH₃OH we found sizes of ~ 4.5" and ~ 5.3", respectively. This result is supported by previous CO (7–6) maps obtained with the Australia Telescope Compact Array (ATCA), for which Bronfman et al. (2008) determined a compact and dense region of ~ 4.3". Merello et al. (2013a,b) mapping the emission traced by SiO (8–7) found a ring-type structure of ≤ 5 ".

4.2. LTE analysis

Rotational diagrams were constructed to estimate the excitation temperatures (T_{exc}) and column densities (N) of A/E-CH₃CN and A/E-CH₃OH. Assuming that the emission is optically thin and uniformly fills the antenna beam, T_{exc} and N can be calculated from

$$\ln\left(\frac{N_u}{g_u}\right) = \ln\left(\frac{N}{Z}\right) - \frac{E_u}{kT_{exc}} \tag{1}$$

where N_u/g_u and E_u are the column density per statistical weight and the energy of the upper level, respectively, and Z is the partition function (Goldsmith & Langer

9



Figure 3. Spectral lines and the best fits assuming non-LTE conditions, plotted as a function of the frequency corrected by the source V_{lsr} , of a) A-CH₃CN (blue dashed-line) and E-CH₃CN (red dash-dotted-line) and b) A-CH₃OH (blue dashed-line) and E-CH₃OH (red dash-dotted-line). For A/E-CH₃CN, the models yielded a kinetic temperature between ~ (140–142) K and column densities between ~ (3.7–4.8) × 10¹⁴ cm⁻². For A/E-CH₃OH, the models yielded a kinetic temperature between ~ (64–85) K and column densities between ~ (8.5–9.8) × 10¹⁵ cm⁻².

1999). In local thermodynamic equilibrium (LTE), Equation 1 represents a Boltzmann distribution whose values $\ln N_u/g_u$ vs E_u/k can be fitted using a straight line, whose slope is defined by the term $1/T_{exc}$.

The rotational diagrams were constructed taking a beam dilution factor associated with a point-like emitter region. Therefore, a correction that is provided by the term $\ln (\Delta \Omega_a / \Delta \Omega_s)$ was introduced on the right-hand of Equation 1 (Goldsmith & Langer 1999). This ratio relates the subtended angle of the source with the solid angle of the antenna beam. The beam dilution factor that we introduced is based on the results derived from the RADEX/MCMC calculations. Then, for CH₃CN and CH₃OH we adopted averaged source sizes of 4.5" and 5.3", respectively.

Since the lines observed in this work fell in a broad band, from 290 to 350 GHz, yielding different beamwidths, the adopted sizes have been used in the comparisons between temperatures and column densities for the different lines.

Depending on the density of the region, the K-ladder structures of CH₃CN and CH₃OH can exhibit an important scattering of $(\ln N_u/g_u, E_u)$ -values in their rotational diagrams. More generally, it is expected that the higher is the gas density, the lower is the K-ladder segregation (Olmi et al. 1993; Cesaroni et al. 1997; Goldsmith & Langer 1999; Remijan et al. 2004; Araya et al. 2005; Cuadrado et al. 2017). In order to check the amount of scatter, we compared the rotational diagrams when they were individually constructed for each J_K level and with those obtained when an unique fit was determined across all the K-levels. As a result, we did not observe considerable discrepancies between the two cases, temperatures and column densities were found within the adopted uncertainty from calibration (§ 2). For instance, in Figure 4 we display the result for $E-CH_3CN$.



Figure 4. Rotational diagram fits of E-CH₃CN across all the K-ladder levels (dashed-dotted line), yielding $N \simeq$ 7.5×10^{14} cm⁻² and $T_{exc} \simeq 172$ K, and within each Kladder structure for the transitions J=16-15 (black triangles), J=18-17 (red squares) and J=19-18 (blue pentagons).

OPTICAL DEPTH CORRECTION

The rotational diagrams obtained from the A/E- CH_3CN and A/E- CH_3OH lines are displayed in Figure 5. They were constructed selecting only the transitions in the ground state and without contaminant emission. So far, the rotational diagrams have been scaled only considering the beam dilution factor. However, they represent a solution under the assumption of emission optically thin, $\tau \ll 1$. In order to revise how different are the LTE solutions from a radiative scenario with finite values of optical depths, we have applied an optical depth correction on the rotational diagrams of A/E-CH₃CN and A/E-CH₃OH. Expressed mathematically, the correction consisted in including the term $-\ln C_{\tau}$, with $C_{\tau} = \tau/1 - \exp^{-\tau}$, on the right side of Equation 1 (Goldsmith & Langer 1999; Gibb et al. 2000; Remijan et al. 2004; Araya et al. 2005). Thus, the ordinates of our rotational diagrams were readjusted considering a term associated with the photon escape probability.

We have performed this adjust based on (1^{st}) the results obtained from the statistical equilibrium calculations. In the case of CH₃CN, transitions with K=0,1,2and 3 were the most affected with $\tau \approx 1$. And, (2^{nd}) considering the flux ratio $C/^{13}C \lesssim 20$ as we described in section 3.3, which suggests finite optical depths.

In order to perform the optical depth correction, we used CASSIS which executes iterative calculations of C_{τ} adjusting the level populations until achieving a consistent solution. With these corrections slight changes are expected in the LTE solutions. For instance, Araya et al. (2005) obtained differences around 10 and 14 % in the temperature and column density of CH₃CN when the optical depth correction was applied, respectively.

The A/E-CH₃CN rotational diagrams (RDs), with the optical depth correction, are displayed in the upper panels of Figure 5. For A-CH₃CN, we obtained N= $(6.4 \pm 2.0) \times 10^{14}$ cm⁻² and $T_{exc} = 156 \pm 25$ K. For E-CH₃CN, we obtained $N = (8.4 \pm 2.0) \times 10^{14}$ cm⁻² and $T_{exc} = 152 \pm 20$ K. The percentage difference of N and T_{exc} , with respect to the RD without the optical depth correction, are ranged between (5–14) % and (2–12) %, respectively.

The A/E-CH₃OH RDs, with the optical depth correction, are displayed in the lower panels of Figure 5. For A-CH₃OH, we obtained $N = (1.9 \pm 0.5) \times 10^{16}$ cm⁻² and $T_{exc} = 71 \pm 10$ K. For E-CH₃OH, we obtained $N = (2.0 \pm 0.1) \times 10^{16}$ cm⁻² and $T_{exc} = 68 \pm 8$ K. The percentage difference of N and T_{exc} , with respect to the RD without the optical depth correction, are ranged between (10-22) % and (9-17) %, respectively.

We detected two lines of ${}^{13}\text{CH}_3\text{OH}$ (Table 5). In the absence of collisional coefficients, those lines were modelled assuming LTE conditions applying the MCMC method leaving various inputs as free parameters: N, T_{exc} , FWHM, and θ . However, we put an upper limit of $N < 5 \times 10^{15} \text{ cm}^{-2}$, which was derived from the column density of the main isotopologue and considering the ratio $C/{}^{13}C \lesssim 20$. The results were consistent with the physical conditions of the main isotopologue. The best LTE models have been displayed in Figure 6, whose solution ($\chi^2=2.82$) yielded $N=(1.01 \pm 0.06) \times 10^{15} \text{ cm}^{-2}$ and $T_{exc}=89.5 \pm 0.4 \text{ K}$ for a modelled source size of 5.7". Additionally, the resulting FWHM values ($\sim 4.4 \text{ km s}^{-1}$) are consistent with the resulting fits reported in Table 5.

5. DISCUSSION

5.1. Molecular abundances

CH₃OH is more abundant and traces a region about 70 K colder than that traced by CH₃CN. From the LTE and non-LTE analysis, we found that the ratio (A + E)-CH₃OH/(A + E)-CH₃CN is about 25, indicating almost the same over abundance of A and E-CH₃OH over A-CH₃CN and E-CH₃CN.

ALMA observations of the ion $\mathrm{H}^{13}\mathrm{CO}^+$ (4–3) probed the existence of a core and various clumpy structures in G331 (Hervías-Caimapo et al. *submitted*). Maps of SiO (8–7) show as well an internal cavity surrounded by molecular emission confined in $\leq 5''$, for which these authors have estimated $N(\mathrm{H}^{13}\mathrm{CO}^+)\approx(1.5-3)\times10^{13}$ cm⁻² (e.g. Bronfman et al. 2008; Merello et al. 2013a,b).



Figure 5. Upper panels. Rotational diagrams of A-CH₃CN and E-CH₃CN showing the least-square fits with the optical depth correction, represented with the asterisk symbols and dashed-dotted lines, and without the optical depth correction, represented with the filled triangles and solid lines. Bottom panels, the same as above for A-CH₃OH and E-CH₃OH. Column densities and excitation temperatures correspond to the results when applied the optical depth correction, numbers in parenthesis represent uncertainties on the last digit.

Table 7. Abundances of CH₃OH and CH₃CN derived from the LTE and non-LTE analysis. We have included the contribution of each spin isomer as a percentage of the total abundance, assumed as the sum A + E of the symmetries.

Method	Hot Component			(Cold Compone	ent
	$[\mathrm{CH}_3\mathrm{CN}]$	$[^{13}\mathrm{CH}_3\mathrm{CN}]$	$\%_{A-\mathrm{CH}_3\mathrm{CN}}$	$[CH_3OH]$	$[^{13}\mathrm{CH}_3\mathrm{OH}]$	$\%_{A-\mathrm{CH}_3\mathrm{OH}}$
	10^{-9}	10^{-9}	\approx	10^{-9}	10^{-9}	\approx
non-LTE	1.8 ± 0.2	_	56	38 ± 10	_	46
LTE	3.1 ± 0.6	$\lesssim 0.08$	43	81 ± 15	$\lesssim 2.1$	49

The column densities of CH₃OH and CH₃CN were divided by $N(H_2)$ values to obtain their abundances. Following the works cited above, in order to estimate a scaled density of H₂ in G331, we applied the ratio $H^{13}CO^+/H_2=3.3 \times 10^{-11}$ of Orion KL (e.g. Blake et al. 1987) on the H¹³CO⁺ column densities derived for G331 (Merello et al. 2013a). As an approximation for the present work, we adopted the H¹³CO⁺/H₂ ratio of the Orion KL system, since this source is an excellent template for comparative studies on abundances and molecular complexity (e.g. Schilke et al. 2001; Beuther et al. 2005). The total abundances of CH₃OH and CH₃CN are summarized in Table 7; as a total, we refer to the A + E contribution from the nuclear spin symmetries (e.g. Fuente et al. 2014). Thus, the CH₃CN emission is linked to a hot core with $T_k \approx 141$ K and $\theta \approx 4.5''$ while CH₃OH traces a cold bulk medium with $T_k \approx 74$ K and $\theta \approx 5.3''$. In the same Table 7, we also included the abundance uncertainties computed from the errors obtained with the radiative models (Bevington & Robinson 2003); although we listed errors of $\leq 25\%$, the abundance uncertainties could be of up to 40% including other sources of errors, such as the calibration uncertainty, adopted here as 20%. In Table 7, the abundances of the ¹³C isotopologues are also listed. Although we obtained an inaccurate result



Figure 6. Spectral lines and LTE models of A^{-13} CH₃OH represented by black histograms and Gaussian curves, respectively. Lines 7₀-6₀ and 1₁-0₀ appeared at the rest frequencies 330252.798 MHz and 350103.118 MHz, respectively. The models were derived from a computation whose best solution provided $N = (1.01 \pm 0.06) \times 10^{15}$ cm⁻² and $T_{exc} = 89.5 \pm 0.4$ K.

based only on two lines of ¹³CH₃OH, we found that it could be not only 20 times lower, as resulting from the ¹³C/C flux ratio, but up to 40 times under abundant than CH₃OH, as the LTE analysis suggested (see Figure 6). Thus, we determined [¹³CH₃OH] $\leq 2.1 \times 10^{-9}$. For the undetected ¹³CH₃CN, we propose upper limits assuming that the ¹³CH₃CN, ¹³CH₃OH ratio works as what we found for the main isotopologues: CH₃CN/CH₃OH $\approx 1/25$. From that conjecture, we conclude that [¹³CH₃CN] $\leq 8 \times 10^{-11}$.

One of the goals of this work was to separately analyse the emission of the A and E spin nuclear isomers; however, the analysis indicated common physical conditions (N and T) for both spin symmetries meaning that they trace the same reservoir. The opposite case would imply spin conversion processes, induced by molecular interactions on molecular ices or collisions in gas phase, favouring an over-abundance in a given spin symmetry. However, this result would be expected at earlier stages or in younger stellar sources (e.g. Minh et al. 1993; Wirström et al. 2011).

Independently of the employed analysis, we find that both the A and E isomers contribute with approximately half of the total emission of CH₃OH and CH₃CN. The percentage derived from both methods are listed in Table 7, as explained above, assuming a total abundance as A + E. As a similar result, MacDonald et al. (1996) determined the ratio $[A-CH_3OH]/[E-CH_3OH]\approx 0.9$ for the hot molecular core G34.3+0.15. In the envelopes around low-mass protostars, Jørgensen et al. (2005) found ortho-to-para ratios close to unity for CH₃OH. Recently, similar results were reported for the pairs A/E-CH₃CN, A/E-CH₃OH and A/E-CH₃CHO in the Orion Bar photodissociation region (PDR) (Cuadrado et al. 2017).

This tendency is similar to another aspect aimed to examine in this study: whether o- and p-H₂ collisional partners may affect the population of the A and E nuclear spin isomers of CH₃OH and CH₃CN. However, the MCMC/RADEX computations yielded H₂ orthoto-para ratios close to unity.

Laboratory studies have offered new measurements about spin conversion, generally assumed as improbable processes. In gas phase, Sun et al. (2015) found that molecular collisions induce the interconversion of spin in molecules of methanol exhibiting a rate that decreases as the pressure increases with the number of collisions. A quantum relaxation mechanism explains the interconversion. In condensed phase, Lee et al. (2006) observed that methanol can suffer a slow conversion from the Eto the A symmetry when it is trapped in a solid matrix of p-H₂ prepared at low temperatures (~ 5 K). In theoretical works, Rabli & Flower (2010) studied the implications of methanol colliding with o-H₂ and p-H₂, finding qualitative and quantitative differences for those cases.

As we summarized in Table 4, emission of torsionally excited methanol was also evidenced, however only one line appeared without blended emission: A-CH₃OH (ν_t =1) at ~337969 MHz. In some studies it has been analysed whether excited transitions may be populated at the same LTE conditions of the ground torsional levels (e.g. Lovas et al. 1982; Menten et al. 1986; Sutton et al. 1995; Ren et al. 2011; Sánchez-Monge et al. 2014). We made a minor examination based on the unique line without contamination; however, we found that, to account for the flux, different regimes are needed with $T_{exc} \gtrsim 90$ K. Complementary observations with APEX will be carried out to determine with a better precision the physical conditions of excited methanol in G331.

5.2. The hot and cold components of G331

In this study we identified two different components that could candidate G331 as a Hot Molecular Core (HMC). Other characteristics also support this hypothesis. For instance, the source is compact, harbours a massive and energetic molecular outflow, and is embedded in a H II region where masers of OH and CH₃OH reveal a high star formation activity. Besides, the temperatures of the gas components are above and below the limits where evaporation of icy mantles plays an important role; for instance, to explain abundances of Complex Organic Molecules and aspects related to the age of the source. The gas densities also support such classification, since we determined H₂ densities typical of a dense medium $(0.7-1) \times 10^7$ cm⁻³.

N- and O-bearing molecules help to diagnose the presence of different gas components in HMCs (e.g. Beuther & Sridharan 2007; Fontani et al. 2007). In this first work, it is proposed that A/E-CH₃CN and A/E-CH₃OH trace a hot ($T_k \simeq 141$ K) and cold ($T_k \simeq 75$ K) component with sizes around 4.5" and 5.3", respectively. Such results are in agreement with ALMA obser-



Figure 7. Comparison of the abundances of CH_3OH (red) and CH_3CN (black) in hot molecular cores, clouds, PDR regions. Values were collected from Sutton et al. (1995); MacDonald et al. (1996); Helmich & van Dishoeck (1997); Gibb et al. (2000); Nummelin et al. (2000); Ren et al. (2011); Crockett et al. (2014) and Cuadrado et al. (2017).

vations of CO (7–6), SiO (8–7) and $\rm H^{13}CO^+$ (4–3). The emission of these species was found to be compacted in ~ 5" (Bronfman et al. 2008; Merello et al. 2013a; Hervías et al. 2015).

In Figure 7, we compare the abundances derived in this work with other sources. Our results are similar with the abundances reported in G34.3+0.15. More generally, our results suggest different abundances, physical components and a chemical differentiation for CH₃OH and CH_3CN in G331. The high sensitivity and spectral resolution of our single-dish spectra have been useful to realise subtle differences in the line profiles of CH₃OH and CH_3CN , inspected via the V_{lsr} and FWHM parameters listed in Tables 2 and 3. In agreement with the radiative models, these spectral signatures reveal clues on the origin and size of the emitter regions. In spite of that, and as a perspective, further observations at higher spatial resolution are needed to study the spatial distribution of different tracers in G331. As well, these observations may reveal density gradients towards the core and the outflow, allowing more accurate determinations of molecular abundances.

The fact of CH_3OH being more abundant than CH_3CN is in agreement with the chemistry that has been modelled in HMCs, where these molecules are usually referred as parent and daughter species, respectively, since CH_3OH is expected to be formed at early stages in grain surfaces, via successive hydrogenation of CO, while CH₃CN appears later via reaction between HCN and CH_3^+ (Millar et al. 1997; Nomura & Millar 2004; Guzmán et al. 2013; Loison et al. 2014). Other alternatives to produce methanol are $CH_3^+ + OH \rightarrow$ CH_3OH and $CH_2 + H_2O \rightarrow CH_3OH$. They can occur when frozen mixtures of CH₄ and H₂O are bombarded with electrons (Hiraoka et al. 2006). In addition, we have found that the physical conditions traced by those molecules might explain the regimes where pre-biotic and complex organic molecules have been detected, such

as CH_3OCH_3 , CH_3CHO , NH_2CHO and the $C_2H_4O_2$ isomers.

Nomura & Millar (2004) performed time-dependent chemical models considering observational aspects of the hot molecular core G34.3+0.15; for instance, that CH₃CN is associated to an inner and hot core region while CH₃OH traces the gas present in clumps (e.g. Hatchell et al. 1998; Millar & Hatchell 1998; van der Tak et al. 2000). Comparing our column densities with the temporal values calculated by these authors, an age of 10^4 yr may represent the epoch for G331, when substantial quantities of daughters molecules (e.g. CH₃CN) are produced without substantially decimate their parents (e.g. CH₃OH). Preliminary results derived by us, obtained with the gas-grain chemical code Nautilus (Ruaud et al. 2016),⁸ shows a CH_3OH/CH_3CN ratio similar to those derived from the observations. However, these results will be presented in a subsequent work depicting the chemistry of O-bearing molecules detected in the source.

6. CONCLUSIONS AND PERSPECTIVES

As an active interstellar laboratory, the G331.512-0.103 system exhibits a rich chemistry in organic and pre-biotic molecules. In this first article, we analysed around 70 lines of A/E-CH₃OH and A/E-CH₃CN towards the central region, abbreviated here as G331. Torsionally excited transitions of methanol were evidenced. Without contaminant emission, we identified the CH₃OH (ν_t =1) 7₁-6₁ transition. Likewise, two lines corresponding to ¹³CH₃OH were detected at 330252.798 MHz and 350103.118 MHz.

The analyses were performed including collisions with H_2 and typical radiative processes under LTE conditions, namely: rotational diagrams. Both analyses coincide in that CH₃OH traces a cold component while

⁸ http://kida.obs.u-bordeaux1.fr/networks.html

CH₃CN traces a hotter core with kinetic temperatures of ~ 74 K and 141 K, respectively. Likewise, the best-fits indicated emitter regions of around 4.5" and 5.3" for these tracers, respectively, with gas density $n(H_2)=(0.7-1) \times 10^7$ cm⁻³.

We treated independently each one of the nuclear spin isomers of CH₃OH and CH₃CN, and determined the ratio (A + E)-CH₃OH/(A + E)-CH₃CN $\simeq 25$. The temperatures and densities of each A and E pair suggest that they trace a same bulk and are equally populated at each local temperature. Considering that, we estimated the CH₃OH and CH₃CN abundances from the total contribution A + E. Under the LTE formalism, we estimated that [CH₃OH] $\approx 8.1 \times 10^{-8}$, [CH₃CN] $\approx 3.1 \times$ 10^{-9} , and the upper limits [¹³CH₃OH] $\approx 2.1 \times 10^{-9}$ and [¹³CH₃CN] $\approx 8 \times 10^{-11}$.

Under the perspective of hot molecular cores, the CH_3OH/CH_3CN ratio could be associated with and epoch between (10^4-10^5) yr, when daughter molecules

like CH₃CN start to be produced from parent molecules, such as CH₃OH.

We thank the anonymous referee for constructive comments and suggestions on the paper. We thank the APEX staff for their helping during the observations. N.U.D. acknowledges support from CON-ICET, projects PIP 00356 and from UNLP, projects 11G/120 and PPID/G002. L.B., R.F. and N.R. acknowledge support from CONICYT project BASAL PFB-06. E.M. and J.R.D.L. acknowledge support from the grant 2014/22095-6, São Paulo Research Foundation (FAPESP).

Facility: Atacama Pathfinder EXperiment, APEX telescope

Software: CASSIS (http://cassis.irap.omp.eu/), GILDAS (https://www.iram.fr/IRAMFR/GILDAS/), RADEX (http://doi.org/10.1051/0004-6361:20066820) NAUTILUS (https://doi.org/10.1093/mnras/stw887)

REFERENCES

- Andersson, M., Askne, J., & Hjalmarson, A. 1984, A&A, 136, 243
- Araya, E., Hofner, P., Kurtz, S., Bronfman, L., & DeDeo, S. 2005, ApJS, 157, 279
- Beuther, H., & Sridharan, T. K. 2007, ApJ, 668, 348
- Beuther, H., Zhang, Q., Greenhill, L. J., et al. 2005, ApJ, 632, 355
- Bevington, P. R., & Robinson, D. K. 2003, Data reduction and error analysis for the physical sciences, 3rd ed., by Philip R. Bevington, and Keith D. Robinson. Boston, MA: McGraw-Hill, ISBN 0-07-247227-8, 2003.
- Blake, G. A., Sutton, E. C., Masson, C. R., & Phillips, T. G. 1987, ApJ, 315, 621
- Boucher, D., Burie, J., Bauer, A., Dubrulle, A., & Demaison, J. 1980, Journal of Physical and Chemical Reference Data, 9, 659
- Bronfman, L., Garay, G., Merello, M., et al. 2008, ApJ, 672, 391
- Caswell, J. L. 1998, MNRAS, 297, 215
- Cesaroni, R., Felli, M., Testi, L., Walmsley, C. M., & Olmi, L. 1997, A&A, 325, 725
- Crockett, N. R., Bergin, E. A., Neill, J. L., et al. 2014, ApJ, 787, 112
- Cuadrado, S., Goicoechea, J. R., Cernicharo, J., et al. 2017, ArXiv e-prints, arXiv:1705.06612
- Cummins, S. E., Green, S., Thaddeus, P., & Linke, R. A. 1983, ApJ, 266, 331

- Flower, D. R., & Watt, G. D. 1984, MNRAS, 209, 25 Fontani, F., Pascucci, I., Caselli, P., et al. 2007, A&A, 470,
- Fontani, F., Pascucci, I., Caselli, P., et al. 2007, A&A, 470, 639
- Foreman-Mackey, D., Hogg, D. W., Lang, D., & Goodman, J. 2013, PASP, 125, 306
- Fuente, A., Cernicharo, J., Caselli, P., et al. 2014, A&A, 568, A65
- Giannetti, A., Leurini, S., Wyrowski, F., et al. 2017, A&A, 603, A33
- Gibb, E., Nummelin, A., Irvine, W. M., Whittet, D. C. B.,& Bergman, P. 2000, ApJ, 545, 309
- Goldsmith, P. F., & Langer, W. D. 1999, ApJ, 517, 209
- Guan, Y., & Krone, S. M. 2007, ArXiv Mathematics e-prints, math/0703021
- Güsten, R., Nyman, L. Å., Schilke, P., et al. 2006, A&A, 454, L13
- Guzmán, V. V., Goicoechea, J. R., Pety, J., et al. 2013, A&A, 560, A73
- Hatchell, J., Thompson, M. A., Millar, T. J., & MacDonald, G. H. 1998, A&A, 338, 713
- Helmich, F. P., & van Dishoeck, E. F. 1997, A&AS, 124, doi:10.1051/aas:1997357
- Hervías, C., Bronfman, L., Merello, M., et al. 2015, in Astronomical Society of the Pacific Conference Series, Vol. 499, Revolution in Astronomy with ALMA: The Third Year, ed. D. Iono, K. Tatematsu, A. Wootten, & L. Testi, 229

- Hiraoka, K., Mochizuki, N., & Wada, A. 2006, in American Institute of Physics Conference Series, Vol. 855, Astrochemistry - From Laboratory Studies to
- Astronomical Observations, ed. R. I. Kaiser, P. Bernath, Y. Osamura, S. Petrie, & A. M. Mebel, 86–99
- Hugo, E., Asvany, O., & Schlemmer, S. 2009, JChPh, 130, 164302
- Jewell, P. R., Hollis, J. M., Lovas, F. J., & Snyder, L. E. 1989, ApJS, 70, 833
- Jørgensen, J. K., Hogerheijde, M. R., Blake, G. A., et al. 2004, A&A, 415, 1021
- Jørgensen, J. K., Schöier, F. L., & van Dishoeck, E. F. 2005, A&A, 437, 501
- Lacy, J. H., Knacke, R., Geballe, T. R., & Tokunaga, A. T. 1994, ApJL, 428, L69
- Lee, Y.-P., Wu, Y.-J., Lees, R. M., Xu, L.-H., & Hougen, J. T. 2006, Science, 311, 365
- Lees, R. M., & Baker, J. G. 1968, JChPh, 48, 5299
- Loison, J.-C., Wakelam, V., & Hickson, K. M. 2014, MNRAS, 443, 398
- Lovas, F. J., Bass, J. E., Dragoset, R. A., & Olsen, K. J. 2009, National Institute of Standards and Technology, Gaithersburg, MD.
- Lovas, F. J., Suenram, R. D., Snyder, L. E., Hollis, J. M., & Lees, R. M. 1982, ApJ, 253, 149
- MacDonald, G. H., Gibb, A. G., Habing, R. J., & Millar, T. J. 1996, A&AS, 119, 333
- Menten, K. M., Walmsley, C. M., Henkel, C., et al. 1986, A&A, 169, 271
- Merello, M., Bronfman, L., Garay, G., et al. 2013a, ApJL, 774, L7
- —. 2013b, ApJ, 774, 38
- Millar, T. J., & Hatchell, J. 1998, Faraday Discussions, 109, 15
- Millar, T. J., MacDonald, G. H., & Gibb, A. G. 1997, A&A, 325, 1163
- Minh, Y. C., Irvine, W. M., Ohishi, M., et al. 1993, A&A, 267, 229
- Müller, H. S. P., Schlöder, F., Stutzki, J., & Winnewisser, G. 2005, Journal of Molecular Structure, 742, 215
- Nomura, H., & Millar, T. J. 2004, A&A, 414, 409
- Nummelin, A., Bergman, P., Hjalmarson, Å., et al. 2000, ApJS, 128, 213

- Nyman, L.-Å., Lerner, M., Nielbock, M., et al. 2001, The Messenger, 106, 40
- Olmi, L., Cesaroni, R., & Walmsley, C. M. 1993, A&A, 276, 489
- Pickett, H. M., Poynter, R. L., Cohen, E. A., et al. 1998, JQSRT, 60, 883
- Rabli, D., & Flower, D. R. 2010, MNRAS, 406, 95
- Remijan, A., Sutton, E. C., Snyder, L. E., et al. 2004, ApJ, 606, 917
- Ren, J. Z., Liu, T., Wu, Y., & Li, L. 2011, MNRAS, 415, L49
- Requena-Torres, M. A., Martín-Pintado, J., Rodríguez-Franco, A., et al. 2006, A&A, 455, 971
- Risacher, C., Vassilev, V., Monje, R., et al. 2006, A&A, 454, L17
- Ruaud, M., Wakelam, V., & Hersant, F. 2016, MNRAS, 459, 3756
- Sánchez-Monge, Á., Beltrán, M. T., Cesaroni, R., et al. 2014, A&A, 569, A11
- Sandell, G., Knee, L. B. G., Aspin, C., Robson, I. E., & Russell, A. P. G. 1994, A&A, 285
- Schilke, P., Benford, D. J., Hunter, T. R., Lis, D. C., & Phillips, T. G. 2001, ApJS, 132, 281
- Solomon, P. M., Jefferts, K. B., Penzias, A. A., & Wilson, R. W. 1971, ApJL, 168, L107
- Sun, Z.-D., Ge, M., & Zheng, Y. 2015, Nature, 6877, 1
- Sutton, E. C., Jaminet, P. A., Danchi, W. C., & Blake, G. A. 1991, ApJS, 77, 255
- Sutton, E. C., Peng, R., Danchi, W. C., et al. 1995, ApJS, 97, 455
- Tahani, K., Plume, R., Bergin, E. A., et al. 2016, ApJ, 832, 12
- van der Tak, F. F. S., Black, J. H., Schöier, F. L., Jansen, D. J., & van Dishoeck, E. F. 2007, A&A, 468, 627
- van der Tak, F. F. S., van Dishoeck, E. F., & Caselli, P. 2000, A&A, 361, 327
- Willacy, K., Williams, D. A., & Minh, Y. C. 1993, MNRAS, 263, L40
- Wilson, T. L., & Rood, R. 1994, ARA&A, 32, 191
- Wirström, E. S., Geppert, W. D., Hjalmarson, Å., et al. 2011, A&A, 533, A24
- Wyrowski, F., Schilke, P., Walmsley, C. M., & Menten, K. M. 1999, ApJL, 514, L43
- Xu, L.-H., & Lovas, F. J. 1997, Journal of Physical and Chemical Reference Data, 26, 17