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26 ABSTRACT

27 The aim of this study was to evaluate changes in polyphenol profile and antioxidant 28 capacity of five soluble coffees throughout a simulated gastro-intestinal digestion, 29 including absorption through a dialysis membrane. Our results demonstrate that both 30 polyphenol content and antioxidant capacity were characteristic for each type of studied 31 coffee, showing a drop after dialysis. Twenty-seven compounds were identified in 32 coffee by HPLC-MS, while only 14 of them were found after dialysis. Green+roasted 33 coffee blend and chicory+coffee blend showed the highest and lowest content of 34 polyphenols and antioxidant capacity before in vitro digestion and after dialysis, 35 respectively. Canonical correlation analysis showed significant correlation between the 36 antioxidant capacity and the polyphenol profile before digestion and after dialysis. 37 Furthermore, boosted regression trees analysis (BRT) showed that only 4 polyphenol 38 compounds (5-p-coumaroylquinic acid, quinic acid, coumaroyl tryptophan conjugated 39 and 5-O-caffeoylquinic acid) appear to be the most relevant to explain the antioxidant 40 capacity after dialysis, being these compounds the most bio-accessible after dialysis. To 41 our knowledge, this is the first report matching the antioxidant capacity of foods with 42 the polyphenol profile by BRT, which opens an interesting method of analysis for future 43 reports on the antioxidant capacity of foods.

44

45 KEYWORDS

46 polyphenol profile, *in vitro* gastro-intestinal digestion, bioaccessibility, chemometrics,

47 boosted regression trees.

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51 INTRODUCTION

52 Coffee is one of the most popular beverages consumed in the world, which has been 53 consumed for its pleasant flavor and aroma as well as its stimulatory properties due to 54 its caffeine content. In recent years, there has been an increasing interest in the possible 55 positive implications of coffee consumption for human health¹.

56 Coffee has been proposed as an important source of antioxidants in the human diet. 57 Epidemiological studies show that moderate coffee consumption may help to prevent chronic diseases such as Type 2 Diabetes Mellitus², Parkinson's³ and liver⁴ diseases. 58 Species (Coffea arabica, Arábica or Coffea canephora, Robusta), cultivars⁵, origin⁶, 59 process and degree of roasting⁷ and different extraction processes⁸ can influence the 60 61 chemical composition and biological activity of coffee infusions. Among the compounds having antioxidant activity in coffee, we can find phenolic compounds¹, 62 melanoidins and other products of Maillard reaction⁹. 63

Mullen et al.¹⁰ found a statistically significant correlation between the content of 64 65 caffeovlquinic acids in coffee berries and its ability to scavenge free radicals. On the other hand, Somoza *et al.*¹¹ demonstrated that chlorogenic acid was the compound with 66 67 the highest influence on the antioxidant activity, evaluated *in vitro* via the inhibition of 68 the peroxidation of linoleic acid. However, some phenolic compounds can be lost in the roasting process¹², forming new compounds derived from Maillard reaction, with 69 antioxidant properties⁸. Nicoli et al.¹³ found that grains with intermediate roasting 70 conditions had high antioxidant capacity. Daglia et al.¹⁴ showed that melanoidins 71 72 resulting from Maillard reaction would be responsible for the antioxidant capacity found 73 in high molecular weight fractions in roasted coffee.

These studies demonstrate that *in vitro* antioxidant capacity of coffee would be influenced by its composition and by the roasting process, but little is known about the

76 bioavailability that these antioxidants have in the human digestive system, as a previous 77 step to their absorption and distribution to exert effects at the cellular level. Thus, only 78 the compounds that are able to tolerate the conditions found throughout the gastro-79 intestinal tract, crossing the intestinal membrane, will be able to produce physiological 80 changes in the human body. So far, models simulating in vitro digestion have been developed to investigate the digestibility and bioaccessibility of polyphenols¹⁵. These 81 82 models simulate the movement of food through the digestive tract, exposing the food to 83 the conditions encountered in the gastric and intestinal canals. This involves the 84 addition of digestive enzymes (pepsin, pancreatin) and bile salts, with pH and 85 temperatures similar to the conditions found *in vivo*. Then dialysis may be performed, 86 where the substances that could potentially pass through intestinal wall, can be 87 measured by one or more chemical-biochemical assays. This methodology has been 88 proposed as an estimation of bioaccessibility of food components in different food matrices^{15–17}. However, to our knowledge, it has been used in only one research work to 89 90 evaluate the bioaccessibility of polyphenols in coffee, but without using the dialysis 91 step¹⁸.

The main goal of this study was to assess the bioaccessibility of polyphenols arising from soluble coffees by *in vitro* digestion, including a final dialysis to simulate intestinal absorption, looking to understand how antioxidants present in coffee and other foods may exert their effect in the human body.

96

97 MATERIAL AND METHODS

98 Coffee samples. Soluble coffee samples (regular, decaffeinated, Arabic, green + roasted
99 blend and chicory + coffee blend) were commercially obtained in supermarkets from

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100 Tres Cantos, Madrid, Spain. Three random samples from each variety, corresponding to 101 different commercial brands, were obtained (n=15).

Chemicals and Materials. Ultra-pure water ($<5 \mu g L^{-1} TOC$) was obtained from a 102 103 purification system Arium 61316-RO plus Arium 611 UV (Sartorius, Germany). 104 Methanol (HPLC grade) and formic acid (puriss. p.a. for mass spectroscopy) were 105 provided by J. T. Baker (Edo. de México, México) and Fluka (Steinheim, Germany), 106 respectively. Commercial standards of ferulic acid and caffeic acid were obtained from 107 Extrasynthèse (Genay, France), 5-O-caffeoylquinic acid and quinic acid were purchased 108 from Sigma-Aldrich (Steinheim, Germany) and p-coumaric acid was provided by Fluka 109 (Dorset, U.K.). Filters (0.45 µm, HVLP04700) were obtained from Millipore (São 110 Paulo. Brazil). ABTS (2,2'-azino-bis-(3-thylbenzothiazolne-6-sulfonic acid) 111 diammonium salt), TPTZ (2,4,6-tripyridyl-S-triazine), Trolox (6-hydroxy-2,5,7,8-112 tetramethyl-chroman-2-carboxylic acid), 1,10-phenanthroline, pepsin (P-7000, from 113 porcine stomach mucosa), pancreatin (P-1750, from porcine pancreas) and bile extract 114 (B-6831, porcine) were provided by Sigma-Aldrich (Switzerland). Folin-Ciocalteu 115 reagent was obtained from Panreac (Barcelona, Spain). Dialysis bag was Medicell 116 7000/2, width 34 mm, 7000 MW cut off. All other reagents were of analytical grade.

Sample preparation. Coffee brews were prepared according to manufacturer's instructions: 3 g of regular (RC), decaffeinated (DC), Arabic (AC), and green + roasted (GRC) coffee samples, and 6 g of chicory + coffee blend (CC) samples, were dissolved with 200 mL of hot ultra-pure water (70-75°C). Coffee brews obtained were then filtered, fractionated and stored at - 20°C until analysis.

122 In vitro digestion. The *in vitro* digestion procedure was performed according to the 123 method described by Ramírez-Moreno *et al.*¹⁷ with minor modifications. Briefly, 25 mL 124 of sample were adjusted to pH 2.0 with 6 M HCl and successively incubated in a

125 shaking water bath for 2 h at 37°C with 120 μ L pepsin solution (40 mg mL⁻¹ in 0.1 M 126 HCl) to simulate gastric digestion. After incubation 1.5 mL pancreatin-bile solution (5 127 mg pancreatin plus 25 mg porcine bile mL⁻¹ in 0.1 M NaHCO₃), was added to simulate 128 intestinal digestion. The digestion products were placed in a dialysis bag and dialyzed in 129 250 mL of sodium bicarbonate solution (pH 7.5) for 3 h. Dialyzed samples were then 130 filtered, fractionated and stored at - 20°C until analysis.

131

Total polyphenol content (TP), polyphenolic profile (PP) and antioxidant capacity of
different coffee samples were studied before and after undergoing *in vitro* gastrointestinal digestion and subsequent dialysis.

135

136 Total Polyphenols Assay. TP of coffee was measured by the Folin-Ciocalteu (FC) method, in accordance to the technique reported by Vignoli *et al.*⁸ and slightly modified 137 according to Parry et al.¹⁹. Coffee brews (100 µL), 10-fold diluted with ultrapure water, 138 139 and 500 μ L of dialyzed sample were added to 3.3 and 2.9 mL of ultrapure water, 140 respectively. Folin-Ciocalteu reagent (200 μ L) was added to each sample and stirred 141 (vortex). After 1 min, 600 μ L of sodium carbonate solution (20% v/v) were added. The 142 samples were stirred and incubated for 2 h at room temperature in the dark. The 143 absorbance was then read at 750 nm. Gallic acid was used as standard; results are 144 expressed as g gallic acid equivalents (GAE) per 100 g fresh weight. All samples were 145 analyzed in triplicate.

LC-MS. Polyphenols were analyzed in coffee and dialyzed samples by HPLC-MS/MS
method, using an Agilent Technologies 1200 Series UPLC equipped with a gradient
pump (Agilent G1312B SL Binary), solvent degasser (Agilent G1379 B) and
autosampler (Agilent G1367 D SL+WP). The chromatographic separation was achieved

150 on a LUNA (Phenomenex, Torrance, CA, USA) C18 column (5 μ m, 250 mm \times 4.60 mm 151 i.d.) at 35°C using a column heater module (Agilent G1316 B). The mobile phase 152 consisted of 0.5% formic acid (v/v, solvent A) and 0.5% acid formic in methanol (v/v, 153 solvent B). The solvent gradient started with 20% B and changed to 50% B along 3 min, 154 kept 5 min, followed by a second ramp to 70% B along 7 min, maintained 5 min, a third 155 ramp to 80% B along 1 min, remaining at this last condition for 9 min before the next run. The flow rate was set at 0.4 mL min⁻¹ and the injection volume was 40 μ L. The 156 157 HPLC system was connected to a photodiode array detector (Agilent G1315 C Starlight 158 DAD) and subsequently to a QTOF mass spectrometer (micrOTOF-QII Series, Bruker), 159 equipped with electro spray ionization (ESI) source. UV-Vis spectra were registered 160 from 200 to 600 nm. Mass spectra were recorded in negative ion mode between m/z 50 161 and 1000. The working conditions for the ionization source were as follows: capillary voltage, 4500 V; nebulizer gas pressure, 4.0 bar; drying gas flow, 8.0 L min⁻¹ and 180°C 162 163 for the drying gas. Nitrogen and argon were used as nebulizer/dryer and collision gases, 164 respectively. The MS detector was programmed to perform MS and alternative MS/MS 165 from the three most abundant ions obtained in MS. MS/MS was performed using 166 collision energy of 13.0 eV. Exact mass was verified by introducing sodium formiate at 167 the beginning and at the end of each chromatographic run through the multipath valve 168 of the MicroQTOF II, located between the DAD and the ESI source. Data acquisition 169 and processing were performed using Compass Version 3.1 software and DataAnalysis 170 Version 4.0 software, respectively (Bruker Daltonics, MA-USA). 171 Polyphenols present in samples were characterized according to their retention times,

Polyphenols present in samples were characterized according to their retention times,
exact mass, UV/Vis spectra, MS and MS/MS spectra in addition to comparison with
authentic standards when available. When authentic standards were not available, a
tentative identification was performed using UV-VIS, exact MS and MS/MS,

175 considering reports from tentative compounds in the literature. Quantification of 176 polyphenols was based on external calibration curves from available phenolic standards, 177 using the mass peak areas obtained from the extracted ion chromatograms, at concentrations between 1 and 100 mg L⁻¹. When the corresponding standards were not 178 179 available, the quantification was performed using an external standard with a similar 180 structure of the tentative compound in question. Samples and standards solutions were 181 filtered (0.45 µm) and injected in HPLC-MS/MS system. All samples were analyzed in 182 duplicate.

183 In vitro antioxidant analysis.

In vitro antioxidant activity was measured using trolox equivalent antioxidant capacity
(TEAC) assay, ferric reducing ability of plasma (FRAP) assay and 1, 10-phenanthroline
(PHEN) assay.

TEAC assay. TEAC assay was performed using adaptations of the methodology 187 described by Re *et al.*²⁰ The ABTS radical was produced by reacting 7 mM ABTS and 188 189 2.45 mM potassium persulfate (final concentration in 10 mL of water), keeping the 190 mixture in the dark at room temperature for 16 h before use. The aqueous ABTS⁺⁺ 191 solution was diluted with PBS (pH = 7.4) to an absorbance of 0.80 ± 0.02 at 734 nm. 192 Hundred fifty microliters of coffee brews (diluted 1:37.5 with ultra pure water) or 193 dialyzed samples were added to 4 mL of the TEAC solution plus 100 µL of methanol, 194 incubated for 30 min in the dark, and measured at 734 nm. The standard curve used was 195 linear between 0 and 0.02 mM trolox. Results are expressed in mmol trolox equivalents 196 (TE) per 100 g fresh weight. All samples were analyzed in triplicate.

FRAP assay. FRAP assay was performed according to Benzie and Strain²¹ with some modifications. The fresh working solution was prepared by mixing 25 mL acetate buffer pH 3.6 (3.1 g $C_2H_3NaO_2.3H_2O$ and 16 mL $C_2H_4O_2$), 2.5 mL of a 10 mM TPTZ solution in 40 mM HCl, and 2.5 mL of a 20 mM FeCl₃.6H₂O solution. Three hundred microliters of coffee brews (diluted 1:60 with ultra pure water) or dialyzed samples were added to 4 mL of the FRAP solution plus 200 μ L of methanol, incubated for 30 min in the dark and measured at 593 nm. A linear dynamic range between 0 and 0.02 mM trolox was observed. Results are expressed in mmol TE per 100 g fresh weight. All samples were analyzed in triplicate.

PHEN assay. PHEN reagent and assay was performed according to Berker *et al.*²². Five hundred microliters of coffee brews (diluted 1:10 with ultra pure water) or 1 mL of dialyzed samples were added to 1 mL of PHEN reagent adding 4.5 mL of ethanol (96%) to coffee samples and 4 mL to dialyzed samples, and diluting to 25 mL with ultra pure water. Samples were incubated for 30 min at 50°C and measured at 510 nm. The method was linear between 0 and 0.04 mM trolox. Results are expressed in mmol TE per 100 g fresh weight. All samples were analyzed in triplicate.

213

214 Statistical Analysis.

215 Results were analyzed using the statistical package Statistica 8.0 from StatSoft Inc.

216 (2007) and the Infostat software package 23 .

217 Analysis of Variance. ANOVA was performed using mixed models²⁴; in the case of 218 significance (P < 0.05), a DGC²⁵ comparison test was performed to reveal paired

- 219 differences between means.
- *Canonical Correlation Analysis* (CCA). CCA was used to study the correlation between
 antioxidant capacity (FRAP, TEAC and PHEN) and polyphenol profile of coffee
 samples, before and after *in vitro* digestion and the subsequent dialysis.
- 223 Boosted Regression Trees Analysis (BRT). Development of predictive models to get
- evidences on the relationship between the polyphenol profile and the antioxidant

activity was carried out using a statistical technique called boosted regression trees²⁶. 225 226 BRT is a powerful modeling method that combines regression trees and boosting 227 algorithm. This method can handle predictor variables with different types and distributional characteristics. Variable selection with this model is robust to co-linearity 228 229 amongst predictors, outliers and lack of data and, therefore, does not require prior variable selection or data reduction. Models were constructed with the "gbm" library²⁷ 230 231 using the R software (version 3.0.3). Model over fitting was avoided by cross validation 232 (CV). In CV, the function selects a fraction of the data provided, according to the 233 parameters set, to build a model. The latter is validated with the fraction of remaining 234 data, allowing the evaluation and confirmation of the predictive quality of the model built²⁶. Three parameters were adjusted to maximize model performance: the proportion 235 236 of data randomly selected at each iteration of the CV procedure (the "bag fraction"), the 237 contribution of each tree to the growing model (the "learning rate") and the number of 238 nodes (interactions) in each tree ("tree complexity"). Model performance was evaluated 239 using the CV correlation (the correlation between predicted and raw data withheld from 240 the model). CV correlations close to 1 indicates good model predictions. The 241 importance of predictor variables in BRT models was evaluated using the function 242 previously described, which calculates the contribution to the model fit attributable to each predictor, averaged across all trees²⁶. 243

244

245 RESULTS AND DISCUSSION

246

247 Polyphenol analysis.

Table 1 shows mean values and standard deviations of total polyphenol content (TP) of
5 types of instant coffees. Green + roasted coffee blend (GRC), Arabic coffee (AC) and

250 regular coffee (RC) samples showed similar TP values (average 15.8 g GAE/100g), 251 higher than decaffeinated coffee (DC) samples (14.8 g GAE/100g), and much higher than chicory + coffee blend (CC) samples, which showed the lowest TP values (8.99 g 252 253 GAE/100g), attributable to the substitution of coffee by chicory (60% of chicory and 38% of coffee). These results are in agreement with other authors. Vignoli et al.⁸ 254 255 showed TP values in instant coffee samples between 14.6 and 15.1 g GAE/100g. Del Pino-García et al.²⁸ showed TP values between 13.2 and 22.2 g GAE/100g in instant 256 regular coffee samples. On the other hand, Alves et al.²⁹ suggested that decaffeination 257 258 process has influence on TP levels. They detected greater amounts of TP in regular 259 coffee than in decaffeinated ones. Additionally, chicory is a plant used in Europe and 260 USA as a coffee substitute because it does not have caffeine. Normally it is used in chicory + coffee blends to reduce dietary caffeine intake³⁰, however this plant has lower 261 262 amount of bioactive constituents (flavonoids, caffeic acid derivatives and other polyphenols) than $coffee^{31,32}$. 263

With respect to individual polyphenol constituents, 27 compounds were identified in coffee samples, which can be divided into 5 groups: 1 quinic acid, 20 free chlorogenic acids, 3 chlorogenic lactones and 3 hydroxycinnamoyl-amino acid conjugates. **Table 2** shows the parameters used for its identification.

Compounds identified in this study agree with those informed by others authors. Mullen *et al.* $(2011)^{10}$ found 16 compounds derived to chlorogenic acids in coffee fruit sample, whereas Rodrigues *et al.* $(2013)^{33}$ found 26 phenolic compounds in coffee brews.

The most abundant compounds in coffee samples were QA, 5-CQA and 5-FQA (**Table** 3). Our current results showed that the content of QA ranged between 576 and 1700 mg/100g, 5-CQA ranged between 353 and 1549 mg/100g, while 5-FQA ranged between 429 and 1327 mg/100g. All the compounds presented significant differences among

studied coffee samples and its values are in agreement with those reported in the literature^{33,34}. GRC, RC and AC showed the highest values in practically all the compounds while CC displayed the lowest ones.

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In vitro digestion model gives an indication as to the availability of coffee antioxidants in a biological system, because this model simulates *in vivo* digestion. It is assumed that the amount of dialyzable polyphenol compounds could be bio-accessible in the intestine.

After *in vitro* digestion, a decrease of TP was observed during dialysis through the semi permeable cellulose membrane (**Table 1**). The dialyzed samples showed TP values 5fold lower than the coffee samples. GRC dialysates showed the highest TP values (3.82 gGAE/100g), followed by AC (3.63 gGAE/100g), and while CC dialysates showed the lowest TP values (1.78 g GAE/100 g), following the same trend that was observed in non-digested samples.

289 Individual results for each of the polyphenol compounds investigated and their recovery 290 percents (R%) after dialysis are presented in Table 3. We observed substantial losses in 291 some of the polyphenol compounds after dialysis in relation to their initial content in 292 coffee samples. Thus, only 14 out of 27 compounds identified in the coffee samples 293 were quantified in dialyzed samples. Nine of them were quantified in all dialysates. 294 Conversely, CA and 5-diMCiQA were only quantified in AC and GRC dialysates; while 295 3,4-diCQA and 4,5-diCQA were quantified in GRC, and 3-diMCiQA was only 296 quantified in AC.

QA (431.6 - 810 mg/100g) was the most abundant compound found in dialyzed
samples, with R% ranging from 47 to 75%, followed by CoT (22 - 40 mg/100g; R%
19-42%). It is worthy to remark that 3-diMCiQA presented the highest R% (84.6) in

300 dialysates of Arabic coffee (AC). Conversely, the greatest drop was observed with 5-

301 FQA, showing an overall R% of 2.4, followed by 5-CQA (R% = 2.62).

302 Several studies have shown that the bioaccessibility of different families of polyphenols in different matrices was lower than 40% when a dialysis bag was used. Gil-Izquierdo et 303 al.¹⁵ observed bioaccessibilities between 11% and 36% in flavanones from orange juice. 304 Vallejo et al.³⁵ obtained a total flavonoid bioaccessibility of 6% in broccoli. Akillioglu 305 and Karakaya¹⁶ reported that the bioaccessibility of TP verified by Folin's method 306 307 ranged 19% to 39%, in bean varieties. However, to our knowledge, there is only one 308 report on the *in vitro* bioaccessibility of polyphenols in instant coffee, using an *in vitro* digestion model and ultrafiltration step¹⁸. Our current work uses dialysis bags, which is 309 more close to physiological conditions, avoiding the use of high pressure 310 311 (ultrafiltration), so our results could be better extrapolated with results from *in vivo* models, in which the bioavailability of polyphenols is studied $^{36-38}$. 312

Then, this methodology constitutes a feasible approach to determine the potential availability of polyphenols. The polyphenols released from the food matrix during the digestive process (named bio-accessible polyphenols) are potentially bio-available to absorption through the gut barrier, and these may be useful for the interpretation of the effects of food polyphenols on health.

318

319 Antioxidant capacity.

Table 4 shows antioxidant capacity of different coffee samples before *in vitro* digestion
and after dialysis, using TEAC, FRAP and PHEN *in vitro* assays.

A similar trend was observed from three assays: RC and AC presented the highest antioxidant capacity, while the lowest values were observed in CC samples. Significant correlation was observed between different antioxidant tests. This was confirmed by a

study of the correlations between them: FRAP/ PHEN (r = 0.911; P < 0.01), FRAP/ 325 326 TEAC (r = 0.745; P < 0.01), and PHEN /TEAC (r = 0.682; P < 0.01). These findings are consistent with those reported in other studies involving TEAC and FRAP assays. 327 Vignoli et al.8 reported similar values of FRAP (76.99-139.99 mmol ET/100g) and 328 329 TEAC (74.99-144.03 mmol ET/100g) in Arabic and Robust coffee with different levels of roasted, while Rufián-Henares et al.³⁹ showed similar ABTS and FRAP values (127 330 331 and 48 mmol ET/100g, respectively) in soluble coffee. On the other hand, Del Pino-García et al.²⁸ found elevated correlation between ABTS and FRAP values in instant 332 333 coffee samples (r = 0.9311). 334 Differences between caffeinated and decaffeinated samples were also reported by others

334 Differences between caffeinated and decaffeinated samples were also reported by others 335 authors. Pellegrini *et al.*⁴⁰ showed differences between caffeinated and decaffeinated 336 espresso coffee samples when they studied antioxidant capacity by TEAC and FRAP 337 assays, while Niseteo *et al.*⁴¹ found the same differences in instant coffee brews.

Chicory + coffee blend (CC) samples showed an antioxidant capacity nearly half less than in coffee samples (RC, AR and GRC) (P < 0.01). These results are consistent with those found by Rautenbach *et al.*⁴² They found that blended coffee-chicory had 2.5-fold less antioxidant capacity than pure coffee (ORAC assay).⁴² To our knowledge, there are not reports showing values of antioxidant capacity using TEAC, FRAP and PHEN assays in chicory + coffee blended.

344

After dialysis a similar trend was also observed. The antioxidant capacity of GRC dialysates was the highest followed by RC and AC, while CC dialysates remained with the lower antioxidant capacity. The correlation between antioxidant capacity assays was also confirmed with dialysates: FRAP/ PHEN (r = 0.537; P < 0.01), FRAP/ TEAC (r =0.824; P < 0.01), and PHEN/TEAC (r = 0.663; P < 0.01). FRAP, TEAC and PHEN showed a drop in all dialyzed samples, regarding the antioxidant capacity in coffee samples. Recovery percents range from 12.7 to 35.7 %, being higher in the mixture with green coffee (GRC), which means a higher antioxidant capacity in the material potentially absorbed from this type of coffee, in agreement with the higher R% obtained in this sample for TP and the content of caffeic acid (**Tables 1** and **3**).

355 Several studies showed that the antioxidant capacity decrease after dialysis in different matrices. Bouayed et al.43 reported a decrease in the total antioxidant capacity of 356 357 dialyzable compounds, compared to those observed in fresh apples (57% and 46% for FRAP and ABTS test, respectively). Rodríguez-Roque et al.⁴⁴ showed that the 358 359 antioxidant capacity after dialysis decreased by 73% in soymilk. Akillioglu and Karakava¹⁶ observed a reduction ranging 1.6 to 2.1-fold in the DPPH test of pinto beans 360 after in vitro gastrointestinal digestion. However, to our knowledge, there are not 361 reports that show values of antioxidant capacity after dialysis step in coffee samples. 362

363

364 Matching between *in vitro* antioxidant capacity and polyphenol profile.

365 The antioxidant potential of coffee is attributed to the presence of polyphenols and 366 melanoidins, but their contribution to the antioxidant capacity varies with the intensity 367 of the roasting process, showing discrepancies in the results obtained by different authors^{7,8,45}. The contribution of high molecular weight melanoidins to the antioxidant 368 capacity of coffee brews was estimated in the range of 26-38 %.³⁹ However Delgado-369 Andrade et al.45 found that the antioxidant capacity of melanoidins depends on the 370 371 presence of low molecular weight compounds (polyphenols), linked non-covalently to the melanoidin skeleton. Other authors⁴⁶ found that only a small proportion of 372 melanoidins could by absorbed through the intestinal wall. 373

374 To evaluate if the antioxidant capacity found in this work is dependent on the 375 polyphenol content (before *in vitro* digestion and after dialysis), we applied simple 376 correlation analysis. We observed statistically significant correlation between TP and 377 the antioxidant capacity determined by FRAP, TEAC and PHEN assays before in vitro digestion (r = 0.753; 0.871 and 0.693, respectively) and after dialysis (r = 0.771; 0.851 378 379 and 0.717, respectively). These results are in agreement with others authors, who evaluated the antioxidant capacity in coffee samples.^{41,47} So, the antioxidant capacity 380 381 found in this work, evaluating instant coffee samples (before in vitro digestion and after 382 dialysis) could be attributed to their polyphenol profile. To confirm this hypothesis, we 383 evaluated the correlation between antioxidant capacity (FRAP, TEAC and PHEN) and 384 the entire polyphenol profile of coffee and dialyzed samples using canonical correlation 385 analysis (CCA). Before in vitro digestion CCA showed significant correlation between antioxidant capacity and polyphenol profile of coffee samples ($r^2 = 0.99$; P < 0.001). 386 387 Additionally, CCA showed significant correlation between FRAP, TEAC and PHEN with the polyphenol profile after dialysis ($r^2 = 0.93$; P < 0.001, using 9 polyphenols 388 389 quantified in all dialyzed samples). So far, CCA evidenced that the antioxidant capacity 390 of coffee samples can be reasonably linked to the polyphenol profiles in both pre-391 digested and dialyzed samples.

We were also interested in evaluating the contribution of individual polyphenols to the antioxidant capacity, looking for evidences on different contribution of individual compounds to the antioxidant capacity. To solve this question, we applied Boosted Regression Trees (BRT). Although BRT methods is applied in various fields including ecology⁴⁸, epidemiology⁴⁹, agriculture⁵⁰ and highway safety⁵¹; to our knowledge, this methodology has never been applied in food science. BRT identifies important predictor variables, enabling complex functions to be modeled (antioxidant capacity), without making assumptions about the type of data. BRT have some advantages over other multivariate statistical techniques, such as multiple regression, because it is robust to missing data, variable outliers, variable co-linearity, focusing on predictive accuracy rather than *P*-values to indicate the significance of model coefficients⁴⁸.

403 The adjusted parameters (bag fraction, learning rate and tree complexity), performance 404 (CV correlation and number of trees) and relative influence of polyphenols for each 405 model (to TEAC, FRAP and PHEN) in coffee and dialyzed samples are presented in 406 Table 5. Before in vitro digestion and after dialysis BRT models showed good 407 performance (CV correlation) in TEAC as well as FRAP and PHEN analysis (Table 5). 408 BRT models showed that 90 % of the variability found in TEAC, FRAP and PHEN 409 analyses before *in vitro* digestion could be explained using 14 or 16 (relative influence 410 in bold, **Table 5**) out of 27 quantified compounds. On the other hand, after dialysis only 5 (relative influence in **bold**, **Table 5**) out of 9 quantified polyphenols are necessary to 411 412 explain 95 % of the variability observed in TEAC, FRAP and PHEN assays.

4-FQA was the most influential variable with a relative contribution of 15.9% to TEAC BRT model in coffee samples, while CQL was the most significant variable for FRAP and PHEN BRT models (27.3 % and 30.1 %, respectively). Additionally, other predictor variables (CoT, FT, 3-FQL, 4-FQA, QA, 5-CQA, FA and 3-CQA) were also important in TEAC, FRAP and PHEN models. Although these techniques explain different mechanisms of action of the polyphenols, these compounds appear to be the most relevant to explain the total antioxidant capacity before *in vitro* digestion.

On the other hand, QA was the most significant predictor for TEAC BRT model (51.8
%) in dialyzed samples, whereas 5-CoQA showed the highest contribution to FRAP and
PHEN BRT models (45.3 % and 37.6 %, respectively). Moreover, 5-CoQA, QA, CoT
and 5-CQA appeared in all models; so these variables appear to be the most relevant to

424 explain the antioxidant capacity after dialysis. In addition, QA, CoT and 5-CoQA were 425 the most bio-accessible compounds after dialysis (Table 3), with R% of 56, 27 and 27, 426 respectively, and QA, CoT and 5-CQA were also important in TEAC, FRAP and PHEN 427 BRT models before *in vitro* digestion, while 5-CoQA was important in both FRAP and 428 PHEN BRT models. 429 The marginal effect of these polyphenols (QA, CoT, 5-CoQA and 5-CQA) on 430 antioxidant capacity could be demonstrated using partial dependence plots. The plots 431 show the association of each compound with the antioxidant capacity (fitted function) while all other compounds have an average effect in the model²⁶. Figure 1 provides the 432 433 partial dependence plots of QA, CoT, 5-CQA and 5-CoQA on TEAC (A), FRAP (B) 434 and PHEN (C) BRT models of dialyzed samples. Plots demonstrate a complex pattern 435 of variation between polyphenols (QA, CoT, 5-CQA and 5-CoQA) and antioxidant 436 capacity (TEAC, FRAP and PHEN). In general, it can be seen that antioxidant capacity

437 (TEAC, FRAP and PHEN) is high when the content of QA is greater than 0.7 g/100g,

438 values of 5-CoQA is equal to 0.0195 g/100g, CoT is approximately 0.03 g/100g and 5-

CQA is greater than 0.0235 g/100g (Figure 1 A, B and C). This condition is satisfied
mostly by GRC samples (Table 3), which would explain the greater antioxidant

441 capacity found in these samples after dialysis.

442

These results demonstrate that the antioxidant capacity of coffee samples before and after *in vitro* digestion and subsequent dialysis can be explained by the polyphenol profile. It is worthy to remark that four compounds (5-CoQA, QA, CoT and 5-CQA) appear to be the most relevant to explain the antioxidant capacity found after dialysis, regardless of the different method used to determine the action of polyphenols (TEAC, FRAP or PHEN). Additionally, three out of these four compounds (5-CoQA, QA and 449 CoT) were the most bio-accessible after dialysis. Assuming that dialysis simulates the 450 absorption of compound in the small intestine, we could affirm that these compounds 451 could be available for absorption *in vivo*, influencing cellular activities that moderate 452 the risk of several diseases and could be potentially beneficial for human health.

453 Our study has also shown that BRT method is a useful analytical tool to study the 454 contribution of polyphenols to the antioxidant capacity. To our knowledge, this is the 455 first report using this approach in food science.

456

457 ABBREVIATIONS

458 Regular (RC), decaffeinated (DC), Arabic (AC), green + roasted coffee blend (GRC) 459 chicory + coffee blend (CC). Total polyphenols content (TP). Quinic acid (QA), 3-O-460 caffeoylquinic acid (3-COA); 4-O-caffeoylquinic acid (4-COA); 1-O-feruloylquinic 461 acid (1-FQA); cis-5-O-caffeoylquinic acid (cis5-CQA); 5-O-caffeoylquinic acid (5-462 CQA); 3-O-feruloylquinic acid (3-FQA); 4-O-feruloylquinic acid (4-FQA); 463 caffeoylquinic lactone (CQL); 5-p-coumaroylquinic acid (5-CoQA); 5 O-feruloylquinic acid (5-FQA); caffeic acid (CA); 3-O-dimetoxicinamoylquinic acid (3-diMCiQA); 3,4-464 465 O-dicaffeoylquinic acid (3,4-diCQA); 3-O-feruloylquinic lactone (3-FQL); 4-O-466 dimetoxicinamoylquinic acid (4-diMCiQA); 5-O-dimetoxicinamoylquinic acid (5-467 diMCiQA); coumaroyl,caffeoylquinic acid (CoCQA); ferulic acid (FA); 3-O-caffeoyl,4-468 O-feruloylquinic acid (3,4-CFQA); 4,5-O-caffeoylquinic acid (4,5-diCQA);469 diferuloylquinic acid (diFQA); 4-O-caffeoyl,5-O-feruloylquinic acid (4,5-CFQA); 470 caffeoyl tryptophan conjugated (CT); dicaffeoylquinic lactone (diCQL); coumaroyl 471 tryptophan conjugated (CoT) and feruloyl tryptophan conjugated (FT). Boosted 472 Regression Trees (BRT).

473

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FIGURE CAPTIONS

Figure 1. Functions fitted for boosted regression trees (BRT) model, showing the influence of QA, 5-CQA, CoT and 5-CoQA (g/100g) and their contribution (between square brackets) to fit the TEAC (A), FRAP (B) and PHEN (C) BRT models after dialysis.

TABLES

 Table 1. Total Polyphenol content expressed in g gallic acid equivalents (GAE)/100g

sample, before *in vitro* digestion and after dialysis. Values are reported as means \pm SD.

	Coffee	Dialysates	R%
Regular Coffee (RC)	15.6 ± 0.4 a	3.33 ± 0.20 C	21.3
Arabic Coffee (AC)	15.8 ± 0.6 a	$3.63\pm0.09~\textbf{B}$	23.0
Green + Roasted Coffee blend (GRC)	16.0 ± 0.5 a	$3.82\pm0.09~A$	23.9
Chicory + Coffee blend (CC)	9.0 ± 0.2 c	$1.78\pm0.07~\textbf{D}$	19.8
Decaffeinated Coffee (DC)	$14.8\pm0.4~\textbf{b}$	3.26 ± 0.22 C	22.0

R%: percent recovery of dialysate with respect to coffee. Different letters (a > b > c > d > e to coffee, or A > B > C > D to dialysates) in the same column indicate significant differences (P < 0.05).

N°	RT (min)	Abreviature	Compound	Molecular formula	[M-H] ⁻ (m/z) experimental	[M-H] ⁻ (m/z) calculated	error (ppm)	MS ² (m/z)	λmáx (nm)
1	7,2	QA*	Quinic acid	C7H11O6	191.0565	191.0561	-1.9	-	223
2	11,2	3-CQA	3-O-caffeoylquinic acid	C16H17O9	353.0883	353.0878	-1.5	179, 191, 173	227, 288sh, 321
3	11,6	4-CQA	4-O-caffeoylquinic acid	C16H17O9	353.0887	353.0878	-2.5	191, 179	228, 292sh, 326
4	12,2	1-FQA	1-O-feruloylquinic acid	C17H19O9	367.1039	367.1035	-1.2	193, 173	-
5	12,2	cis5-CQA	putative cis-5-O-caffeoylquinic acid	C16H17O9	353.0893	353.0878	-4.1	191, 179, 173	227, 289sh, 325
6	12,6	5-CQA*	5-O-caffeoylquinic acid	C16H17O9	353.0887	353.0878	-2.6	191, 179, 173	230, 301sh, 326
7	12,7	3-FQA	3-O-feruloylquinic acid	C17H19O9	367.1045	367.1035	-2.9	193	-
8	13,8	4-FQA	4-O-feruloylquinic acid	C17H19O9	367.1043	367.1035	-2.3	173, 191, 193, 335	229, 325
9	13,8	CQL	Caffeoylquinic lactone	C16H15O8	335.0798	335.0772	-7.7	161, 173, 179	229, 325
10	13,9	5-CoQA	5-p-coumaroylquinic acid	C16H17O8	337.0905	337.0929	7.1	191, 173	-
11	14,1	5-FQA	5-O-feruloylquinic acid	C17H19O9	367.1064	367.1035	-7.9	191, 173	228, 326
12	14,2	CA*	Caffeic acid	C9H7O4	179.0353	179.0350	-1.8	-	225, 300sh, 325
13	14,6	3-diMCiQA	3-O-dimetoxicinamoylquinic acid	C18H21O9	381.1187	381.1191	1.2	207, 337	-
14	15,2	3,4-diCQA	3,4-O-dicaffeoylquinic acid	C25H23O12	515.1207	515.1195	-2.2	353, 179, 173	226, 292sh, 323
15	16,0	3-FQL	3-O-feruloylquinic lactone	C17H17O8	349.0933	349.0929	-1.3	175, 193	227, 287sh, 317
16	17,1	4-diMCiQA	4-O-dimetoxicinamoylquinic acid	C18H21O9	381.1192	381.1191	-0.1	173, 207, 337	226, 285, 320
17	17,5	5-diMCiQA	5-O-dimetoxicinamoylquinic acid	C18H21O9	381.1192	381.1191	-0.3	173, 207, 193	226, 288, 317
18	17,8	CoCQA	Coumaroyl,caffeoylquinic acid	C25H23O11	499.1280	499.1246	-6.9	191, 173, 353, 311	226, 287sh, 322
19	17,8	FA*	Ferulic acid	C10H9O4	193.0501	193.0506	2.8	-	226, 291sh, 322
20	18,1	3,4-CFQA	3-O-caffeoyl, 4-O-feruloylquinic acid	C26H25O12	529.1360	529.1351	-1.6	367, 173, 335, 179, 193	226, 300sh, 327
21	18,2	4,5-diCQA	4,5-O-caffeoylquinic acid	C25H23O12	515.1201	515.1195	-1.1	353, 173, 179	226, 294sh, 324
22	20,3	diFQA	diferuloylquinic acid	C27H27O12	543.1523	543.1508	-2.9	349, 193, 367, 173	-
23	20,7	4,5-CFQA	4-O-caffeoyl, 5-O-feruloylquinic acid	C26H25O12	529.1364	529.1351	-2.4	353, 367, 173, 179, 191	227, 287sh, 320
24	21,3	СТ	Caffeoyl tryptophan conjugated	C20H17N2O5	365.1144	365.1143	-0.4	229, 186, 203, 161	226, 286, 317
25	22,2	diCQL	putative dicaffeoylquinic lactone	C25H21O11	497.1128	497.1089	-7.7	335	226, 283sh, 319
26	22,9	СоТ	Coumaroyl tryptophan conjugated	C20H17N2O4	349.1184	349.1194	2.7	229, 186, 203	226, 282, 316
27	23,1	FT	Feruloyl tryptophan conjugated	C21H19N2O5	379.1286	379.1299	3.7	203, 229	227, 281, 318

 Table 2. Polyphenols identified in coffee samples.

RT, retention time; $[M-H]^-$ (m/z), negatively charged molecular ion; M^2 (m/z), daughter ions produced from $[M-H]^-$ fragmentation; λ_{max} , maximum absorbance for compound identification by UV-VIS spectra; sh, shoulder. Compounds marked with * were identified and quantified using authentic standards. Other compounds are tentatively proposed based on RT, accurate MS and MS/MS in according to data from the literature.

Table 3. Polyphenolic compounds identified in coffee samples before *in vitro* digestion and after dialysis. Values (mgstandard/100g sample) are reported as means \pm SD.

	Regular coffee			Arabic coffee		Green and roasted coffee blend		Chicory and coffee blend			Decaffeinated coffee				
	Coffee	Dialysate	R%	Coffee	Dialysate	R%	Coffee	Dialysate	R%	Coffee	Dialysate	R%	Coffee	Dialysate	R%
QA ¹	1700 ± 230 a	$810\pm80\mathbf{A}$	48	$1440\pm230\textbf{b}$	$780 \pm 100 \mathbf{A}$	54	$1314 \pm 24\mathbf{b}$	$708\pm18\textbf{B}$	54	$576 \pm 4c$	$431.6 \pm 1.7\mathbf{C}$	75	1700 ± 300 a	$800\pm80\mathbf{A}$	47
3-CQA	$92\pm12\textbf{b}$	< LOD	0	104 ± 14 a	< LOQ	0	$98.7\pm0.5\textbf{a}$	< LOQ	0	$26.6\pm0.3 \textbf{d}$	< LOD	0	$77 \pm 3c$	< LOD	0
4-CQA	$550\pm120\textbf{b}$	16.9 ± 1.1 C	3,1	$720\pm140\textbf{a}$	$18.1\pm0.6\textbf{B}$	2,5	$787 \pm 16a$	$20.60\pm0.11\mathbf{A}$	2,6	$136.0 \pm 1.9 \textbf{d}$	16.7 ± 0.4 C	12	460 ± 40 c	16.7 ± 0.9 C	3,6
1-FQA	$28 \pm 4\mathbf{a}$	< LOD	0	$25\pm 4 {\bf b}$	< LOD	0	21.0 ± 0.3 c	< LOD	0	$13.55\pm0.09 \textbf{d}$	< LOD	0	$25\pm3\textbf{b}$	< LOD	0
cis5-CQA	$134\pm10\textbf{b}$	$8\pm8\mathbf{C}$	6,0	$134\pm13\textbf{b}$	$15.43\pm0.23\textbf{B}$	12	$145 \pm 4a$	$15.63\pm0.25\textbf{B}$	11	$38.4 \pm 1.1 \text{d}$	$16.06\pm0.21\mathbf{A}$	42	$111 \pm 9c$	$10 \pm 8\mathbf{C}$	9
5-CQA	$1050\pm200\boldsymbol{c}$	19.7 ± 1.9 C	1,9	$1340\pm240\textbf{b}$	$23.1 \pm 1.7 \textbf{B}$	1,7	$1549 \pm 19a$	$32.40\pm0.25\mathbf{A}$	2,1	$352.9 \pm 0.3 \mathbf{e}$	18.7 ± 0.3 C	5,3	$940 \pm 100 \textbf{d}$	$19.4 \pm 2.1 \mathrm{C}$	2,1
3-FQA	$280\pm80 \textbf{a}$	$16.0\pm0.5\mathbf{A}$	5,7	$250\pm90\textbf{b}$	$16.0\pm0.5\mathbf{A}$	6,4	$195.2\pm1.7\textbf{b}$	$16.09\pm0.17\mathbf{A}$	8,2	104.1 ± 1.9 c	$16.1\pm0.5\mathbf{A}$	16	$220\pm 60 \textbf{b}$	$15.67\pm0.21\textbf{B}$	7,1
4-FQA	110 ± 30 a	$15.85\pm0.22\mathbf{A}$	14	$90\pm40\textbf{b}$	$15.8\pm0.4\mathbf{A}$	18	49.5 ± 0.5 c	$15.61 \pm 0.24 \mathbf{B}$	32	$46.5\pm1.2\textbf{d}$	$16.1 \pm 0.5 \mathbf{A}$	35	$69 \pm 16 \mathbf{b}$	$15.90\pm0.12\mathbf{A}$	23
CQL	$430\pm210\textbf{b}$	< LOQ	0	$740 \pm 140\mathbf{a}$	< LOQ	0	674 ± 16 a	< LOQ	0	$233.8\pm~0.5\mathbf{c}$	< LOQ	0	$220 \pm 90c$	< LOD	0
5-CoQA	$90 \pm 22c$	$18.9 \pm 1.2 \textbf{B}$	21	120 ± 23 b	$20.7\pm0.6\mathbf{A}$	17	136.7 ± 1.9 a	$20.6\pm0.5\mathbf{A}$	15	$33.9\pm0.8\boldsymbol{e}$	16.7 ± 0.3 C	49	$51 \pm 12 \mathbf{d}$	17.2 ± 1.0 C	34
5-FQA	$1150\pm250\textbf{b}$	$19.0 \pm 1.1 \textbf{B}$	1,6	$1000\pm 300 \textbf{b}$	$19.9 \pm 2.5\mathbf{B}$	2	$1327 \pm 6a$	$20.89\pm0.20\textbf{A}$	1,6	$429\pm7\textbf{d}$	$18.23\pm0.18\mathbf{C}$	4,2	660 ± 90 c	$17.3\pm0.4\textbf{D}$	2,6
CA ²	$19 \pm 7\mathbf{a}$	< LOQ	0	17.2 ± 1.8 a	$6.4\pm0.4\textbf{B}$	37	$14.77\pm0.12\boldsymbol{b}$	$6.97\pm0.12\mathbf{A}$	47	$4.131\pm0.014 \textbf{d}$	< LOD	0	13.7 ± 1.8 c	< LOQ	0
3-diMCiQA	24 ± 15 a	< LOQ	0	18.2 ± 1.2 a	15.39 ± 0.23	85	$15.54\pm0.08 \textbf{a}$	< LOQ	0	< LOD b	< LOD	0	$19 \pm 7a$	< LOQ	0
3,4-diCQA	$120\pm40\boldsymbol{c}$	< LOD	0	168 ± 23 b	< LOD	0	$466 \pm 6a$	16.04 ± 0.25	3,4	$28.4392\pm0.0006\boldsymbol{e}$	< LOD	0	89 ± 15 d	< LOD	0
3-FQL	$79\pm23\textbf{b}$	< LOD	0	108 ± 19 a	< LOD	0	65.5 ± 0.3 c	< LOD	0	$65.90\pm0.13 \textbf{c}$	< LOD	0	70 ± 30 c	< LOD	0
4-diMCiQA	$24 \pm 16a$	< LOQ	0	16.9 ± 1.0 a	< LOQ	0	$14.69\pm0.12 \textbf{b}$	< LOQ	0	7.15 ± 0.05 c	< LOD	0	$20\pm 8a$	< LOQ	0
5-diMCiQA	$40\pm40 \bm{a}$	< LOQ	0	$26 \pm 4a$	$17.0\pm0.6\mathbf{A}$	65	$19.53\pm0.11 \textbf{b}$	$15.8\pm0.4\textbf{B}$	81	$8.49\pm0.4\bm{c}$	< LOD	0	$29 \pm 16a$	< LOQ	0
CoCQA	19.7 ± 2.3 c	< LOD	0	$22.4 \pm 1.8 \textbf{b}$	< LOD	0	$28.2\pm0.3 \textbf{a}$	< LOD	0	$8.25\pm0.10\textbf{e}$	< LOD	0	$17.8\pm2.0\textbf{d}$	< LOD	0
FA ³	$90\pm 30 \textbf{b}$	< LOD	0	130 ± 60 a	< LOD	0	$146.0 \pm 1.4 \mathbf{a}$	< LOD	0	$68 \pm 4c$	< LOD	0	$90\pm40\textbf{b}$	< LOD	0
3,4-CFQA	$67 \pm 22\mathbf{b}$	< LOD	0	$58\pm25\mathbf{b}$	< LOD	0	111.3 ± 1.5 a	< LOD	0	$18.7\pm0.3\textbf{d}$	< LOD	0	$39.8\pm1.3\text{c}$	< LOD	0
4,5-diCQA	140 ± 50 c	< LOD	0	$180\pm60 \textbf{b}$	< LOD	0	$675.4\pm0.6\textbf{a}$	15.68 ± 0.25	2,3	$30.441\pm0.013 \textbf{e}$	< LOD	0	$89\pm18 \textbf{d}$	< LOD	0
diFQA	$32 \pm 16a$	< LOD	0	$29 \pm 6a$	< LOD	0	$23.4\pm0.6\textbf{b}$	< LOD	0	10.7 ± 0.3 c	< LOD	0	$21\pm5\textbf{b}$	< LOD	0
4,5-CFQA	$43\pm14\textbf{b}$	< LOD	0	$39\pm12\textbf{b}$	< LOD	0	$105.6\pm0.3\mathbf{a}$	< LOD	0	$12.55\pm0.11 \textbf{d}$	< LOD	0	$27.6\pm0.9\mathbf{c}$	< LOD	0
CT ²	$160\pm40\textbf{b}$	< LOD	0	$120\pm70\textbf{b}$	< LOD	0	$287\pm4\textbf{a}$	< LOQ	0	$52.2\pm0.4\textbf{d}$	< LOD	0	59 ± 11 c	< LOD	0
diCQL	$19 \pm 3\mathbf{b}$	< LOD	0	$23 \pm 4a$	< LOD	0	$21.25\pm0.03 \textbf{a}$	< LOD	0	$8.16\pm0.11 \textbf{d}$	< LOD	0	16.2 ± 1.6 c	< LOD	0

CoT ⁴	200 ± 60 a	$40\pm8\mathbf{A}$	20	160 ± 110 b	$30\pm30\mathbf{B}$	19	$123 \pm 8\mathbf{b}$	$30.9 \pm 1.4 \textbf{B}$	25	$113.6\pm0.6\textbf{b}$	$33.4\pm0.3\boldsymbol{B}$	29	$52 \pm 16c$	$22 \pm 9C$	42
FT ³	$53\pm8\boldsymbol{b}$	< LOD	0	$40 \pm 30c$	< LOD	0	61.0 ± 0.6 a	< LOD	0	$25.47\pm0.12\boldsymbol{c}$	< LOD	0	$25 \pm 4c$	< LOD	0

R%: percent recovery of dialysate with respect to coffee. Different letters (a > b > c > d > e to coffee, or A > B > C > D to dialysate) in the same row indicate significant differences (P < 0.05). All compound were quantified using 5-O-caffeoylquinic acid as reference compound, except: ¹quantified with quinic acid; ²quantified with caffeic acid; ³quantified with ferulic acid; ⁴quantified with *p*-coumaric acid. < LOD, below limit of detection. < LOQ, below limit of quantification. IDL = 4 mg/100g to 3-CQA, CQL, 3-diMCiQA, 3,4-diCQA, 4-diMCiQA, 5-diMCiQA, CoCQA, 3,4-CFQA, 4,5-diCQA, 4,5-CFQA and diCQL. IDL = 0.6 mg/100g to 1-FQA, 3-FQL, FA, diFQA and FT. IDL = 2 mg/100g to CA and CT. IQL = 15 mg/100g to 3-CQA, CQL, 3-diMCiQA, - IQL = 6 mg/100g to CA and CT.

 Table 4. Antioxidant capacity (mmoles eq TROLOX/100g sample), for coffee samples

 before *in vitro* digestion and after dialysis.

		FRAP			TEAC	PHEN				
	coffee	dialysates	R (%)	coffee	dialysates	R (%)	coffee	dialysates	R (%)	
RC	71 ± 7 a	$14.7\pm0.9\;\textbf{B}$	20.7	133 ± 11 a	$27 \pm 4 \mathbf{B}$	20.3	113 ± 18 a	22 ± 9 B	19.5	
AC	76 ± 12 a	13.8 ± 1.3 B	18.2	132 ± 18 a	$23.4\pm0.7~\text{C}$	17.7	124 ± 15 a	28 ± 4 B	22.6	
GRC	$65.5\pm1.7~\textbf{b}$	$17.20\pm0.08~A$	26.2	$121 \pm 4 \mathbf{b}$	$33.3 \pm 1.1 \text{ A}$	27.5	112 ± 6 a	40 ± 3 A	35.7	
CC	$42.1\pm1.0~\textbf{d}$	$6.88\pm0.19~\textbf{D}$	16.3	72.3 ± 2.4 c	$11.5 \pm 0.9 \text{ E}$	15.9	$66 \pm 3 c$	8.4 ± 0.4 C	12.7	
DC	61 ± 8 c	10.5 ± 1.3 C	17.2	128 ± 9 a	$22.7\pm0.7~\textbf{D}$	17.7	102 ± 14 b	$24 \pm 10 \text{ B}$	23.5	

R%: percent recovery of dialysate with respect to coffee. Different letters (a > b > c to coffee, or A > B > C to dialysates) in the same column indicate significant differences (P < 0.05).

Table 5. Adjusted parameters, performance, and relative influence (%) of Boosted Regression Trees models for antioxidant capacity (FRAP, TEAC and PHEN) of coffee samples before *in vitro* digestion and after dialysis.

		Coffee		Dialysates				
	TEAC	FRAP	PHEN	TEAC	FRAP	PHEN		
Adjusted parameters								
bag fraction	0.55	0.55	0.50	0.55	0.60	0.60		
learning rate	0.0006	0.0008	0.0006	0.001	0.0006	0.0005		
tree complexity	2	2	2	2	2	2		
Model performance								
CV correlation	0.969	0.979	0.926	0.985	0.970	0.982		
optimal number of trees	48,250	39,500	29,750	38,250	32,250	60,500		
Relative influence of polyphenols (%)								
QA	9.5	3.4	5.6	51.8	20.9	10.0		
3-CQA	2.0	2.9	2.0	-	-	-		
4-CQA	4.5	1.0	2.0	8.9	0.3	0.7		
1-FQA	7.3	3.8	2.4	-	-	-		
cis5-CQA	0.8	0.4	0.8	2.7	1.5	12.8		
5-CQA	9.2	3.0	12.3	5.1	2.3	14.3		
3-FQA	13	2.5	1.9	1.1	2.1	2.2		
4-FQA	15.9	2.7	2.3	0.3	1.9	0.6		
CQL	5.6	27.3	30.1	-	-	-		
5-CoQA	1.1	5.3	2.8	20.3	45.3	37.6		
5-FQA	1.0	0.8	1.1	1.0	15.5	1.4		
СА	0.8	0.5	0.7	-	-	-		
3-diMCiQA	0.4	0.4	0.6	-	-	-		
3,4-diCQA	1.4	3.1	5.1	-	-	-		
3-FQL	1.8	5.2	5.3	-	-	-		
4-diMCiQA	0.3	0.3	0.3	-	-	-		
5-diMCiQA	0.2	0.3	0.6	-	-	-		
CoCQA	0.1	0.1	1.1	-	-	-		
FA	2.5	2.9	2.1	-	-	-		
3,4-CFQA	1.6	3.0	5.3	-	-	-		
4,5-diCQA	0.4	0.2	1.1	-	-	-		
diFQA	4.4	0.8	0.7	-	-	-		
4,5-CFQA	0.4	0.5	1.6	-	-	-		
СТ	11.6	1.7	1.5	-	-	-		
diCQL	0.2	3.8	4.7	-	-	-		
СоТ	2.3	13.4	3.6	8.8	10.3	20.3		

FT	1.7	10.7	2.1	-	-	-
Cumulative influence (%)	100	100	100	100	100	100

FIGURE GRAPHICS

Figure 1.







GRAPHIC FOR TABLE OF CONTENTS.

